

1 **A hydro-economic modeling framework for optimal**  
2 **management of groundwater nitrate pollution from**  
3 **agriculture**

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10

11 **Abstract:** A hydro-economic modeling framework is developed for determining  
12 optimal management of groundwater nitrate pollution from agriculture. A holistic  
13 optimization model determines the spatial and temporal fertilizer application rate that  
14 maximizes the net benefits in agriculture constrained by the quality requirements in  
15 groundwater at various control sites. Since emissions (nitrogen loading rates) are what  
16 can be controlled, but the concentrations are the policy targets, we need to relate both.  
17 Agronomic simulations are used to obtain the nitrate leached, while numerical  
18 groundwater flow and solute transport simulation models were used to develop unit  
19 source solutions that were assembled into a pollutant concentration response matrix.  
20 The integration of the response matrix in the constraints of the management model  
21 allows simulating by superposition the evolution of groundwater nitrate concentration  
22 over time at different points of interest throughout the aquifer resulting from multiple  
23 pollutant sources distributed over time and space. In this way, the modeling framework  
24 relates the fertilizer loads with the nitrate concentration at the control sites. The benefits  
25 in agriculture were determined through crop prices and crop production functions. This

26 research aims to contribute to the ongoing policy process in the Europe Union (the  
27 Water Framework Directive) providing a tool for analyzing the opportunity cost of  
28 measures for reducing nitrogen loadings and assessing their effectiveness for  
29 maintaining groundwater nitrate concentration within the target levels. The management  
30 model was applied to a hypothetical groundwater system. Optimal solutions of fertilizer  
31 use to problems with different initial conditions, planning horizons, and recovery times  
32 were determined. The illustrative example shows the importance of the location of the  
33 pollution sources in relation to the control sites, and how both the selected planning  
34 horizon and the target recovery time can strongly influence the limitation of fertilizer  
35 use and the economic opportunity cost for meeting the environmental standards. There  
36 is clearly a trade-off between the time horizon to reach the standards (recovery time)  
37 and the economic losses from nitrogen use reductions.

38

39 Key words: *nitrogen management; diffuse groundwater pollution; hydro-economic*  
40 *modelling; optimization; Water Framework Directive.*

41

## 42 **INTRODUCTION**

43 Nitrate is among the most common and widespread pollutants in groundwater. Diffuse  
44 pollution from agricultural activities and livestock are often the main sources of  
45 elevated nitrate concentrations in groundwater (Nolan et al., 1997; EEA, 2003).  
46 Nitrogen is a vital nutrient to enhance plant growth, which has motivated intensive use  
47 of nitrogen-based fertilizers to boost up the crop production. But increased fertilizer use  
48 also has social and environmental costs. When the nitrogen fertilizer application  
49 exceeds plant demand and the denitrification capacity of the soil nitrogen can leach to  
50 groundwater, usually as nitrate, a highly mobile form with little sorption. Nitrate in

51 drinking water has been linked to human health problems like methemoglobinemia in  
52 infants and stomach cancer in adults (Hatch et al., 2002; Wolfe and Patz, 2002),  
53 although the evidence for nitrates as a cause of these diseases remains controversial  
54 (Powlson et al., 2008). Excess nitrates in ecosystems can cause serious environmental  
55 damages, leading to eutrophication of connected surface water bodies that can  
56 eventually provoke algal blooms and fish kills. Agricultural non-point source pollution  
57 is the primary cause of water quality deterioration in many European watersheds (EEA,  
58 1999 and 2003). Although the control of point source emissions improved the quality of  
59 many water bodies across Europe, nitrate concentrations in rivers from diffuse sources  
60 have remained relatively stable in Europe's rivers and groundwater, reflecting the large  
61 nitrogen surplus in agricultural soils and high livestock densities (EEA, 2003).

62

63 Water pollution has given rise to the development of an extensive legal framework. In  
64 Europe, the Nitrates Directive (Directive 91/676/EEC) was established in 1991 to  
65 reduce nitrate water pollution from agricultural sources, and involved the declaration of  
66 Nitrate Vulnerable Zones in which constraints are placed on inorganic fertilizer and  
67 organic slurry application rates. The Drinking Water Directive (80/778/EEC and its  
68 revision 98/83/EC) sets a maximum allowable concentration for nitrate of 50 mg/l. The  
69 EU Water Framework Directive (Directive 2000/60/EC; WFD), enacted in 2000,  
70 proclaims an integrated management framework for sustainable water use, and requires  
71 that all water bodies reach a good status by 2015. The good groundwater status implies  
72 both a good quantitative and a good chemical status. In addition to the groundwater  
73 status, any significant upward trend in the concentration of any pollutant should be  
74 identified and reversed (Directive 2006/118/EC, Groundwater Directive). The WFD  
75 explicitly recognizes the role of economics in reaching the environmental and

76 ecological objectives. Different studies have been conducted to identify economically  
77 efficient groundwater pollution thresholds values (e.g. Brouwer et al., 2006).

78  
79 Nitrate groundwater contamination results from several and complex processes from  
80 pollution sources to water bodies, including pollution formation (nitrogen leaching) and  
81 pollution reactions, fate and transport. Different methods have been reported to analyze  
82 the effects of policies on groundwater nitrate concentration and to find optimal levels of  
83 nitrogen use. Some studies focus on integrating of nitrate leaching into an economic  
84 framework to design nitrogen pollution abatement policies (e.g., Yadav, 1997; Martinez  
85 and Albiac, 2004 and 2006; Kim et al., 1996; Lee and Kim, 2002; Knapp and Schwabe,  
86 2008). In these cases, nitrogen leaching is estimated using a wide range of soil-plant and  
87 nitrogen balance models, but nitrate transport and fate in groundwater is not considered.  
88 Therefore, the natural aquifer's ability to attenuate nitrate concentration is not taken into  
89 account. These approaches do not assess the resulting nitrate concentrations in  
90 groundwater, which are needed to assess if the standards are met or not. Other studies  
91 have applied a compartmental approach, in which the results of a nitrogen management  
92 model are tested using groundwater flow simulation models (e.g., Bernardo et al., 1993;  
93 Mapp et al., 1994). In this case, also the attenuation of nitrate concentrations within the  
94 aquifer is not considered.

95  
96 A more detailed modelling of the bio-physico-chemical processes involved in nitrate  
97 transformation and fate and transport in groundwater is of great importance when  
98 designing optimal nitrogen abatement policies to control groundwater pollution in order  
99 to satisfy certain environmental constraints. Despite the considerable advances in the  
100 development of integrated tools for nitrate transport simulation at the catchment scale

101 (ex. Refsgaard et al., 1999; Lasserre et al., 1999; Birkinshaw and Ewen, 2000) these  
102 modelling frameworks are not usually suitable for integration into management  
103 optimization models for identifying optimal policies. A few studies have proposed  
104 integrated economic-biophysical simulation approaches to assess the evolution of  
105 groundwater quality under different agriculture policies or protection measures, linking  
106 agricultural economic models with soil-plant, nitrogen balance, and groundwater flow  
107 and transport models (e.g., Gömann et al., 2005; Graveline and Rinaudo, 2007a;  
108 Graveline et al., 2007; Almasri and Kaluarachchi, 2007). In Almasri and Kaluarachchi  
109 (2005), a “black-box” statistical modelling approach (artificial neural networks) is used  
110 to relate on-ground nitrogen loadings with nitrate concentrations at specific control sites  
111 in a multicriteria decision framework.

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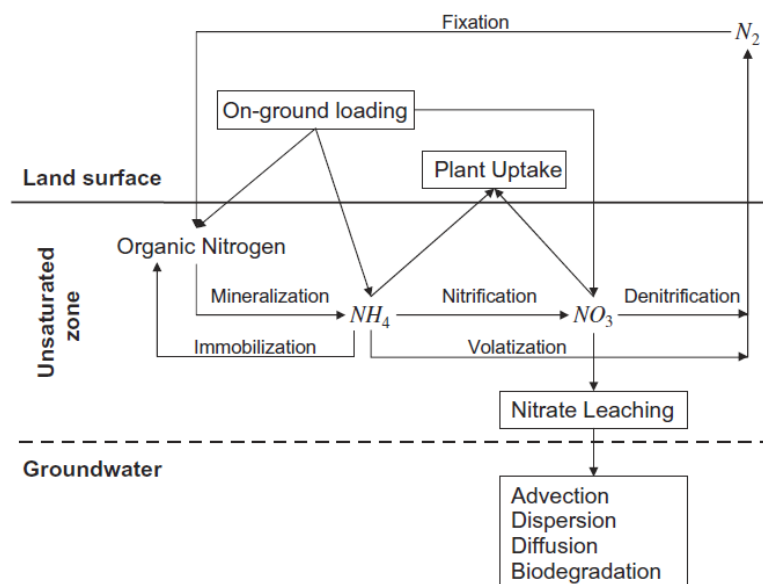
113 The objective of this study is to develop a hydro-economic modelling framework for  
114 optimal management of groundwater nitrate pollution from agriculture. The  
115 optimization modelling framework explicitly integrates nitrate leaching and fate and  
116 transport in groundwater with the economic impacts of nitrogen fertilizer restrictions in  
117 agriculture. This research aims to contribute to the ongoing policy process in the Europe  
118 Union (the Water Framework Directive) by analyzing the cost of measures for reducing  
119 nitrogen loadings and their effectiveness on maintaining groundwater nitrate  
120 concentration within the target levels. With this method we contribute to the  
121 development of the programme of measures to be established by 2012.

122

## 123 **NITRATE GROUNDWATER POLLUTION**

124 Once nitrogen enters the soil, it undergoes several biochemical transformations before  
125 leaching to groundwater mostly as nitrate (Fig. 1). Losses in modern agriculture

126 commonly account for 10-30% of the nitrogen additions (Meisinger et al., 2006). The  
 127 transport and fate of nitrogen in the subsurface environment depends upon the form of  
 128 entering nitrogen and the biochemical and bio-physico-chemical processes involved in  
 129 transforming one form of nitrogen into others. Depending on the sources, nitrogen can  
 130 enter the subsurface environment in organic or inorganic forms; nitrogen from chemical  
 131 fertilizers will typically be in ammonium or nitrate form. The major sources of nitrates  
 132 in groundwater include irrigated and rainfed agriculture and intensive animal operations  
 133 (EEA, 1999). Septic tanks and other sources as landfills can leach nitrates in localized  
 134 areas (Meisinger et al., 2006).



135 **Fig. 1.** Nitrogen groundwater pollution processes.

136 More than 90 % of the nitrogen in soil is organic, either in living plants and animals or  
 137 in humus originating from decomposition of plant and animal residues (Canter, 1996).  
 138 The nitrate content is generally low because it is taken up in synthesis, leached by water  
 139 percolating through the soil, or subjected to denitrification activity below the aerobic  
 140 top layer of the soil. However, synthesis and denitrification rarely remove all nitrates  
 141 added to the soil from fertilizers and nitrified wastewater effluents (Tesoriero et al.,  
 142 2000). Accordingly, nitrates leached from soils are a major groundwater quality

143 problem. Accurate quantification of nitrate leaching to groundwater is difficult due to  
144 the complex interaction between land use practices, on-ground nitrogen loading,  
145 groundwater recharge, soil nitrogen dynamics and soil characteristics. Therefore it is  
146 important to understand the interaction of the aforementioned factors to account for the  
147 transient and spatially variable nitrate leaching to groundwater.

148

149 When nitrogen in the form of nitrate reaches groundwater, it becomes very mobile  
150 because of its solubility. Nitrates can move with groundwater with minimal  
151 transformation and can migrate long distances from input areas if there are highly  
152 permeable subsurface materials that contain dissolved oxygen. This process can be  
153 affected by a decline in the redox potential of groundwater that can lead to a  
154 denitrification process (Tesoriero et al., 2000). Groundwater fate and transport models  
155 are essential for assessing the impact of protection alternative measures that protect  
156 groundwater quality and reduce contamination.

157

## 158 **METHOD**

### 159 *Management Model*

160 An optimization model is developed to define efficient fertilizer allocation in  
161 agriculture: when, where and by how much fertilizer reductions have to be applied to  
162 meet the ambient standards (groundwater quality) in specific control sites in the aquifer.

163

164 The efficient allocation maximizes the present value of the net social benefit. The net  
165 social benefit equals the benefit received from the use of the resource minus external  
166 costs imposed on the society, including costs of damage from pollutants in the  
167 environment. Unless the level of pollution is very high indeed, the marginal damage

168 caused by a unit of pollution increases with the amount emitted, and the marginal  
169 control cost increases with the amount controlled. Efficiency is achieved when the  
170 marginal cost of control is equal to the marginal damage caused by the pollution for  
171 each emitter. The optimal level of pollution is not necessarily the same for all locations.  
172 One way to achieve this equilibrium is to impose legal limits on the pollution allowed  
173 from each emitter, for the level of pollution where marginal control cost equals marginal  
174 damage. Another approach would be to internalize the marginal damage caused by each  
175 unit of emission by a tax or charge on each unit of emissions. To implement these  
176 policy instruments, we must know the level of pollution at which the two marginal cost  
177 curves cross for every emitter, which requires an unrealistically high information burden  
178 on control authorities (Tietenberg, 2002). Another approach is to select ambient  
179 standards, legal upper bounds on the concentration level of specified pollutants in water,  
180 based on some criterion such as adequate margins of safety for human or ecological  
181 health. The allocation of the necessary reduction of emissions for meeting the ambient  
182 standards can be achieved through cost-effective policies. A cost-effective policy results  
183 in the lowest cost allocation of control responsibility consistent with ensuring that the  
184 predetermined ambient standards are met at specified locations called “control sites”.  
185 Since emissions are what can be controlled, but the concentration at the receptor sites  
186 are the policy targets, it is necessary to relate both through the proper numerical  
187 simulation of the pollutants leaching, transport and fate within the aquifer.

188

189 In the proposed hydro-economic modelling framework, the non-point pollution  
190 abatement problem was stated as the maximization of welfare from crop production  
191 subject to constraints that control the environmental impacts of the decisions in the  
192 study region. Welfare was measured as the private net revenue, calculated through crop



193 production functions and data on crops, nitrogen and water prices. The hydro-economic  
 194 model integrates the environmental impact of fertilization by simulation of soil nitrogen  
 195 dynamics and fate and transport of nitrate in groundwater with the economic impact  
 196 (agricultural income losses) of water and fertilization restrictions, assessed through  
 197 agronomic functions representing crop yields and crop prices. The decision variables of  
 198 the problem are the sustainable quantities of nitrogen per hectare applied in the different  
 199 crop areas (pollution sources) to meet the environmental constraints.

200

201 The management model for groundwater pollution control is formulated as:

$$202 \quad Max \Pi = \sum_{c=1}^n \sum_{t=1}^t \frac{1}{(1+r)^t} A_c (p_c \cdot Y_{c,t} - p_n \cdot N_{c,t} - p_w \cdot W_{c,t}) \quad (1)$$

203 subject to:

$$204 \quad [RM]\{cr\} \leq \{q\} \quad (2)$$

205 where  $\Pi$  is the objective function to be maximized and represents the present value of  
 206 the net benefit from agricultural production (€) defined as crop revenues minus fertilizer  
 207 and water variable costs (other costs are not included);  $A_c$  is the area cultivated for the  
 208 crop  $c$ ;  $p_c$  is the crop price (€/kg);  $Y_{c,t}$  is the production yield of crop  $c$  at year  $t$  (kg/ha),  
 209 that depends on the nitrogen fertilizer and irrigation water applied;  $p_n$  is the nitrogen  
 210 price (€/kg);  $N_{c,t}$  is the fertilizer applied to the crop  $c$  at year  $t$  (kg/ha),  $p_w$  is the price of  
 211 water (€/m<sup>3</sup>), and  $W_{c,t}$  is the water applied to the crop  $c$  at year  $t$  (m<sup>3</sup>);  $r$  is the annual  
 212 discount rate,  $[RM]$  is the unitary pollutant concentration response matrix;  $\{q\}$  is a  
 213 column vector of water quality standard imposed at the control sites over the simulation  
 214 time (kg/m<sup>3</sup>);  $\{cr\}$  is a vector of  $n$  elements which corresponds to the nitrate  
 215 concentration recharge (kg/m<sup>3</sup>) reaching groundwater from each crop area, whose  
 216 components are given by:

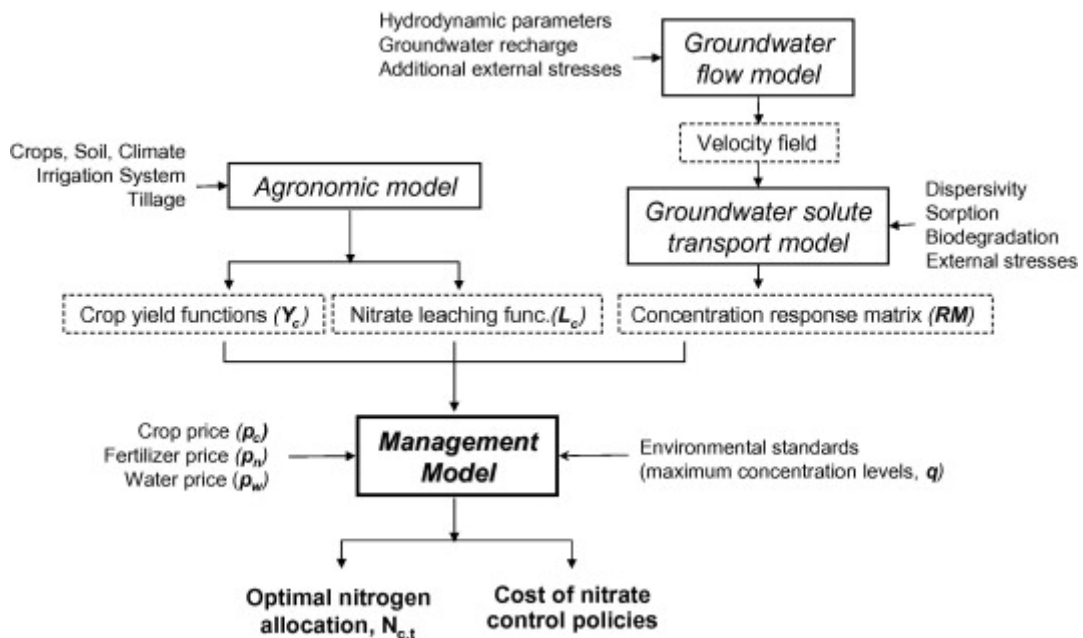
217 
$$cr_t = \frac{L_{c,t}}{r_t} \tag{3}$$

218 where  $r_t$  is the water that recharges the aquifer ( $m^3/ha$ ) at time  $t$ , and  $L_{c,t}$  is the nitrogen  
 219 leached from each crop area ( $kg/ha$ ) at time  $t$ . The sub-index  $t$  in the formulation refers  
 220 to the year within the planning horizon or the number of successive years in which the  
 221 fertilizer is applied.

222

223 The application of the optimization management model requires the integration of the  
 224 soil nitrogen dynamics simulation (to define nitrate leaching) with the simulation of  
 225 groundwater flow and nitrate fate and transport, so that on-ground nitrogen loadings can  
 226 be translated into groundwater nitrate concentrations (Fig 2). Groundwater flow and  
 227 transport governing equations are represented within the management model through  
 228 the pollutant concentration response matrix  $[RM]$ .

229



230

231

Fig. 2. Schematic describing the modelling framework

232 The method of embedding a numerical groundwater simulation model in an  
233 optimization management model as a series of constraints was first described by  
234 Aguado and Remson (1972). The number of model constraints defined using classic  
235 numerical methods can be excessively high, especially in hardly discretized aquifers  
236 (Peralta et al., 1995). When linearity of a system performance can be accepted, the  
237 principles of superposition and translation in time are applicable. Under the assumption  
238 of linear groundwater flow equations (linear boundary conditions and transmissivity  
239 values that do not depend on the hydraulic head), influence functions, discrete kernels  
240 or response matrices have been applied to embed distributed-parameter simulation of  
241 aquifers into conjunctive use management models (Maddock, 1972; Schwarz, 1976,  
242 Morel-Seytoux and Daly, 1975). The main advantage of response matrices is their  
243 condensed representation of external simulation models. The response functions are  
244 incorporated into constraints, coupling the hydrologic simulation with the management  
245 optimization. Gorelick et al. (1979) and Gorelick and Remson (1982) first applied a  
246 response matrix approach in the development of a management model of a groundwater  
247 system with a transient pollutant source.

248 To apply superposition, we need to assume linearity of the system with regard to the  
249 decision variables. For this purpose, in the application of the response matrix approach  
250 to groundwater pollution problems, groundwater flow has to be considered as steady  
251 state, while nitrate transport can be simulated as time dependent (transient) (Gorelick et  
252 al., 1979).

253 Consistently with the steady state assumption, we assume that each crop area provides a  
254 constant recharge to the aquifer and therefore, the groundwater velocity field is time  
255 invariant. The concentration recharge is the quotient of the amount of nitrate leaching  
256 over the volume of water recharge. Treating both factors as unknowns would create a

257 non-linearity with respect to the advective and dispersive transport, both of which  
 258 depends on concentration and velocity. To overcome this, groundwater recharge is  
 259 considered as constant in time. The use of the steady state flow assumption may not be  
 260 suitable for sites with significant hydraulic head variations in time, because of the  
 261 transport simulation errors introduced by ignoring flow transient

262

263 *Nitrate fate and transport and groundwater flow*

264 Solute transport and fate in groundwater depends on the velocity of groundwater flow,  
 265 which can be obtained solving the groundwater flow equation for steady-state flow  
 266 through a saturated anisotropic porous medium (Freeze and Cherry, 1979):

$$267 \quad \frac{\partial}{\partial x} \left( K_x \frac{\partial H}{\partial x} \right) + \frac{\partial}{\partial y} \left( K_y \frac{\partial H}{\partial y} \right) + \frac{\partial}{\partial z} \left( K_z \frac{\partial H}{\partial z} \right) + W = 0 \quad (4)$$

268 where  $K_x$ ,  $K_y$  and  $K_z$  are the hydraulic conductivity values (L/T) in the x, y and z  
 269 directions;  $H$  is the hydraulic head (L) and  $W$  is the flux term (L/T) that accounts for  
 270 pumping, recharge or other sources and sinks.

271

272 The solute concentration throughout the aquifer can be described by the general  
 273 equation for advective-dispersive transport, incorporating equilibrium-controlled  
 274 sorption and first-order irreversible reactions (Zheng and Bennett, 2002):

$$275 \quad R \frac{\partial C}{\partial t} = \frac{\partial}{\partial x_i} \left( D_{ij} \frac{\partial C}{\partial x_j} \right) - \frac{\partial}{\partial x_i} (v_i C) + \frac{q_s}{\theta} C_s - \lambda \left( C + \frac{\rho_b}{\theta} \bar{C} \right) \quad (5)$$

276 where  $C$  is the dissolved concentration ( $M/L^3$ );  $t$  is the time (T);  $\bar{C}$  is the sorbed  
277 concentration ( $M/L^3$ );  $v_i$  is the pore water velocity (L/T);  $q_s$  is the volumetric flow rate  
278 per unit volume of aquifer and represents fluid sources and sinks ( $T^{-1}$ );  $C_s$  is the  
279 concentration of the fluid sources or sink flux ( $M/L^3$ );  $\lambda$  is the reaction rate constant ( $T^{-1}$ );  
280  $\rho_b$  is the bulk density of the porous medium ( $M/L^3$ );  $\theta$  is the porosity  
281 (dimensionless); and  $R$  is the retardation factor.

282

### 283 *Pollutant Concentration Response Matrix*

284 The response matrix describes the influence of pollutant sources upon concentrations at  
285 the control sites over time. Dynamic management of pollutant sources affecting  
286 groundwater quality has been examined by Gorelick et al. (1979), Gorelick and Remson  
287 (1982), Gorelick (1982) or Ahlfeld (1988). The pollutant concentration response matrix  
288  $[RM]$  is a rectangular ( $m \times n$ ) matrix. The number of columns,  $n$ , equals the number of  
289 crop areas (pollution sources) times the number of years within the planning horizon.  
290 The number of rows,  $m$ , equals the number of control sites times the number of  
291 simulated time steps in the frame of the problem (Fig 3). The simulated time horizon  
292 corresponds to the time for the solute to pass all the control sites, and it is independent  
293 of the length of the planning period.

294

		Sources x Planning horizons									
		$S_{1,1}$	$S_{2,1}$	...	$S_{s,1}$	$S_{1,2}$	$S_{2,2}$	...	$S_{s,2}$	...	$S_{s,m}$
Control sites x Time	$O_{1,1}$	$C_{1,1,1,1}$	$C_{1,1,2,1}$		$C_{1,1,s,1}$	$C_{1,1,1,2}$	$C_{1,1,2,2}$		$C_{1,1,s,2}$		$C_{1,1,s,m}$
	$O_{2,1}$	$C_{1,2,1,1}$	$C_{1,2,2,1}$		$C_{1,2,s,1}$	$C_{1,2,1,2}$	$C_{1,2,2,2}$		$C_{1,2,s,2}$		$C_{1,2,s,m}$
	$\vdots$	$\vdots$	$\vdots$		$\vdots$	$\vdots$	$\vdots$		$\vdots$		$\vdots$
	$O_{e,1}$	$C_{1,1,1,1}$	$C_{1,1,2,1}$		$C_{1,1,s,1}$	$C_{1,1,1,2}$	$C_{1,1,2,2}$		$C_{1,1,s,2}$		$C_{1,1,s,m}$
	$O_{1,2}$	$C_{2,1,1,1}$	$C_{2,1,2,1}$		$C_{2,1,s,1}$	$C_{2,1,1,2}$	$C_{2,1,2,2}$		$C_{2,1,s,2}$		$C_{2,1,s,m}$
	$O_{2,2}$	$C_{2,2,1,1}$	$C_{2,2,2,1}$		$C_{2,2,s,1}$	$C_{2,2,1,2}$	$C_{2,2,2,2}$		$C_{2,2,s,2}$		$C_{2,2,s,m}$
	$\vdots$	$\vdots$	$\vdots$		$\vdots$	$\vdots$	$\vdots$		$\vdots$		$\vdots$
	$O_{e,2}$	$C_{2,1,1,1}$	$C_{2,1,2,1}$		$C_{2,1,s,1}$	$C_{2,1,1,2}$	$C_{2,1,2,2}$		$C_{2,1,s,2}$		$C_{2,1,s,m}$
	$\vdots$	$\vdots$	$\vdots$		$\vdots$	$\vdots$	$\vdots$		$\vdots$		$\vdots$
	$O_{of}$	$C_{o,1,1,1}$	$C_{o,1,2,1}$		$C_{o,1,s,1}$	$C_{o,1,1,2}$	$C_{o,1,2,2}$		$C_{o,1,s,2}$		$C_{o,1,s,m}$

295

296

297

Fig. 3. Schematic representation of the pollutant concentration response matrix.

298

Numerical simulation models based on the flow and solute transport governing

299

equations were used to develop the pollutant concentration response matrix.

300

MODFLOW (McDonald and Harbough, 1988), a 3D finite difference groundwater flow

301

model, and MT3DMS (Zheng and Wang, 1999), a 3D solute transport model, were

302

applied to ensemble the pollutant response matrix. First, the field of groundwater

303

velocities is computed using the calibrated groundwater flow model. With the velocity

304

field and the calibrated mass transport model, MT3DMS computes the nitrate

305

concentrations over time (breakthrough curve) at each control site resulting from unit

306

nitrate concentration recharges at each pollution source. These concentration values are

307

assembled as columns to conform the pollutant concentration response matrix.

308

309

For advection-dominated problems, the solution of the transport equation presents two

310

types of numerical problems: numerical dispersion and artificial oscillations (Zheng and

311

Bennett, 2002). The MT3DMS has several solution techniques, the one used here is the

312

third-order TVD scheme based on the ULTIMATE algorithm which is mass

313 conservative, without excessive numerical dispersion, and essentially oscillation-free  
314 (Zheng and Wang, 1999).

315

### 316 *Agronomic simulation*

317 Crop production and nitrogen leaching functions can be derived from agronomic  
318 simulation models like EPIC (Williams, 1995; Liu et al., 2007). GLEAMS (Knisel et  
319 al., 1995; De Paz and Ramos, 2004) and NLEAP (Shaffer et al., 1991; Shaffer et al.,  
320 2008) are also popular models for simulating nitrate leaching. In EPIC, a crop growth  
321 /chemical transport simulation model help defines functions relating crop yield, and  
322 groundwater nitrate leaching to water applied, on-ground nitrogen fertilization and  
323 nitrogen stock in the soil. These functions will depend on local conditions on soils,  
324 climate, irrigation water, tillage, and other operations.

325

326 The crop yield can be defined through crop production functions with the following  
327 polynomial equation:

$$328 \quad Y_c = a + b \cdot W_c + c \cdot W_c^2 + d \cdot N_c + e \cdot N_c^2 + f \cdot W_c \cdot N_c \quad (6)$$

329 where  $Y_c$  is the crop yield (kg/ha),  $W_c$  is the water applied to the crop (m<sup>3</sup>/ha) and  $N_c$  is  
330 the fertilizer applied to the crop (kg/ha). Flexible quadratic function forms are often  
331 used to characterize crop yields (Doorenbos and Kassam, 1979; Vaux and Pruitt, 1983;  
332 Zhengfei et al., 2006). The coefficients of the equation (a, b, c, d, e, and f) are calibrated  
333 for the best fit to the values obtained through an external agronomic simulation model.

334

335 The amount of leaching and hence the amount of nitrates in groundwater is a function of  
336 the timing of fertilizer application, vegetative cover, soil porosity, fertilizer application  
337 method, and irrigation rate (Canter, 1996). After the plant uptake and transformation,

338 some of that nitrogen applied is converted into nitrate that can leach to the aquifer. The  
339 amount of nitrate leached is then introduced into the management model through  
340 quadratic functions of water applied and nitrogen fertilization, also this functions are  
341 often used to characterize nitrate leaching (Calatrava and Garrido, 2001; Martinez and  
342 Albiac, 2004;) as follows:

$$343 \quad L_c = g + h \cdot W_c + i \cdot W_c^2 + j \cdot N_c + k \cdot N_c^2 + l \cdot W_c \cdot N_c \quad (7)$$

344 where  $L_c$  is the nitrogen leached (kg/ha),  $W_c$  is the water applied to the crop (m<sup>3</sup>/ha) and  
345  $N_c$  is the fertilizer applied to the crop (kg/ha). The coefficients of the equation (g, h, i, j,  
346 k, and l) are calibrated for the best fit to the values obtained through an external  
347 agronomic simulation model.

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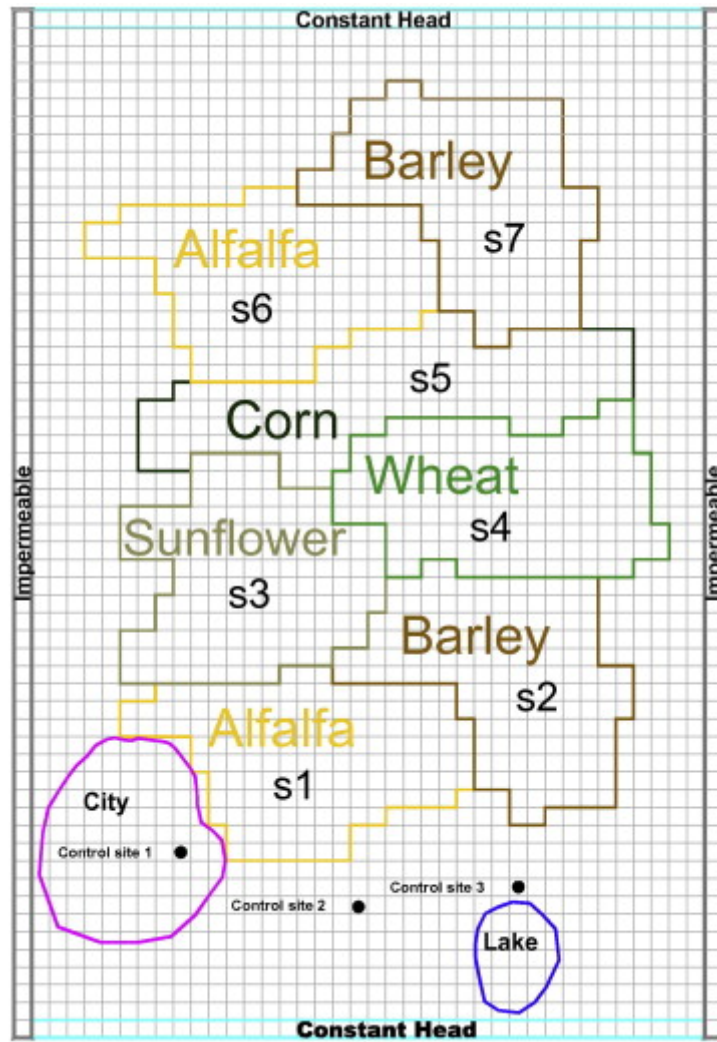
## 349 **APPLICATION OF THE MODELLING FRAMEWORK**

### 350 **Illustrative example**

351 The modelling framework was applied to a hypothetical groundwater system (Fig. 4).  
352 The aquifer has impermeable boundaries and steady flow from the top to bottom of the  
353 Figure. The finite difference grid is 500 x 500 meters. The system parameters are  
354 hydraulic conductivity of 40 m/day, aquifer thickness of 10 meters, effective porosity of  
355 0.2, and dispersivity of 10 meters. The natural recharge is 500 m<sup>3</sup>/ha. There are 70 stress  
356 periods, each of one year (365 days). Seven crop zones with five different crops are  
357 considered. For each crop a quadratic production function and a leaching function have  
358 been defined. Each source is related to a crop as shown in Figure 4. The coefficients  
359 used for the production and nitrate leaching functions are shown in Table 1. Three  
360 control sites with concentration upper bounds (maximum of 50 mg/l of nitrates) are  
361 defined.

362





363

364 Fig. 4. Aquifer system

**Table 1**  
Production function and nitrogen leaching coefficients.

Crop	a	b	c	d	e	f
<i>Production functions coefficients</i>						
Alfalfa	4.43E+00	2.63E-02	-1.62E-05	4.68E-02	-3.45E-04	0.00E+00
Barley	-3.68E-01	6.06E-03	-1.02E-05	1.88E-02	-5.15E-05	0.00E+00
Sunflower	4.37E-01	6.80E-04	-9.70E-06	3.12E-02	-1.40E-04	5.40E-05
Wheat	6.11E-01	3.90E-03	-3.40E-05	4.60E-02	-1.30E-04	5.00E-05
Corn	-1.30E+01	3.80E-02	-2.40E-05	6.70E-03	-7.20E-05	5.17E-05
	g	h	i	j	k	l
<i>Leaching functions coefficients</i>						
Alfalfa	-7.04E+00	-3.69E-03	1.36E-05	9.69E-03	1.02E-03	0.00E+00
Barley	-1.96E+01	-1.15E-03	2.20E-04	-2.04E-02	5.06E-04	0.00E+00
Sunflower	0.00E+00	0.00E+00	-3.44E-04	7.68E-01	-2.25E-03	1.34E-03
Wheat	0.00E+00	4.36E-02	0.00E+00	3.05E-01	1.30E-04	-1.17E-04
Corn	0.00E+00	4.40E-03	-6.69E-05	3.96E-01	0.00E+00	0.00E+00

365

366

**Table 2**  
Sources, crops and irrigation.

Source	Crop	Area (ha)	Water applied (m <sup>3</sup> /ha)	Crop price (€/kg)
S1	Alfalfa	3600	950	0.09
S2	Barley	3600	300	0.12
S3	Sunflower	3600	400	0.30
S4	Wheat	3600	250	0.13
S5	Corn	3600	700	0.12

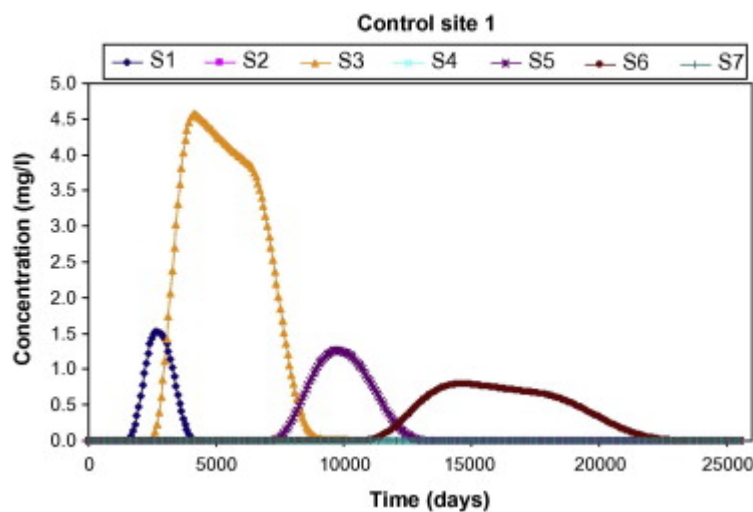
367

368 The irrigation water applied was kept constant at the level where the crop yield is  
 369 maximum (Table 2). The fertilizer price is 0.60 €/kg.

370

371 **Pollutant concentration response matrix and breakthrough curves**

372 The response matrix is generated by simulating the effects of a fertilizer application of  
 373 200 Kg/ha and an annual recharge of 500 m<sup>3</sup>/ha. Using the corresponding concentration  
 374 recharge as “unit” recharge rate at each source, the breakthrough curves (nitrate  
 375 concentration time series) for the different sources were generated using MODFLOW  
 376 and MT3DMS. For the solute transport simulation only advection and dispersion were  
 377 considered, and the simulation time horizons were determined by the time for which the  
 378 solute completely passed the control sites. Breakthrough curves were obtained for each  
 379 crop area and for the three different control sites (Fig. 5).



380

381

382

*Fig. 5. Breakthrough curve for the control site 1.*

383 *Crop area S3* (sunflower) is the nitrate source with the greatest influence on control  
384 sites 1 and 2, followed by *S1*. Source *S3* has greater influence than sources *S1* and *S2*,  
385 despite these areas are closer to the control sites (Fig. 4), since nitrate leaching  
386 concentration from *S3* is higher than from the other crop areas. *S5* (corn) is the only  
387 pollution source with a significant impact on the three control sites.

388

### 389 **Scenarios and results**

390 Five different scenarios have been considered to illustrate the applicability of the  
391 proposed approach. In the scenario 0 or base case, no ambient standards are considered,  
392 and the fertilizer applied is the one that yields the highest benefit. In scenarios 1 to 4, a  
393 maximum nitrate concentration of 50 mg/l is imposed at the three control sites as  
394 follows:

- 395       ▪ Scenario 1. The initial solute concentration in groundwater is zero, and the  
396       fertilizer application can vary in space and time.
- 397       ▪ Scenario 2. The initial solute concentration in groundwater is zero and the  
398       fertilizer application is restricted to be the same over the planning horizon.
- 399       ▪ Scenario 3. The initial solute concentration is 55 mg/l throughout the aquifer,  
400       and the fertilizer application can vary in time and space. For this scenario four  
401       different recovery times were considered: 10, 20, 30 and 40 years.
- 402       ▪ Scenario 4. The initial concentration is 55 mg/l and the fertilizer application is  
403       restricted to be the same for all the management periods.

404

405 For each scenario, four planning horizons (10, 20, 30 and 40 years) were considered to  
406 test the influence of the planning horizon on the optimal nitrate management and its  
407 economic and environmental impacts.

408

409 The model was coded in GAMS, a high-level modelling system for mathematical  
410 programming problems (GAMS, 2008). The non-linear problem to be solved has 1681  
411 variables and 2939 constraints. The MINOS solver was used to find the optimal  
412 solution.

413

#### 414 **Scenario 0. No nitrate standard**

415 This scenario is a reference case with no nitrate standard and the aquifer not initially  
416 polluted. Therefore, the resulting fertilizer application is the one that yields the  
417 maximum aggregated net benefit, without constraining nitrate pollution. The optimal  
418 fertilizer distribution in space and time was calculated for 10, 20, 30 and 40 year  
419 planning horizons. The longer the considered planning horizon, the higher the peak  
420 concentration of nitrate.

421

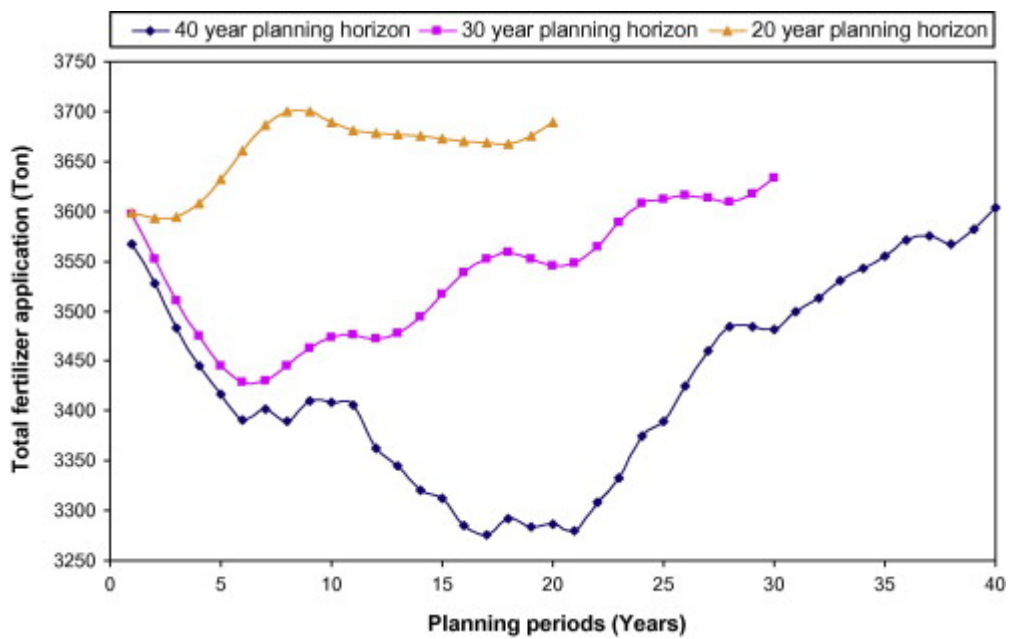
422 While for the 10 year planning horizon the maximum concentration is below the current  
423 standard, the nitrate standard is exceeded for 20 year and longer planning horizons (64  
424 mg/l would be reached in the 40 year planning horizon case). Since in all the planning  
425 horizons the optimal fertilizer application would be the same (3731 ton/year on  
426 average), an equal annual benefit (20.96 M€/year) would be obtained.

427

#### 428 **Scenario 1. Variable fertilizer application.**

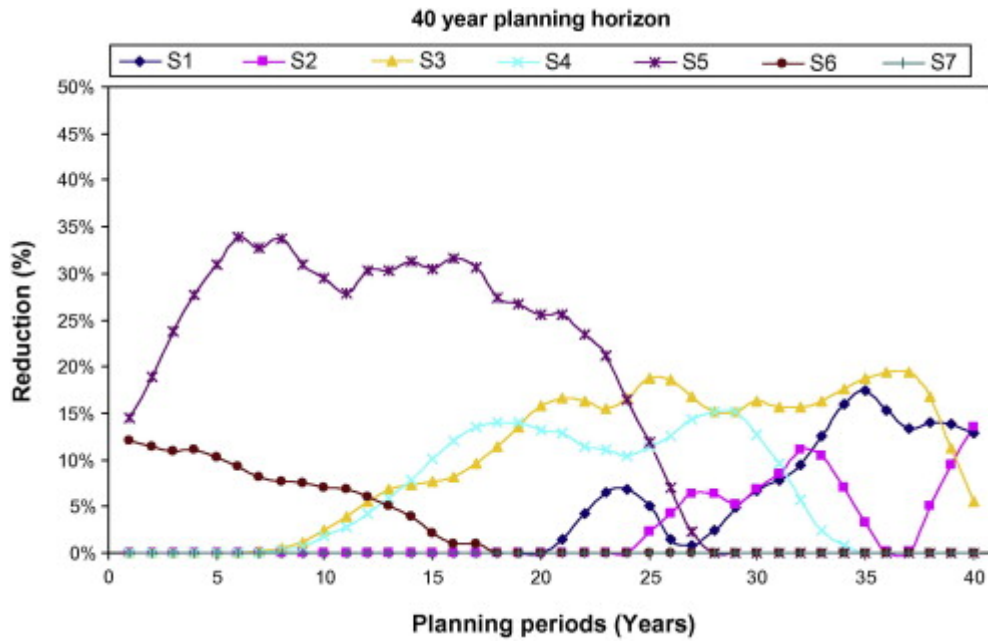
429 For the 10 year planning horizon, the fertilizer application was the same as that  
430 providing the maximum benefits, since the ambient standard was not reached at any of  
431 the control sites. However, for longer planning horizons (20, 30, and 40 years) the  
432 fertilizer application was reduced to keep nitrate concentrations at the control sites

433 below 50 mg/l. Figure 6 shows the optimal fertilizer application for the different  
 434 planning horizons, showing the application is further reduced as the planning horizon  
 435 increases, since there is an extension in time of the application of the fertilizer loading.  
 436 From here on, only the results for the 40 year management period will be shown, a  
 437 representing long-term management.  
 438



439  
 440 Fig. 6. Total fertilizer application for different planning horizons. Scenario 1.

441  
 442 Figure 7 shows the reduction of fertilizer application corresponding to each source with  
 443 regards to the fertilizer application of maximum crop yield. The level of sustainable  
 444 fertilizer loading reduction differs with location depending on its influence upon the  
 445 nitrate concentration at the control sites and the economic losses from crop yield  
 446 reduction. According to this Figure, crop area S5 (corn) requires the most fertilizer  
 447 reduction, reaching a 30% reduction during the first 30 years. As shown in Figure 5, this  
 448 crop area strongly influences nitrate concentration at the 3 sites.



449

450

Fig 7. Spatial and temporal reduction of fertilizer application. Scenario 1.

451

452

453 The arrival time of the peak nitrate concentration to the control sites differs for each

454 source; therefore, the optimal timing and magnitude of fertilizer reduction to meet the

455 environmental targets will differ for each source. Figure 8 shows the times series of

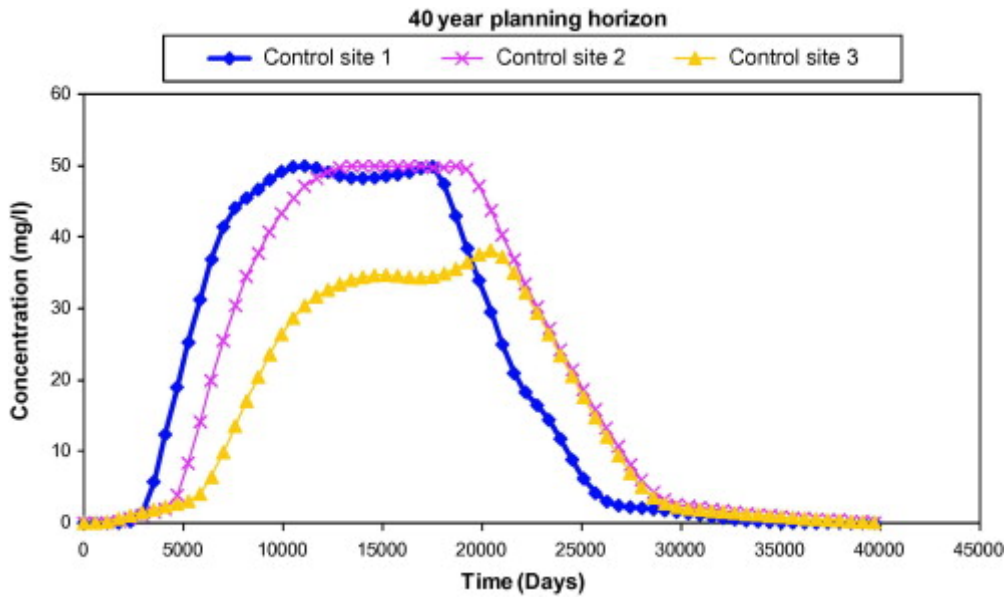
456 nitrate concentration for the optimal fertilizer application at the 3 control sites. Figure 8

457 shows that nitrate concentrations are maintained below the ambient standard of 50 mg/l.

458 While the concentrations at control site 1 and 2 are close to the limit, the values at

459 control site 3 are notably below.

460



461

462

Fig. 8. Time series of nitrate concentration. Scenario 1.

463

464 Table 3 shows the economic impacts of different planning horizons. The longer the  
 465 planning horizon, the higher the reduction in fertilizer application, with lower average  
 466 benefits per year.

**Table 3**  
 Fertilizer application and benefit for different planning horizons. Scenario 1.

Planning horizon (years)	Total annual fertilizer application (ton/year)	Total benefit (Me/year)
10	3731	20.96
20	3660	20.93
30	3533	20.83
40	3429	20.76

467

468 **Scenario 2. Constant fertilizer application.**

469 Scenario 2 illustrates the case where the fertilizer application is kept constant through  
 470 the years, which is obviously not the economically optimal solution but represents a  
 471 simpler management alternative. Table 4 shows the fertilizer application and the  
 472 percentage of fertilizer reduction from the loading that produces the maximum crop  
 473 yield that is required to meet the ambient standards. Crop area *S5* (corn) again has the  
 474 highest fertilizer reduction, followed by *S3* (sunflower).

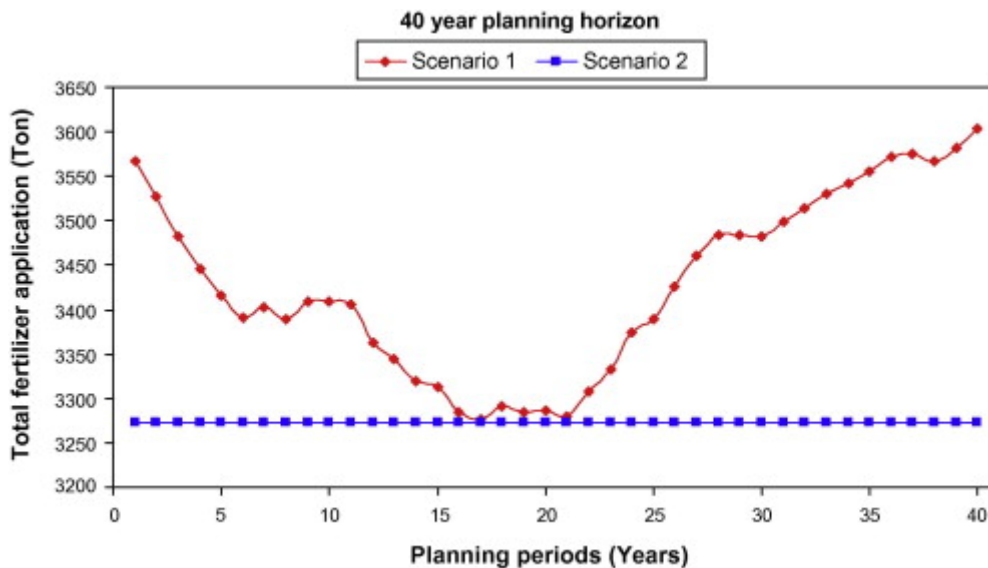
**Table 4**  
Constant fertilizer application and percentage of fertilizer reduction. Scenario 2.

Source	Crop	Fertilizer application (kg/ha)	Fertilizer reduction (%)
S1	Alfalfa	50.1	13.9
S2	Barley	124.1	7.5
S3	Sunflower	151.9	16.2
S4	Wheat	180.3	13.0
S5	Corn	183.7	30.2
S6	Alfalfa	55.8	4.1
S7	Barley	134.1	0.0

475

476 Comparing the fertilizer application in scenarios 1 and 2 (Fig. 9) we conclude that when  
 477 the fertilizer application is constant over time (scenario 2) the total fertilizer application  
 478 has to be reduced to meet the constraints. Over time, both curves get closer up to the  
 479 point in which the minimal fertilizer application in scenario 1 reaches the value in  
 480 scenario 2. Since scenario 2 presents the highest reductions in fertilizer applications, the  
 481 benefits for agriculture are consequently lower (20.50 against 20.96 M€/year).

482



483

484

Fig. 9. Comparison between scenarios 1 and 2.

485

**Scenario 3. Recovery from pollution.**

486

The EU Water Framework Directive requires determining the most cost-efficient  
 487 combination of measures to reduce nitrate concentration in polluted groundwater bodies  
 488 below the standard (50 mg/l). In this scenario, an initial uniform nitrate concentration of  
 489 55 mg/l was considered, and the objective was to find the optimal fertilizer application

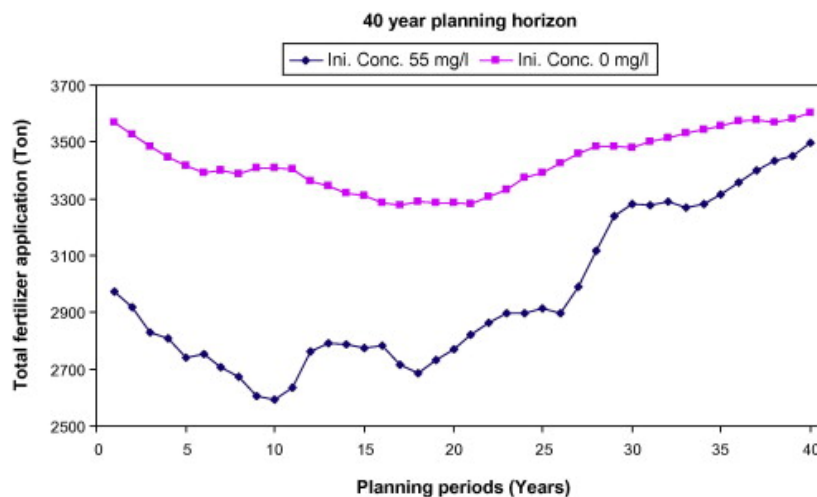


490 to reduce nitrate groundwater concentrations to 50 mg/l for different recovery time  
491 horizons (10, 20, 30 and 40 years). The recovery time horizons were imposed in the  
492 management model by setting the maximum concentration constraint at the specific  
493 recovery time and beyond.

494

495 Figure 10 shows the fertilizer application for the scenarios 1 (initially unpolluted  
496 aquifer) and 3 (initially polluted aquifer) with a 40 year recovery time horizon. The  
497 fertilizer application is higher for scenario 1 than for scenario 3 to reduce the initial  
498 nitrate concentrations. However, both applications converge over time, once the effect  
499 of the initial concentration has been lowered by natural attenuation.

500



501

502 Fig. 10. Comparison between scenarios 1 and 3.  
503

504 Table 5 shows the benefits for the different recovery times. The difference in benefits  
505 between the more constrained case (10 year recovery time) and the 40 years of recovery  
506 is €230,000/year.

507

**Table 5**  
Total benefits for different recovery times. Scenario 3.

Recovery time (years)	Total fertilizer application (ton/year)	Total annual benefits (M€/year)
10	2898	19.43
20	2917	19.45
30	2921	19.53
40	2964	19.66

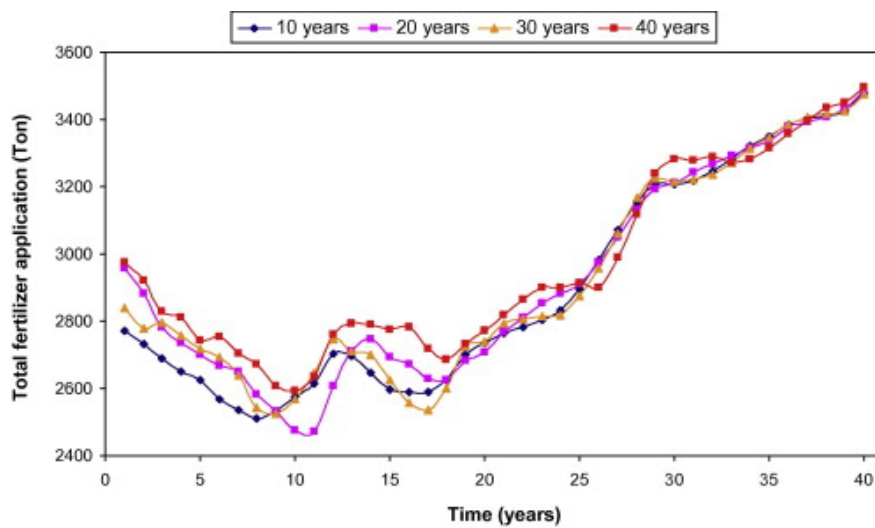
508

509

510 Figure 11 depicts the total fertilizer application that corresponds to the different  
511 recovery time horizons.

512

513 Longer recovery time horizons increase total fertilizer application (concentrations must  
514 be reduced faster for shorter recovery times). However, the differences decrease over  
515 time.



516

517 Fig 11. Total fertilizer application for different recovery times. 40 year planning horizon.  
518 Scenario 3.

519

520 **Scenario 4. Constant fertilizer application with initial pollution.**

521 In this scenario the aquifer is considered polluted with an initial uniform concentration  
522 of 55 mg/l, and the fertilizer application is kept the same throughout the planning  
523 horizon.

524

525 Comparing scenarios 3 and 4 for the 40 year planning period case, there is a significant  
526 reduction in the benefits from agriculture (€580,000/year) when the fertilizer is kept  
527 constant, although the difference in the average fertilizer application is only 15 kg/ha-  
528 year.

529

530 Some researchers (e.g., Yadav, 1997; Martinez and Albiac, 2004) have performed cost-  
531 effectiveness analysis of groundwater pollution control policies as if the ambient  
532 standards were imposed at every location in the aquifer, and therefore, the pollutant  
533 concentration recharge is implicitly limited to 50 mg/l. The same case was simulated  
534 and compared with the results previously obtained imposing nitrate concentration limits  
535 only at the three control sites. Table 6 shows the total fertilizer reduction required for  
536 maintaining nitrate concentration below 50 mg/l throughout the aquifer, showing that no  
537 fertilizer reductions are required for some crops, since the quantity of fertilizer that  
538 yields the highest crop production can be applied without exceeding the ambient  
539 standard. However, other crops (sunflower, wheat, corn) require a big reduction in  
540 fertilizer loads. With these fertilizer application rates, the maximum nitrate  
541 concentration at the control points stays below 20 mg/l, far from the limit of 50 mg/l.  
542 Because of the further reduction in fertilizer application, the average benefits are  
543 considerable smaller (17.09 M€/year versus 19.08 M€/year).

**Table 6**  
Fertilizer application and fertilizer reduction for the case where the concentration  
recharge is below 50 mg/l.

Source	Crop	Fertilizer application (kg/ha)	Fertilizer reduction (%)
S1	Alfalfa	58.2	0
S2	Barley	134.1	0
S3	Sunflower	69.8	62
S4	Wheat	50.0	76
S5	Corn	138.1	48
S6	Alfalfa	58.2	0
S7	Barley	134.1	0

544

545

546 **CONCLUSIONS**

547 In recent decades, nitrate concentrations in groundwater have increased due to the  
548 intensive use of fertilizers in agriculture. In Europe, the EU water legislation establishes  
549 a limit of nitrate concentration in groundwater bodies of 50 mg/l, and requires that  
550 groundwater bodies reach a good quantitative and chemical status by 2015. To control  
551 groundwater diffuse pollution is necessary to analyse and implement management  
552 decisions.

553

554 This paper describes the development and application of a method for exploring optimal  
555 management of groundwater nitrate pollution from agriculture. The model suggests the  
556 spatial and temporal fertilizer application rate that maximizes the net benefits in  
557 agriculture constrained by the quality requirements in groundwater at specific control  
558 sites. The analysis accounts for key underlying biophysical processes linked to the  
559 dynamics of nitrogen in the soil and the aquifer, as well as the crop yield responses to  
560 water and fertilizer application. External soil-plant agronomic models, and groundwater  
561 flow and solute transport simulation models are used to obtain influence or response  
562 functions that are integrated into the optimization model, translating nitrogen applied on  
563 the surface into nitrates at wells or other points of interest throughout the aquifer, so the  
564 effectiveness of measures can be assessed in terms of reduction of nitrate concentrations  
565 within the groundwater body. Unlike simulation approaches, the management model  
566 automatically generates optimal solutions for a very complex problem. Instead of  
567 resorting to black-box statistical models, the fate and transport of nitrates within the  
568 aquifer is explicitly simulated in the optimization model using a pollutant concentration  
569 response matrix under the assumption of steady-state flow. The concentration response

570 matrix shows the concentration over time at different control sites throughout the  
571 aquifer resulting from multiple pollutant sources distributed over time and space.

572

573 The method was applied to an example under five scenarios. Optimal solutions to  
574 problems with different initial conditions, planning horizons and recovery times were  
575 found. The case study shows how both the selected planning horizon and the target  
576 recovery time can strongly influence the limitation of fertilizer use and the economic  
577 opportunity cost for reaching the environmental standards. There is clearly a trade-off  
578 between the time horizon to reach the standards (recovery time) and the economic  
579 losses from nitrogen use reductions.

580

581 This method can contribute to implementing the EU Water Framework Directive by  
582 providing insights for the definition of cost-efficient policies or program of measures to  
583 control diffuse groundwater pollution. The modelling framework allows estimation of  
584 the opportunity cost of measures to reduce nitrogen loadings and their effectiveness for  
585 maintaining groundwater nitrate concentration within the target levels. The method also  
586 can be applied to identifying economically efficient “good quality status” threshold  
587 values. Finally, it can be used to justify less stringent environmental objectives based on  
588 the existence of disproportionate cost (for cases in which opportunity costs surpass the  
589 expected benefits) or to ask for deadline extensions when it is not feasible or the  
590 objectives cannot “reasonably” be achieved within the required timescales.

591

592 Additional work to assess the influence of uncertainty in the different parameters of the  
593 model would be required. A stochastic modelling framework can be derived from the  
594 proposed methodology. The modelling framework can be used to test the effects of

595 different policies such as water prices, nitrogen taxes, nitrogen standards, subsidies, etc.  
596 Finally, the method can be extended to consider other sources of nitrate pollution such  
597 as animal farming, landfills, and septic tanks. Although the method and tools are  
598 suitable for simulating the effects of these sources on nitrate concentration at the control  
599 sites, further research would be required for modelling the economics of abating the  
600 pollution from these other sources.

601

602

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