

Document downloaded from:

<http://hdl.handle.net/10251/153841>

This paper must be cited as:

Nigar, H.; Sturm, GSJ.; García-Baños, B.; Penaranda-Foix, FL.; Catalá Civera, JM.; Mallada, R.; Stankiewicz, A.... (2019). Numerical analysis of microwave heating cavity: Combining electromagnetic energy, heat transfer and fluid dynamics for a NaY zeolite fixed-bed. *Applied Thermal Engineering*. 155:226-238.  
<https://doi.org/10.1016/j.applthermaleng.2019.03.117>



The final publication is available at

<https://doi.org/10.1016/j.applthermaleng.2019.03.117>

Copyright Elsevier

Additional Information

1 **Numerical Analysis of Microwave Heating Cavity: Temperature distribution in**  
2 **time and space within a NaY zeolite fixed-bed**

3  
4 Nigar H<sup>a</sup>, Sturm G.S.J.<sup>b</sup>, Garcia-Baños B.<sup>c</sup>, Peñaranda-Foix F.L.<sup>c</sup>, Catalá-Civera J.M.<sup>c</sup>,  
5 Mallada R.<sup>a,d\*</sup>, Stankiewicz, A.<sup>b</sup>, Santamaría J.<sup>a,d\*</sup>

6 <sup>a</sup> Nanoscience Institute of Aragon and Chemical and Environmental Engineering  
7 Department, University of Zaragoza, 50018 Zaragoza, Spain

8 <sup>b</sup> Process & Energy department, Delft University of Technology, Leegwaterstraat 39,  
9 2628 CB Delft, The Netherlands

10 <sup>c</sup> Instituto ITACA, Universidad Politecnica de Valencia, Camino de Vera, 46022,  
11 Valencia, Spain

12 <sup>d</sup> Networking Research Centre CIBER-BBN, 28029 Madrid, Spain

13 \*Corresponding authors: [rmallada@unizar.es](mailto:rmallada@unizar.es), [jesus.santamaria@unizar.es](mailto:jesus.santamaria@unizar.es)

14 **Keywords**

15 Modelling and numerical simulation, Dielectric properties, Microwave heating, Power  
16 dissipation, Transient temperature profiles

17 **Abstract**

18 Three-dimensional mathematical model was developed for the rectangular TE<sub>10n</sub> cavity  
19 system. Energy/heat, momentum equations were solved together with the Maxwell's  
20 electromagnetic field equations using COMSOL Multiphysics® simulation  
21 environment. The dielectric properties,  $\epsilon'$  and  $\epsilon''$ , of NaY zeolite were evaluated as a  
22 function of temperature. Considering these values, the microwave heating, i.e.,  
23 temperature profile along the bed, and temperature evolution with time, of a fixed-bed  
24 made of dry porous NaY zeolite was simulated, and temperature results were compared  
25 to the experimental values for the validation of the model. Furthermore, the prediction  
26 of thermal runaway and heating behavior of other materials were studied.

## 27 **1. Introduction**

28           Microwave heating, MWH, directly converts the microwave energy into heat in  
29 contrast to conventional heating, CH, transfer mechanisms that occur through a surface.  
30 The electromagnetic waves directly interact with the charges in the molecules and  
31 solids (i.e., dipoles, ions or delocalized electrons) which result in a volumetric heating  
32 of the material. MWH is characterized by a non-contact, rapid and selective heating  
33 that leads to shorter processing times. MWH is considered as an energy efficient  
34 process; although not always as it has been reported by several authors [1, 2].  
35 Nevertheless the importance and efficiency of MWH technology is endorsed by many  
36 different processing industries, starting from food technology, material processing and  
37 sintering, coal and mining that routinely heat by microwaves as a well-established  
38 technology [3, 4]. In the last fifteen years the advances in microwave assisted  
39 synthesis/chemistry including organic synthesis and heterogeneous catalysis are  
40 becoming more important and microwave energy is presented as a tool for green  
41 chemistry [2] and process intensification [5].

42           Microwave driven chemical reactors often report better performance compared  
43 to its counterparts heated by conventional methods, but most of the times the  
44 differences are related with the so-called “thermal effects”, and several studies point  
45 out the non-uniform heat generation with microwave irradiation due to the spatial  
46 variations of the electromagnetic field within the sample [6-8] or non-homogeneous  
47 distribution of the material [9]. The non-uniform heating and temperature distribution  
48 can be overcome in the case of liquids basically by stirring; however, the problem still  
49 remains in the case of continuous flow reactors, and heterogeneous solid-gas phase  
50 catalytic systems. Especially, the temperature distribution within the fixed/packed-bed  
51 configuration is important for the conversion of the reactants of interest. Durka *et al.*  
52 [6] reported significant two-dimensional temperature gradients in a fixed-bed of  
53 CuZnO/Al<sub>2</sub>O<sub>3</sub> catalysts both in axial and radial directions. They observed 60°C  
54 difference from bottom to top in 12 mm length, and 25-45°C from center to the wall in  
55 24 mm of diameter. In a recent work of Horikoshi *et al.* [10] focused on the selective  
56 heating of Pd/Activated Carbon catalyst in heterogeneous systems via microwave  
57 heating for continuous evolution of hydrogen from organic hydrides. They observed a  
58 large temperature distribution along the 50 mm length of the catalyst bed. Almost a

59 50% drop in temperature below the 15 mm of the catalytic bed, where the  
60 dehydrogenation reaction could not take place due to the lower temperature. They claim  
61 that a more uniform distribution of electromagnetic field should lead more uniform  
62 temperature within the catalyst bed and could result in significant improvement in  
63 process efficiency while giving a high conversion yields and considerable energy  
64 saving compared to the conventional heating.

65 The spatial electric and magnetic field distribution inside a microwave cavity,  
66 as well as in the material processed, are the major factors for materials processing to  
67 obtain the temperature profiles of the heated sample. The study of temperature  
68 distribution in microwave heated solids has been addressed by several authors, focused  
69 on the non-uniform temperature distribution, hot and /or cold spots inside the heated  
70 material by microwave [11-14]. The numerical models include the simultaneous  
71 solution of differential equations for the electromagnetic field distribution, heat transfer  
72 and fluid dynamics with a finite element analysis with software based on finite element  
73 analysis such as COMSOL Multiphysics® [15]. To perform the simulation, the exact  
74 geometry of the cavity and material together with its three-dimensional position within  
75 the cavity are essential. The temperature gradients simulated in the case of heating  
76 pinewood, carbon, Pyrex and combinations of thereof could be as high as 800K in the  
77 case of carbon cylinders inside a wood cube of 86 mm side, from the outer side to the  
78 inner part [14]. Unfortunately, none of the works were able to validate the simulated  
79 temperature profiles with experimental temperature distribution in a 2D map.  
80 Furthermore, in all the cases constant values for the dielectric properties with  
81 temperature have been considered, but this could be of special relevance in the case of  
82 important changes on these properties that could lead to a runaway.

83 In the present work, microwave heating of a dry porous zeolite fixed-bed in a  
84 quartz tube was studied both experimentally and numerically in a mono-mode  
85 rectangular resonant cavity. The focus was on the study of the electric field distribution,  
86 heat generation, heat transfer, and temperature distribution. The COMSOL  
87 Multiphysics® simulation environment was used to perform the three-dimensional  
88 modelling of a mono-mode (TE<sub>10</sub>) rectangular waveguide microwave heating cavity.  
89 The temperature distribution in time and space within the sample in a fixed-bed  
90 configuration was simulated in the basis of experimentally measured dielectric  
91 properties ( $\epsilon'$  and  $\epsilon''$ ) of the NaY zeolite as a function of temperature. The simulated  
92 temperature profiles were then validated with experimental data obtained with a fiber

93 optic and thermographic camera and the model was used to predict possible runaway  
94 during microwave heating of zeolites as well as the heating behavior of other materials  
95 inside the cavity.

## 96 2. Experimental System

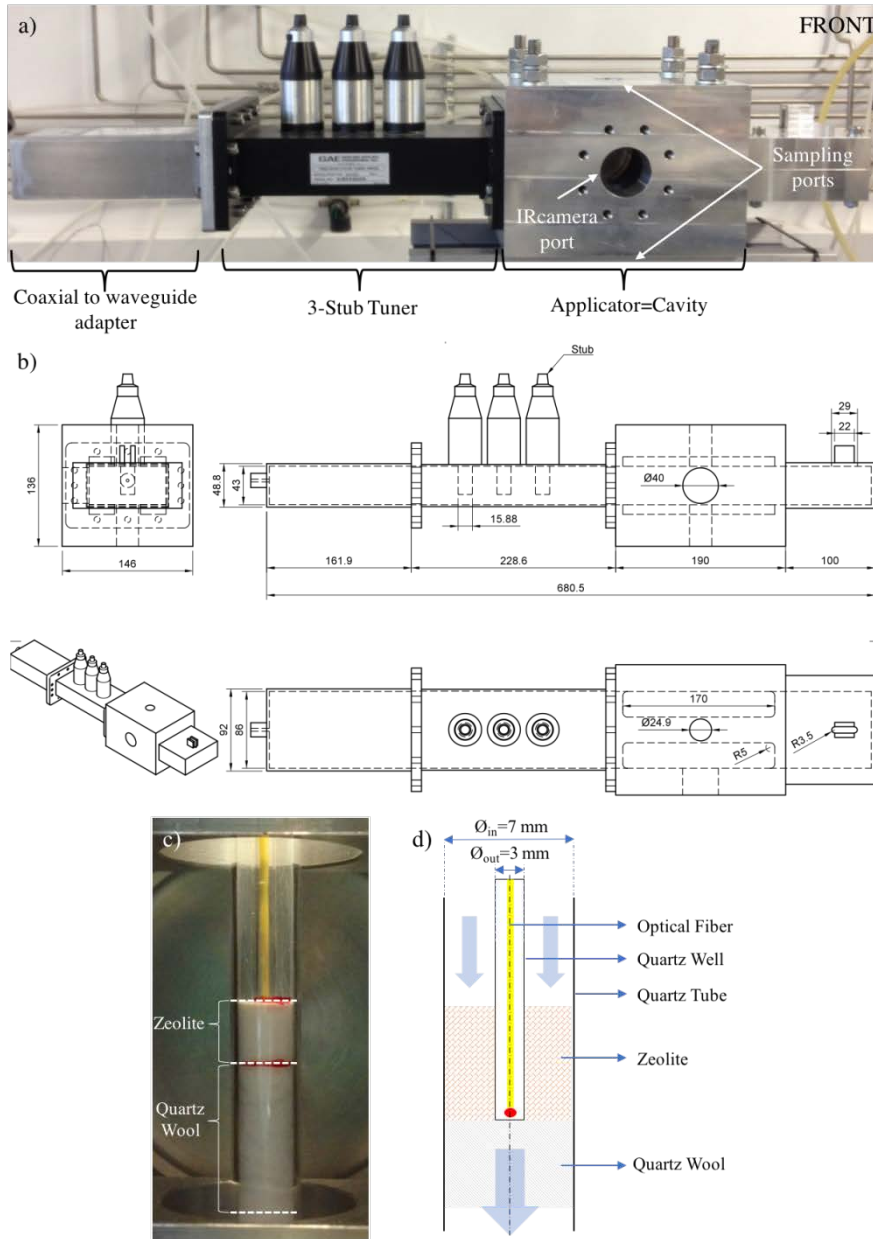
97 The microwave heating cavity, see **Figure 1 a**, was supplied by Sairem Iberica  
98 and consist of a solid-state microwave generator functioning at the range of 2.43 – 2.47  
99 GHz with a 0.1 MHz step and a maximum power of 150 W, and a TE<sub>10</sub> mode  
100 microwave cavity with a WR420 waveguide. The cavity was further modified with a  
101 precision 3-Stub Tuner (GA1002 model, Gerling Applied Engineering, Inc., USA) to  
102 reduce the reflected power below than 10% of the forward power. Before starting the  
103 experiments, the cavity was tuned and the mismatch was analyzed by the evaluation of  
104 S-parameters, S<sub>11</sub>, which are obtained by a Network Analyzer (Agilent E5061B 5 Hz –  
105 3 GHz) within the frequency range of 2.43 – 2.47 GHz with 0.1 MHz steps.

106 A fixed-bed quartz tube ( $\varnothing_{in-out}=7-9$  mm) was located inside the cavity through  
107 two circular sampling ports (top and bottom). The temperature in the fixed-bed was  
108 measured with a fiber optic (range: -80 to 250°C,  $\varnothing$ :1 mm, Neoptix T1 Probe) inside a  
109 capillary quartz well, see **Figure 1 c**. The temperatures reported in this chapter for the  
110 fixed bed experiments correspond to the optical fiber readings at the bottom of the  
111 fixed-bed. The surface temperature of the quartz tube was also measured by an infrared  
112 thermographic camera, (range: 0-500°C, InfraTec, GmbH, quartz emissivity: 0.9)  
113 located in front of the side window of the cavity. The Infrared camera captures the  
114 infrared images every 3 seconds. The images are further processed with an algorithm  
115 that has been developed with the Image Processing Toolbox® of MATLAB [16] to  
116 extract the quartz tube surface temperature, e.g., transient average, maximum or both.  
117 **Figure 1** shows in detail the experimental setup of the microwave heating cavity with  
118 the dimensions and the fixed-bed configuration.

119 The heated solid was a commercial zeolite Y powder, CBV100, supplied by  
120 Zeolyst, containing 13.0 wt.% of Na<sup>+</sup> as an extra-framework cation. The fine powder,  
121 without any post-treatment, was first pelletized with a laboratory press using a stainless-  
122 steel mold (13 mm in diameter). Then the pellets were crushed and ground to 80-150  
123  $\mu$ m in order to prevent compaction of the fixed bed (200 mg, 80-150  $\mu$ m, L=10 mm,  
124  $\varnothing_{in}=7$  mm). Before the heating experiments, the fixed-bed was regenerated by

125 microwave heating while passing N<sub>2</sub> (100 ml/min, 99.9999% pure, Praxair),  
 126 meanwhile, the water concentration was registered by an on-line quadrupole mass  
 127 spectrometer (OmniStar, GSD 320, Pfeiffer Vacuum) to ensure that total dehydration  
 128 of the sample was reached for a given set of conditions.

129



130

131 **Figure 1.** Experimental setup, a) Microwave heating cavity b) The first-angle projection of 3D modelled  
 132 empty Microwave Heating Cavity, the dashed lines indicate hidden edges and corners, measurement in  
 133 mm c) Fixed-bed quartz tube and its corresponding d) schematic view

## 2.1. Dynamic measurement of dielectric properties

The ability of a material to absorb microwave energy and transfer it into heat is governed by its dielectric properties [17]. The complex relative permittivity is defined as:

$$\epsilon_r = (\epsilon_r' - j\epsilon_r'') \quad (1)$$

where  $\epsilon_r'$  is the real part of the complex relative permittivity is known as dielectric constant, which characterizes the polarization of a material in response to an applied external electric field and it is also a measure of dielectric materials to ability to store electrical energy [18-20], whereas the imaginary part,  $\epsilon_r''$ , is the dielectric loss factor also known as dissipation factor, which reflects the loss in the medium due to the damping vibrating dipole moment [20], which generates heat. These properties are crucial to the prediction of the spatial distribution and dissipation of electromagnetic wave fields. It is a challenging task to obtain sufficiently accurate values since they are highly dependent on temperature, moisture, frequency, the physical state either solid or liquid and composition [3, 19].

Dielectric properties ( $\epsilon_r'$  and  $\epsilon_r''$ ) of zeolites were measured with a dual-mode microwave system recently developed by Catala-Civera et al. [17]. The equipment allows dielectric measurements of materials during microwave heating in real time. Two separate microwave sources are used for simultaneously heating and measuring, and a cross-coupling filter is used to isolate the two modes ( $TE_{111}$  and  $TM_{010}$ ) from each other.

As the dielectric properties of the samples depend on temperature, the resonant frequency and quality factor of the cavity varies with temperature. From these two values, the dielectric properties of the sample are computed with an enhanced Cavity Perturbation Method [17]. To ensure an efficient power delivery and to maintain the desired heating rate a control loop ensures that, the sweep frequency bandwidth of the heating source is adjusted continuously to track the resonant peak of the cavity during the heating cycle.

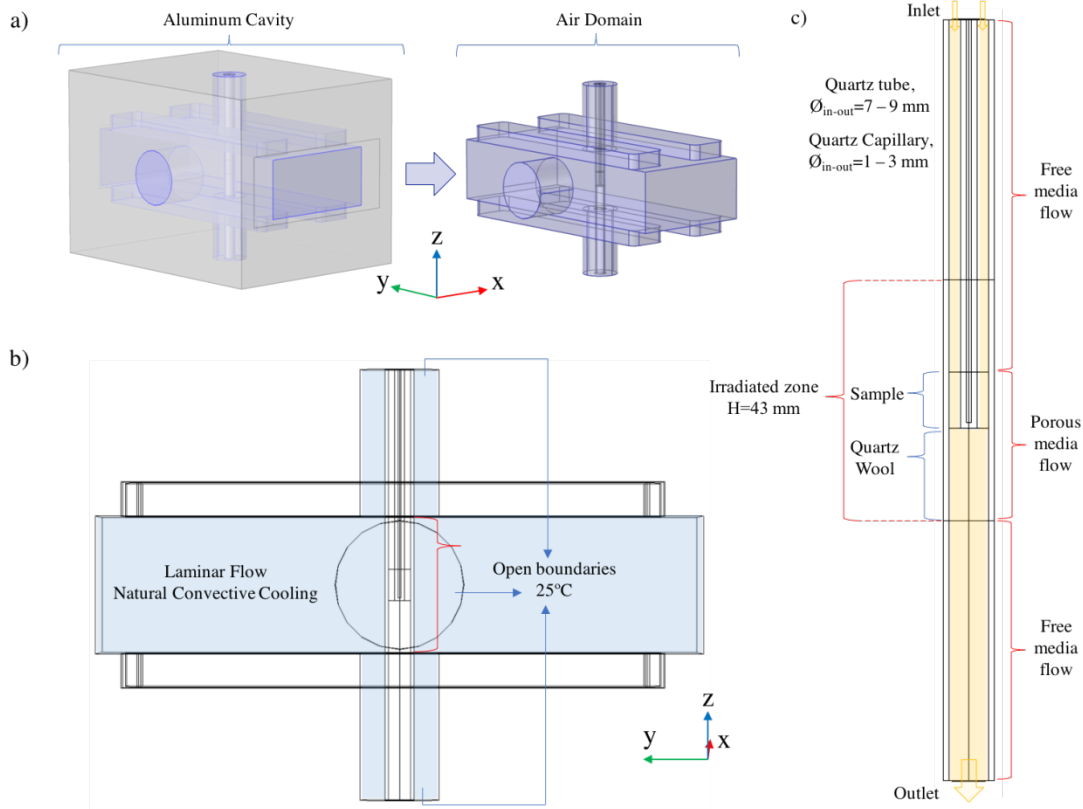
The sample is inserted into the cavity through a cut-off hole at the central plane of the top wall. The test sample is placed in a quartz vial with inner diameter 10 mm and external diameter 12 mm, which can handle temperatures of up to 1300°C. The zeolite sample has a bed size of 10 mm diameter and 15 mm height to ensure a uniform

167 electric field distribution in all the sample volume and thus uniform processing. The  
168 temperature of the sample under test is measured by an infrared radiation (IR)  
169 thermometer with 0.1°C accuracy, positioned outside of the cavity and connected to it  
170 via a window in the sidewall. Since the IR thermometer measures the holder surface  
171 temperature, a calibration process was applied to find the relationship with the bulk  
172 temperature of the sample [17]. Before any measurement, the zeolite was dried  
173 overnight in an oven at 250°C; then the measurements were done with a cycle of heating  
174 and cooling of the sample under nitrogen flow to minimize the effect of adsorbed  
175 ambient moisture. An additional second cycle of heating and cooling was performed to  
176 ensure complete drying.

### 177 **3. Mathematical Model**

178 The 3D-finite element model couples together electromagnetic waves, heat  
179 transfer and fluid dynamics. **Figure 2** shows the perspective and cross-sectional views  
180 of the model, implemented flow models, and dimensions of the quartz tube and **Table**  
181 **1** represents the characteristics of the simulated system. Flanges and the outer parts of  
182 the stub tuner were not included into the model in order to reduce the complexity of the  
183 simulation, and because their effect on the overall heat balance is expected to be  
184 negligible.





185  
186 **Figure 2.** a) Perspective and b) cross-sectional views of the cavity and c) quartz tube and applied flow  
187 models, i.e., laminar, free and porous media flows

188  
189 Following governing equations were used for each physic module:

### 190 3.1. Electromagnetic Waves

191 Maxwell's electromagnetic field distribution and the general volumetric power  
192 dissipation are calculated by solving the following equations [20-22]:

$$193 \nabla^2 \mathbf{E} + \omega^2 \varepsilon \mu \mathbf{E} = 0 \quad (2)$$

$$194 Q_{MW} = \pi f \varepsilon_0 \varepsilon_r'' \mathbf{E}^2 \quad (3)$$

193 where  $\mathbf{E}$  (V/m) is the electric field vector,  $\omega = 2\pi f$  ( $s^{-1}$ ) is the angular frequency, and  
194  $\varepsilon$ ,  $\mu$  stands for the permittivity and permeability of the media.

195 The wave equation, Eq. (2), is solved inside the microwave heating cavity (air  
196 domain), zeolite fixed-bed and quartz tube, in this physic interface some assumptions  
197 have been made:

- 198 i. Since there is no magnetic material, the magnetic permeability of all the materials  
199 is assigned as free-space,  $\mu_r = 1$ , which gives no magnetic field contribution for  
200 the volumetric power dissipation.

- 201 ii. Since the electromagnetic field penetrates a negligible distance into the metallic  
 202 walls, the 3D wave equation is only applied in the air domain inside of the  
 203 waveguide and inside the fixed-bed and quartz tube. It was found by preliminary  
 204 modeling that the energy balance was not correctly represented if a perfect electrical  
 205 conductor (PEC) boundary condition was employed due to the relatively low  
 206 electromagnetic dissipation inside the fixed-bed. Thus, an impedance boundary  
 207 condition was assigned to the metal walls of the microwave heating cavity to  
 208 account for the electric surface current present in them.
- 209 iii. Experimentally measured dielectric properties,  $\varepsilon'$  and  $\varepsilon''$ , of the sample are  
 210 introduced as a function of temperature.

### 211 3.2. Heat Transfer

212 This module is applied inside the microwave heating cavity (air domain), as  
 213 well as the fixed-bed and the quartz tube. Heat transport equation incorporates the  
 214 conversion of microwave energy to thermal energy, as well as the thermal losses to the  
 215 environment through the quartz tube. In this physic interface following transient  
 216 equation is solved for the porous fixed-bed (solid-fluid system), quartz tube and  
 217 surrounding air [23] simultaneously:

$$\rho C_p \frac{\partial T}{\partial t} + \rho C_p \mathbf{u} \cdot \nabla T + \nabla \cdot (-k \nabla T) = Q_{MW} \quad (4)$$

218 where  $\rho$ ,  $C_p$ ,  $k$  are the density, heat capacity and thermal conductivity of the fluid/solid,  
 219 respectively. The symbol  $\mathbf{u} = (u, v, w)$  stands for the flow velocity vector field for the  
 220 x,y,z directions. First term on the left side is the rate of heat accumulation, and the  
 221 second and third terms are convective and conductive contributions to heat transfer. On  
 222 the right-hand side is the volumetric power dissipation, which was calculated with the  
 223 electromagnetic waves module. The following assumption in this module have been  
 224 made:

- 225 i. Effective average volumetric heat capacity,  $C_{p_{eff}}$ , at constant pressure and  
 226 effective average thermal conductivity,  $k_{eff}$ , of the solid-fluid system can be used  
 227 to describe heat transfer in porous media, hence Eq. (4) was modified accordingly.
- 228 ii. In the quartz tube, only conduction phenomenon was considered,  $\mathbf{u} = 0$
- 229 iii. Material properties such as thermal conductivity and the heat capacity of the zeolite  
 230 are considered temperature independent.

- 231 iv. A continuity condition was applied to the interface between two different domains  
 232 1 and 2:  $-n \times (-k_1 \nabla T_1) - n \times (-k_2 \nabla T_2) = 0$  [24].  
 233 v. The heat flux by radiation is negligible ( $<0.2\text{W}$ ).  
 234 vi. The convective velocity field around the fixed-bed quartz tube can be simulated  
 235 with fluid dynamics to account for natural convection.  
 236 vii. Heat loss of the metallic walls can be neglected.

### 237 3.3. Fluid dynamics

238 Two physics interfaces were used in this module; *Free and Porous Media Flow*  
 239 was adapted for the tubular fixed-bed due to nitrogen flow through the porous sample  
 240 and *Laminar Flow* for the air domain around the quartz tube due to the natural  
 241 convective cooling. The fluid dynamics are represented as time-dependent  
 242 compressible fluid flow according to the following equations [25]:

$$\rho \frac{\partial}{\partial t} \mathbf{u} + \rho (\mathbf{u} \cdot \nabla) \mathbf{u} = \nabla \cdot \left[ -p \mathbf{I} + \mu (\nabla \mathbf{u} + (\nabla \mathbf{u})^T) - \frac{2}{3} \mu (\nabla \cdot \mathbf{u}) \mathbf{I} \right] + \mathbf{F} \quad (5)$$

$$\frac{\partial}{\partial t} \rho + \nabla \cdot (\rho \mathbf{u}) = 0 \quad (6)$$

$$\frac{\rho}{\epsilon_p} \left( \frac{\partial}{\partial t} \mathbf{u} + (\mathbf{u} \cdot \nabla) \frac{\mathbf{u}}{\epsilon_p} \right) = \nabla \cdot \left[ -p \mathbf{I} + \frac{\mu}{\epsilon_p} (\nabla \mathbf{u} + (\nabla \mathbf{u})^T) - \frac{2\mu}{3 \epsilon_p} (\nabla \cdot \mathbf{u}) \mathbf{I} \right] - (\mu \kappa^{-1}) \mathbf{u} \quad (7)$$

243 Eq. (5) stands for the free flow in the quartz tube and the laminar flow for the  
 244 air domain, see **Figure 2 b-c**. The first and second terms on the left-hand side are  
 245 acceleration forces, and on the right side, there are pressure gradient and viscous forces.  
 246 Conservation of mass is expressed by the continuity equation, Eq. (6), which is derived  
 247 by considering a unit volume of the medium and it states that the rate of increase of the  
 248 mass of the fluid within an elementary unit equals to the net mass flux into the volume.  
 249 Eq. (7) was used to describe the flow in the porous region, known as the *Brinkman*  
 250 *equation* which is a combination of the continuity equation and the momentum  
 251 equation. In this equation,  $\epsilon_p$  and  $\kappa$  stand for the bed porosity and permeability,  
 252 respectively, see **Table 1** [25, 26]. Following assumptions have been made:

- 253 i. The nitrogen flow rate, 100 mL/min, and the pressure, 1 atm, are fixed at the inlet  
 254 of the quartz tube.  
 255 ii. No slip conditions, on the cylindrical wall of the quartz tube,  $\mathbf{u} = 0$   
 256 iii. The convective velocity field around the quartz tube is computed by applying a  
 257 laminar flow model, taking into account the Rayleigh number

258 criteria ( $Ra=Gr\cdot Pr < 1\times 10^9$ ), where Gr and Pr are the Grashof and Prandtl  
 259 numbers, respectively.

260 iv. A vertical buoyancy force term related to the thermal expansion is also included in  
 261 eq. (2.4) for the air domain around the quartz tube:  $\mathbf{F} = -g(\rho_{air} - \rho_{air,ref})$  [N/  
 262  $m^3$ ], where g is the gravity ( $9.81 m/s^2$ ) and  $\rho$  and  $\rho_{ref}$  are the density and reference  
 263 density of air, respectively (at atmospheric pressure and 25°C).

264 v. Compressible flow,  $\rho$  is variable, model is assigned for air and nitrogen domains  
 265 [27].

266

*Table 1. Characteristics of the simulated system*

<b>Fixed-bed</b>		<b>Reference</b>
Height, H (m)	0.010	-
Inner diameter, $D_i$ (m)	0.007	-
Outer diameter, $D_o$ (m)	0.009	-
Bed density, $\rho_{bed}$ ( $kg\ m^{-3}$ )	672	-
Bed porosity, $\epsilon$ (-)	0.375 *	[33]
Bed permeability, $\kappa$ ( $m^2$ )	$1.15\times 10^{-11}$ **	[34]
<b>Quartz tube</b>		
Quartz density, $\rho_w$ ( $kg\ m^{-3}$ )	2200	[35]
Dielectric constant, $\epsilon'(-)$	3.78	[35]
Heat capacity, $C_{p,w}$ ( $J\ kg^{-1}\ K^{-1}$ )	712	[35]
Thermal conductivity, $k_w$ ( $W\ m^{-1}\ K^{-1}$ )	1.96	[36]
<b>Particles, NaY zeolite</b>		
Average diameter, $d_{NaY}$ (m)	$1.15\times 10^{-4}$	-
Dielectric constant, $\epsilon'(-)$ and loss factor $\epsilon''(-)$	f(T) ***	-
Heat capacity, $C_{p,s}$ ( $J\ kg^{-1}\ K^{-1}$ )	836	[37]
Thermal conductivity, $k_s$ ( $W\ m^{-1}\ K^{-1}$ )	0.15	[38]
<b>Air and Nitrogen</b>		
Dielectric constant, $\epsilon'(-)$ and loss factor $\epsilon''(-)$	1 and 0	-

267

\*calculated using empirical expression of Pushnov for sphere grains

268

\*\* calculated using empirical expression of Rumpf and Gupte for sphere packings

269

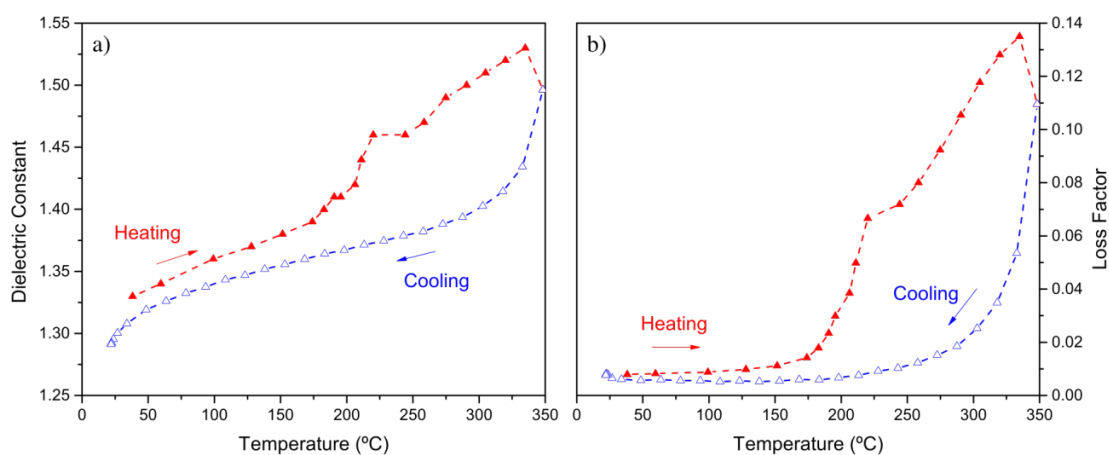
\*\*\* experimentally measured at 2.45 GHz as a function of temperature

## 270 4. Results and Discussions

### 271 4.1. Dynamic measurement of dielectric properties

272 The measured dielectric properties at 2.45 GHz of NaY zeolite as a function of  
273 temperature is shown in the **Figure 3**. When the zeolite is dehydrated ( $>250^{\circ}\text{C}$ ), the  
274 relaxation mechanism related to microwave heating is linked to the mobility of extra-  
275 framework cations to different ion exchange positions [28-30]. In this manner, Legras  
276 *et al.* [28] studied the dielectric properties for partially and fully hydrated FAU type  
277 zeolites in the 0.5 to 20 GHz microwave region at a fixed temperature of  $20^{\circ}\text{C}$ . At low  
278 water loading, they observed a relaxation mechanism at peak frequencies between 1.4  
279 to 1.6 GHz, which could be extended to 2.5 GHz with 15% changes in the loss factor.  
280 This mechanism is related to the phenomenon called space charge polarization that can  
281 be produced by the separation of mobile positively and negatively charged particles  
282 under an applied electric field. In the case of zeolites, the negative charge is in the  
283 oxygen atoms of the structure and the positive corresponds to extra-framework cations.  
284 Since the NaY zeolite contains 13.0 wt.% of  $\text{Na}^+$  as an extra-framework cation, its  
285 contribution to the dielectric properties of low silica zeolite is the major phenomenon  
286 at 2.45 GHz. According to the analyses, see **Figure 3**, the loss factor increased  
287 exponentially with temperature ( $>300^{\circ}\text{C}$ ), as a consequence of the fact that, as  
288 temperature increases, the motion of  $\text{Na}^+$  increases. This corresponds to thermal  
289 runaway and is observed during experimentation. The applied microwave power needs  
290 to be controlled properly in order to avoid the thermal runaway [28, 31, 32].

291

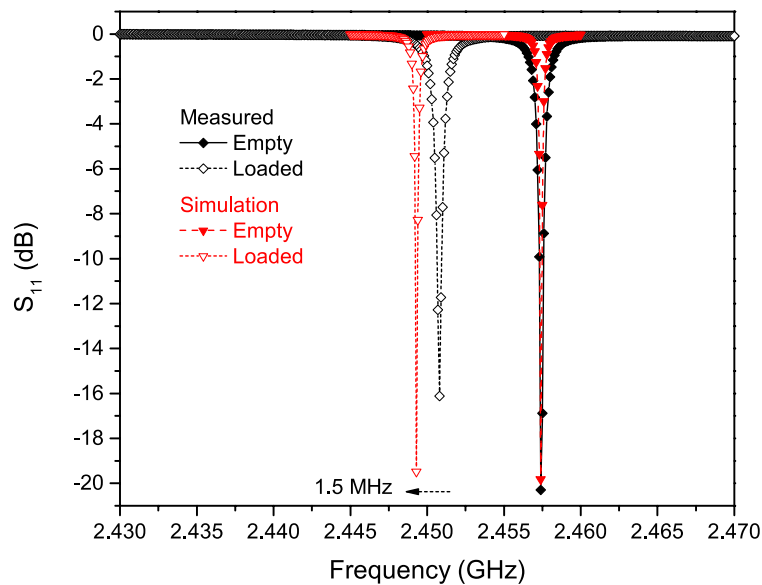


292

293 **Figure 3.** Dielectric properties of NaY zeolite a) Dielectric constant b) Loss factor as a function of  
294 temperature.

#### 4.2. 3D-finite element model, electromagnetic field distribution

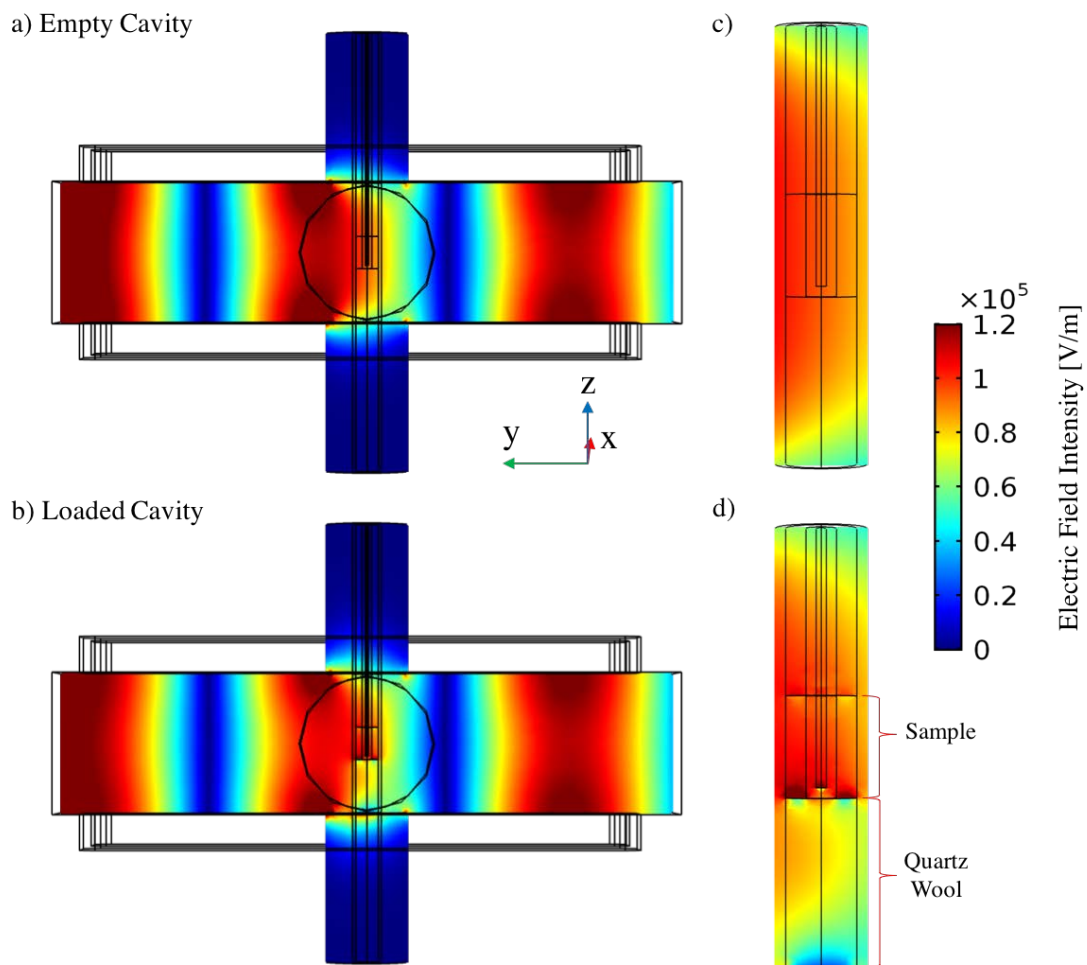
Reflection spectrums,  $S_{11}$  parameters, were experimentally measured and simulated with the regarding immersion depths of  $stub\ 1=25.44\text{ mm}$ ,  $stub\ 2=27.57\text{ mm}$  and  $stub\ 3=8.48\text{ mm}$ , see **Figure 1**. The comparison between the measured and simulated data of  $S_{11}$  parameters are presented in **Figure 4**. A shift in resonance frequency in the reflection spectrum is apparent between the empty and the cavity loaded with the quartz tube; resonance occurs at the frequency of minimum reflection. As the quartz tube was introduced inside the cavity it interacts with the electromagnetic field and changes its resonant behavior, the effect of which is apparent throughout the cavity and can therefore be registered at the cavity port as a change in the  $S_{11}$  parameter. The same phenomena were also observed in the simulation with only a slight difference (magnitude and the frequency) in the case of loaded cavity. These matching spectral response and impedance matching characteristics show that the behavior of the microwave model agrees well with the physical setup which results in the validation of the model from the electromagnetic point of view.



**Figure 4.** Measured and Simulates  $S_{11}$  parameters (reflection spectrum) of empty and loaded cavity

The cross-section views of normalized electric field distributions along the  $y$ -direction inside the empty and loaded cavity are presented in **Figure 5**. The introduction of the fixed-bed quartz tube affects the electric field distribution but the effect is not large due to the small size of the fixed-bed and the low values of the dielectric constant of the sample. This is also apparent in the relatively small shift in resonance frequency,

319 see **Figure 4**. The highest electric field intensity was observed at the frontier between  
 320 the sample and the quartz wool (see **Figure 5 d**). This simulation also reveals that the  
 321 maximum electric field intensity of the standing wave practically coincides with the  
 322 fixed-bed quartz tube location because of the physical design of the cavity. This  
 323 maximal region cannot be displaced to the center of the sampling windows, due to the  
 324 absence of a movable shorting plate/reflector and microwave generator limitations  
 325 (spectral band: 2.43-2.47 GHz).  
 326



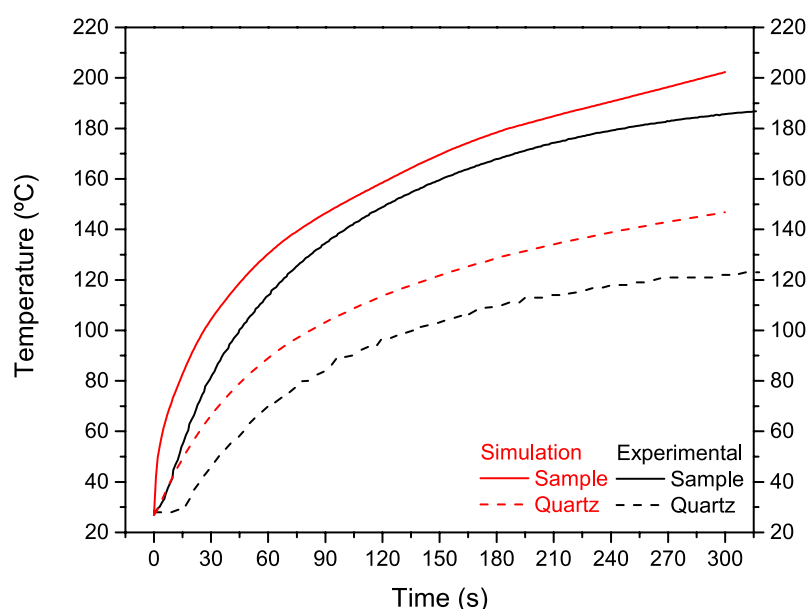
327  
 328 **Figure 5.** Simulated electric field distribution inside the a) empty and b) loaded cavity and close look to  
 329 corresponding quartz tube locations inside the cavity when it is c) empty and d) loaded. For clarity, in  
 330 b) and d) quartz tube and fixed-bed domains are maintained in the case of empty cavity with the  
 331 properties of air ( $\epsilon'=1$ ,  $\epsilon''=0$ ,  $\mu_r=1$ , and  $\sigma=0$ ) to have the same number of elements.

### 332 4.3. 3-D finite element model, temperature distribution

333 The experimental fixed-bed temperature data was registered by the fiber optic  
 334 at the bottom of the sample, where the temperature was highest, and the infrared camera  
 335 was used to measure the outer surface temperature of the quartz wall. Transient

336 temperature profiles show that the simulated data are in good agreement with the  
337 experimental ones, as shown in **Figure 6**, where the predicted maximum temperatures  
338 for the fixed-bed and the quartz wall are represented, respectively. The average  
339 percentage relative errors between the simulation and the experimental data of the  
340 fixed-bed and the surface temperatures were  $\pm 10.3\%$  and  $\pm 13.4\%$ , respectively.  
341 Simulated temperatures data are higher than the experimental ones; this discrepancy  
342 could be attributed to the material properties such as thermal conductivity of the zeolite  
343 and quartz tube. In the case of zeolites, various thermal conductivities have been  
344 reported by many researchers, some of them ranges in between 0.15-0.30 W/(m.K) [33-  
345 36] which results in simulated final temperatures of 203 and 173°C, respectively, see  
346 **Figure S1**-Supporting information.

347



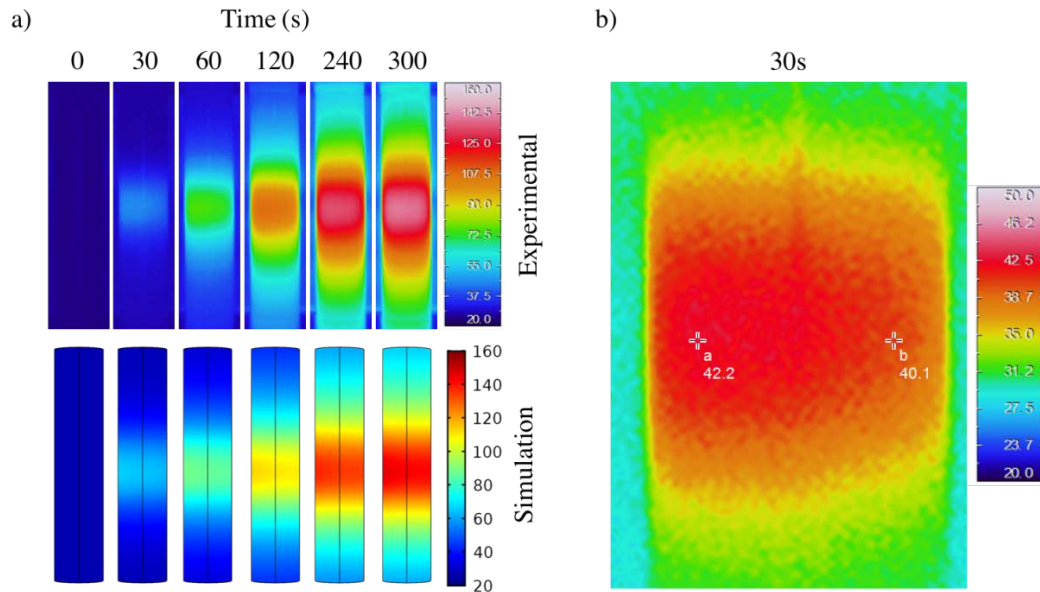
348

349 **Figure 6.** Comparison of the experimental measurements and simulation predictions of transient  
350 maximum temperature profiles of the fixed-bed (measured with the fiber optic) and the quartz wall  
351 (measured with the infrared thermographic camera)

352

353 **Figure 7** shows the surface temperature of the quartz wall measured by an  
354 infrared camera at different time periods and the corresponding simulated data. The  
355 predictions are in good agreement with experimental data; in both cases, the sample  
356 starts heating from the center and then it expands axially. Even though the infrared  
357 camera has a limited resolution (320x240 pixel), it can be observed that the left side of  
358 the quartz tube has a slightly higher temperature (**Figure 7 b**, at 30s, ca. 2°C).





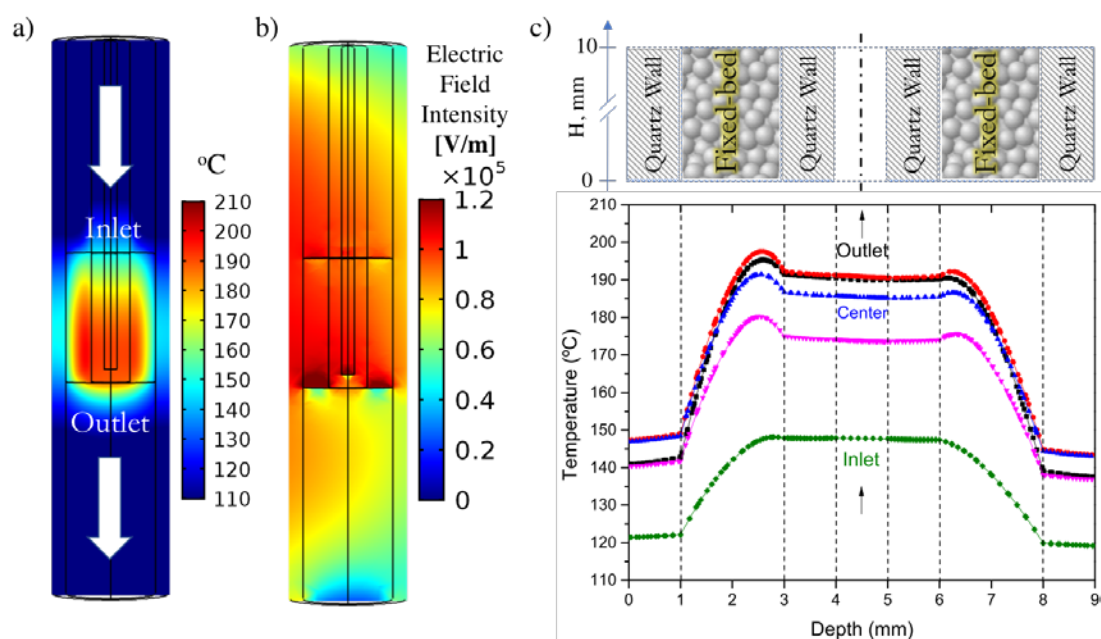
359  
 360 **Figure 7.** Infrared images of the quartz surface temperature at different time periods and corresponding  
 361 simulated data, b) an infrared image at 30s, color legends are in °C.

362

363 **Figure 8** shows the simulated radial temperature distribution inside the quartz  
 364 tube as well as along the sample at steady-state. The radial temperature profiles show  
 365 that the temperature is considerably lower at the outer layers, compared to the center of  
 366 the quartz tube. This is because of heat is absorbed in the volume of the sample under  
 367 microwave irradiation and then lost by natural convection along the surface of the  
 368 quartz tube to the surroundings. **Figure 8** also shows the variation of radial temperature  
 369 profiles at different heights in the fixed-bed quartz tube. There is a slight temperature  
 370 increase, ca. 2-4°C, at the left side (1-3 mm, towards the generator side) of the sample  
 371 in comparison to the right side (6-8 mm). This is in accordance with the electrical field  
 372 distribution in the cavity, which is not centered within the fixed-bed, explained  
 373 previously, see **Figure 8 c**. Despite a distribution that tends towards the generator side,  
 374 the overall distribution is roughly symmetrical, i.e., the asymmetry of the microwave  
 375 field is smoothed out by heat transfer. The temperature difference is less pronounced at  
 376 the top (H=10 mm) compared to the bottom of the sample (H=0 mm), and the bottom  
 377 part is hotter than the top, ca. 40 °C. The maximum radial temperature is reached near  
 378 the fiber optic location, 4-5mm. This temperature gradient is linked to the non-uniform  
 379 electric field distribution inside the sample, and the heat transfer dynamics in the  
 380 system. It should be noted that the high-temperature region corresponds to the relatively  
 381 high electric field region, see **Figure 8 b** [18, 37]. Despite this electric field gradient, it

382 was observed that 58% of the fixed bed has the temperature range in between 160-  
383 197°C.

384



385

386 **Figure 8.** Simulated cross-sectional view of the a) quartz tube fixed-bed and its corresponding b) electric  
387 field distribution c) spatial temperature distribution at steady-state.

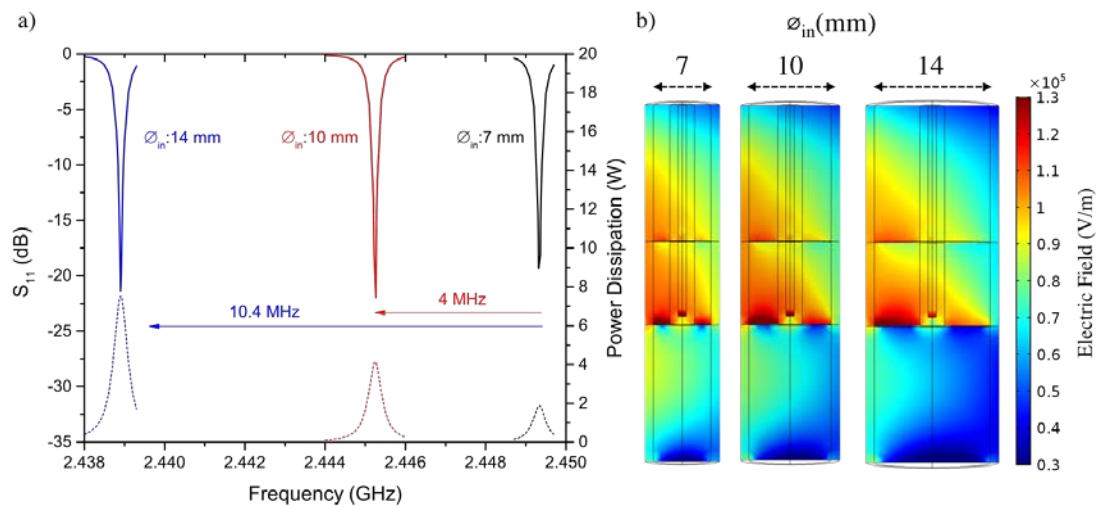
388

#### 4.4. Energy Balance

389

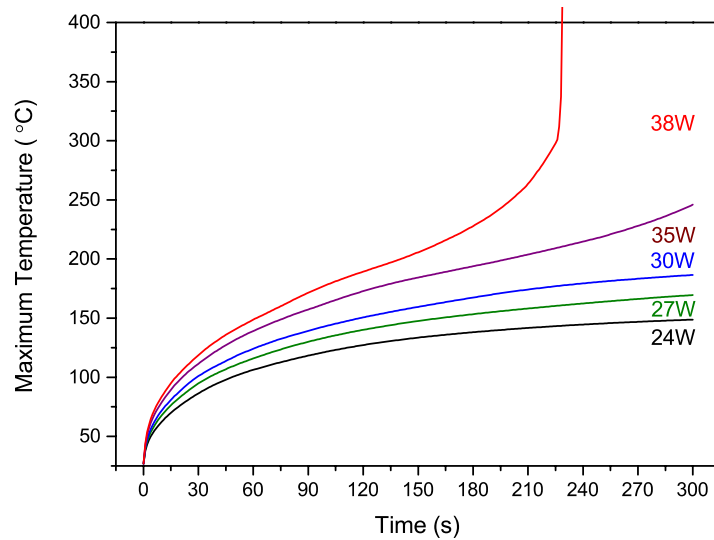
390 The overall energy balance in the microwave heating cavity and the fixed-bed  
391 shows that only 6.2% of the input power was dissipated within the NaY zeolite fixed-  
392 bed, with a reflection loss of 1.2%. This means that the rest of the microwave energy  
393 was dissipated in the metallic cavity wall due to electric currents. The efficiency of this  
394 microwave heating cavity is very low, and this could be due to the small volume of the  
395 NaY zeolite fixed-bed (0.31 mL with respect to the whole cavity volume, 2.8 L) in the  
396 microwave heating cavity. A similar conclusion was reached by Coss et al. [38], only  
397 27% of the input power was absorbed in their case, even though they used a larger bed  
398 of a good microwave absorber, granulated activated carbon and the adsorbent  
399 volume=18.8 mL. The authors expected that this efficiency could become higher with  
400 the larger volume of adsorbent. Also, Cherbanski [39] calculated the efficiency factor  
401 using 13X zeolite in a multi-mode cavity, giving a 21% efficiency (adsorbent  
402 volume=253 mL). We decided to simulate larger volumes of the same adsorbent, in our  
403 microwave heating cavity, to predict its efficiency. The inner diameter of the quartz  
tube was increased from 7 mm (V:0.31mL) to 10 (V:0.71 mL) and 14 mm (V:1.47 mL).

404 **Figure 9** represents the parametric analysis, power dissipation and corresponding  
 405 electric field distributions of different volumes of adsorbent. The results show that 13.8  
 406 and 25.1 % of the input power could be dissipated if the adsorbent volume was  
 407 increased to 0.71 and 1.47 mL, respectively. Even though the resonance frequencies of  
 408 larger volumes of adsorbent were shifted to the lower frequencies, see **Figure 9 a**, they  
 409 were still in the solid-state microwave generator's spectral band, which is from 2.43 to  
 410 2.47 GHz. Thus, the matching basically can be done by simply modifying the supplied  
 411 frequency to the cavity. It was observed that less than 1 MHz deviation from the  
 412 nominal frequency is sufficient to change the dissipated power by 14-27%. Cherbanski  
 413 *et al.* [40] also observed the same behavior for the microwave heating of water in a  
 414 single-mode applicator; they concluded that a minor 5 MHz deviation from the nominal  
 415 frequency caused about 20% variation of the absorbed microwave power.



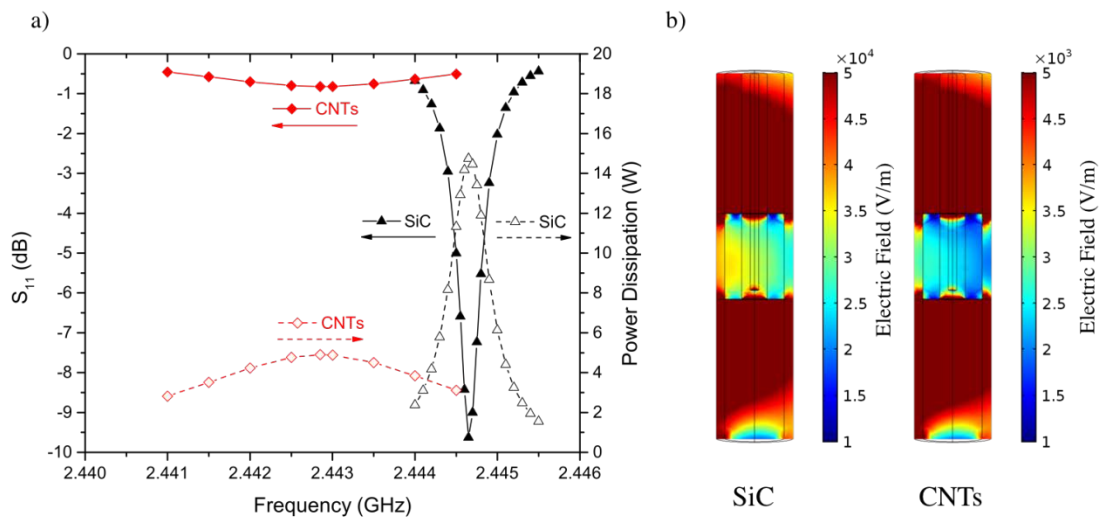
416  
 417 **Figure 9.** Simulated a) reflection spectrum ( $S_{11}$  parameters) and power dissipation as a function of the  
 418 excitation frequency with b) corresponding electric field distribution of larger volumes of adsorbent with  
 419 an input power of 30 W.

420  
 421 As it was mentioned in the introduction the evolution of the dielectric properties  
 422 with temperature would result in a temperature runaway in the sample that it could be  
 423 predicted with the simulation. In **Figure 10** it could be observed that when the input  
 424 power of the microwave cavity increases up to 38 W the temperature rises sharply after  
 425 210 s heated. This was also observed experimentally (not shown).



426  
427 **Figure 10.** Simulation of temperature evolution with time for different MW power.  
428

429 Different materials such as SiC, ( $\epsilon'=16.9$  and  $\epsilon''=0.77$ , V:0.31mL) and CNTs  
430 ( $\epsilon'=17.8$  and  $\epsilon''=26.4$ , V=0.31mL) were also employed in the simulation due to their  
431 relatively higher dielectric properties to NaY zeolite ( $\epsilon'=1.3$  and  $\epsilon''=0.01$ , V:0.31mL).  
432 **Figure 11** represents the parametric sweep of two different samples. It was observed  
433 that 16.3 and 49.2% of the input power were dissipated in the CNTs and SiC fixed-  
434 beds, respectively. The electric field intensity decreased by two orders of magnitude,  
435 especially in the case of CNTs, see **Figure 11 b**. This is due to the high dielectric  
436 properties, which reduce the penetration depth of the electromagnetic waves into the  
437 load.



438  
439 **Figure 11.** Simulated a) reflection spectrum ( $S_{11}$  parameters) and power dissipation and b)  
440 corresponding electric field distribution of SiC and CNTs with an input power of 30 W ( $\phi$  in: 7mm).

## 441 **5. Conclusions**

442 In this study, microwave heating of dry porous NaY zeolite was investigated  
443 numerically and experimentally. A three-dimensional mathematical model of the  
444 microwave heating system with the fixed-bed tubular configuration was developed.  
445 Energy and momentum equations were solved together with Maxwell's equations using  
446 COMSOL Multiphysics® software. The electric field distribution in the microwave  
447 heating cavity, as well as in the fixed-bed, and the reflection spectrum were obtained.  
448 Furthermore, the transient temperature profiles of the zeolite and quartz tube were  
449 simulated on the basis of experimentally measured dielectric properties ( $\epsilon'$  and  $\epsilon''$ ) as  
450 a function of temperature.

451 The numerical results of the reflection spectra and transient temperature profiles  
452 of the fixed-bed and quartz wall surface matched with the experimental data  
453 satisfactorily. An average percentage relative error of  $\pm 10.3\%$  for the fixed-bed and  
454  $\pm 13.4\%$  for the quartz wall was observed. Due to the non-uniform electric field  
455 distribution within the sample, a temperature gradient was observed. Despite this  
456 gradient, 58% of the fixed bed has a temperature between 160-197°C.

457 The overall energy balance shows that only 6.2% of the input power was  
458 dissipated within the sample. This lower efficiency was linked to the small volume of  
459 sample (0.31 mL) with regards to the whole cavity volume (2.8 L). The simulation  
460 results show that the efficiency of microwave heating cavity could be increased even  
461 further by using larger volumes of the NaY zeolite load, as well as heating materials  
462 with higher dielectric loss compared to zeolite such as SiC and CNTs.

## 463 **Acknowledgements**

464 Financial support from the European Research Council ERC-Advanced Grant  
465 HECTOR is gratefully acknowledged. Hakan Nigar acknowledges financial support  
466 from the Spanish Ministry of Education for the FPU grant (Formación del Profesorado  
467 Universitario – FPU12/06864), and also for the academic short stay grant (Estancia  
468 Breve – FPU2016) at the Delft University of Technology, Delft, The Netherlands.

469  
470  
471  
472  
473  
474

475 **REFENRECES**

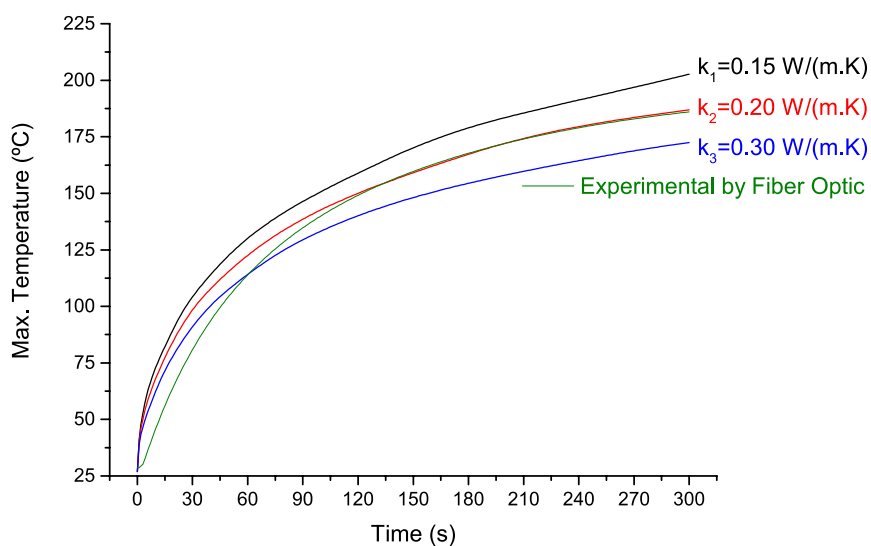
- 476 1. Bermúdez, J.M., et al., *Energy consumption estimation in the scaling-up of*  
477 *microwave heating processes*. Chemical Engineering and Processing: Process  
478 Intensification, 2015. **95**: p. 1-8.
- 479 2. Moseley, J.D. and C.O. Kappe, *A critical assessment of the greenness and*  
480 *energy efficiency of microwave-assisted organic synthesis*. Green Chemistry,  
481 2011. **13**(4): p. 794-806.
- 482 3. Meredith, R.J. and E. Institution of Electrical, *Engineers' Handbook of*  
483 *Industrial Microwave Heating*. Energy Engineering Series. 1998: Institution of  
484 Electrical Engineers.
- 485 4. Xu, L., et al., *Study on characteristics of microwave melting of copper powder*.  
486 Journal of Alloys and Compounds, 2017. **701**: p. 236-243.
- 487 5. Stefanidis, G.D., et al., *A helicopter view of microwave application to chemical*  
488 *processes: reactions, separations, and equipment concepts*. Reviews in  
489 Chemical Engineering, 2014. **30**(3): p. 233-259.
- 490 6. Durka, T., et al., *Microwave-activated methanol steam reforming for hydrogen*  
491 *production*. International Journal of Hydrogen Energy, 2011. **36**(20): p. 12843-  
492 12852.
- 493 7. Sturm, G.S.J., et al., *On the effect of resonant microwave fields on temperature*  
494 *distribution in time and space*. International Journal of Heat and Mass Transfer,  
495 2012. **55**(13–14): p. 3800-3811.
- 496 8. Sturm, G.S.J., et al., *On the parametric sensitivity of heat generation by*  
497 *resonant microwave fields in process fluids*. International Journal of Heat and  
498 Mass Transfer, 2013. **57**(1): p. 375-388.
- 499 9. Ramírez, A., et al., *Ethylene epoxidation in microwave heated structured*  
500 *reactors*. Catalysis Today, 2016. **273**: p. 99-105.
- 501 10. Horikoshi, S., et al., *Selective heating of Pd/AC catalyst in heterogeneous*  
502 *systems for the microwave-assisted continuous hydrogen evolution from*  
503 *organic hydrides: Temperature distribution in the fixed-bed reactor*.  
504 International Journal of Hydrogen Energy, 2016. **41**(28): p. 12029-12037.
- 505 11. Mimoso, R.M.C., et al., *Simulation and control of continuous glass melting by*  
506 *microwave heating in a single-mode cavity with energy efficiency optimization*.  
507 International Journal of Thermal Sciences, 2017. **111**: p. 175-187.
- 508 12. Lin, B., et al., *Sensitivity analysis on the microwave heating of coal: A coupled*  
509 *electromagnetic and heat transfer model*. Applied Thermal Engineering, 2017.  
510 **126**: p. 949-962.
- 511 13. Hong, Y.-d., et al., *Three-dimensional simulation of microwave heating coal*  
512 *sample with varying parameters*. Applied Thermal Engineering, 2016. **93**: p.  
513 1145-1154.
- 514 14. Farag, S., et al., *Temperature profile prediction within selected materials heated*  
515 *by microwaves at 2.45GHz*. Applied Thermal Engineering, 2012. **36**: p. 360-  
516 369.
- 517 15. COMSOL Multiphysics® v. 5.2. [www.comsol.com](http://www.comsol.com). COMSOL AB, Stockholm,  
518 Sweden.
- 519 16. MATLAB R2014b, The MathWorks, Inc. Natick, MA.
- 520 17. Catala-Civera, J.M., et al., *Dynamic Measurement of Dielectric Properties of*  
521 *Materials at High Temperature During Microwave Heating in a Dual Mode*  
522 *Cylindrical Cavity*. Microwave Theory and Techniques, IEEE Transactions on,  
523 2015. **63**(9): p. 2905-2914.

- 524 18. Chen, W.-H., T.-C. Cheng, and C.-I. Hung, *Numerical predictions on thermal*  
525 *characteristic and performance of methanol steam reforming with microwave-*  
526 *assisted heating*. International Journal of Hydrogen Energy, 2011. **36**(14): p.  
527 8279-8291.
- 528 19. Sturm, G.S.J., et al., *Microwaves and microreactors: Design challenges and*  
529 *remedies*. Chemical Engineering Journal, 2014. **243**: p. 147-158.
- 530 20. Pozar, D.M., *Microwave Engineering*. 2012, Wiley: Hoboken :.
- 531 21. *RF Module User's Guide*, pp. 80-82. COMSOL Multiphysics® v. 5.2.  
532 [www.comsol.com](http://www.comsol.com). COMSOL AB, Stockholm, Sweden.
- 533 22. Sturm, G.S.J., et al., *Exploration of rectangular waveguides as a basis for*  
534 *microwave enhanced continuous flow chemistries*. Chemical Engineering  
535 Science, 2013. **89**: p. 196-205.
- 536 23. *Heat Transfer Module User's Guide*, pp. 105-110. COMSOL Multiphysics® v.  
537 5.2. [www.comsol.com](http://www.comsol.com). COMSOL AB, Stockholm, Sweden.
- 538 24. Legras, B., et al., *About using microwave irradiation in competitive adsorption*  
539 *processes*. Applied Thermal Engineering, 2013. **57**(1-2): p. 164-171.
- 540 25. *CFD Module User's Guide*, pp. COMSOL Multiphysics® v. 5.2.  
541 [www.comsol.com](http://www.comsol.com). COMSOL AB, Stockholm, Sweden.
- 542 26. Nield, D.A. and A. Bejan, *Convection in Porous Media*. 2012: Springer New  
543 York.
- 544 27. Çengel, Y.A. and J.M. Cimbala, *Fluid Mechanics: Fundamentals and*  
545 *Applications, SI Version*. McGraw-Hill series in mechanical engineering. 2006:  
546 McGraw-Hill Education.
- 547 28. Legras, B., et al., *Mechanisms Responsible for Dielectric Properties of Various*  
548 *Faujasites and Linde Type A Zeolites in the Microwave Frequency Range*. The  
549 Journal of Physical Chemistry C, 2011. **115**(7): p. 3090-3098.
- 550 29. Gracia, J., et al., *Heating of Zeolites under Microwave Irradiation: A Density*  
551 *Functional Theory Approach to the Ion Movements Responsible of the*  
552 *Dielectric Loss in Na, K, and Ca A-Zeolites*. The Journal of Physical Chemistry  
553 C, 2013. **117**(30): p. 15659-15666.
- 554 30. Simon, U. and U. Flesch, *Cation-cation interaction in dehydrated zeolites X and*  
555 *Y monitored by modulus spectroscopy*. Journal of Porous Materials, 1999. **6**(1):  
556 p. 33-40.
- 557 31. Ohgushi, T. and M. Nagae, *Durability of Zeolite Against Repeated Activation*  
558 *Treatments with Microwave Heating*. Journal of Porous Materials, 2005. **12**(4):  
559 p. 265-271.
- 560 32. Tatsuo, O. and W. Akiko, *Simple suppressing method of thermal runaway in*  
561 *microwave heating of zeolite and its application*. PhysChemComm, 2001. **4**(3):  
562 p. 18-20.
- 563 33. Leong, K.C. and Y. Liu, *Numerical modeling of combined heat and mass*  
564 *transfer in the adsorbent bed of a zeolite/water cooling system*. Applied  
565 Thermal Engineering, 2004. **24**(16): p. 2359-2374.
- 566 34. Griesinger, A., K. Spindler, and E. Hahne, *Measurements and theoretical*  
567 *modelling of the effective thermal conductivity of zeolites*. International Journal  
568 of Heat and Mass Transfer, 1999. **42**(23): p. 4363-4374.
- 569 35. Jakubinek, M.B., B.-Z. Zhan, and M.A. White, *Temperature-dependent thermal*  
570 *conductivity of powdered zeolite NaX*. Microporous and Mesoporous Materials,  
571 2007. **103**(1): p. 108-112.
- 572 36. Liu, Z.Y., et al., *Fast simple and accurate measurement of zeolite thermal*  
573 *conductivity*. Zeolites, 1990. **10**(6): p. 565-570.

- 574 37. Chen, W.-H., H.-J. Liou, and C.-I. Hung, *A numerical approach of interaction*  
575 *of methane thermocatalytic decomposition and microwave irradiation.*  
576 *International Journal of Hydrogen Energy*, 2013. **38**(30): p. 13260-13271.
- 577 38. Coss, P.M. and C.Y. Cha, *Microwave regeneration of activated carbon used for*  
578 *removal of solvents from vented air.* *Journal of the Air and Waste Management*  
579 *Association*, 2000. **50**(4): p. 529-535.
- 580 39. Cherbański, R., *Calculation of critical efficiency factors of microwave energy*  
581 *conversion into heat.* *Chemical Engineering and Technology*, 2011. **34**(12): p.  
582 2083-2090.
- 583 40. Cherbański, R. and L. Rudniak, *Modelling of microwave heating of water in a*  
584 *monomode applicator – Influence of operating conditions.* *International Journal*  
585 *of Thermal Sciences*, 2013. **74**: p. 214-229.

## 586 Supporting Information

587



588

589 *Figure S1. Simulated transient maximum temperature profiles of the fixed-bed with different thermal*  
590 *conductivities and measured data with the fiber optic*