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This paper must be cited as:

Jorge Amorim, R.; Novella Rosa, R.; García Martínez, A.; Molina, S. (2017). Study on LTC for light duty engines Part 2 Spray enhancements. Fuel. 193:206-219. doi:10.1016/j.fuel.2016.12.050



The final publication is available at http://doi.org/10.1016/j.fuel.2016.12.050

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Additional Information

Study on LTC for light duty engines – Part 2 – Spray

2 enhancements

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DEFINITIONS, ACRONYMS, ABBREVIATIONS 17 18 (A/F)STOICH Stoichiometric Air-Fuel ratio 19 A/F Air-Fuel ratio 20 aTDC After Top Dead Center 21 C_{a} nozzle orifice area contraction coefficient 22 **CAD** Crankshaft Angle Degree **CFD** Computational Fluid Dynamics 23 CO Carbon Monoxide 24 Carbon Dioxide 25 CO₂26 CR Compression ratio 27 dBa decibel Direct Injection Engine Simulation software 28 **DIES** 29 d_{nozzle} nozzle orifice diameter **Exhaust Gas Recirculation** 30 **EGR** 31 EU European Union FC Fuel consumption 32 33 **FSN** Filter Smoke Number 34 HLift-off length Hydrocarbons 35 HC 36 HRL Heat Release Law **High-speed Direct Injection** 37 **HSDI** Indicated Mean Effective Pressure **IMEP** 38 IP**Injection Pressure** 39 40 LL Liquid Length 41 LTC Low Temperature Combustion intake air mass flow 42 m_{air} 43 EGR mass flow m_{EGR} 44 fuel mass flow per cycle m_f intake total air mass flow 45 mint 46 NO_X Nitrogen Oxide 47 O_2 Oxygen Original Equipment Manufacturer 48 OEM Polycyclic Aromatic Hydrocarbons 49 **PAH**

Premixed Compression Ignition

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PCI

51	p_{int}	intake air pressure
52	ppm	parts per million
53	RoHR	Rate of Heat Release
54	rpm	rotations per minute
55	SFC	Specific Fuel Consumption
56	SoE	Start of Energizing
57	SoI	Start of Injection
58	T_{AD}	Adiabatic Flame Temperature
59	T_{Ad_Max}	Maximum Adiabatic Flame Temperature
60	T_{AMB}	ambient air temperature
61	$tan(\theta/2)$	jet spreading half-angle
62	TDC	Top Dead Center
63	tinj	Injection Duration
64	Tint	intake air temperature
65	U	Injection Velocity
66	XO_2	Oxygen concentration
67	X_{O2_air}	ambient air volumetric oxygen concentration
68	X_{O2_int}	intake air volumetric oxygen concentration
69	Z_{st}	stoichiometric fuel mixture fraction
70	hoair	in-cylinder air density
71	$ ho_f$	fuel density
72	$ au_{ig}$	Ignition Delay
73	$oldsymbol{\Phi}_H$	Equivalence ratio at the lift-off length cross-section

75 ABSTRACT:

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- This paper presents the 2nd part of a study on low temperature diffusive combustion in a small HSDI (High Speed Direct Injection) engine with the objective of avoiding simultaneously NOx and soot formation, aiming to extend the range of operation conditions where this combustion strategy has been achieved and. Also, it was intended to reduce CO and HC emissions, considered very high in the 1st part of the tests. New hardware configuration was chosen in order to improve air entrainment in the lift-off length such as increasing injection pressure, reducing fuel injection rate and nozzle hole diameter. Also, a parametric study on the behavior of in-cylinder air thermodynamic characteristics towards the diffusion-flame low temperature combustion was also carried out in order to evaluate the important aspects of each one and observe how the LTC range has been extended.
- The strategy is based mainly in a significant reduction of the equivalence ratio at the lift-off length cross section and the combustion temperature, maintaining the conventional mixing controlled diesel jet structure. In this 2nd part of the study, injection pressure was increased and new nozzles with reduced holes were used to enhance air entrainment and the injection rate was reduced to extend the duration of diffusion-flame combustion phase.
- The new proposal allowed extending Diffusion-flame LTC range, being possible to achieve this combustion modality with higher air temperature which lead to considerable reduction in HC, CO and combustion noise while NOx and Soot remained insignificant.
 - KEYWORDS: LTC; Diffusion-flame Combustion; Light duty engine; Diesel engine; Diffusion flame

1. INTRODUCTION

Diesel engine technology development, much of them regulatory driven, was the main responsible for the increasing diesel market share from around 20% in 1995 to more than 50% in 20011 in Europe, and the total EU fleet from 12% to 35%. The increase market share is mostly attributed to tax incentives for diesel fuel, tax advantages to support Diesel-fueled car sales and provisions to taken by the OEM's in order to invest in Diesel engine development [1,2]. A very strict regulatory is going to be effective soon in Europe, USA and Japan, a very exigent market in terms of reliability, performance and cleaner vehicles, and a great demand for fuel economy engines turned into the great challenges for the diesel engine manufacturers. Diesel developers have reacted using ultimate technology such as advanced injection systems, EGR (Exhaust Gas Recirculation) control, downsizing and hybridization, etc. [3,4]

In a classic definition, a diffusion flame is considered a flame which fuel and oxidizer are initially separated. For a Diesel engine, burning rate is much faster than the mixing rate, so the mixing controls the combustion process. Since the mixing process depends on the injection event, for a diffusion flame, combustion and injection events must happen simultaneously. If the injection ends prior than the start of combustion, this combustion process is considered a premixed combustion, even if it is not homogeneous.[5]

Combustion process in a Diesel engine is a very complex phenomenon. In 1997, Dec proposed a new Diesel spray conceptual model which became a reference on diesel combustion and on which the development of this work is based. This work is based initially on the Diesel spray conceptual model introduced by him, which became a standard on diesel combustion. In this model, he presented a schematic of diesel combustion including the zones where NOx (Nitrogen Oxide) and soot were initially formed. [6,7]



Figure 1. Schematic view of Diesel spray conceptual model proposed by Dec. [6]

Figure 1 presents the conceptual model proposed by Dec. From the injector nozzle until the beginning of the flame, the fuel entrains some air that surrounds the jet (in light blue). Fuel-air mixture reaches the auto-ignition temperature and first reactions start in the fuel-rich premixed flame zone (dark blue). In this zone, an almost ideal environment for soot formation is created by the high ambient temperature and the fuel-rich mixture. These particles tend to enlarge (yellow) as they go towards the flame vortex where the largest soot particles are found (red zone). At the surroundings of the flame, the remaining fuel burns in contact with the external air oxygen. The mixing-controlled diffusion flame at the periphery of the jet significantly increases flame temperature and provides thermal NO_X formation. Also, in the diffusion flame, soot particles formed inside the jet might be oxidized by the effects of high temperatures (in black) [6].

Some investigations were elaborated on conceptual models of reactive free diesel sprays and those models have been improved continuously due to the use of more modern technology applied to the reactive and non-reactive spray visualization. [6,8–14] Those visualizations permitted to identify zones where cool flames, soot precursors and high temperature reactions happen. Also, some investigations have been carried out to identify different zones where NO_X, soot precursors or soot are found in the flame and where they are first formed. Therefore, the influence of wall-jet and jet-jet interactions was studied to develop a combustion structure model. [15,16] Although they were of vital importance, it is important to point-out that those works were carried out in a constant volume vessel to study the spray behavior independently of the engine dynamics.

Extended Zeldovich mechanism describes the chemistry of the NO_X formation process that consists of chain reactions involving O and N atoms. Thermal NO_X formation is considered insignificant in local ambient

temperatures below 2000K and mostly happens in near-stoichiometric mixing-controlled reaction zones in the combustion chamber. [17,18]

Moreover, engine-out soot is the final result of a competition between soot formation and soot oxidation. Soot formation origins from a chemical process called Pyrolysis when soot precursors, PAH (polycyclic aromatic hydrocarbons) and acetylenes, are formed due to changes of the fuel organic compounds molecular structure. In the sequence, Nucleation, or soot particle inception, takes place in a temperature range from 1400 to 1600K, consisting of radical additions of small hydrocarbons to larger aromatic molecules. The next step in soot formation is the mass addition to the surface of a nucleated soot particle, known as Surface growth. Lastly, two processes of particle association happen simultaneously: Coalescence and agglomeration, forming new bigger particles with spherical shapes or chain-like structure. In fact, soot is really formed during the nucleation. The other processes can be described only as enlarging effects defining final soot particle size. On the contrary, soot oxidation is the conversion of carbon or hydrocarbons to combustion products, and, once the carbon present in the fuel is partially involved in oxidation to CO (Carbon Monoxide), it will no longer form soot. In diffusion flames, soot formation rate and soot oxidation rate increase with temperature. Soot formation starts above 1400K whereas its oxidation ceases below 1300K. Consequently, combustion temperature is the most influent parameter in the process. [12]

Basically, soot formation in a diesel jets depends on three factors: equivalence ratio at the lift-off length (Φ_H), residence time of the fuel inside the flame and local temperature inside the flame. On the other hand, soot oxidation depends mainly on the combustion process temperature. [13]

In addition, combustion temperatures must be reduced to avoid thermal NO_X formation. However, the observed effect of the combustion temperature reduction is the immediate extinction of soot oxidation process leading to a substantial engine-out soot increase. Nevertheless, a possibility of reducing NO_X and engine-out soot emissions simultaneously might occur by reducing combustion temperature so that NO_X and soot formation would be avoided. [9]

A viable and novel solution is the Diffusion-flame Low Temperature Combustion (LTC), presented as the previous part of this investigation, reduces NO_X and soot formations to negligible amounts, preserving the mixing-controlled diesel jet structure [19]. The concept is based on a significant reduction of the combustion temperature and the equivalence ratio at the lift-off length cross section. This zero-NO_X, non-sooting combustion strategy was first attained using a free-jet configuration in a constant-volume vessel. [20] Afterwards, the Diffusion-flame LTC concept was successfully transferred to a real engine, as seen in the previous work. In that investigation, although soot and NO_X levels were negligible, CO and HC (Hydrocarbons) emissions were inacceptable. Also, IMEP (Indicated Mean Effective Pressure) presented a reduction around 10% due to the combustion process deterioration [19].

It is important to highlight that Diffusion-flame LTC presents other advantages such as not having problems with combustion ignition control and low combustion noise if compared to regular mixing-controlled combustion process [19]. However, in a real engine situation, in-cylinder gas thermodynamic characteristics vary cycle-by-cycle and according to the operating condition, which could be a major challenge to obtain this combustion modality [21]. Moreover, lift-off length and the equivalence ratio at this point can be affected by jet-jet and jet-wall interactions. For longer lift-off lengths, wall-impingement is likely to happen and might cause undesirable increase in HC and CO emissions. Finally, it is necessary to use very high intake boost pressures to achieve very low oxygen concentrations (very high EGR rates) with suitable values for A/F (Air/Fuel ratio) in order to reduce substantially combustion temperatures to avoid NO_X and Soot formation.

2. OBJECTIVES

The previous part of this study succeeded in attaining Diffusion-flame LTC in a DI Diesel engine although it only happened for engine operating condition with a very low volumetric oxygen concentration of 9%, high air density of 40 kg/m³ and cooled intake air temperature (air + EGR) of 40°C. This is considered an extreme and not realistic

- operational condition. However, very high CO and HC emissions were observed associated to a significant reduction in IMEP [19].
- 183 The main objective of this study is to extend Diffusion-flame LTC range using a small HSDI diesel engine in
- order to avoid both NO_X and soot formation by reducing O₂ concentration and equivalence ratio at the same time.
- The extension of the range would be fundamental to allow this combustion strategy to work in less several
- operational conditions (e.g., higher T_{int} and lower EGR rate) in which HC and CO emissions are reduced and the
 - penalty on IMEP are less important, demonstrating its feasibility to be applied on the market in a near future.
 - This combustion strategy is studied in a real engine in order to:
 - investigate the possibility of extending Diffusion-flame LTC range;
 - study the influence of different parameters separately and their interactions;
 - identify any restriction and opportunity on applying and/or extend this concept in a HSDI engine.

3. METHODOLOGY

 This section elaborates the experimental facility and the development of the test plan through the engine maps used in the investigation.

3.1. EXPERIMENTAL FACILITY AND TOOLS

The engine used was a single cylinder, HSDI diesel engine with high capability of controlling engine parameters and measurement accuracy. For this engine, boost pressure could be set up independently of the engine operational condition, presenting compatibility between high intake boost pressures and very high EGR rates. The engine main characteristics are shown in table 1.

Table 1. Engine characteristics

Characteristic	Description
CR	15:1
Displacement	0.399 dm^3
Bowl diameter	46.35 mm
Nozzle	6 x 129 μm

The engine was installed in a fully equipped test cell, with all the auxiliary devices required for operating and controlling the facility. It was designed to work with high concentrations of cooled EGR (up to 70%) with no water condensation in the intake, being possible to achieve EGR+air intake temperatures in the order of 40° C. The EGR flow has its temperature controlled in a cooler and then, in a settling chamber, the condensed water is drained to avoid water accumulation in the intake system.

For in-cylinder pressure measurement, a KISTLER 4076A pressure transducer and a KISTLER 4618A amplifier were used. A total of 50 consecutive engine cycles per each tested point was acquired, averaged and filtered to avoid cycle-by-cycle variations during engine test. An in-house combustion diagnostic software, CALMEC, was used for heat release law calculation. The 2-zone, 0-D model, based on the first law of thermodynamics applied to a diesel engine and adiabatic flame temperature, assumes constant pressure and temperature along the combustion chamber, being possible to calculate unburned and burned gas temperatures considering that the gases inside the cylinder could be separated in 2 zones. [22–24]. Combustion noise was calculated by CALMEC by means of an estimation proposed by Payri et al. using an in-cylinder pressure trace decomposition technique. The

pressure signal is converted with a Fast Fourier Transform in a cylinder pressure spectrum. Then, using a filter (U-filter), the pressure spectrum is attenuated to a frequency range that is audible to human ear and converted to dBa. [25,26].

The actual fuel injection rate was measured by an IAV-EVI injection analyzer and calculated using the equation by Bosch [27]. In order to evaluate the non-reactive diesel spray development inside the cylinder, an in-house CFD tool called DIES (Direct Injection Engine Simulation) predicts the behavior of non-reactive diesel sprays according to operational conditions such as air density, temperature, injection pressure and nozzle characteristics when introducing the injection rate, engine geometry and in-cylinder ambient conditions [28].

Exhaust gases were analyzed by a HORIBA 7100D which is able to measure HC, CO, CO₂, NO_x and O₂. In order to increase the quality of data, measurements were carried out using samples collected during 60 seconds. All emission results obtained in ppm during the tests were converted to specific values in g/kg_{fuel}. Soot emissions were measured with an AVL 415 smokemeter, providing results directly in FSN (Filter Smoke Number) averaged from 3 consecutive measurements under the same operating conditions. The FSN value was converted into mg/m³ by means of the correlation presented by Christian et al. [29]

The measurement uncertainties were defined according to the equipment calibration and statistical procedures [30]. The uncertainties for each parameter is presented in Table 2 and they are expressed in a percentage of the measured value.

Uncertainty Equipment Parameter NO_X, CO, CO₂, HC Horiba 7100D ± 3% Soot **AVL 415** $\pm 3\%$ Fuel mass flow $\pm 0.1\%$ AVL KMA4000 Air mass flow Elster RVG G-65 $\pm 2\%$ In-cylinder pressure Kistler 6055 $\pm 0.1\%$

Table 2 Measurement System Uncertainty

Besides the conventional $6 \times 129 \mu m$, 2 new injector nozzles have been developed and constructed for this study, both with reduced fuel injection rate around 25% if compared to the original one. The description of the two new nozzles can be found in table 3:

Table 3 Injector nozzle characteristics

Characteristics	Unity	Nozzle 1	Nozzle 2	Nozzle 3
Orifice number	[]	6	6	9
Orifice diameter	[µm]	129	101	83
Hydraulic delivery rate	$[cm^3/30s]$	284 - 295	187 - 195	187 - 195
Injection angle	[°]	150	150	150

3.2. ENGINE MAP

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The engine maps are very important as it can correlate various engine controllable operating parameters, such as intake air pressure, A/F and EGR rate to in-cylinder ambient conditions, like O_2 concentration and air density. This method was developed to control a new combustion approach with oxygen concentration. [19,31,32]. Firstly,

engine test load condition is defined through fuel consumption per cycle (m_f) . It can be directly linked to intake air mass flow $(m_{int}=m_{air}+m_{EGR})$ in function of A/F and EGR ratio by the equation:

$$A/F = (1 - EGR) \cdot m_{int}/m_f \tag{1}$$

where m_{int} is intake total air mass flow and m_f is fuel mass flow. For the hypothesis of complete combustion, it is possible to represent the O_2 volumetric concentration (X_{O2_int}) isoline by:

$$X_{O2 int} = X_{O2 air} \cdot (1 - EGR \cdot (A/F) s_{TOICH} / (A/F))$$
(2)

where X_{O2_int} and X_{O2_air} are intake and ambient air volumetric O_2 concentrations, respectively, and $(A/F)_{STOICH}$ is the stoichiometric air fuel ratio. Thus, a 2^{nd} hypothesis that at constant values of intake pressure and temperature, volumetric efficiency and gas constant remain unchanged, independently of EGR ratio, is accepted. Once the engine map for a specific engine load is defined, engine geometry and engine operating conditions, such as clearance volume, volumetric efficiency, engine speed and intake air temperature (T_{int}) , can be associated to intake total air mass flow to obtain values for air density (ρ_{air}) and intake air pressure (p_{int}) .

The engine map calculated for T_{int} = 75°C used in this investigation is shown in figure 2. It is noticeable that air density and intake air pressure are parallel. In other words, air density is easily controlled by changing intake air pressure.

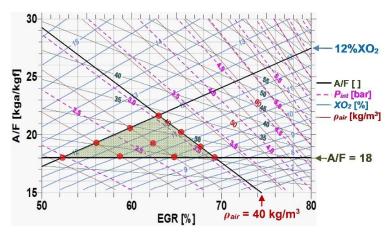


Figure 2. Engine map calculated for intake air temperature of 75°C. Light blue lines are volumetric O₂ concentration, pink lines are intake air pressure and red lines are air density and green lines are air-fuel ratio. Note that air density and intake air pressure are parallel lines. The green triangle is the limited area designed to be the test plan and the red dots are the points to be tested.

3.3. ENGINE TEST PLAN

In 1998, Siebers investigated the influence of various parameters on liquid length (Figure 1) [33]. Moreover, García proposed an empirical equation that associated the liquid length to nozzle orifice diameter (d_{nozzle}), air temperature (T_{AMB}) and air density (ρ_{air}) [34]:

$$LL = (d_{nozzle}/150) \cdot 7.2455 \times 10^{5} \cdot T_{AMB}^{-1.269} \cdot \rho_{air}^{-0.53}$$
(3)

Reducing the liquid-phase and increasing lift-off length (Figure 1) would increase significantly the amount of entrained air downstream the lift-off length. [35] Also, an equivalence ratio value at the lift-off cross section around or below two was affirmed to reduce soot formation to negligible amounts [36]. In more recent studies, the effects of ambient conditions, nozzle orifice diameter, injection pressure and oxygenated fuels on lift-off length and global air entrainment were measured. Based on those, Siebers, Higgins and Picket developed two equations to estimate lift-off length (H) and equivalence ratio at the lift-off (Φ_H), respectively:

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$$H = 7.04 \times 10^8 \cdot T_{AMB}^{-3.74} \cdot \rho_{air}^{-0.85} \cdot d_{nozzle}^{0.34} \cdot U \cdot Z_{st}^{-1}$$
 (4)

$$\phi_{H} = \frac{2 \cdot \left(\frac{A}{F}\right)_{st}}{\sqrt{\frac{1+16 \cdot \left(\frac{P_{f}}{\rho_{air}}\right) \cdot \sqrt{C_{a}} \cdot d_{nozzle}}{0.75 \cdot tan(\theta/2)}}} - 1}$$

$$(5)$$

where U is injection velocity, Z_{st} is stoichiometric fuel mixture fraction, ρ_f is fuel density, C_a is the nozzle hole area contraction coefficient and $tan(\theta/2)$ is the jet spreading half-angle [16,20,35,36]. Pickett, Siebers and Idicheria also researched into the relationship between ignition processes and the lift-off. They observed that there was not a general rule, but there was a trend that the longer the lift-off the longer the ignition delays (τ_{ig}). They affirmed that ignition delay tends to increase by decreasing temperature, air density and O_2 concentration:

$$\tau_{ig} \propto e^{(1/T_{AMB})} \cdot \rho_{air}^{\ n} \cdot Z_{st}^{\ m} \tag{6}$$

where the value of n \approx -1.3 and m \approx -1.0 for diesel fuel. [37]

Each parameter presents a different effect on diesel jet according to the several studies carried out by Sandia National Laboratory in a constant volume vessel with controlled ambient conditions. In the Table 4, the parameter effects on the spray are exposed. The Trend is the parameter change that is adopted in this study that leads to spray enhancements or enhances Diffusion-flame LTC conditions. The arrows indicate if the trend or the other spray characteristics increases (↑) or decreases (↓):

Table 4: Parameter Effects on the Diesel Spray

Parameter	Trend	LL	Н	$oldsymbol{\phi}_H$	Observations
Air Density	↑	\	\	Global net result is a slight increment	Increasing air density shortened combustion delay and increased combustion efficiency. [32]
Air temperature	\	↑	↑	\	It reduces the whole cycle temperature that can lead to a reduction in NOx formation and affect soot formation and oxidation rates.
O ₂ Concentration	\	-	↑	The rate of air entrainment decreases so the final net effect is a constant amount of entrained air.	The dilution slows O ₂ -fuel mixture ratio that has great influence in the flame temperature, directly correlated to soot formation and oxidation and NOx formation. [12,33–36]
Injection Pressure	↑	-	1	\	Improved air fuel mixing properties can help reduce HC and CO emissions [12,38].
Orifice diameter	\	\	\	\	The amount of air entrained downstream the lift-off cross section decreases, but injection rate is reduced more intensively. Thus, air entrainment per mass unit increases and equivalence ratio at the lift-off length cross section decreases. [12,38]

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The extensive use of EGR combined with very high intake boost pressures was studied by Noehre et al and Colban et al. [39,40] However, they did not have the compromise to obtain strictly a Diffusion-flame LTC. They observed that reducing intake O₂ concentration to very low levels at high air densities could lower soot and NO_x simultaneously. The reduction in combustion temperatures caused by lower O₂ concentration avoids soot and NO_x formation. On the contrary, CO, HC and SFC (Specific Fuel Consumption) presented considerable losses.

The test plan was developed to obtain three fundamental purposes: (1) Diffusive combustion; (2) LTC and (3) The lowest possible equivalence ratio at the lift-off length cross section. Ambient air density, A/F and O₂ concentration operating range were determined based on: (a) Engine operating conditions was 2250 rpm, midload, single injection with constant fuel mass flow of 19 mg/cc and targeting 9.0 bar IMEP (not fixed). Fuel injection SoE (start of energizing) was -10.0 CAD aTDC, equivalent to a SoI (start of injection) at -5.0 CAD aTDC due to the fuel system injection delay. (b) The A/F value could not be lower than 18 to avoid excessive HC and CO. (c) Air density could not be higher than 40 kg/m³ because of equipment intake pressure safety restrictions. (d) Once X_{O2 int} would not affect equivalence ratio at the lift-off length, it could be set as the lowest range possible in order to obtain a very low combustion temperature, aiming to avoid both NO_X and soot formation. Other investigations obtained very good results from 12% until 8% of XO₂. However, as it can be seen in the engine map (Figure 2), 8% of XO₂ could not be set due to A/F restriction or intake pressure limitations. Reaching XO₂ of 8% would demand an intake air pressure of approximately 4.0 bar. Finally, XO₂ higher than 12% would avoid LTC due to high combustion temperatures [8,9,14,36,39,40]. (e) In-cylinder air temperature range was defined firstly based on the range from 850K and 1000K investigated by Sandia National Laboratory. Good results were obtained using $T_{AMB} = 800$ K, but in a realistic small engine with $r_c = 15.1$ such as the engine used in this work it would mean an intake air temperature around 0°C. Also in the previous work by Benajes et al., with a 850K (20°C) T_{AMB}, wall impingement was imminent because the lift-off length for those points was longer than the physical limit (bowl radius) [19]. This range indicated an intake air temperature range (air+EGR) from 40°C ($T_{AMB} = 900$ K) to 75°C ($T_{AMB} = 1000$ K). Each intake air temperature requires a different engine map due to the fact that varying intake air temperature changes the relationship between ρ_{air} and p_{int} . Obs.: Ambient air temperature and in-cylinder air temperature are referred as the same parameter T_{AMB} . Nomenclature changes because the former refers to a constant volume vessel air temperature whereas the latter refers to an engine incylinder air temperature. (f) Furthermore, engine test plan range was set as it is presented in figure 2 (green triangle). The red dots on the area are the testing points. They are the intersection between air density and XO_2 isolines and make it possible to read their respective values for intake air pressure and EGR rate which are the real controlled engine parameters.

In a small-bored diesel engine, the bowl diameter is very short. To evaluate wall impingement likelihood, for each point liquid length and lift-off length were calculated using equations (3) and (4) and compared to the distance from the injector tip to the piston bowl wall. The final test plan is exposed in the table 5.

Injection pressure (IP) values were fixed in 1200 bar, as in the previous study [19], and 1400 bar in order to observe the benefits of increasing injection pressure avoiding shortening injection duration significantly.

Two new nozzles were designed to reduce fuel injection rate around 25%. The 6x101 µm configuration reduced only nozzle orifice diameter. The other one, 9x083 µm, had the orifice diameter reduced while the number of orifices increased. Reducing fuel injection rate and orifice diameter simultaneously were supposed to enhance Diffusion-flame LTC conditions although using 9 orifices distance between sprays is shorter and jet-jet interaction is more likely to happen.

Table 5. Engine test plan

Parameter	Value
mf	19 mg/cc
Engine Speed	2250 rpm
Engine Load	Around 9 bar IMEP (mid-load)
IP	1200 bar - 1400 bar
SoE	-10 CAD
Tint (TAMB)	40°C (900K) - 55°C (950K) - 75°C (1000K)
XO_2	9% - 10% - 11% - 12%
Air Density	$26kg/m^3 - 30kg/m^3 - 35kg/m^3 - 40kg/m^3$

Oxygen concentration and air density were ranged as shown in figure 2, in a total of 10 points per intake air temperature. The total amount of test points in this parametric study was thirty so that for each engine tested point, only one engine operating parameter was varied. A predicted Φ_H calculation was performed based on equation (5) and the lowest predicted Φ_H average value was around 2.4 for 9x083 µm and 2.7 for 6x101 µm, close to the proposed value that was around 2.0 to have all soot-precursors oxidized.[7,36,41]

4. RESULTS AND ANALYSIS

This section shows the obtained results from experimental engine test carried out ranging oxygen concentration, air density, intake air temperature, injection pressure and nozzle design. The influence of each parameter in terms of pollutant emissions and combustion characteristics are studied separately.

4.1. AIR DENSITY

Air density has no significant influence on equivalence ratio. However, increasing it tends to shorten lift-off length and reduce ignition delay. Based on this, it was expected to obtain similar NO_x and Soot results among all tested ρ_{air} . Increasing air density is a necessity in order to guarantee that the combustion process is mixing controlled. In figure 3, NO_x and soot graphs compare different air densities using the same 9 x 83 µm nozzle for 75°C intake air temperature for 1200 and 1400 bar injection pressure. All air densities had approximately the same values for the same oxygen concentration. Emissions of NO_x are the same for both IP's, indicating that adiabatic flame temperatures are similar. Pickett stated that similar equivalence ratios at the lift-off length for different ρ_{air} do not precisely mean that soot formation and soot oxidation rates would be similar [14]. On the contrary, he observed that probably soot formation rate was affected. It means that, to have the same engine-out soot, oxidation rate might be affected at the same intensity than formation rate. It is not possible to know if soot formation and oxidation rates were affected.

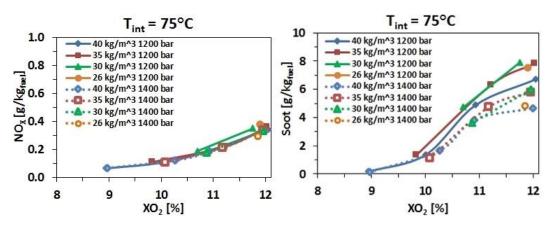


Figure 3 Air density comparison for 75°C using 9x083 μm nozzle for both injection pressures.

4.2. AIR TEMPERATURE

Reducing in-cylinder air temperature at TDC effectively reduced soot emissions. This temperature was controlled through intake air temperature. The gaps among different T_{int} curves are decreased with the reduction in X_{O2_int} until the point that there were no soot emissions. Figure 4 (a) and (b) present T_{int} influence with 40 kg/m³ ρ_{air} using 6x101 µm nozzle. There is a noticeable reduction when using 1400 IP but the curve trends are the same. For both IP's, there was a reduction in soot values for each T_{int} . Also, differences among curves tend to disappear for the lowest oxygen concentrations. Soot emissions are very close to 0 FSN with oxygen concentrations around 10% and extinguish at 9%. Under lower X_{O2_int} there is no evidence of T_{int} influence on combustion process, and consequently, soot formation.

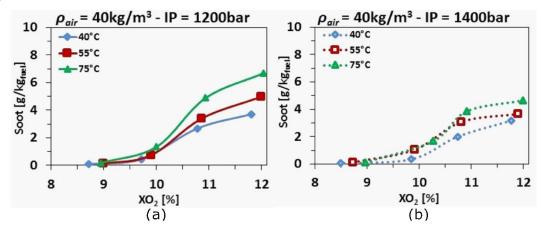


Figure 4 Intake air temperature influence with 40kg/m^3 using $9 \text{x} 083~\mu\text{m}$ and $6 \text{x} 101~\mu\text{m}$ nozzle for both injection pressures

Figure 5 shows maximum adiabatic flame temperature (T_{Ad_Max}) using the same configuration from Figure 4 (b). In the graph, T_{Ad_Max} is directly related to intake air temperature. The lower intake air temperature the more reduced are engine-out soot emissions.

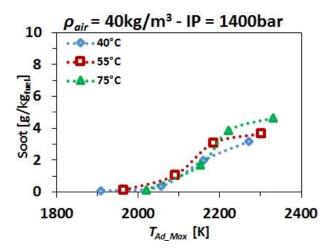


Figure 5 Maximum adiabatic flame temperature using 9x083 μm nozzle

Soot emissions are intimately connected to soot formation and soot oxidation rates which depend directly on adiabatic flame temperature. Moreover, equivalence ratio at lift-off length (\mathcal{O}_H) is one of the main factors that controls soot precursor formation before the flame and effectively avoids soot formation. Remembering, \mathcal{O}_H is reduced by the decrease in T_{int} due to the lengthening of H. Figure 6 presents soot emissions vs \mathcal{O}_H graph. \mathcal{O}_H values presented in the graph were calculated using equation (5) shown previously in section 3.3. Grey arrow indicates oxygen concentration reduction. Straight lines connect results for 1200 bar IP and dotted lines for 1400 bar IP. For the highest O_2 concentration, line slopes are different. while reducing $XO_2\%$ line slopes tend to equalize for both IP. For 9%, soot emissions are insignificant.

Comparing lines with the same T_{int} and different IP's, in general, there is a reduction in soot emissions that is attributed to the increased IP since it causes a reduction on \mathcal{O}_H . Comparing lines with the same IP for distinct T_{int} , soot emission reduction is even more intense. It happened because decreasing T_{int} tends to reduce Adiabatic Flame Temperature (T_{AD}) and \mathcal{O}_H simultaneously. It is not possible to distinguish the soot reduction fraction due to the enhancement of \mathcal{O}_H or reduction of T_{AD} despite of the fact it is known both factors have substantial influence on this phenomenon.

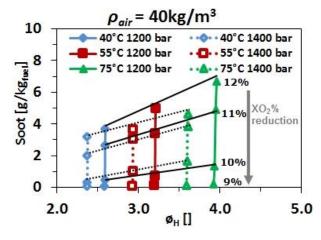


Figure 6 Soot vs Ø_H using 9x083 μm nozzle

Combustion noise and CO, HC emissions felt the effect of T_{int} reduction. Figure 7 shows combustion noise vs Ignition delay graph using 9 x 083 µm and 1200 bar IP. The black lines indicate the results with the same oxygen concentration. Comparing the 3 T_{int} , combustion noise increases around 1dBa with its increasing. T_{int} is one of

the parameters that controls unburned gas temperature before combustion ignition. Unburned gas temperature is related to ignition delay and its reduction tends to increase it. For similar conditions (constant ρ_{air} , IP and XO_2), increasing ignition delay cause higher combustion noise because the amount of fuel available to be burnt during premixed combustion is increased. It was observed that variation of pressure rise rate caused by intake temperature was not significant to explain this reduction in combustion noise, being the ignition delay the most relevant parameter. However, combustion noise levels are very low in all the studied cases if compared to an ordinary HSDI, mainly due to high EGR rates used in this investigation.

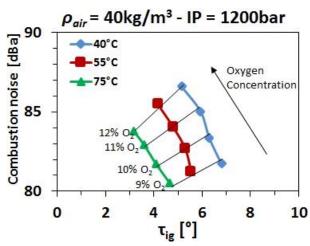


Figure 7 Combustion noise vs Ignition Delay for 9x083 μm nozzle

4.3. O₂ CONCENTRATION

Oxygen concentration was reduced from 12% to 9%. Reducing XO_2 also reduces T_{AD} with no penalty to \mathcal{O}_H . In figure 8, it is presented RoHR (Rate of Heat Release) curves for ρ_{air} of 40kg/m³ and T_{int} of 40°C for the 3 tested nozzles. Dotted lines over the graphs determine injection rate for each nozzle with its respective IP. Analyzing these graphs, it is possible to observe the effects of XO_2 reduction on combustion process.

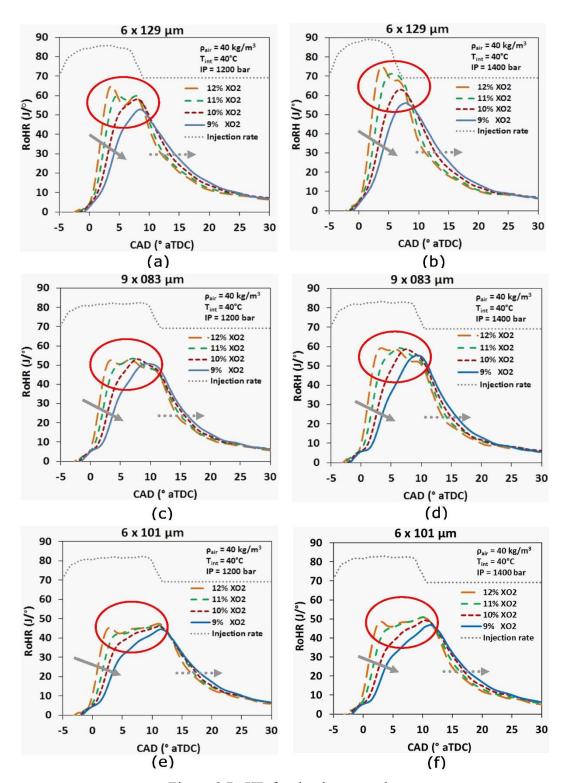


Figure 8 RoHR for the three nozzles

First of all, premix combustion slope reduces due the reduction of reaction speed (grey arrows). Second, premix combustion peak is reduced and moves towards mixing controlled combustion (red circles). Also, slowing combustion process causes a higher of heat release rate at the end of the combustion process, when combustion is decelerating, after the end of injection (dotted arrows). At last, other phenomena that can be observed were the increase of ignition delay and presence of cold flame reactions, also observed by Bruneaux in his studies on LTC

[15,17]. Those observations occurred for all studied cases of this investigation, independently of injection rate, nozzle diameter and injection pressure.

Figure 9 presents T_{AD} curves in relation to oxygen concentration. Test set-up was similar to the exposed in previous figure, with 6x101 µm nozzle and 1200 bar IP. As a consequence of EGR increase, T_{AD} gradually decreased. This reduction in adiabatic flame temperature is very significant and important to avoid thermal NO_X mechanism and soot formation simultaneously.

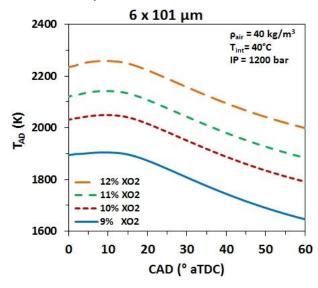


Figure 9 Adiabatic flame temperature for 6 x 101 μm nozzle and 1200 bar IP

In figure 10, it is possible to find NO_X, HC, CO, combustion noise and IMEP. Test conditions were 40°C and 40 kg/m³ for T_{int} and ρ_{air} , respectively. The observed trends are very similar, independent of nozzle. NO_X graph shows a significant reduction with XO_2 reduction. The decrease in T_{AD} inhibits thermal NO_X production, reaching very low level with 9% O₂. Undoubtedly, XO_2 range for this study was very low and, although the NO_X reduction was really intensive, their values are insignificant.

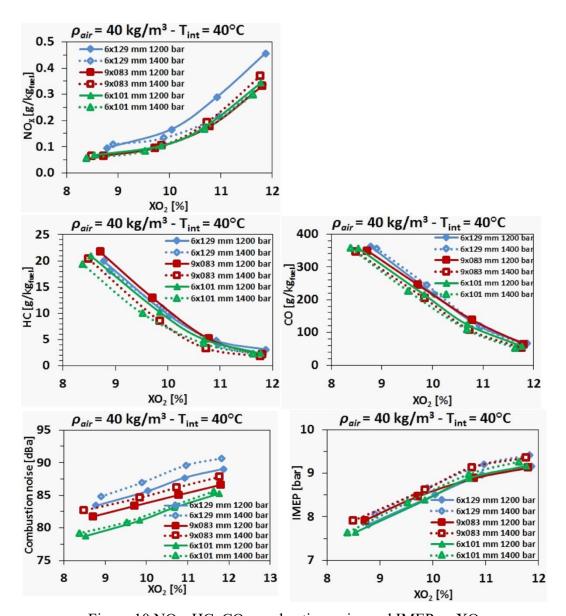


Figure 10 NOx, HC, CO, combustion noise and IMEP vs XO₂

Soot emissions reduce by decreasing oxygen concentration (Figure 11). Also, with $9\% XO_2$, soot emissions were practically zero. The very low T_{AD} affected soot formation so intensely that no soot was formed in those tests. Concerning to HC and CO, comments are very similar. Despite the fact that reducing XO_2 causes great benefits to NOx and soot formation, HC and CO were affected negatively and reached unacceptable values. The reason can be observed in RoHR curves (figure 8). The reduction in XO_2 and its consequent combustion process reaction deceleration cause a deterioration of the process. This can also be confirmed by IMEP graph. At last, combustion noise decreased around 5dBa, from 12% to $9\% XO_2$. Combustion noise is intimately related to reaction velocity during premixed combustion. In figure 8, it is seen that premixed combustion changed its trend and it affected positively combustion noise, producing a smoother combustion if compared to the points with higher O₂ concentrations. Also there is an increase in ignition delay when oxygen concentration is reduced (Figure 7), that might have contributed to a combustion noise rise.

4.4. INJECTION PRESSURE

Increasing injection pressure increases lift-off length and more air can be absorbed upstream it, enhancing mixing process and, consequently, reducing soot precursor formation. However, it increases injection rate and shortens

injection duration that, in some cases, could make it more difficult to reach mixing controlled combustion and flame stabilization.

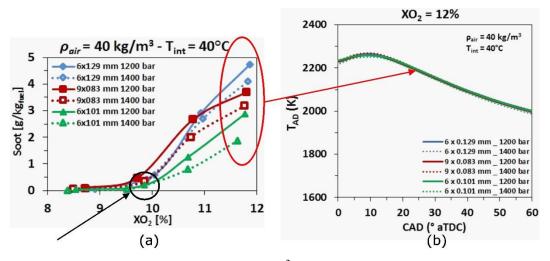


Figure 11 (a) Soot vs XO_2 with ρ_{air} and T_{int} of 40kg/m^3 and $40 ^\circ\text{C}$, respectively and (b) T_{AD} for 12 % XO_2 .

In Figure 11, it is shown the Soot vs XO_2 graph (a) and Adiabatic flame temperature for 12% of oxygen concentration (b). In the graph (a), there is a soot emission reduction when IP is increased from 1200 bar to 1400 bar, mainly for higher oxygen concentrations, due to the reduction of equivalence ratio at the lift-off length cause by IP increment. For the same oxygen concentration ($XO_2 = 12\%$), adiabatic flame temperature is constant, independently of IP or nozzle (red circle) (Figure 11(b)). As it was seen, reducing XO_2 causes soot formation extinction due to T_{AD} reduction effect. It would cancel any effect of IP on soot formation.

Results for NOx, HC, CO, combustion noise and IMEP are presented on the graphs of Figure 10. The enhancement of mixing capability caused by IP increment helps reducing HC and CO emissions. The behavior is similar for the 3 nozzles. However, the new nozzles with reduced injection rates were more effective than 6 x 129 µm nozzle. The enhancement is also observed in IMEP result graph that presents a 5% increase in all XO_2 range. Nonetheless, NOx emissions are not affected once injection pressure have any influence on T_{AD} . Emissions and IMEP had a noticeable improvement, but combustion noise increased slightly. The increased fuel mass injected in the cylinder before the combustion ignition associated to the better mixing capability contributed to a faster and more violent premixed combustion, and consequently, more noise.

4.5. INJECTOR NOZZLE

In this investigation, the new nozzle development aimed to enlarge the duration and improve diffusive combustion process through which it would be easier to obtain Diffusion-flame LTC. Both nozzles were designed in order to: (1) Reduce fuel injection rate; (2) Reduce orifice diameter and (3) Keep or increase number of orifices with no loss by jet-jet interaction.

Figure 11 shows a comparison among the 3 nozzle with T_{int} and ρ_{air} of 40° and 40kg/m³, respectively, for both IP's (1200 bar and 1400 bar). Soot emissions are not inversely proportional to orifice diameter reduction, as suggested previously [38]. Despite of the fact that 9x083 µm nozzle equivalence ratio was the smallest among the 3 nozzles, its soot emissions was not the lowest. Equivalence ratio equation (eq. 5) is valid for controlled thermodynamic condition environment in a constant volume vessel, indicating that they are just an approximation for the development and may be used more qualitatively than quantitatively when comparing the nozzles. The higher soot formation of 9x083 µm probably was caused by an undesirable jet-jet interaction, which causes lift-off length shortening and, proportionally, reduces the amount of entrained air downstream lift-off length, as observed by Musculus [16] and Garcia [21].

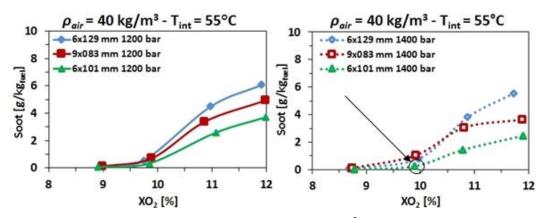


Figure 12 Soot vs XO_2 for ρ_{air} and T_{int} of 40kg/m^3 and $55 ^{\circ}\text{C}$, respectively.

In figure 12, soot emissions are presented for each *IP* separately, this time with 55°C *T_{int}*. The behavior is very similar if compared to the figure 11 results where 6x101 μm nozzle presented the lowest emissions. On the right graph, set to 1400 bar *IP*, soot extinction occurred even with 10% oxygen concentration when using 6x101 μm nozzle (black arrow). This is very important once associating 6x101 μm nozzle with 1400 bar *IP*, besides bringing benefits respect to soot, also helps to improve other parameter results, such as HC and CO, as will be shown later.

The graphs of HC, CO and NO_X emissions in figure 10 compares the behavior of the 3 nozzles. For HC emissions, all nozzles seem to have similar behavior and values. They seem to be affected by the injection pressure, following this pattern: For 1200 bar IP, $6x101~\mu m$ nozzle have better results than $6x129~\mu m$ nozzle while $9x083~\mu m$ nozzle presents insignificant differences from the conventional $6x129~\mu m$ nozzle. With a IP of 1400 bar, both nozzles present better results than the conventional nozzle even though $6x101~\mu m$ nozzle is still slightly better than $9x083~\mu m$ nozzle.

The effect of the nozzles on CO emissions is very alike to HC. The nozzle $9x083 \mu m$ does not present the best result though it has the smallest orifice diameter. Besides, NOx emissions do not seem to be sensitive to the nozzle characteristics probably because adiabatic flame temperature is more dependent from oxygen concentration than to the characteristics of the spray.

Jet-jet interaction in $9x083~\mu m$ nozzle, caused by the increased proximity between two adjacent sprays, might be the reason of combustion process worsening. However, increasing injection pressure slightly reduces CO and HC to levels near to $6x101~\mu m$ nozzle. In spite of this little enhancement, soot emission was not significantly affected to be compared to $6x101~\mu m$ nozzle results.

For the 3 nozzles, IMEP are very similar, such as CO and HC, and cannot be considered a factor that fells intensively nozzles characteristics although 6x101 µm nozzle presents little loss comparing to the others.

Moreover, injection rate seems to have an important influence over combustion noise. If it is higher, the amount of fuel injected prior to combustion ignition is higher, heat release is more intense due to more energy available. Reducing injection rate cause exactly the opposite effect and combustion noise tend to be reduced. It can be seen on the combustion noise graph in figure 10, where noise curves follow exactly injection rate order (Figure 13).

Figure 13 presents 2 comparisons among RoHR of all nozzles with both injection pressures. In all the cases, Diffusion-flame LTC was achieved. As it was expected, Diffusion-flame LTC phenomenological concept with no soot and NOx formation simultaneously reveals a smooth heat release rate. There is no premixed combustion peak and it is noticeable a small cold flame reaction area. Diffusion-flame combustion is more evident, mainly using the new nozzles especially designed to reduce injection rate. In spite of being designed to have the same injection rate (hydraulic injection rate at 100 bar are very similar), with 1200 bar or 1400 bar it is possible to notice a significant difference, around 4%. Furthermore, in the cases which NOx and soot were despicable, combustion process confirmed to be diffusive indeed.

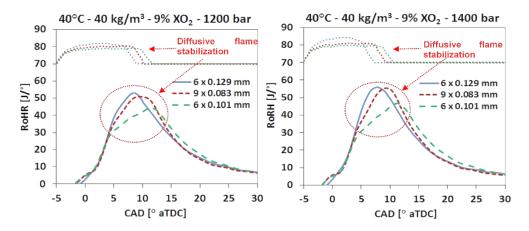


Figure 13 RoHR for 3 tested points of Diffusion-flame LTC with 1200 bar (top) and 1400 bar (below) IP.

4.6. JET-WALL INTERACTION

Some solutions proposed in this investigation to improve Diffusion-flame LTC can increase the likeliness of a jet-wall interaction. However, reducing orifice diameter tends to avoid it. Jet-Wall interaction, if occurred, may have affected some results and, in order to try to reduce its effect in a future work, it was investigated.

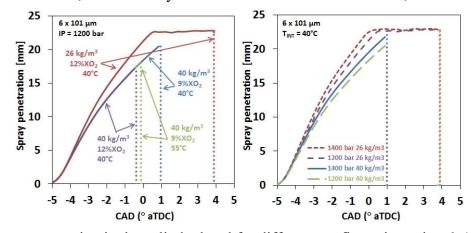


Figure 14 Spray penetration in the cylinder bowl for different configuration using 6x101 μm nozzle.

In this combustion concept, spray penetration does not vary with oxygen concentration and TDC unburned gas temperature but both parameters have a significant effect on ignition delay. In the non-reactive spray simulation using 6x101 µm nozzle, shown in figure 14, vertical dotted lines indicate start of combustion (SoC). Spray penetration, represented by the continuous line, is interrupted at SoC because spray model is non-reactive.

In the left graph, spray penetration and ignition delay for 26 kg/m³ are much longer than for 40 kg/m³ ρ_{air} , indicating that higher air densities help reduce the probability of wall-impingement. Also, reducing T_{int} to 40°C and XO_2 from 12% to 9% makes ignition delay longer. Nonetheless, ρ_{air} has more influence on ignition delay than XO_2 or T_{int} , or both simultaneously, in the range of this study, as had been stated by Picket et al [37] through equation (6). At the LTC configuration ($\rho_{air} = 40 \text{ kg/m}^3$, $T_{int} = 40^{\circ}\text{C}$ and $XO_2 = 9\%$), ignition delay is increased very dangerously until almost reach the time needed to have wall impingement, as it is possible to observe. Since ignition does not occur instantly, it is coherent to consider that wall impingement may have happened.

In the right graph, penetration increases with injection pressure because increasing IP leads to a higher injection velocity. However, ignition delay was not affected by IP as can be seen in figure 15 which compares ignition delay for various XO_2 , ρ_{air} and IP. Based on this graph, oxygen concentration affects ignition delay but with a lower intensity than air density. Maybe, due to fact that air density is extremely high and oxygen concentration too low, IP influence on ignition delay was cancelled in this range of operation. In other words, the fact that IP

increases penetration with no change in ignition delay indicates that the risk of wall impingement is higher comparing to 1200 bar.

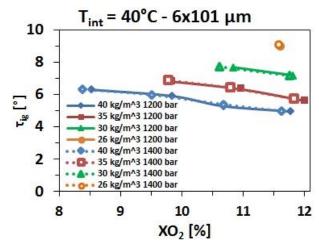


Figure 15 Ignition delay for various ρ_{air} and IP

Figure 16 presents HC emissions using 6 x 101 μ m nozzle. Left graph contrasts it against ρ_{air} and IP and the right one against intake air temperature and injection pressure. Concerning to XO_2 , HC emission rise gradient is remarkably higher for lower concentrations. Contrasting this information with figure 14 graphs, it is possible to observe that wall impingement likely occurred. Since XO_2 also causes combustion process deterioration, it is difficult to quantify the influence of each effect on HC. With 26 kg/m³, HC emissions increased if compared to the other densities what could mean wall impingement once oxygen concentration and intake air temperatures are the same. The same behavior is observed when comparing different T_{int} 's: decreasing it leads to an increase in ignition delay and possibly increase wall impingement or the amount of fuel that impinges. For injection pressure, although penetration graphs indicated wall impingement was prone to occur and HC would increase, these pollutant emissions decreased which means mixing process enhancement influence on HC was stronger than the possible wall impingement's.

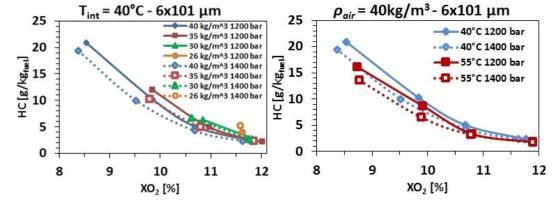


Figure 16 HC emissions using 6x101 μm nozzle.

4.7. DISCUSSION

 Based on the results, an analysis is carried out on the new proposals to enhance Diffusion-flame LTC concept. In figure 11 (a) and figure 12, there are 2 points pointed out with a black arrow. They had similar configurations, with 40 kg/m³ air density and 1400 bar injection pressure, using the same nozzle, 6 x 101 μ m, differing only by T_{int} , 40°C (figure 11) and 55°C (figure 12). For both points, soot emissions were despicable although oxygen configuration was 10%. It was really remarkable that clean combustion concept, with no soot and NO_X formation, was achieved with an oxygen concentration 1% higher than the original configuration. It means that Diffusion-

flame LTC was achieve with a EGR rate reduced from 70% (with 40°C) to 67% (55°C) [19]. The development of the new 6x101 µm nozzle, with micro-orifices, associated to a higher injection pressure and a reduced unburned gas temperature, enhanced conditions to achieve an improved Diffusion-flame LTC.

Once Diffusion-flame LTC concept had been achieved, other factors such as HC and CO emissions, IMEP and combustion noise were examined. Figure 17 shows a comparison between 9% and 10% XO₂ using 6x101 µm nozzle. Circles point out results with the same XO₂ values. HC and CO emissions obtained a 50% result comparing to 9% oxygen. Also, IMEP increased around 0.5 bar, which means approximately a 7% increase in engine performance. Considering that FC was constant, it means a similar improvement in specific fuel consumption. Combustion noise increased from around 79-80 dBa (9% XO₂) to 81-82 dBa (10% XO₂). Although it increased, combustion noise remained in a very low level, still under the best values obtained using the conventional nozzle. Based on this, developing a nozzle with reduced nozzle orifice diameter and reduced hydraulic mass flow aiming the extension of Diffusion-flame LTC range really led to very significant results.

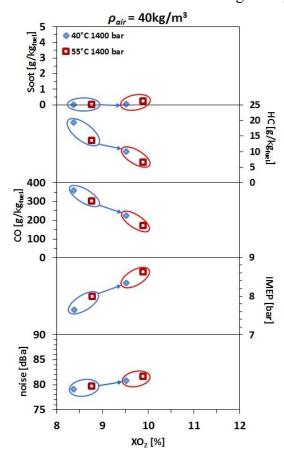


Figure 17 Comparison between 9% and 10% *XO*₂ in order to evaluate combustion noise, IMEP, HC and CO emission improvement.

Diffusive LTC was tested only at mid load and mid speed with the purpose to prove its feasibility. It needs to be tested in other conditions to go to a production engine. Some consideration must be done for other operational conditions. For low loads, achieving Diffusive LTC might not be possible due too very short injection duration. In this case, other type of premixed combustion would be more suitable.

On the contrary, Diffusive LTC was achieved using a micro-hole nozzle and increased injection pressures. For high loads, it might be tested with higher injection pressures, in order to avoid very long fuel injections. One solution might be an injection system with variable pressure to fulfill all load requirements. Moreover, this

579 concept demands a very high intake pressure in order to ensure the appropriate air density. Thus, the turbocharging system must be rethought in order to achieve very high pressures on mid-loads and high-load. 580

5. **CONCLUSIONS**

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- An experimental study on the improvement of Diffusion-flame LTC concept obtained by Benajes et al. [19] 582 aiming to extend its range was carried out. The conclusions are pointed out below: 583
 - It was possible to extend Diffusion-flame LTC range avoiding NO_X and soot formation simultaneously, using the new 6x101 µm nozzle associated to the increased injection pressure. The proposed modifications made possible to reach this combustion mode with 10% oxygen concentration and an intake air temperature of 55°C, which enhanced significantly CO, HC, IMEP if compared to the previous results obtained by Benajes et al.
 - Air density does not affect soot emissions but allows very low oxygen concentration, essential for Diffusionflame LTC to be reached. Also, its effect on reducing ignition delay helps avoid wall impingement and avoid the combustion to be purely premixed.
- 591 Decreasing intake air temperature reduced simultaneously adiabatic flame temperature and equivalence ratio at the lift-off length, contributing to a lower soot formation. Nevertheless, adiabatic flame temperature decrease did 592 not affect NOx, HC, CO and IMEP significantly. 593
 - Oxygen concentration was confirmed to be the fundamental to achieve Diffusion-flame LTC. Its intensive reduction is the main cause for adiabatic flame temperature decrease. Despite the fact NO_X emissions achieved very low level, its complete extinction did not happen. This effect points to a change in NO_X formation mechanism from thermal to prompt. When reducing oxygen concentration, IMEP decreases while HC and CO is kept in very high levels, really unacceptable for a small diesel engine. It occurs due to combustion process deterioration caused by excessive lack of oxygen, reducing global efficiency. The influence of reducing oxygen concentration can be observed on the premixed combustion phase, before flame stabilization. Consequently, there is a reduction in reaction velocity. The slope of RoHR and its peak during premixed combustion tend to reduce. This effect lead to a less intense pressure rise rate and lower combustion noise.
 - Increasing injection pressure enhances mixing process and reduces HC and CO. This improvement is also observed on IMEP. On the contrary, the increase of fuel injected before SoC associated to better mixing capability contributes to a more violent and noisier premixed combustion.
 - Nozzle hydraulic mass flow rate reduction is very effective to achieve Diffusion-flame LTC. Enlarging injection duration increases the time available for the stabilization of a mixing controlled flame. Besides, reducing the amount of delivered fuel before combustion ignition helps reducing combustion noise.
 - Smaller orifice diameters decrease equivalence ratio at the lift-off length. Consequently, soot precursor formation is reduced, affecting engine-out soot emissions. Mixing process improvement is reflected also in CO and HC, with slight reduction on their values.
- The effects of jet-jet interaction have been observed when using 9x083 µm nozzle. In spite of having the smallest 612 orifice diameter, soot emissions were not the lowest, remaining between the conventional nozzle and the 6x101 613 um nozzle. Reducing orifice diameter is a very effective solution, however, if it increases the number of orifices 614 615
 - its effect may be cancel or even reversed.

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