

Investigation on the Stability and Efficiency of MAPbI₃ and MASnI₃ Thin Films for Solar Cells

Julia Marí-Guaita, Amal Bouich, Muhammad Aamir Shafi, Asmaa Bouich, and Bernabé Marí*

Hybrid organic–inorganic halides are considered as outstanding materials when used as the absorber layer in perovskite solar cells (PSCs) because of its efficiency, relieve of fabrication and low-cost materials. However, the content of lead (Pb) in the material may origin a dramatic after effect on human's health caused by its toxicity. Here, we investigate replacing the lead in MAPbI₃ with tin (Sn) to show its influence on the growth of the film nucleation and stability of the solar device based on MASnI₃. By analysing the manufactured perovskite films by scanning electron microscopy (SEM), transmission electron microscopy (TEM), X-ray diffraction (XRD), UV–visible absorption, photoluminescence (PL) and atomic force microscopy (AFM), the properties of the thin films when lead is replaced by tin are reported. The simulation run for the case of MAPbI₃ is reported, where $V_{oc} = 0.856$ V, $J_{sc} = 25.65$ mA cm⁻², FF = 86.09%, and ETA = 18.91%, and for MASnI₃, $V_{oc} = 0.887$ V, $J_{sc} = 14.02$ mA cm⁻², FF = 83.72%, and ETA = 10.42%. In perovskite-based devices using MASnI₃ as absorber, it was found to be more stable despite of its lower efficiency, which could be improved by enhancing the bandgap alignment of MaSnI₃. The results of this paper also allow the development of a new, reliable production system for PSCs.

various low-cost techniques have been used. Among these techniques, spin coating with one-step process is usually used as it is easy to control and quick to produce.^[4,5] Also, the two-step process is widely used due to its ease of control for the preparation of perovskite thin layers.^[6] Other PSCs' manufacturing techniques are the two-step vapor-assisted deposition, which is advantageous for controlling the thin film morphology, thickness, and grain size,^[7] and thermal vapor deposition normally used to manufacture films with consistent thickness.

Indeed, the highest conversion efficiency in PSCs is reached when MAPbI₃ is used as an absorber. Moreover, the content of lead (Pb) in the material may lead to a dramatic aftereffect on human health caused by its toxicity. This is why the required study of substituting the lead in MAPbI₃ by tin (Sn) is highly suitable and must be done. In the article, we report on the influence on the growth of film nucleation and stability of the solar

1. Introduction

Perovskite materials possess excellent optical and electrical properties, such as a direct bandgap ranging from 1.5 to 2.3 eV and the potential to transport electrons and holes.^[1–3] For manufacturing perovskite thin films with good morphology and high quality,


device based on MASnI₃ perovskite (instead of MAPbI₃ absorbers), having elaborated both absorbers with the same one-step deposition technique. This technique exhibits wide-scale manufacturing for PSCs with high performance. Typically, the fabrication of the device is associated with the regulation of the surface shape and the capability to be developed scaled-up. The surface morphology has shown to be highly controllable and that it is the primary factor influencing the efficiency of PSCs. For this reason, many techniques and treatments for PSCs have been revised to improve the surface uniformity and the perovskite crystal formation.

In this work, we looked at the impact of replacing Pb by Sn on the properties of perovskite MAPbI₃ thin film. We report complex changes of the device by observing the morphology and the composition of the films. The obtained perovskite devices were characterized by X-ray diffraction (XRD), scanning electron microscopy (SEM), atomic force microscopy (AFM), transmission electron microscopy (TEM), photoluminescence (PL), and UV–vis absorption analysis to inquire into the electrical and optical properties of the deposited films and the performance of solar cells which were estimated by SCAPS simulator.^[8]

J. Marí-Guaita, A. Bouich, A. Bouich, B. Marí
Institut de Disseny, Fabricació i Producció Automatitzada
Universitat Politècnica de València
València 46022, Spain
E-mail: bmari@fis.upv.es

A. Bouich
Department of Computer Science Engineering
Ibn Tofail University
Kenitra 14000, Morocco

M. A. Shafi
Department of Electrical Engineering
COMSATS University Islamabad
Islamabad 45550, Pakistan

 The ORCID identification number(s) for the author(s) of this article can be found under <https://doi.org/10.1002/pssa.202100664>.

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2. Perovskite Manufacture

Perovskite absorbers ABX₃, where A = methylammonium (MA), B = (Pb or Sn), and X = Iodide (I), were manufactured employing different materials purchased from Sigma-Aldrich and Alfa

Caesar. Specifically, methylammonium iodide (MAI), thin II iodide 99% (SnI₂), and lead iodide (PbI₂) were used as precursors, and *N*-dimethylformamide (DMF) and dimethyl sulfoxide (DMSO) were used as solvents.

First, the prepared solutions of MAPbI₃ and MASnI₃ were dropped onto fluorine-dope tin oxide (FTO) glass substrates. The FTO glass was spun at 4000 rpm for 50 s during the dripping of the perovskite solution and the addition of toluene as antisolvent. Later, the samples were annealed at 60 °C for 5 min, followed by an annealing at 100 °C for 10 min. **Figure 1** shows a diagram of the elaboration process carried out.

3. Characterization Techniques

The structural properties of the thin films were examined by XRD with RIGAKU Ultima IV diffractometer using Cu K α radiation ($\lambda = 1.5418 \text{ \AA}$). SEM was used to examine the surface morphology of the perovskite layers using 1.5 kV at different magnifications. AFM measurements were performed using Nano Surf with a voltage cell from -1.5 to 1.5 V at a scan rate of 0.5 Hz. TEM (JEO-JEM-1010) analysis was carried out with 2.5 kV at different magnifying tools. UV-vis absorption analysis was made with a Si charge couple device, and PL emission was excited with a 405 nm semiconductor laser. The output

parameters of the solar cell, namely, in the short-circuit current density (J_{sc}), the open-circuit voltage (V_{oc}), the fill factor (FF), and the efficiency (η), were assessed by SCAPS simulator.

4. Results and Discussion

The XRD scanning of MAPbI₃ and MASnI₃ thin films (**Figure 2**) reveals various diffractions peaks located at 14°, 28°, and 52° matching the characteristics peaks of the perovskite materials (110), (220), and (303) respectively. Previously published structures of MAPbI₃^[9] are in good agreement with the patterns of this work. Furthermore, it was observed that the replacement of Pb by Sn leads to an increase in the intensity of the peak (110). All perovskite films form a tetragonal crystal structure with the space group *I4/mcm*.

Table 1. The structural parameters of MAPbI₃ and MaSnI₃ thin films.

Sample ID	Grain size [nm]	Roughness [nm]	Dislocation density [10^{-05} nm^{-1}]	Lattice strain (ϵ)
MaSnI ₃	403	46.7	0.62	0.39
MaPbI ₃	302	37.0	1.11	0.37

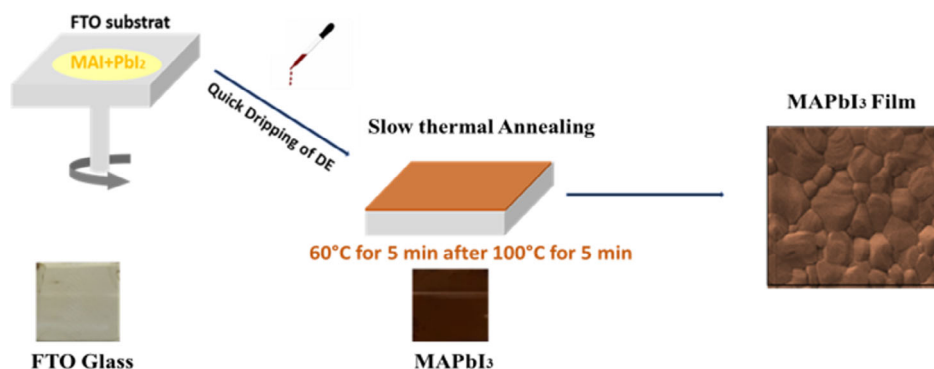


Figure 1. Elaboration process of MAPbI₃ and MASnI₃ thin films.

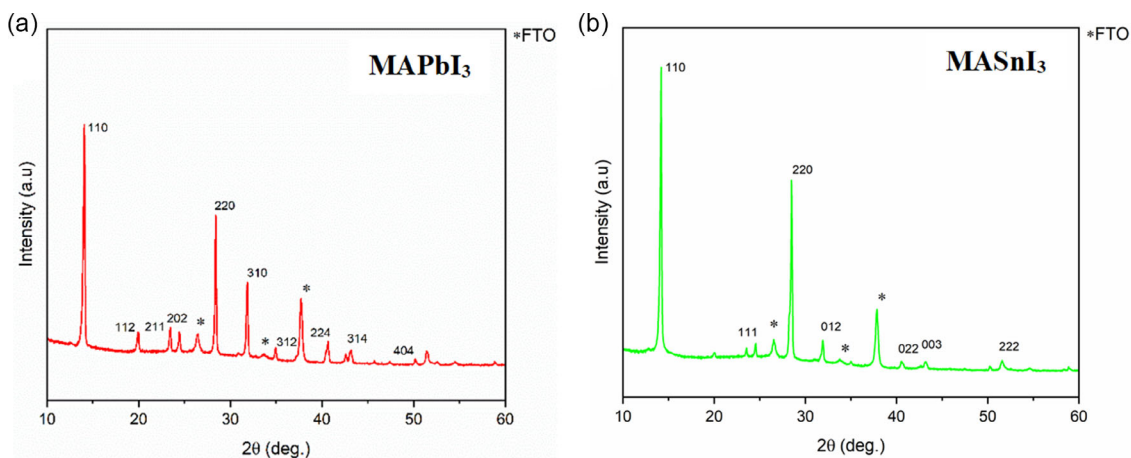


Figure 2. XRD patterns of MAPbI₃ and MASnI₃.

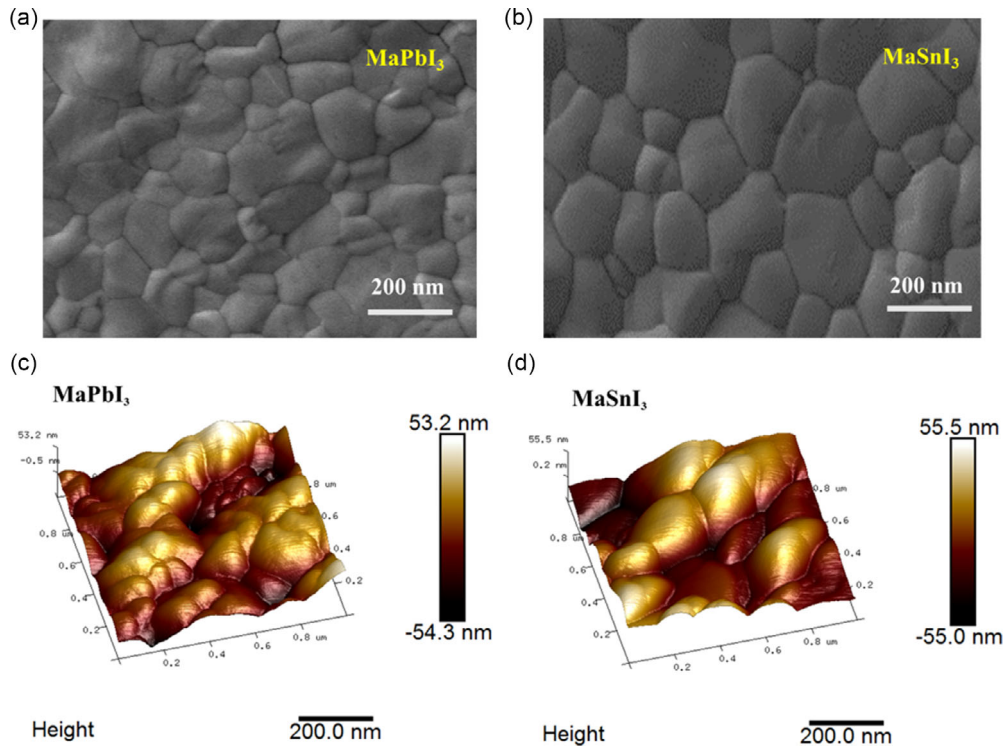


Figure 3. SEM images of a) MaPbI_3 and b) MASnI_3 . AFM images of c) MaPbI_3 and d) MASnI_3 .

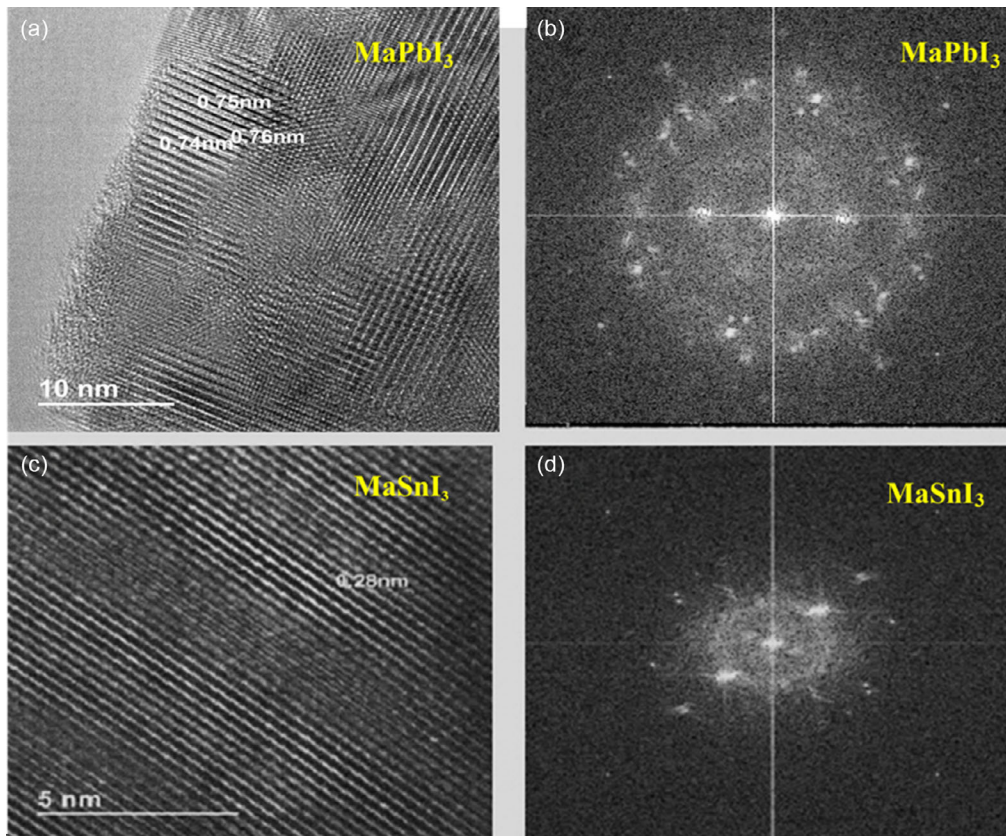


Figure 4. a,c) High resolution transmission electron microscopy images of MAPbI_3 and MASnI_3 , respectively. b,d) SAED of MAPbI_3 and MASnI_3 , respectively.

The (110) XRD peak is wider for MASnI_3 than for MAPbI_3 , as revealed by the values of full width at half maximum (FWHM) for this XRD peak ($\text{FWHM (110)} = 0.23$ for MASnI_3 and $\text{FWHM (110)} = 0.13$ for MAPbI_3), which means bigger crystallite sizes for MAPbI_3 , according to the well-known Scherrer equation.^[10]

The calculated grain size, dislocation density, lattice strain, and effective lattice strain data are represented in **Table 1**. The grain size for MASnI_3 (403 nm) thin films was found to have a substantial bigger grain size than MAPbI_3 , which was 302 nm. So, according to XRD and AFM analysis, when Pb is replaced by Sn, grain sizes increase even if crystallite sizes decrease. Further, dislocation density and effective lattice strain have been calculated to have an idea about scarcity and deformations of the grains in the film.

Figure 3a,b shows SEM images of MAPbI_3 and MASnI_3 , and the surface morphology of both thin films was smooth and contained random grain boundaries. Furthermore, MASnI_3 showed improvement of the surface morphology like previously reported investigations.^[11,12] It can be clearly observed from the homogeneity of the surface that MASnI_3 had large grain size around 403 nm, whereas for MAPbI_3 film, the grain size was about 302 nm. Also, the addition of antisolvent toluene, which evaporates the solvent and therefore creates supersaturation to speed up the crystallization process, resulted in excellent film

development. Moreover, temperature annealing treatment had a significant impact on the surface of the films. Results indicate that when annealing temperature increases, the crystal as well as the grain size grow considerably. **Figure 3c,d** shows the AFM analysis of MAPbI_3 and MASnI_3 over a $2\ \mu\text{m} \times 2\ \mu\text{m}$ region. Computed roughness of MAPbI_3 was 37.0 nm, and when Pb was replaced by Sn, roughness was 46.7 nm (**Table 1**). The surface of MASnI_3 is rougher due to the huge size of hills and troughs.

TEM examinations of MASnI_3 and MAPbI_3 thin films with lattice fringe spacings of 0.28 and 0.75 nm, respectively, corresponding to (110) or (220) of the tetragonal perovskite phases, are shown in **Figure 3a,c**. **Figure 3** exhibits the selected-area electron diffraction (SAED) spectrum and revealed that MAPbI_3 and MASnI_3 thin films are polycrystalline; validated findings (**Figure 4**).^[13–15]

PL measurement was recorded in the range of 500–1100 nm, as represented in **Figure 5**. The PL peak intensity in the region of 700–900 nm is in good agreement with the previously reported study of MAPbI_3 .^[16] The measured intensity of the PL peak in MAPbSn_3 films is about 30% higher than in MAPbI_3 films.

It is suggested that tin can be placed at the optimal level for absorbing light. This finding also boosts the enhancement crystallinity of the perovskite thin film. The UV–vis absorption spectrum of MAPbI_3 and MASnI_3 was obtained between 400 and 900 nm. The optical bandgap was calculated through the Tauc

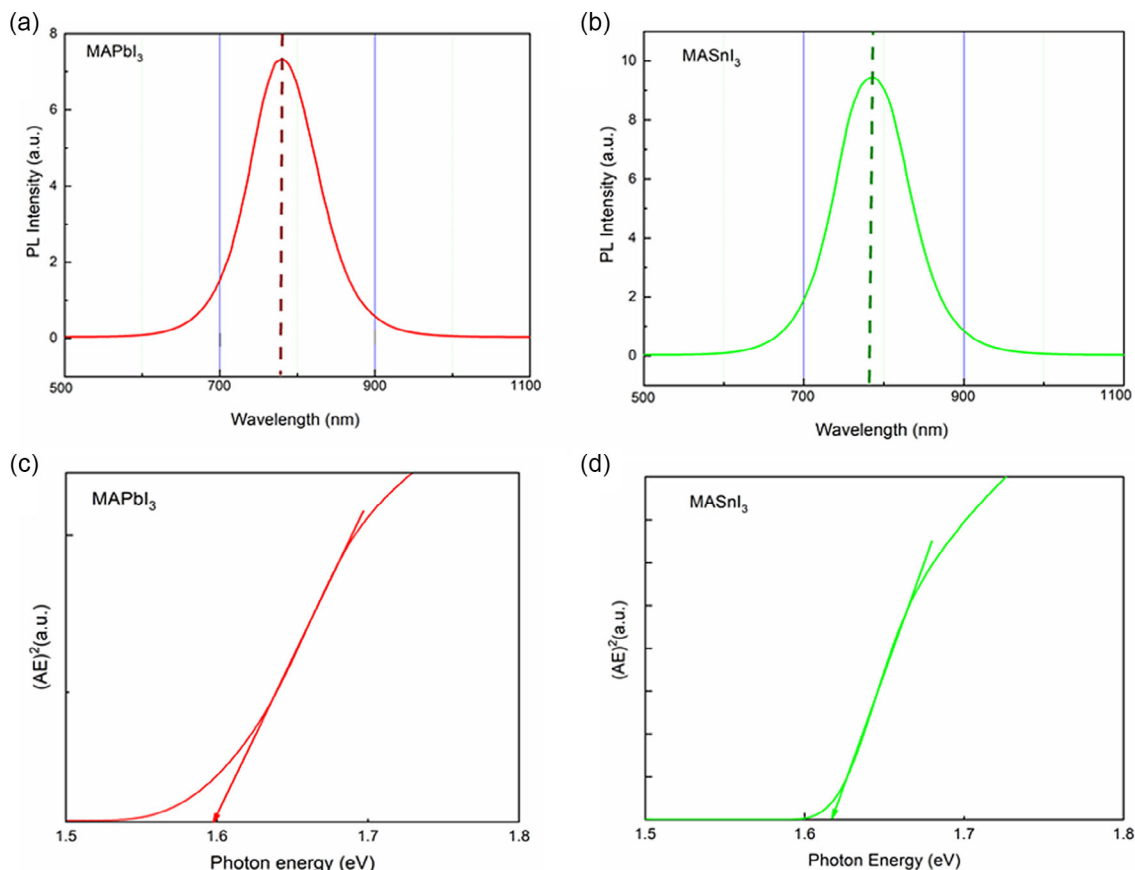


Figure 5. a) PL emission for MAPbI_3 perovskite thin films, b) PL emission for MaPbI_3 perovskite thin films, c) optical bandgap of MAPbI_3 , and d) optical bandgap of MASnI_3 .

Table 2. PL and optical bandgap for MAPbI₃ and MASnI₃.

Sample ID	PL		Optical absorption		Stokes shift [meV]
	λ [nm]	E_g [eV]	λ [nm]	E_g [eV]	
MaPbI ₃	780	1.59	775	1.60	10
MaSnI ₃	786	1.58	765	1.62	40

plot for the absorbance spectrum, from the equation $(Ah\nu)^2 = B(h\nu - E_g)$. An optical bandgap of 1.60 eV and 1.62 was found for MAPbI₃ and MASnI₃ films respectively (Figure 5c,d).^[17–19]

Table 2 summarizes the optical bandgap and PL emission peaks according to PL and UV–vis absorption measurements. The energy difference between the edge of the optical absorption and the energy of the PL emission, known as Stokes shift, is found to be higher for MASnI₃ (40 meV) than for MAPbI₃ (10 meV), meaning that the bottom of the conduction band is more filled with electrons in MASnI₃ films.

5. Degradation Study

Environmental components like oxygen and humidity have a significant impact on the photovoltaic stability of perovskite solar devices. The degradation mechanism of methylammonium-based perovskites when exposed to the environment has been attributed to a reduction of these compounds in PbI₂, CH₃NH₂, and HI. To evaluate the degradation of MAPbI₃ and MaSnI₃ samples, we performed XRD, SEM, and PL measurements on fresh and 4 weeks-aged samples, kept under 60% of humidity and in the dark (Figure 6).

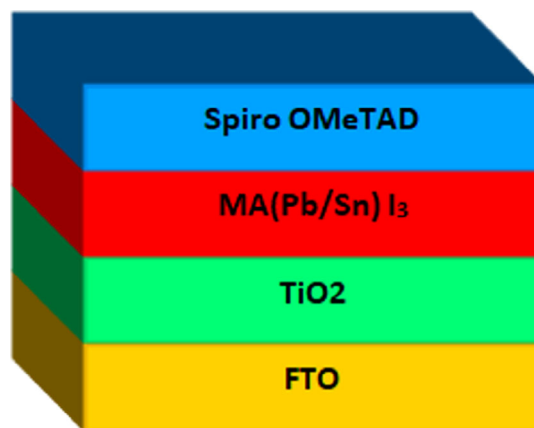


Figure 7. Schematic diagram of the proposed solar used in SCAPS.

For the 4 weeks-aged MAPbI₃ samples, XRD patterns reveal lesser intensity of characteristic peaks (Figure 6a) and a reduction of the intensity of PL emission (Figure 6c). In the case of 4 weeks-aged MASnI₃ samples, both XRD peaks and PL emission decrease in intensity, but this reduction in intensity is minor for MASnI₃ than for MAPbI₃ samples (Figure 6d,e). This fact means that MASnI₃ samples are more stable than MAPbI₃ ones. Similar results are reported in literature.^[20–24] Further, SEM images (Figure 6b,e) support this finding. New grain boundaries, as presented in Figure 6b, appear in aged MAPbI₃ films. In the case of aged MASnI₃ films, some new defects consisting of pin-holes and changes in the surface morphology with respect to no fresh samples (Figure 6e) are observed in SEM images. As a

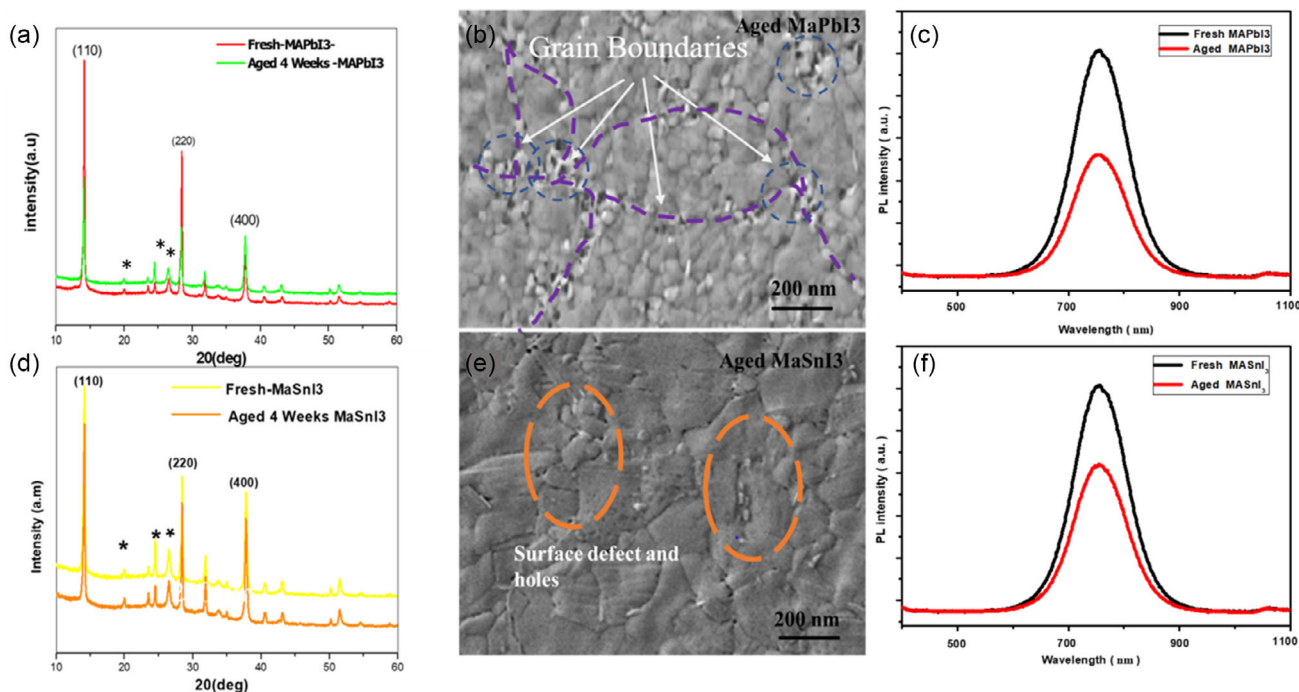


Figure 6. Degradation study of MAPbI₃ a) XRD pattern of the fresh and aged MAPbI₃ samples after 4 weeks. b) SEM image of aged MAPbI₃. c) PL emission of aged fresh and aged MAPbI₃ samples.

result, the degradation process seems to be different for both types of films.

6. Device Manufacture and Numerical Simulation

For the study of the solar cell performance, we performed the simulation of MAPbI₃ and MASnI₃-based Perovskites solar cells using SCAPS simulator software. The structure of solar spiro OMeTAD/MA(Pb/Sn)I₃/TiO₂/FTO used in the simulation is shown in the diagram (Figure 7).

The simulation was run two times for OMeTAD/MAPbI₃/TiO₂/FTO and OMeTAD/MASnI₃/TiO₂/FTO proposed models separately.^[25,26] Figure 8 and 9 show the *J-V* and *P-V* characteristics curves, and from these curves we observed that MASnI₃-based solar cell is less efficient as compared with MAPbI₃ based solar cell. The bandgap of MAPbI₃ is more favorable for light absorption as compared with the other one.

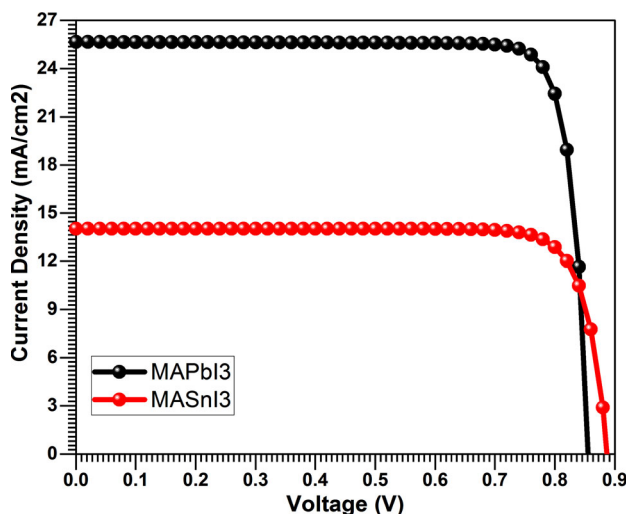


Figure 8. *J-V* characteristics curve for both MAPbI₃ and MASnI₃.

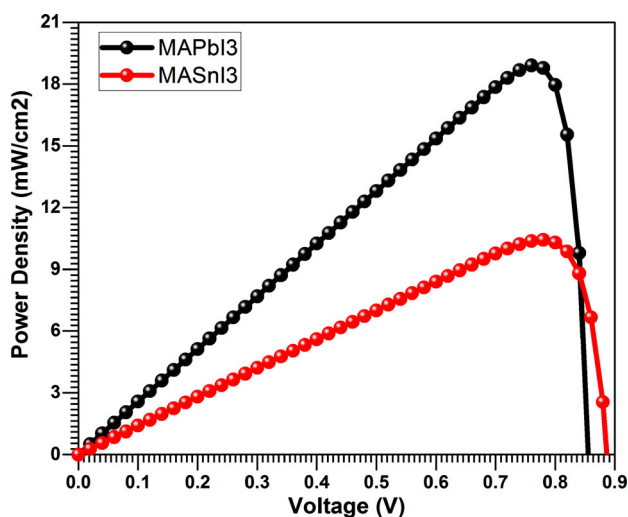


Figure 9. *P-V* characteristics curve for both MAPbI₃ and MASnI₃.

Table 3. Characteristics parameters of MAPbI₃ and MASnI₃ obtained by simulation.

Absorber	V_{oc}	J_{sc}	FF	ETA
	[V]	[mA cm ⁻²]	[%]	[%]
MAPbI ₃	0.856	25.65	86.09	18.91
MASnI ₃	0.888	14.02	83.72	10.42

The results of simulation are summarized in Table 3, where the photovoltaic characteristic parameters V_{oc} (open-circuit voltage), J_{sc} (short-circuit current density), FF, and ETA (conversion efficiency) are shown. When the simulation was run for the case of MAPbI₃, we find the V_{oc} of 0.856 V, J_{sc} of 25.65 mA cm⁻², FF of 86.09%, and ETA of 18.91%. For MASnI₃, we report V_{oc} of 0.888 V, J_{sc} of 14.02 mA cm⁻², FF of 83.72%, and ETA of 10.42%. It is worthy to notice that there is about 9% difference in the efficiencies of both devices. For Sn-based PSCs, this efficiency could be improved by enhancing the bandgap alignment of MASnI₃ with respect the hole transport layer and by adjusting film thickness.

To better show the differences of the photovoltaic parameters of MAPbI₃ and MASnI₃, a bar graph displaying the four photovoltaic parameters for both proposed devices is shown in Figure 10. This comparison shows a well-defined difference between both devices' parameters.

7. Conclusion

In summary, the use of Sn in spite of Pb to increase the stability and decrease toxicity of perovskite solar cells (PSCs) has been demonstrated. Results show a higher intensity of the characteristic peaks of perovskite when using Sn. Replacement of Pb by Sn has a significant impact on the crystallization process of

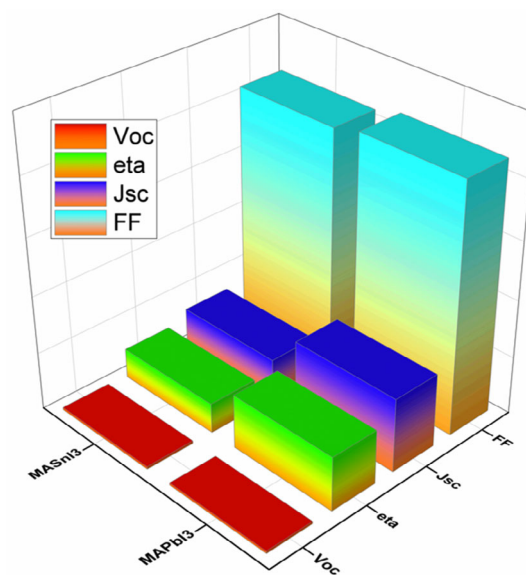


Figure 10. Graphical comparison of characteristics parameters of MAPbI₃ and MASnI₃.

perovskite materials and produces an increase in the grain size of the perovskite thin film, as seen by SEM pictures. The fact that Sn increases the grain size may result in an increase in the light absorption of the perovskite layer. Optimizing the inclusion of Sn in perovskite-based solar cells can significantly enhance stability, according to our research. Simulation was run for both films prepared. On the one hand, for MAPbI₃, we obtained $V_{oc} = 0.856$ V, $J_{sc} = 25.65$ mA cm⁻², FF = 86.09%, and ETA = 18.91%. On the other hand, for MASnI₃, we calculated $V_{oc} = 0.888$ V, $J_{sc} = 14.02$ mA cm⁻², FF = 83.72%, and ETA = 10.42%. Here, we report that there is around 9% difference in efficiencies of both devices. However, in case of Sn-based perovskite, this efficiency could be enhanced by improving the bandgap alignment of the material.

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Conflict of Interest

The authors declare no conflict of interest.

Data Availability Statement

The data that support the findings of this study are available from the corresponding author upon reasonable request.

Keywords

MAPbI₃, MASnI₃, perovskite solar cells, stability, toxicity

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[1] A. Bouich, S. Ullah, H. Ullah, M. Mollar, B. Marí, M. E. Touhami, *JOM* **2020**, *72*, 615.

- [2] N. J. Jeon, H. Na, E. H. Jung, T. Y. Yang, Y. G. Lee, G. Kim, H.-W. Shin, S. Il Seok, J. Lee, J. Seo, *Nat. Energy* **2018**, *3*, 682.
- [3] A. Bouich, S. Ullah, B. Marí, L. Atourki, M. E. Touhami, *Mater. Chem. Phys.* **2021**, *258*, 123973.
- [4] J. H. Im, H. S. Kim, N. G. Park, *Appl. Mater.* **2014**, *2*, 081510.
- [5] M. J. Carnie, C. Charbonneau, M. L. Davies, J. Troughton, T. M. Watson, K. Wojciechowski, H. Snaith, D. A. Worsley, *Chem. Commun.* **2013**, *49*, 7893.
- [6] D. Bi, S. J. Moon, L. Häggman, G. Boschloo, L. Yang, E. M. Johansson, M. K. Nazeeruddin, M. Grätzel, A. Hagfeldt, *RSC Adv.* **2013**, *3*, 18762.
- [7] Q. Chen, H. Zhou, Z. Hong, S. Luo, H. S. Duan, H. H. Wang, Y. Liu, G. Li, Y. Yang, *J. Am. Chem. Soc.* **2013**, *136*, 622.
- [8] A. Bouich, B. Hartiti, S. Ullah, H. Ullah, M. E. Touhami, D. M. F. Santos, B. Mari, *Optik* **2019**, *183*, 137.
- [9] W. Kong, Z. Ye, Z. Qi, B. Zhang, M. Wang, A. Rahimi-Iman, H. Wu, *Phys. Chem. Chem. Phys.* **2015**, *17*, 16405.
- [10] U. Holzwarth, N. Gibson, *Nature Nanotechnol.* **2011**, *6*, 534.
- [11] Z. Xiao, Q. Dong, C. Bi, Y. Shao, Y. Yuan, J. Huang, *Adv. Mater.* **2014**, *26*, 6503.
- [12] S. Luo, W. A. Daoud, *Materials* **2016**, *9*, 123.
- [13] N. G. Park, *CrystEngComm* **2016**, *18*, 5977.
- [14] X. Zheng, B. Chen, C. Wu, S. Priya, *Nano Energy* **2015**, *17*, 269.
- [15] Q. Jeangros, M. Duchamp, J. Werner, M. Kruth, R. E. Dunin-Borkowski, B. Niesen, A. Hessler-Wyser, *Nano Lett.* **2016**, *16*, 7013.
- [16] D. Liu, J. Yang, T. L. Kelly, *J. Am. Chem. Soc.* **2014**, *136*, 17116.
- [17] M. Zhang, H. Yu, M. Lyu, Q. Wang, J. H. Yun, L. Wang, *Chem. Commun.* **2014**, *50*, 11727.
- [18] J. J. Choi, X. Yang, Z. M. Norman, S. J. Billinge, J. S. Owen, *Nano Lett.* **2013**, *14*, 127.
- [19] A. Halder, R. Chulliyil, A. S. Subbiah, T. Khan, S. Chattoraj, A. Chowdhury, S. K. Sarkar, *J. Phys. Chem. Lett.* **2015**, *6*, 3483.
- [20] Z. Xie, S. Sun, Y. Yan, L. Zhang, R. Hou, F. Tian, G. G. Qin, *J. Phys. Condens. Matter* **2017**, *29*, 245702.
- [21] C. C. Homes, T. Vogt, S. M. Shapiro, S. Wakimoto, A. P. Ramirez, *Science* **2001**, *293*, 673.
- [22] G. Abdelmageed, L. Jewell, K. Hellier, L. Seymour, B. Luo, F. Bridges, J. Z. Zhang, S. Carter, *Appl. Phys. Lett.* **2016**, *109*, 233905.
- [23] J. Li, Q. Dong, N. Li, L. Wang, *Adv. Energy Mater.* **2017**, *7*, 1602922.
- [24] A. Bouich, B. Mari, L. Atourki, S. Ullah, M. E. Touhami, *JOM* **2021**, *73*, 551.
- [25] X. Wu, H. Wang, Y. Song, X. Ma, Z. Zeng, J. Wu, Y. Liu, *Improving the Performance of Organic Lead-tin Laminated Perovskite Solar Cells from the Perspective of Device Simulation*, Springer **2021**.
- [26] H. Alipour, A. Ghadimi, *Opt. Mater.* **2021**, *120*, 111432.