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Additional Information

1 **Monitoring honey adulteration with sugar syrups using an automatic**
2 **pulse voltammetric electronic tongue**

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12 **ABSTRACT**

13 The new tendency to detect adulterated honey is the development of affordable
14 analytical equipment that is in-line and manageable, enabling rapid on site screening.
15 Therefore, the aim of this work was to apply an electronic tongue based on potential
16 multistep pulse voltammetry, in combination with multivariate statistical techniques to
17 detect and quantify syrup in honey. Pure monofloral honey (heather, orange blossom
18 and sunflower), syrup (rice, barley and corn), and samples simulating adulterated honey
19 with different percentages of syrup (2.5, 5, 10, 20 and 40) were evaluated. An
20 automatic, electrochemical system for cleaning and polishing the electronic tongue
21 sensors (Ir, Rh, Pt, Au) significantly improved the repeatability and accuracy of the
22 measurements. PCA analysis showed that the proposed methodology is able to
23 distinguish between types of pure honey and syrup, and their different levels of
24 adulterants. A subsequent PLS analysis successfully predicted the level of the

25 adulterants in each honey, achieving good correlations considering the adjusting
26 parameters. The best results being for sunflower honey adulterated with corn syrup and
27 heather honey with barley syrup ($r^2=0.997$), and heather with corn ($r^2=0.994$) whereas
28 the weakest was found for heather honey adulterated with brown rice syrup ($r^2=0.763$)
29 and orange blossom honey with corn syrup ($r^2=0.879$). The measurement system here
30 proposed could be a very quick and effective option for the honey packaging sector with
31 the finality of providing information about a characteristic as important as the
32 adulteration of honey.

33 **Keywords:** honey adulteration, syrups, pulse voltammetry, electronic tongue

34 **1. Introduction**

35 Honey is a nutritional natural sweetener highly valued for its healing properties (Naila
36 et al., 2018, Cabanero et al., 2006; Padovan et al., 2003; Ruiz-Matute et al., 2010; Bázár
37 et al., 2016). The European Commission has stipulated that nothing should be added to
38 honey (European Commission, 2002), but the limited availability and its price have
39 provided major incentives for adulteration (Anklam, 1998). Honey is adulterated mainly
40 with cheaper sweeteners such as sugar syrups that simulate its own sugar composition
41 (Naila et al., 2018, Li et al., 2017; Sobrino-Gregorio et al., 2017, Cai et al., 2013).
42 Adulterated honey affects the international honey market and the economy of the
43 producers. In addition, it could have negative effects on consumer's nutrition and health
44 (Tosun, 2013). Therefore, guaranteeing the authenticity of honey has become a very
45 important issue for everyone involved in the food chain (Sobrino-Gregorio, 2017).

46 In recent years, a large number of analytical methods have been used to differentiate
47 genuine honey from adulterated. Among them, the most recognized are NMR
48 spectroscopy (Bertelli et al., 2010; Boffo et al., 2012; Davide and Massimo, 2010;
49 Ohmenhaeuser et al., 2013; De Oliveira et al., 2014) and stable carbon isotopic ratio

50 mass spectrometry (SCIRA) (Elfleing & Raezke, 2008; Adnan et al., 2012; Simsek et
51 al., 2012; Tosun, 2013). Another commonly used method is the reflectance-Fourier
52 transforms infrared spectroscopy (Oroian & Ropciuc, 2017, Rios-Corripio et al., 2012;),
53 high performance liquid chromatography (HPLC) to detect starch syrups (Wang et al.,
54 2015), enzymatic activity (diastase, invertase) (Serra et al., 2000), specific markers (Xue
55 et al., 2013) and differential scanning calorimetry (DSC) (Cordella et al., 2002 y 2003;
56 Sobrino-Gregorio et al., 2017).

57 However, using these techniques individually the results obtained are not always
58 conclusive, therefore, to guarantee the purity in honey the combination of several of
59 them is required. Moreover, these techniques are very expensive, they require highly
60 specialized equipment and are time-consuming (Sobrino-Gregorio et al., 2017).

61 To identify the authenticity of honey the industry needs to have simple, fast and easy to
62 handle techniques without the need for expensive equipment and highly skilled workers
63 (Bougrini et al., 2016; Juan-Borrás et al., 2017). Furthermore, the honey sector does not
64 require data of exact levels of adulteration of honey, since any type of addition is
65 prohibited. Only with a screening technique that able is to detect the slightest
66 adulteration is enough.

67 Among the most promising techniques that fulfill this requirement, in addition to being
68 more environmentally friendly than the usual methods, the electronic tongue has the
69 advantage, as it can be an alternative tool to the traditional analytic methods (Bougrini
70 et al., 2016). Unlike the traditional methods, electronic tongues do not obtain
71 information about the nature of the compounds under consideration, but only present a
72 digital fingerprint of the food material (Ghasemi-Varnamkhasti et al., 2010). It is also a
73 qualitative analytical technique that permits recognition, classification or identification

74 of samples, depending on the composition of the sensor array and the mathematical
75 procedure adopted for data treatment.

76 Electronic tongue systems are based on an array of sensors with low selectivity while
77 being sensitive to several components in the measured sample (cross-sensitivity). The
78 signals collected by these sensors are processed by means of pattern recognition tools in
79 order to generate prediction models that allow the classification of the samples and the
80 quantification of some of their physicochemical properties (Gutés et al., 2007).

81 There are several alternatives to electronic tongue systems, the voltammetric being one
82 of the most used (Martínez-Mañez et al., 2005; Lvova et al., 2006; Winqvist et al.,
83 2005), which has different advantages: high sensitivity, versatility, simplicity,
84 robustness and good signal to noise ratio (Winqisk, 2008).

85 These techniques, using arrays of electrodes, is at present the most popular for the
86 design of electronic tongue systems, which include linear voltammetry, differential
87 pulse voltammetry (Bataller et al., 2013), stripping voltammetry and over all cyclic
88 voltammetry (Campos et al., 2010). Cyclic voltammetry is the most widely used
89 technique (Bollo et al., 2004; De Beer et al., 2004; Dogan et al., 2005) and the obtained
90 voltammogram permits the characterisation of electrochemical processes (oxidation–
91 reduction) over a wide potential range. On the other hand, pulse voltammetry is used
92 when higher sensitivity and resolution are required, allowing the detection of lower
93 concentrations of compounds (Escobar et al., 2013). In all cases the enormous amount
94 of data generated by these systems must be processed using appropriate multivariate
95 analysis techniques such as PCA (principal component analysis), LDA (linear
96 discriminant analysis) or CA (cluster analysis) (Benedetti et al., 2004; Dias et al., 2008;
97 Wei et al., 2009).

98 Electronic tongue systems are capable of identifying and classifying liquid samples such
99 as wine, beer, coffee, milk, juices, teas and vegetable oils (Schreyer & Mikkelsen, 2000;
100 Parra et al., 2004; Apetrei et al., 2005; Huang et al., 2007; Chen et al., 2008; Moreno-
101 Codinachs et al., 2008; Rodríguez-Méndez et al., 2008; He et al., 2009; Oliveri et al.,
102 2009; Gutiérrez et al., 2010; Gutierrez-Capitan et al., 2013; Apetrei & Apetrei, 2014;
103 Veloso et al., 2016). Moreover, it is used in quality assessment of solid foods such as
104 meat, fish, fruit and vegetables (Han et al., 2008; Rodríguez-Méndez et al., 2009;
105 Campos et al., 2010; Labrador et al., 2010). In the context of discriminating different
106 classes within the same food type, it has been successfully used in honey, specifically
107 focused on its differentiation according to its botanical and geographical origin (Dias et
108 al., 2008; Wei et al., 2009; Wei & Wang, 2011; Major et al., 2011; Escriche et al., 2012;
109 Garcia-Breijo et al., 2013; Tiwari et al., 2013; Sousa et al., 2014; Bougrini et al., 2016;
110 Juan-Borrás et al., 2017).

111 One of the most promising applications of the electronic tongue systems is the detection
112 of food adulterations. Good results have been reported in the identification of sunflower
113 oil introduced in argan oil (Bougrini et al., 2014) or in the case of goat milk adulterated
114 with bovine milk (Dias et al., 2009). However, little research about the use of electronic
115 tongue systems to identify adulteration of honey has been reported using pulse
116 voltammetry (Men et al., 2014) or cyclic voltammetry (Bougrini et al., 2016; Ropciuc et
117 al., 2017). However, the application of pulse voltammetry, in the above context, could
118 provide important advantages since it has higher sensitivity and resolution (Bataller et
119 al., 2013).

120 Nevertheless, an important problem that limits the use of electronic tongues as a
121 technique for on-line quality controls is that of sensor system cleaning. In the aim to
122 solve this, a mechanical system was developed by Swedish Sensor Center for polishing

123 the electrodes of the electronic tongues (Olsson et al., 2006). However, it significantly
124 limits the utility of the technique in controlling automated processes, as it requires high
125 maintenance and costs. As a cheaper and easier alternative, the same group proposed the
126 use of electropolishing to clean the electronic tongues (Holmin et al., 2004). The
127 procedure consists of applying a high enough voltage to oxidize the surface of the
128 electrodes, and to apply proper cathodic voltage to regenerate the different metal
129 surfaces. Although studies using this technique were promising, the methodology was
130 not optimized for systems that have high concentrations or high levels of contaminant
131 load. Honey falls within the group of substances for which there is no well-defined
132 electropolishing methodology.

133 Taking this into consideration, the aim of this study was to optimize an adequate
134 electropolishing system to investigate the capacity of a pulse voltammetric electronic
135 tongue, which consisted of a set of metal electrodes, to differentiate the presence of
136 syrups in honey samples simulating various levels of adulteration.

137 **2. Materials and methods**

138 **2.1. Samples preparation**

139 Three types of raw honey harvested in 2016, provided by the company Melazahar
140 (Montroy, Valencia), were used in this study: sunflower (*Helianthus annuus*); orange
141 blossom (*Citrus spp.*) and heather (*Erica spp.*). They were selected based on their
142 different physicochemical characteristics (Juan-Borrás et al., 2015). The botanical
143 categorization of all the batches was carried out by means of pollinic analysis following
144 the recommendations of the International Commission for Bee Botany (Von Der Ohe et
145 al., 2004). Microscopic examination, identification and the interpretation of pollen types
146 were carried out by an experienced pollen analyst, using pollen slides and references
147 (Sáenz & Gómez, 2000; Persano-Oddo & Piro, 2004). Furthermore, three kinds of

148 syrups from different origins were used: barley (La Finestra sul Cielo, Italy); corn
149 (Roquette Laissa, Spain) and brown rice (Mitoku Macrobiotic, Japan).

150 The samples evaluated in the present work were: three pure syrups, three pure honeys
151 and a mixture of both in different percentages (40, 20, 10, 5 and 2.5%, respectively)
152 simulating the adulteration of honey. In each case, 8 g of sample (considered on a dry
153 basis) were used. For this, the moisture content was obtained by using a refractometer
154 (Abbe-type model T1; Atago, Bellevue, WA, USA) and the Chataway tables in
155 accordance with the Harmonized Methods of the European Honey Commission
156 (Bogdanov, 2009). All samples were analysed three times achieving four repetitions for
157 each replication.

158 **2.2. Equipment**

159 The measuring equipment is based on a potentiostat designed in the Institute of
160 Molecular Recognition and Technological Development (IDM) at the Universitat
161 Politècnica de València (Campos et al., 2013). This device allows performing pulse
162 voltammetry measurements where the potentials and lengths of the pulses can be
163 configured for each specific application. In this particular work, 40 pulses of 50 ms are
164 applied. The voltages distribution is similar to a stair case voltammetry in increasing (or
165 decreasing) steps of 200 mV between +1 V and -1 V (to avoid water electrolysis), and
166 the potential is set to zero after each increment (Figure 1).

167 The voltammetry tests are measured with four working electrodes (Ir, Rh, Pt and Au)
168 housed inside a stainless-steel cylinder used as the electronic tongue body. A stainless
169 steel circular piece is used as counter electrode and a calomel electrode is used as
170 reference.

171 An integrated system of solenoid valves and a pump permit the automatic injection of
172 liquid samples into a specifically designed measurement chamber. This complete
173 system allows the implementation of an innovative electrochemical polishing of the
174 working electrodes. For each metal, a configuration of basic or acidic solution is used
175 when a sequence of cathodic and anodic pulse (or reversed) is applied to them. The aim
176 is desorbing the organic material accumulated at the surface of the electrodes, and
177 detaching any oxide layer that may have been formed (Table 1).

178 An in-house design of a specific software manages both the measuring equipment and
179 the pumping system. It performs a complete set of measurements with the same setup,
180 and stores the results for a later statistical analysis.

181 This system was patented in 2016, under the name “Sistema y método de control de la
182 calidad del agua en plantas de tratamiento”, which translation would be “System and
183 method to control water quality in treatment plants”, property of the company Fomento
184 Agrícola Castellonense, S.A. and the by the Interuniversity Research Institute for
185 Molecular Recognition and Technological Development (IDM) of the Universitat
186 Politècnica de València, with reference number P201631405 (Bataller et al., 2016).

187 **2.3. Statistical analysis**

188 Multivariate statistical analysis techniques were used to analyse the data gathered for
189 this study. Principal Components Analysis (PCA) was used to discriminate between
190 samples and Partial Least Square (PLS) to quantify the content of honey adulterant in
191 the analysed samples. The PLS model was calibrated with 66% of the data set and
192 validated with the remaining 33%. Model’s assessment is done by comparing real
193 versus predicted adulteration levels. The parameters used are the correlation coefficient
194 (r^2), a, b (from the simplest linear model: $y = ax + b$) and the root mean square error of

195 prediction (RMSEP) as the most common metric obtained to measure accuracy of this
196 methodology (Bataller et al., 2012).

197 All these statistical studies have been performed with Solo 8.6 software (Eigenvector
198 Research, Inc., Wenatchee, Washington, DC, USA).

199 **3. Results and discussion**

200 **3.1. Differentiation of pure honeys and syrups**

201 A PCA analysis was applied (from the data generated by the four electrodes of the
202 electronic tongue) in order to show if there was a classification of the different types of
203 pure samples (honeys and syrups). Figure 2 shows the score plot of this analysis, in
204 which the first two principal components together explain 80.01% of the data
205 variability, specifically 59.59% by PC1 and 20.42% by PC2. Discrimination between
206 honeys and syrups is mainly determined by the X axis (PC1), where the honey samples
207 are in the centre of the score graph and the syrups are placed on both sides (on the left
208 the barley and brown rice syrups, on the right the corn). Since, proximity between
209 samples indicates similar behaviour in terms of the electrochemical response of the
210 sensors, small differences between barley and brown rice syrups with respect to corn
211 syrups were found. On the contrary, the type of honey is differentiated by PC2, where
212 heather honey is in the upper half and sunflower honey in the lower, whereas orange
213 blossom honey is in the middle.

214 Once proven that this methodology could differentiate between all types of pure
215 samples analysed, the next phase was to verify if this type of electronic tongue was able
216 to discriminate honeys in which syrups have been added.

217 **3.2. Differentiation by adulteration levels**

218 Figure 3 shows, as an example, the behaviour of the signal obtained by applying the
219 corresponding potential pulse pattern to sunflower honey adulterated with barley syrup.
220 There is a clear differentiation between the signals obtained for syrup, pure honey and
221 the different percentages of adulteration (40, 20, 10, 5 and 2.5%, respectively). It is
222 evident that the signals are affected by the samples and the adulteration levels. The
223 highest signal corresponds to pure syrup which progressively decreases to pure honey,
224 going through its different and ordered stages of adulteration. This behaviour was in
225 most cases constant, regardless of the type of syrup and honey.

226 For the purpose of evaluating from a descriptive point of view, the global effect of the
227 electronic tongue response in the pure honeys, syrups and their corresponding
228 adulteration levels, different Principal Component Analysis (PCA) were carried out.
229 Figure 4 shows, as an example, the PCA performed in the case of adding barley syrup
230 (40, 20, 10, 5 and 2.5%) to the three pure honeys. In this figure (4.A to 4.C), the two
231 principal components represent 86.5% (PC1: 52.50%; PC2: 34.02%); 83.98% (PC1:
232 67.41%; PC2: 16.57%) and 78.87% (PC1: 61.89%; PC2:19.98%) for sunflower; orange
233 blossom and heather, respectively. Pure barley syrup and honey with 40% of barley
234 syrup are in all cases on the left side of the plots (but in opposite quadrants), whereas
235 the rest of the samples are placed on the right side. In all these PCA plots, a progressive,
236 ordered and clear tendency is observed in relation to the adulteration level. The higher
237 level (40%) is farther away from the pure honey, whereas the lower level (2.5%) is
238 closer to it.

239 Similar results were reported by Bougrini et al., (2016) using cyclic voltammetry for
240 adulteration detection, from 2 to 20%, of pure honey although glucose and sacharose
241 syrups were added. The values of adulteration detected by these authors in honey are
242 even better than those described by them in the case of adulteration in argan oil

243 adulterated with different proportions of sunflower oil (10 to 70%) (Bougrini et al.,
244 2014). Ropciuc et al. (2017), using cyclic voltammetry (with Ag and Au as working
245 electrodes) differentiated honeys adulterated with inverted sugar and malt wort only
246 when they did not exceed 20%.

247 The present work confirms that using pulse voltammetry allowed for further
248 possibilities by designing a specific pulse pattern. In addition to the automation of the
249 electrochemical cleaning process while providing good reproducibility of the sensors
250 and good classification results, it also permitted the detection of a wider adulteration
251 range (up to 40%). Moreover, among other advantages it was observed that the time
252 required to analyse one sample using the pulse voltammetry technique is considerably
253 less: 8 seconds to scan with the 4 electrodes; 40 seconds for 5 iterations; 12 seconds for
254 electropolishing per electrode; 4 minutes for the final cleaning of the sensor system if 5
255 consecutive cleanings are performed. However, considering the protocol described by
256 Bougrini et al. in 2016, the cleaning alone (disassembled, manual cleaning of
257 electrodes, electrochemical cell re-assembled, etc.) takes at least 20 min.

258 **3.3. PLS analysis: correlation of pulse voltammetric data with the level of** 259 **adulteration**

260 In order to verify whether the data provided by the electronic tongue could be useful in
261 predicting the adulteration of pure honeys (sunflower, orange blossom, heather) with
262 syrup (barley, corn, brown rice) at different percentages (40, 20, 10, 5 and 2.5%), a
263 Partial Least Square (PLS) analysis was applied. Nine PLS models of prediction were
264 created (3 honeys multiplied by 3 syrups) with the voltammetric experimental data
265 obtained from the four metallic electrodes (Ir, Rh, Pt, Au). Figure 5 shows one of these
266 PLS graphs (heather honey adulterated with barley syrup) in which measured vs.
267 predicted values of the adulteration levels are plotted together in order to evaluate the

268 performance of the created prediction linear model. Table 2 shows the PLS prediction
269 results (slope, intercept, the regression coefficient, number of latent variables and
270 RMSEP) for the nine models obtained. In most cases, there is a good result, the best one
271 being for sunflower-corn and heather-barley with correlation coefficients of 0.997 and
272 heather-corn with 0.994. The weakest correlation was for heather-brown rice (0.763)
273 and orange blossom-corn (0.879). In order to quantitatively describe the accuracy of
274 model outputs obtained, the RMSEP were shown (Table 2). The best model in terms of
275 capability of prediction corresponded to that obtained for heather-barley (0.834)
276 followed by sunflower- barley (1.252). The worst model was for orange blossom-corn
277 (5.261) and header brown-rice (5.159).

278 Cai, et al., in 2013, applied cyclic voltammetry in Chinese Angelica honey adulterated
279 with rice syrup (from 20% to 50%). They were able to prove that in quantitative
280 analysis of honey adulteration, a multiple linear regression (MLR) model fitted and
281 predicted well with the square of the correlation coefficients ($R_c=0.921$ and $R_p=0.898$).
282 Other authors proposed the combination of PLS with Fuzzy ARTMAP tools to improve
283 the classification of honey adulterated in different proportions (from 0 to 70%) when a
284 voltammetric electronic tongue system is applied (Men et al, 2014).

285 **4. Conclusions**

286 This paper has presented for the first time that an innovative automatic pulse
287 voltammetry can be applied to quantify the presence of syrups in honey. The outcome is
288 the possibility that this type of electronic tongue (with automatic, electrochemical
289 system for cleaning and polishing the electronic tongue sensors) permits detecting this
290 kind of fraud in a bee product to which no addition of substances is allowed. PCA
291 analysis demonstrated that this automatic pulse voltammetry electronic tongue system,
292 made of four metallic electrodes (Ir, Rh, Pt, Au) was capable of not only differentiating

293 between types of pure honey and pure syrups but also to discriminate honeys to which
294 syrups have been added at different levels. The PLS models are capable of predicting
295 the additions of adulterants in different types of honey, and therefore provides a
296 powerful tool to quantify their level of incorporation.

297 The present findings might help to solve the necessity to have a manageable and in-line
298 analytical equipment that enables rapid on site screening and also more affordable for
299 the apiculture sector. However, future studies on the current topic are recommended in
300 order to create a wide and comprehensive data base of pure types of honey from
301 different botanical and geographical origin.

302 **Conflicts of interest**

303 The authors declare that they have no conflict of interest.

304 **Acknowledgment**

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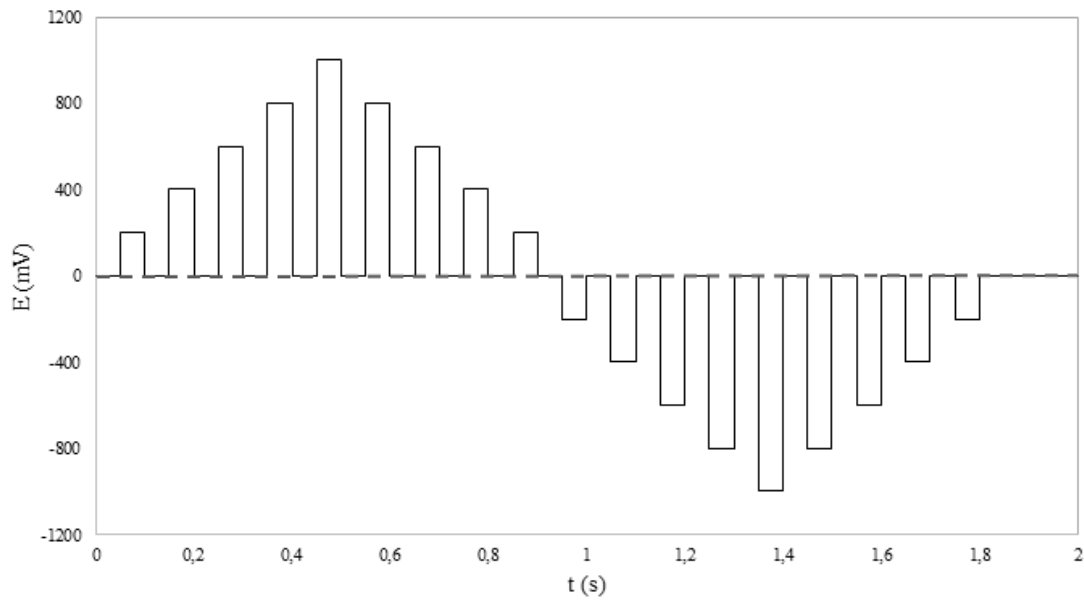
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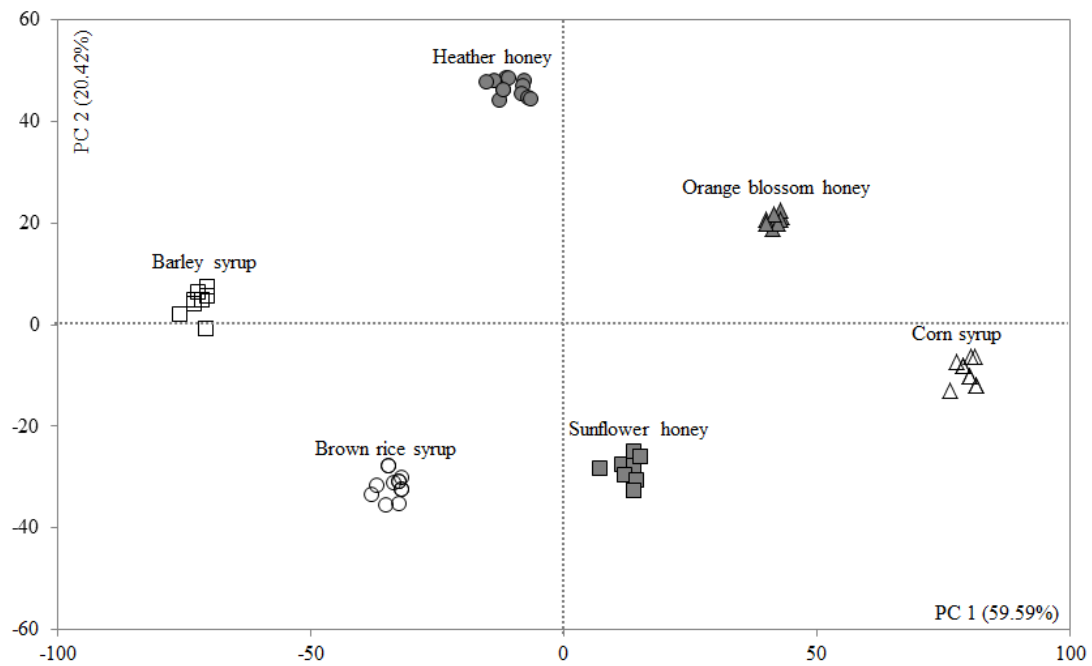
553 **Figure caption**

554 **Figure 1.** Voltammetric pulse pattern.



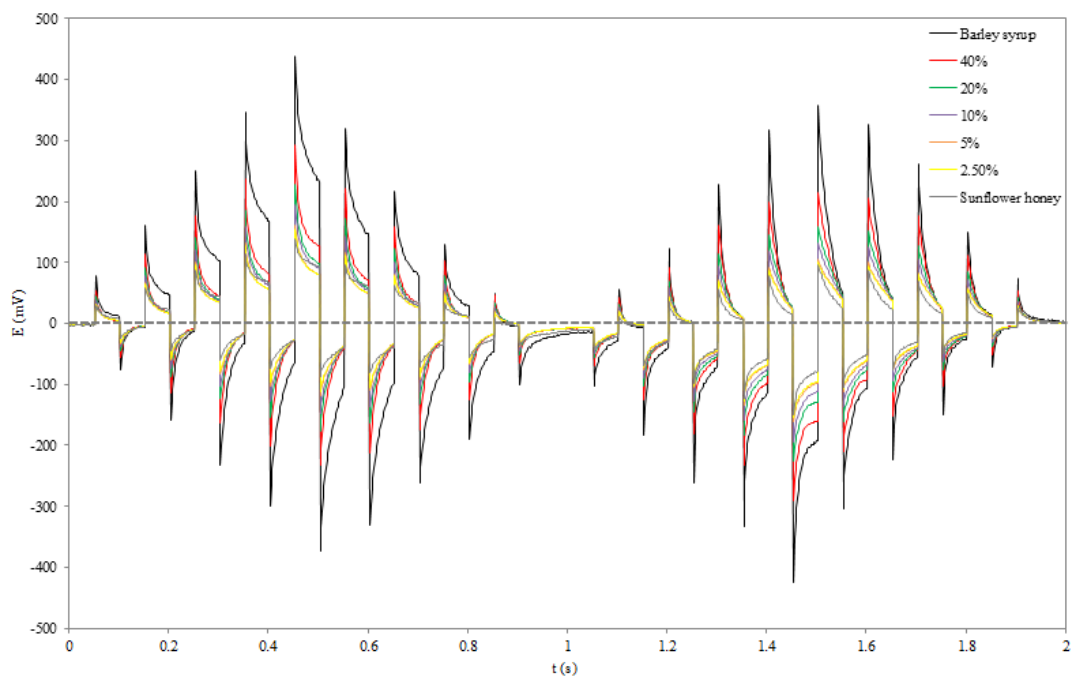
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556 **Figure 2.** Score plot of the PCA performed on pure honeys (sunflower, orange blossom,
557 heather) and pure syrups (barley, corn, brown rice) samples.



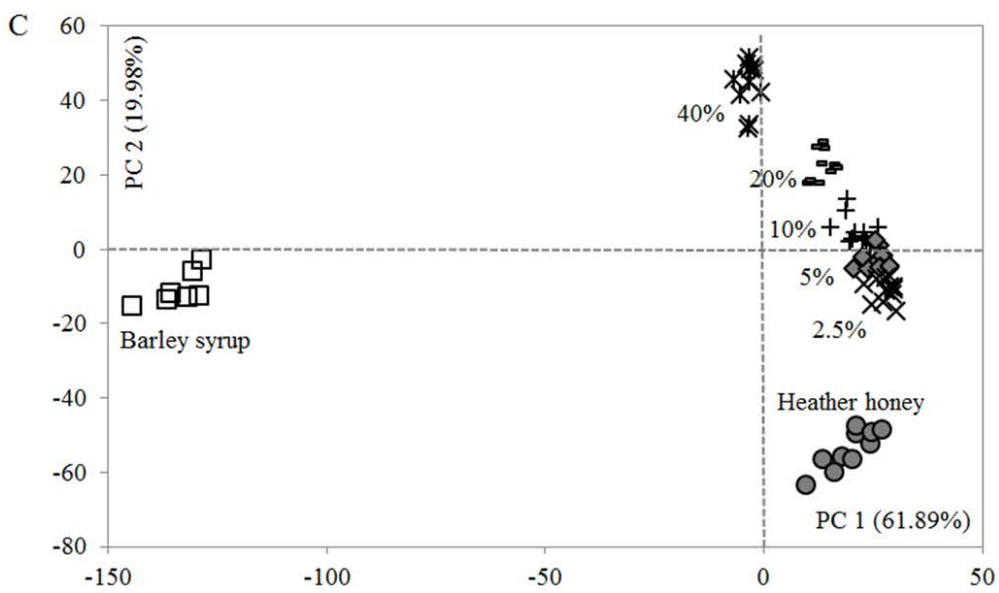
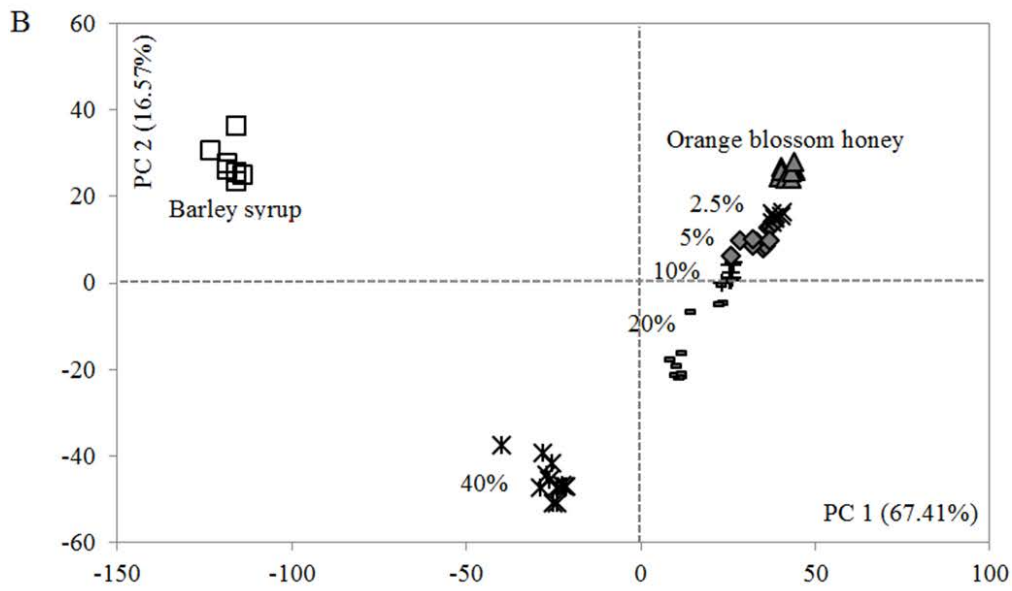
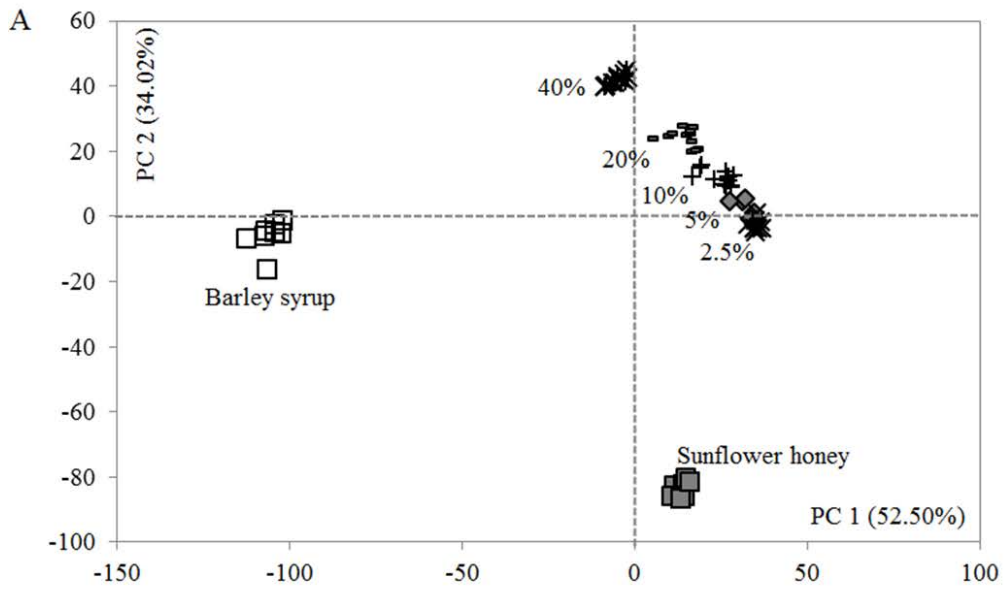
558

559 **Figure 3.** Electrochemical trace of the layered sequence of the potential for sunflower
560 honey adulterated with barley syrup at different levels (40, 20, 10, 5 and 2.5%).

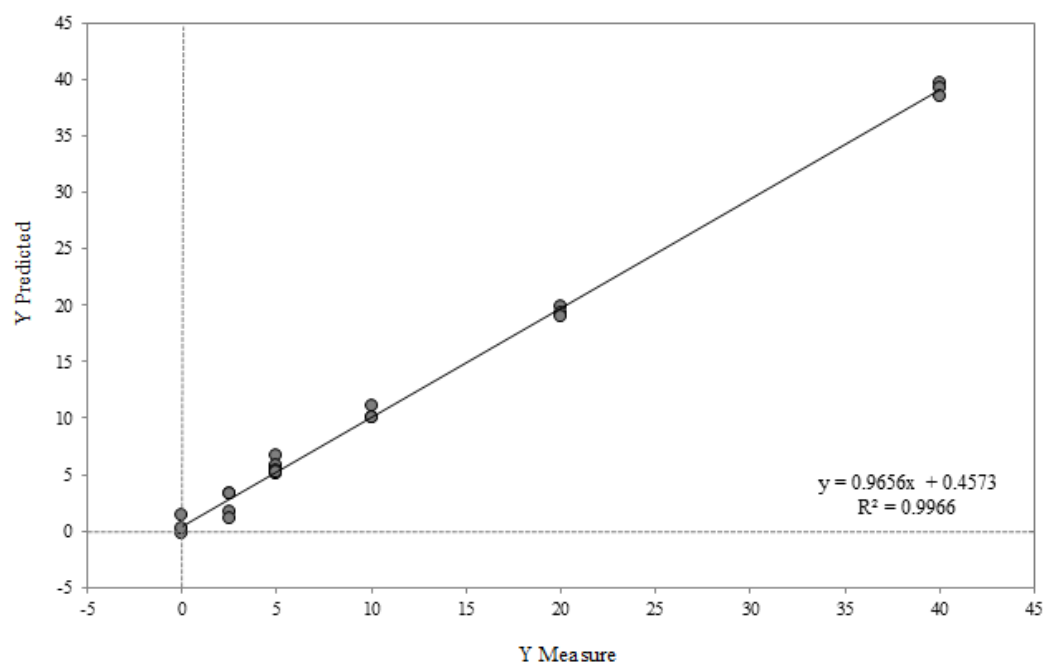


561

562 **Figure 4.** Scores plot of the PCA performed on barley syrup and pure honey (A:
 563 sunflower; B: orange blossom; C: heather hone), and mixtures of both in different
 564 percentages (40, 20, 10, 5 and 2.5 %) simulating the adulteration of honey.



566 **Figure 5.** Predicted versus measured values of heather honey adulterated with barley
 567 syrup given by PLS model.



568

569 **Highlights**

570 -Pulse voltammetry can be useful in detecting and quantifying syrups in honey

571 -This methodology can distinguish between pure honeys and syrups

572 -This methodology can distinguish among different levels of adulterants

573 -PLS analysis can predict the level of adulteration with syrups in honeys

574 Table 1. Electrochemical polishing of the working electrodes: configuration of basic or

575 acidic solution in the sequence of cathodic and anodic pulse applied to the different

576 metals.

Electrodes	Cathodic pulse (mV)	Anodic pulse (mV)	Rest pulse (mV)	Polishing media
Ir	+900	-500	0	Acidic

Rh	+1500	-1500	0	Basic
Pt	+1800	-500	0	Acidic
Au	+1500	-1500	0	Basic

577

578 **Table 2.** PLS prediction results obtained from the validation data for the adulteration of
579 pure honeys (sunflower, orange blossom, heather) with syrup (barley, corn, brown rice)
580 at different percentages (40, 20, 10, 5 and 2.5%).

581

Adulterations	No. latent variables	Correlation coefficient	Slope	Intercept	RMSEP
Sunflower-barley	4	0.991	0.999	0.206	1.252
Sunflower-corn	5	0.997	0.937	1.858	2.622
Sunflower-brown rice	2	0.949	0.909	1.073	3.489
Orange blossom- barley	7	0.993	0.983	0.589	1.336
Orange blossom-corn	6	0.879	0.847	1.234	5.261
Orange blossom- brown rice	6	0.988	1.029	0.203	1.681
Heather- barley	5	0.997	0.966	0.457	0.834
Heather- corn	5	0.994	1.012	0.997	1.479
Heather- brown rice	4	0.763	0.823	3.936	5.159

582