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Additional Information

1	"Faba bean protein films reinforced with cellulose nanocrystals as edible food packaging
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# 29 Abstract

30 In the present work, transparent films were obtained by the solution casting method from faba 31 bean protein isolate (FBP), reinforced with different cellulose nanocrystals (CNCs) content (1, 3, 32 5 and 7 wt%), obtained by acid hydrolysis of pine cone, and using glycerol as plasticizer. The 33 influence of different CNCs loadings on the mechanical, thermal, barrier, optical, and 34 morphological properties was discussed. Microstructurally, the FTIR and FESEM results 35 corroborated the formation of intramolecular interactions between the CNCs and proteins that 36 lead to more compact and homogeneous films. These interactions had a positive influence on the 37 mechanical strength properties, which is reflected in higher tensile strength and Young's modulus in reinforced films with respect to the control film, resulting in stiffer films as the CNCs content 38 39 increases. Thermal stability of the FBP films was also improved with the presence of CNCs, by 40 increasing the characteristic onset degradation temperature. In addition, the linkages formed between the CNCs, and proteins reduced the water affinity of the reinforced films, leading to a 41 42 reduction in their moisture content and water solubility, and an increase in their water contact 43 angle, obtaining more hydrophobic films as the CNCs content in the matrix increased. The 44 addition of CNCs in the FBP film also considerably improved its barrier properties, reducing its water vapour transmission rate (WVTR) and oxygen transmission rate (OTR). The present work 45 46 shows the possibility of obtaining biobased and biodegradable films of CNC-reinforced FBP with 47 improved mechanical, thermal and barrier properties, and low water susceptibility, which can be 48 of great interest in the food packaging sector as edible food packaging material.

- 49
- 50 Keywords: Faba beans; proteins films; cellulose nanocrystals (CNCs); food packaging; pine
  51 cone.
- 52

# 53 **1. INTRODUCTION**

54 In recent years, the worldwide consumption of plastics has increased in a remarkable way, mainly 55 due to its high versatility and low price. Only in Europe there was a total production of 57.9 56 million tonnes in 2019, with the packaging industry being the main consumer with a 39.6% of the 57 total (Plastics Europe, 2020). The high consumption of plastics generates a large amount of waste 58 that causes serious environmental problems, especially when used as single-use packaging 59 materials. This has led to an increase in environmental concern in society which, together with 60 petroleum depletion, has led to a considerable increase in the research and development of new, 61 more environmentally friendly materials capable of replacing traditional synthetic plastics 62 (Liminana, Garcia-Sanoguera, Quiles-Carrillo, Balart, & Montanes, 2018). Among the sustainable materials that have received the most attention, biodegradable polymeric films from 63 64 renewable sources that are abundant in nature such as proteins, lipids and polysaccharides, have 65 positioned as an actual alternative to synthetic polymer films, and are characterized by their 66 biodegradability, wide availability, and low cost (Zhang, et al., 2016). Protein films are 67 characterized by their non-toxicity, biodegradability and good barrier properties to oxygen, lipids 68 and flavourings (Reddy, Jiang, & Yang, 2012). However, they have a number of drawbacks 69 compared to synthetic polymers that limit their massive use in sectors such as packaging. Among 70 these disadvantages, it is worthy to note their low mechanical strength, poor water vapour barrier 71 properties and high water sensitivity due to the highly hydrophilic nature of the proteins 72 (Bourtoom, 2009). Although these properties highly depend on the protein structure and the 73 method of film preparation (Salgado, Ortiz, Petruccelli, & Mauri, 2010).

Legume seeds are a cheap source of protein with high nutritional value, which makes them an interesting raw material for use in the manufacture of bio-based materials for the packaging sector due to their sustainability (Makri, Papalamprou, & Doxastakis, 2005). One of the most consumed legumes worldwide is faba bean (*Vicia Faba L.*), which is characterized by its low cost and high quality protein content, which can contain up to 27.34% protein (dry weight), depending on the cultivation conditions and the variety. (Samaei, et al., 2020). Faba bean has long been a basic food

80 staple in Middle Eastern and Southeast Asian countries, however, its main use has been as a 81 livestock feed, although it has also been used as a cover crop to restore nitrogen content and 82 prevent soil erosion (Nivala, Nordlund, Kruus, & Ercili-Cura, 2020; Vioque, Alaiz, & Girón-83 Calle, 2012). This makes it an excellent raw material for protein extraction for use as films in the food packaging industry. However, currently, the use of faba bean proteins to produce 84 85 biodegradable films is still very limited. Some works such as those carried out by Saremnezhad 86 et al. (Saremnezhad, Azizi, Barzegar, Abbasi, & Ahmadi, 2011) and Montalvo-Paquini et al. 87 (Montalvo-Paquini, Rangel-Marrón, Palou, & López-Malo, 2014) analyzed the effect of different pH values, and plasticizer content of film forming solution on the physical and chemical 88 89 properties of faba bean protein films. On the other hand, Hopkins et al. (Hopkins, Stone, Wang, 90 Korber, & Nickerson, 2019) observed that faba bean protein films with 50% glycerol content, had 91 better tensile strength and opacity and lower water permeability than protein films obtained from 92 other legumes such as peas, lupins, lentils, and soybeans.

93 On the other hand, the formulation of protein films requires the use of plasticizers to reduce their brittleness and increase their functionality. The plasticizer molecules interact by hydrogen 94 95 bonding with amino-acid groups in protein chains, thus reducing the intermolecular forces 96 between them, resulting in an increase in their mobility, which translates into an increase in the 97 flexibility, toughness and tear strength of the films (Aguirre, Borneo, & León, 2013; Tian, Guo, Xiang, & Zhong, 2018). Some of the most commonly used plasticizers for the manufacture of 98 99 protein films are sorbitol, polyethylene glycol and glycerol, being glycerol one of the most 100 effective for the production of stable, flexible and less brittle films (Pérez, Piccirilli, Delorenzi, 101 & Verdini, 2016; Ramos, et al., 2013). Glycerol is one of the most important by-products of 102 biodiesel production, which is generated in large amounts due to the remarkable increase in 103 biodiesel production, becoming a waste product with an associated economic and environmental 104 cost (Rosa, et al., 2010; Tong, Luo, & Li, 2015). Glycerol represents approximately 10 wt% of 105 total biodiesel production (Yang, Hanna, & Sun, 2012). Therefore, the use of glycerol as 106 plasticizers in protein films, in addition to increasing the ductile mechanical properties of the films, can be an effective alternative to convert a by-product of the biodiesel industry into a high
value-added product (Ye, Xiu, Shahbazi, & Zhu, 2012).

109 One of the approaches to overcome the above-mentioned drawbacks of protein films is the use of 110 nanoparticles. The use of small amounts of nanoparticles has shown good potential to improve 111 the mechanical and barrier properties of protein films (Calva-Estrada, Jiménez-Fernández, & 112 Lugo-Cervantes, 2019). Inorganic nanoparticles such as montmorillonite (Azevedo, et al., 2018; Azevedo, Silva, Pereira, da Costa, & Borges, 2015; Kumar, Sandeep, Alavi, Truong, & Gorga, 113 114 2010; Pereira, Carneiro, Assis, & Borges, 2017), TiO<sub>2</sub> (Fathi, Almasi, & Pirouzifard, 2019; Li, et 115 al., 2011; Y. Liu, et al., 2019), halloysite nanotubes (Kang, Liu, Zhang, & Li, 2017; X. Liu, et al., 116 2017), or carbon nanotubes (Xiang, Guo, & Tian, 2017) have been successfully employed to reduce the drawbacks of protein films. However, the use of renewable and biodegradable 117 118 nanoparticles has gained relevance in recent years due to their greater environmentally 119 friendliness. In this way, cellulose nanocrystals (CNCs) obtained from agroforestry residues are 120 considered one of the most promising reinforcing agents due to their biodegradability, abundance, 121 variety, chemical modification capacity and price (Garcia-Garcia, Lopez-Martinez, Balart, 122 Strömberg, & Moriana, 2018). Cellulose nanocrystals consist of highly crystalline rod-like 123 particles, obtained from biomass, usually by acid hydrolysis, with dimensions of around 10-20 124 nm in width and several hundred nanometers in length, giving rise to high aspect ratios (Xu, et 125 al., 2013). CNCs are characterized by high strength, high elastic modulus and low density 126 (Moriana, Vilaplana, & Ek, 2016), despite their physico-chemical properties depend on the 127 cellulose source, and on synthesis process conditions (Silvério, Neto, Dantas, & Pasquini, 2013; 128 Slavutsky & Bertuzzi, 2014). CNCs are also characterized by good compatibility with hydrophilic 129 polymers (Han, Yu, & Wang, 2018). The effect of CNC incorporation has been previously studied 130 in soy (González & Igarzabal, 2015; Han, et al., 2018; Yu, et al., 2018; Zhang, et al., 2016), whey 131 (Qazanfarzadeh & Kadivar, 2016), amaranth (Condés, Añón, Mauri, & Dufresne, 2015), or canola 132 (Osorio-Ruiz, Avena-Bustillos, Chiou, Rodríguez-González, & Martinez-Ayala, 2019) protein 133 films, showing that the presence of nanoparticles in the protein matrix leads to intramolecular

interactions between both phases, resulting in more compact films with better mechanicalproperties and better barrier properties.

The main objective of the present work is to evaluate the effect of the incorporation of different CNC content extracted from pine cone (1, 3, 5 and 7 wt%) on the mechanical properties, thermal stability, barrier properties, optical properties, water susceptibility (moisture content, contact angle and water solubility) and the microstructure on glycerol-plasticized faba bean protein-based biodegradable films.

141

#### 142 2. EXPERIMENTAL

143 **2.1 Materials** 

Faba beans used in this work (*Gloria* variety) were supplied by RISE (Research Institutes of
Sweden). Pine cones (*Pinus Pinea*) were collected from a local pine forest in Alicante (Spain).
Sodium hydroxide (NaOH, 99.0% purity) and hydrochloric acid (HCl, 36% purity) were supplied
by VWR (Darmstadt, Germany). Glycerol was purchased from POCH S.A. (Gliwice, Poland).

148

# 149 2.2 Faba bean protein isolation

150 The faba bean protein (FBP) isolation was done following the method described by Langton et 151 al. (Langton, et al., 2020) with some modifications. First, the faba beans were dehulled with a 152 dehuller, and the beans were milled in a rotary mill from Brabender (Duisburg, Germany) with a 153 1.5 mm mesh screen. Subsequently, faba bean flour was mixed in distilled water at a ratio 1:10 154 (w/v) and the pH of the obtained solution was adjusted to 9.0 using 2 M NaOH to increase 155 solubility of the protein. The suspension was stirred at room temperature for 1 h, and then was 156 centrifuged in a Sorvall Lynx 6000 centrifuge from Thermo Scientific (Langenselbold, Germany) 157 at 3700 G for 30 min and 18 °C. The pH of supernatant was adjusted to 4.0 using 1 M HCl to 158 precipitate the protein, stirred at room temperature for 1.5 h and centrifuged at 3700 G for 30 min 159 and 18 °C. The precipitates were collected and re-dispersed in distilled water at a ratio 1:10 (w/v). 160 The suspension was adjusted to pH 4 and centrifuged again with the same described conditions. 161 Finally, the precipitate protein was frozen and lyophilized to obtain FBP.

#### 163 **2.3 Preparation of pine cone nanocrystals**

164 Pine cone nanocrystals were obtained by acid hydrolysis according to our previous work (García-165 García, Balart, Lopez-Martinez, Ek, & Moriana, 2018). Briefly, grinded pine cone particles were 166 first subjected to an alkaline treatment in a 4.5% NaOH solution for 2 h at 80 °C followed by 167 bleaching process using a solution made up of 1.7 wt% aqueous sodium hypochlorite, acetate 168 buffer (0.2 M, pH 4.8) and water (1:1:1) for 4 h at 80 °C, in order to remove extractives, 169 hemicellulose and lignin of the raw material. Subsequently, the bleached fibres were hydrolysed with sulfuric acid (65 wt%) for 45 min at 45 °C. The excess acid was then removed from the 170 171 solution by washing with Milli-Q water by repetitive centrifugation and finally by dialysis. The 172 resulting suspension was sonicated to promote CNCs dispersion and then centrifuged to remove 173 the remaining unhydrolyzed fibres. The collected suspension was then stored in a refrigerator for 174 further use.

175

# 176 2.4 Preparation of reinforced protein films

177 Preparation of FBP films was done by solution casting process. Control protein films were 178 prepared dissolving FBP (5% w/v), glycerol (50% w/w based on the dry weight of FBP) and 179 distilled water. The pH of the solution was adjusted to 10.5 using a 1M NaOH solution and was 180 mechanically stirred for 1 h. After this, the solution was denatured by heating at 85 °C for 30 min 181 in a water bath and cooled down at room temperature. For the CNCs-reinforced films, the same 182 procedure was followed as for the control film, however in this case the appropriate amounts of 183 CNCs were incorporated (1, 3, 5 and 7 wt% on the dry weight of FBP) after the denaturation 184 process, keeping the solution in a stirrer for 30 minutes. The resulting solutions were casted into 185 a polystyrene Petri dish and dried at 40 °C for 24 h. Then, the dried films were peeled off and 186 conditioned at 52% relative humidity (RH) and 25 °C for 48 h prior to testing. All films were prepared in triplicate. Table 1 summarizes the composition of all the developed films. 187

189 **Table 1.** Composition and codes for faba beans proteins (FBP) films with different pine cone

	Content					
Samples	Proteins (g)	Glycerol (g)	Water (g)	CNC (wt% on dry proteins)		
FBP	5	2.5	100	0		
FBP_CNC-1	5	2.5	100	1		
FBP_CNC-3	5	2.5	100	3		
FBP_CNC-5	5	2.5	100	5		
FBP_CNC-7	5	2.5	100	7		

cellulose nanocrystals content.

191

190

#### **192 2.5 Characterization techniques**

#### 193 2.5.1 Film thickness

194 The thickness of the FBP films was determined at ten random points around each film sample 195 using a manual micrometer Mitutoyo No. 2109S-10 (Tokyo, Japan) with 0.001 mm sensitivity. 196 The average value of each film was determined and used in calculations for mechanical properties, 197 oxygen transmission rate, water vapour transmission rate and transparency.

198

# 199 2.5.2 Mechanical properties

The mechanical tensile properties, namely the tensile strength (TS), the Young's modulus (E) and the elongation at break ( $\varepsilon$ ) of FBP films were determined following the standard method ASTM D882-02, using a Linkam TST-350 tensile stage (Linkam Scientific Instruments, Ltd., U.K.) with a 20 N load cell. Preconditioned FBP films were cut into rectangular strips (30 x 4 mm<sup>2</sup>) and placed between the tensile grips. The initial grip distance was set at 15 mm and the crosshead speed at 5 mm min<sup>-1</sup>. At least five specimens from each film were tested at room temperature and the average results were obtained.

207

208 2.5.3 Thermogravimetric analysis (TGA)

209 Thermal behaviour of different FBP films obtained was analyzed by thermogravimetric analysis

210 using a Mettler-Toledo TGA/DSC 1 thermobalance (Schwerzenbach, Switzerland). Samples with

213	min <sup>-1</sup> ). All samples were analyzed in triplicate to ensure reproducibility.
212	to 800 °C at a constant heating rate of 10 °C min <sup>-1</sup> under nitrogen atmosphere (flow rate 50 mL
211	a total weight between 8-10 mg were placed in an alumina crucible ( $70 \mu$ L) and heated from 30

a total mainte between 9, 10 ma more placed in an alumina amailta (70 mJ) and basted from 20

214

711

#### 215 *2.5.4 Moisture content*

216 Moisture content of FBP films samples was determined gravimetrically by measuring the weight 217 loss of films, with a size of  $1 \times 1 \text{ cm}^2$ , before and after drying in an oven at 105 °C for 24 h, to 218 attain a constant weight. The moisture content of each film was calculated according to the 219 following Equation (1):

220

221 Moisture content (%) = 
$$100 \times \left[\frac{W_i - W_f}{W_i}\right]$$
 (Eq. 1)

222

223 Where  $W_i$  and  $W_f$  are the initial and dried weight of the samples respectively. Three replications 224 of each film were used for calculating the moisture content average.

225

226 *2.5.5 Solubility* 

227 The solubility of FBP films in water was determined according to the method proposed by K. 228 Masamba et al. (Masamba, et al., 2016) with minor modifications. Rectangular pieces of films (10 x 10 mm<sup>2</sup>) were dried in an oven at 105 °C for 24 h, until constant weight. The weight of the 229 230 dried films was measured, and then the films were placed into a test tube with 10 mL of distilled 231 water. The tubes were slowly and periodically stirred for 24 h at 25 °C. After that period, the nonsolubilized fraction was dried in the oven at 105 °C for 24 h in order to determine the weight of 232 dry matter not dissolved in water. The % solubility of each film was determined using the 233 234 following equation (2):

235

236 
$$Solubility (\%) = 100 \times \left[\frac{W_i - W_f}{W_i}\right]$$
(Eq. 2)

238 Where  $W_i$  is the initial weight of the sample and  $W_f$  is the sample weight after drying. The 239 measurements were made in triplicate.

240

# 241 2.5.6 Static contact angle measurements

The static contact angle ( $\theta$ ) of each preconditioned FBP films was measured by an optical goniometer model FM140 (110/220 V, 50/60 Hz) from KRÜSS GmbH (Hamburg, Germany) at room temperature. This goniometer is equipped with a video capture kit and analysis software (Drop Shape Analysis SW21; DSA1). Five different water droplets, using distilled water as contact liquid, were deposited onto the film surface with a micro syringe and ten measurements were obtained for each drop and averaged. The contact angle was measured 30 s after depositing the drop to obtain stabilized values.

249

#### 250 2.5.7 Transparency

FBP films transparency was determined using a UV-Vis spectrophotometer Perkin Elmer model Lambda 2 (Massachusetts, EEUU) according to ASTM D1746. Preconditioned film specimens were cut into rectangle shape and directly placed in the spectrophotometer cell. The percent transmittance of light at 600 nm in each film was measured by triplicate, and the transparency value was calculated by the following equation (3):

256

257

$$Transparency = \left[\frac{LogT_{600}}{x}\right]$$
(Eq. 3)

258

259 Where  $T_{600}$  is the transmittance at 600 nm of each film and x is the film thickness (mm).

260

# 261 2.5.8 Colour of films

The effect of CNCs in the colour properties of FBP films was studied in a colorimeter model
KONICA CM-3600d COLORFLEX-DIFF2 from Hunter Associates Laboratory (Virginia,
EEUU) calibrated using a white standard. The CIELAB colour space was used to measure the

degree of  $L^*$  (lightness),  $a^*$  (red-green) and  $b^*$  (yellow-blue) of each film. Five measurements were made at random positions over each film surface and average values were calculated. Total colour difference ( $\Delta E$ ) was calculated using the following equation (4):

268

270

Where  $\Delta L^*$ ,  $\Delta a^*$  and  $\Delta b^*$  are the differences between the corresponding colour parameters of the samples and the colour parameter values of the control film ( $L^* = 66.6$ ,  $a^* = 5.5$ ,  $b^* = 37.0$ ).

(Eq. 4)

 $\Delta E = \sqrt{(\Delta L^{*})^{2} + (\Delta a^{*})^{2} + (\Delta b^{*})^{2}}$ 

- 273
- 274 2.5.9 Oxygen Transmission Rate (OTR)

The oxygen transmission rate (OTR) of the FBP films was obtained using an OTR Permeation Analyzer Mocon OX-TRAN model 2/20 from Modern Controls Inc. (Minneapolis, USA) equipped with a coulometric oxygen sensor in accordance with ASTM D3985-95. The area of measurement of the samples was 5 cm<sup>2</sup> and the OTR measurements were performed at 23 °C and 50% RH. Before the analysis, the samples were conditioned for 2 days at 23 °C and 50% RH. At least two samples of each film were tested.

281

#### 282 2.5.10 Water vapor permeability (WVP)

283 The WVP test was conducted gravimetrically in triplicate following ASTM E96-95. Conditioned 284 circular films were sealed on permeability cups TQC Sheen B.V. model VF2200 (Capelle aan 285 den IJssel, Netherlands) containing 2 g of CaCl<sub>2</sub> (0% RH) with an exposed area of 10 cm<sup>2</sup>. Later, 286 the cups were placed in a desiccator at  $25 \pm 1$  °C containing a saturated Mg(NO<sub>3</sub>)·6H<sub>2</sub>O solution 287 to reach a relative humidity of  $54\% \pm 3\%$ . The cups were weighed each hour for a total period of 288 8 hours using an analytical balance. The changes in cup weight were plotted as a function of time 289 and the slope was calculated by linear regression. Finally, the WVTR was calculated according to the following equation (5) (Trifol, et al., 2016): 290

$$WVTR = \frac{n \times l}{s}$$

292

Where *n* is the slope of the straight line, l is the thickness of the film and *S* is the exposed area of the film.

(Eq. 5)

296

#### 297 2.5.11 Field Emission Scanning Electron Microscopy (FESEM)

The morphology of the cross section of nitrogen cryofractured films was studied in a field emission scanning electron microscope (FESEM) ZEISS model ULTRA55 (Eindhoven, The Netherlands) with an operating voltage of 1 kV. Prior to the morphological characterization, the cryofracture surface of films were coated with a thin layer of platinum in a high vacuum sputter coater EM MED20 from Leica Microsystems (Milton Keynes, United Kingdom).

303

# 304 2.5.12 Attenuated total reflection-Fourier transform infrared (ATR-FTIR) spectroscopy

The FBP films spectra were recorded by attenuated total reflection method (ATR) in an infrared spectrometer Perkin-Elmer Spectrum BX (Perkin-Elmer Spain S.L., Madrid, Spain). Each sample was subjected to 20 scans between 4000 and 600 cm<sup>-1</sup> with a resolution of 4 cm<sup>-1</sup>. After attenuation of total reflectance and baseline correction, spectra were normalized with a limit ordinate of 1.5 absorbance units.

310

# 311 3. RESULTS AND DISCUSSION

# 312 **3.1 Mechanical Properties**

The mechanical properties of the control film (FBP) and films reinforced with different CNCs content are shown in Fig. 1. As can be seen, the incorporation of low CNCs content (1 and 3 wt%) to the FBP film hardly affects the strength and tensile modulus, obtaining values very similar to those of the unreinforced FBP film. However, it is observed that the addition of higher CNCs amounts (5 and 7 wt%) results in a considerable increase in tensile strength, with values of 5.3 MPa and 6.5 MPa for the films reinforced with 5 and 7 wt% of CNCs respectively, which 319 represents a percentage increase of 22.3% and 50.5% respectively, with respect to the 320 unreinforced film (4.3 MPa). The tensile strength values obtained with the film reinforced with 7 321 wt% CNCs are higher than those obtained with other protein films reinforced with similar or 322 higher amounts of CNCs (González, et al., 2015; Osorio-Ruiz, et al., 2019; Sukyai, et al., 2018). 323 This tendency is also observed in Young's modulus, where it can be seen that the addition of 5 324 and 7 wt% of CNCs into the FBP film resulted in an increase of 43.7% and 76.1% respectively 325 with respect to the unreinforced protein film (67.1 MPa). Therefore, the incorporation of CNCs 326 into the FBP film resulted in an increase in its stiffness, which increases with the amount of 327 reinforcement in the matrix. This is due to the increased hydrogen bonds formed between the 328 protein matrix and the reinforcement as CNCs loading increases, thus increasing the cohesion of 329 the films and reducing their flexibility (González, et al., 2015; Sukyai, et al., 2018). This same 330 trend with respect to mechanical strength properties was observed by Sukyai et al. (Sukyai, et al., 331 2018) who studied the effect of different content of cellulose nanocrystals from sugarcane bagasse 332 (0, 2, 5 and 8 wt%) in whey protein films. In this case, the authors observed that the addition of 8 333 wt% CNCs to the whey protein film resulted in an increase in tensile strength from 2.30 MPa to 334 4.93 MPa and Young's modulus from 57.56 MPa to 187.42MPa with respect to the unreinforced 335 whey protein film.

336 With respect to the elongation at break, Fig. 1 shows how the addition of CNCs results in a 337 decrease of ductile mechanical properties of the FBP film. In this case, it is observed that the 338 control film shows an elongation at break of about 105%. This elongation at break is significantly 339 higher than the elongation at break obtained in other protein films with similar plasticizer content, such as soy protein films (González, Gastelú, Barrera, Ribotta, & Igarzabal, 2019) or whey protein 340 341 films (Galus & Kadzińska, 2016). The addition of 1 wt% of CNCs considerably reduces the 342 elongation at break of the control film, changing from 104.4% in FBP film to 63.9%. For higher 343 amounts of CNCs (3 and 5 wt%) the elongation at break decreases to around 54%, while for films reinforced with 7 wt% CNCs the elongation decreases to 48.9%, which represents a decrease of 344 345 around 53.1% with respect to the elongation at break of the control film. This decrease in elongation at break is typical of protein films reinforced with cellulose nanocrystals, since the 346

347 interactions of the CNCs with the proteins lead to a reduction in the polymer chains' mobility, 348 thus reducing their ductility (Shabanpour, Kazemi, Ojagh, & Pourashouri, 2018).

349 Comparing the mechanical properties of unreinforced FBP films and films reinforced with 350 different CNCs content with those of synthetic films used in the food packaging sector, it can be 351 observed that the elongation at break obtained in the control film is higher than that of widely used materials such as PET (70%) or HDPE (20-50%). However, these materials have higher 352 353 tensile strength values than FBP films. On the other hand, it is observed that the tensile strength 354 achieved by the FBP film reinforced with 7 wt% CNCs has a tensile strength value similar to that 355 of LDPE (7-25 MPa) and EVA (6-19 MPa) (Bastarrachea, Dhawan, & Sablani, 2011).

356



Fig. 1.- Tensile properties of unreinforced (control) FBP film and FBP film reinforced with 358 359 different CNCs content: a) stress-strain curves and b) tensile properties.

360

#### 361 **3.2 Thermal properties**

362 The thermal stability of unreinforced FBP film and films reinforced with different CNCs content 363 was studied by thermogravimetric analysis. Fig. 2 shows the mass loss curves of the FBP films 364 with respect to temperature and their corresponding first derivative curves (DTG). Additionally, 365 Table 2 gathers the most relevant thermal degradation properties, namely the onset degradation 366 temperature of each film and the maximum degradation temperatures in each of the different 367 degradation stages, as well as the percentage of mass loss obtained in each of them. As can be

seen in all the curves, three main stages can be identified. The weight loss that occurs at low 368 369 temperatures (30-125°C) is related to the moisture loss absorbed by the films. The second 370 degradation stage (125-280°C) is attributed to the loss of low molecular weight protein fractions 371 and glycerol evaporation. Finally, the third stage (280-500°C), is related to the protein and CNCs 372 degradation (Osorio-Ruiz, et al., 2019). As shown in Table 2, the control film (FBP) has a higher moisture content than the CNCs-reinforced films. This is a clear evidence of the increased 373 374 hydrophobicity of the film due to the presence of the CNCs, as their interaction with the proteins 375 reduces the number of active sites capable of reacting with the water molecules, thus reducing the 376 moisture content of the film. It can also be observed that the FBP film onset degradation temperature increases with the presence of CNCs, with higher values as the CNCs content 377 378 increases. In this case, the highest onset degradation temperature is obtained for the film reinforced with 7 wt% CNCs, with an increase of about 15 °C with respect to the control film. On 379 380 the other hand, if the third degradation stage is observed, related to the degradation of the proteins 381 and CNCs, it can be seen that the reinforced films present higher thermal stability with increasing 382 CNCs content. This increase in degradation temperature in reinforced films may be due to the 383 higher thermal stability of CNCs compared to proteins, as well as the formation of strong 384 intermolecular bonds between CNCs and the matrix phase resulting in improved thermal stability 385 compared to the control film (Chang, Jian, Zheng, Yu, & Ma, 2010; Osorio-Ruiz, et al., 2019). In 386 this case, the highest degradation temperature ( $T_{max3}$ ) was obtained in the film reinforced with 7 387 wt% of CNCs, with a value of 318.9 °C. This is due to the higher number of interactions 388 established between the proteins and CNCs, resulting in an increase in their thermal stability.



**Fig. 2.** a) TGA curves and b) DTGA curves of unreinforced (control) FBP film and FBP film

391

# reinforced with different CNCs content.

392

**Table 2.** Thermal parameters of unreinforced (control) FBP film and FBP film reinforced with

394

different	<b>CNCs</b>	content.
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Same la s	T <sub>0</sub> <sup>[a]</sup> (°C)	Stage 1 30 - 125 °C	Stage 2 125-280 °C		Stage 3 280-500 °C		Residue	
Samples		Mass loss (%)	Mass loss (%)	T <sub>max2</sub> (°C)	Mass loss (%)	T <sub>max3</sub> (°C)	Mass (%)	
FBP	$155.9\pm0.1$	$2.8\pm0.2$	34.2 ± 2.9	$245.9\pm1.2$	44.1 ± 3.3	$311.9\pm1.5$	$18.9\pm0.3$	
FBP_CNC-1	$165.4\pm1.9$	$2.1\pm0.0$	$45.6\pm1.6$	$237.9 \pm 1.7$	35.3 ± 1.3	$311.7\pm0.7$	$17.1\pm0.3$	
FBP_CNC-3	$166.7\pm1.4$	$2.0\pm0.1$	$39.6 \pm 1.6$	$235.0\pm2.7$	$40.0\pm1.2$	$316.4\pm0.4$	$18.4\pm0.3$	
FBP_CNC-5	167.1 ± 1.0	$2.1\pm0.0$	$43.7\pm1.2$	$232.8\pm1.9$	$36.3\pm0.9$	$318.5\pm0.5$	$17.9\pm0.3$	
FBP_CNC-7	$170.8\pm2.1$	$1.9\pm0.2$	43.2 ± 3.4	$243.7\pm3.7$	36.3 ± 3.0	$318.9\pm0.6$	$18.6\pm0.1$	

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[a]  $T_0$  is calculated at 5% weight loss.

396

# **397 3.3 Water susceptibility**

398 Moisture content, water solubility and contact angle values of different FBP films are shown in 399 Table 3. As can be seen, the addition of CNCs to the protein film results in a slight reduction in 400 the moisture content of the films, being this reduction greater for higher CNCs contents (5 and 401 7%). This decrease in the moisture content of the reinforced films may be due to the interactions 402 formed between the hydroxyl groups of the CNCs and the free functional groups of the proteins, 403 which reduce the number of available active sites capable of interacting with the water molecules, 404 resulting in an increase in the hydrophobicity (Qazanfarzadeh, et al., 2016). This reduction of 405 hydrophilic groups in the protein film due to interactions with CNCs also leads to an increase in 406 the surface hydrophobicity. As can be seen in Table 3, the contact angle of the control film (56.7°) 407 increases after the addition of CNCs, being higher as the CNCs concentration in the film increases. 408 In this case, the maximum contact angle is obtained for the FBP film reinforced with 7 wt% 409 CNCs, with a contact angle of 66.7°, which represents an increase of 21.4% with respect to the control film (Condés, et al., 2015). This increase in FBP film hydrophobicity after CNCs 410 incorporation is of great interest for food packaging applications. However, the contact angle is 411

412 still lower than that of materials used as food packaging such as LDPE ( $100.2^{\circ}$ ), HDPE ( $92.4^{\circ}$ )

413 or PP (90°), which is evidence of their hydrophilic nature (Fávaro, Rubira, Muniz, &
414 Radovanovic, 2007; Karbowiak, Debeaufort, Champion, & Voilley, 2006).

415 Table 3 shows the water solubility values of the different films. Water solubility is an important 416 property to be considered in food packaging applications where the film has to be in contact with 417 food that contains a large amount of water. In this case it is observed that the unreinforced FBP 418 film shows a low water solubility compared to other protein films, such as soy protein films 419 (González, et al., 2019; González, et al., 2015), whey protein (Sukyai, et al., 2018) or Amaranth 420 protein films (Condés, et al., 2015). This low solubility of the films was evidenced by the good 421 structural integrity obtained after 24 hours of immersion in water, which showed that only small 422 amounts of low molecular weight peptides and plasticizer were dissolved in the water during the 423 test (Wang, et al., 2017). In this case, it is observed that the incorporation of CNCs in the FBP 424 film decreases its water solubility, obtaining a reduction in solubility of around 10% for films 425 with CNCs contents of 3, 5 and 7 wt% compared to the control film. This reduction in solubility 426 is due to the strong hydrogen bonding interactions established between the hydroxyl groups of 427 CNCs and the carboxyl and amino groups of proteins, which improve their cohesion and reduce 428 their susceptibility to water molecules (Abdollahi, Alboofetileh, Behrooz, Rezaei, & Miraki, 429 2013; Shabanpour, et al., 2018). Some authors have reported that this reduction in water solubility 430 may also be influenced by the high crystallinity of the CNCs (Deepa, et al., 2016).

431

**Table 3.** Moisture content, water solubility and contact angle for unreinforced (control) FBP
film and FBP film reinforced with different CNCs content.

Somelas			Sample		
Samples	FBP	FBP_CNC-1	FBP_CNC-3	FBP_CNC-5	FBP_CNC-7
Moisture content (%)	$15.9\pm0.5$	$15.4\pm0.6$	$15.3\pm0.7$	$14.2\pm0.3$	$14.3\pm0.9$
Water solubility (%)	$34.6\pm0.5$	$33.6\pm0.2$	$31.7\pm0.5$	$31.1\pm1.1$	$31.5\pm0.8$
Contact angle (°)	$56.7 \pm 1.3$	$57.4\pm2.1$	$61.6 \pm 1.4$	$64.6\pm3.1$	$66.7 \pm 1.0$

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# 436 **3.4 Chemical structural properties**

The chemical interactions between CNCs and the proteins during film production were assessed 437 438 by ATR-FTIR. Fig. 3 shows the FTIR spectra of the unreinforced FBP film and the films 439 reinforced with different CNCs content. As can be seen, the control film spectra (FBP) shows the main absorption bands of the peptide linkage which are the peak at 1630 cm<sup>-1</sup> corresponding to 440 the amide I band (C=O stretching), the peak at 1538 cm<sup>-1</sup> attributed to the amide II band (N-H 441 442 bending and C-N stretching) and the peak at 1238 cm<sup>-1</sup> corresponding to the amide III band (N-H 443 bending and C-N stretching) (Montalvo-Paquini, et al., 2014). Other characteristic peaks appearing in the spectra are the peak at 3272 cm<sup>-1</sup> related to the N-H and O-H groups of the 444 proteins, and the peak at 1038 cm<sup>-1</sup> associated with C-C and C-O stretching vibrations. On the 445 446 other hand, the peaks located in the range between 800 and 1150 cm<sup>-1</sup> are attributed to the glycerol 447 used as a film plasticizer (Shabanpour, et al., 2018; Sogut, 2020; Sukyai, et al., 2018). After the 448 incorporation of CNCs, no significant changes in the spectra were observed. However, it is observed that in the reinforced films a new peak appears at 1158 cm<sup>-1</sup> related to the saccharide 449 450 structures of CNCs (Yu, et al., 2018), as well as a change in the intensity of the bands associated 451 to the amide I, II, and III compared to the control film. These differences are most evident in the 452 case of the film reinforced with 7 wt% CNCs. This suggests the formation of interactions between 453 the main polypeptide chains of proteins and the CNCs, consisting mainly of hydrogen bonds between the amino and carboxyl groups of the proteins and the hydroxyl groups of the CNCs 454 455 (Martelli-Tosi, et al., 2018; Shabanpour, et al., 2018).



CNCs content.

458 Fig. 3. FTIR spectra of unreinforced (control) FBP film and FBP film reinforced with different

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3.5 Morphological properties

The FESEM images of the cryofractured cross section of the different films are gathered in Fig. 461 462 4. As can be seen, the control film has a smooth and homogeneous surface, with the presence of 463 small microcracks and numerous surface pores. After the CNCs incorporation, a modification of 464 the fracture surface is observed. In this case, CNCs-reinforced films have a higher cross-section 465 roughness, which increases as the CNCs content in the film increases. Furthermore, it is observed 466 that the CNCs incorporation in the film leads to a reduction in the size and pore number on the 467 surface, resulting in a more homogeneous and uniform surface. This is most noticeable in samples 468 reinforced with small CNCs content (1% and 3%). This reduction in the size and number of pores 469 on the surface evidences the good interaction between the functional groups of the proteins, the 470 glycerol and the CNCs, which results in more compact and resistant films, but with a lower 471 molecular mobility, which is reflected in the mechanical properties with an increase in the tensile 472 strength and a decrease in the elongation at break (Mirpoor, et al., 2020; Osorio-Ruiz, et al., 2019; 473 Yu, et al., 2018).



475

476 Fig. 4. Field emission scanning electron microscopy (FESEM) images at 5000× of
477 cryofractured cross section of FBP-based films: (a) FBP (control); (b) FBP\_CNC-1; (c)
478 FBP\_CNC-3; (d) FBP\_CNC-5 and (e) FBP\_CNC-7.

# 480 **3.6 Barrier properties**

Fig. 5 shows the OTR and WVTR values of unreinforced FBP film (control), and films reinforced 481 482 with different CNCs content. As can be seen in Fig. 5a, the CNCs addition to the control FBP 483 film results in a reduction of the OTR values, being this reduction greater as the CNCs content in the film increases. In this case, the lowest OTR value was obtained in the film with a CNCs 484 content of 7 wt% (3.67×10<sup>3</sup> mL µm m<sup>-2</sup> day<sup>-1</sup>), obtaining an OTR value 36.2% lower than the 485 control film  $(5.75 \times 10^3 \text{ mL } \mu\text{m } \text{m}^{-2} \text{ day}^{-1})$ . This OTR decrease is due to the formation of a dense 486 487 network structure by the interaction between proteins and CNCs which results in an increase in the tortuosity of the diffusion pathway of the oxygen molecules, hindering their transfer through 488

the film and thus increasing its oxygen barrier property (Xie, et al., 2020). This reduction of
oxygen permeability is of great importance in the food packaging sector, as it allows to increase
the product's shelf life.

492 In the same way, it can be seen in Fig. 5b that the WVTR is also reduced by the presence of CNCs, 493 improving the water barrier property of the FBP films. In this case, the addition of 1 wt% CNCs 494 hardly affects the water vapour permeability. However, higher CNCs content results in a decrease 495 in WVTR with respect to the control film, leading to a greater decrease as the CNC content in the film increases. The WVTR of the control film was  $2.93 \times 10^4$  g µm m<sup>-2</sup> day<sup>-1</sup>, reducing to  $2.27 \times 10^4$ 496 g µm m<sup>-2</sup> day<sup>-1</sup> after incorporating 7 wt% CNCs, a decrease of about 22.7%. This improvement 497 498 of the water vapour barrier property of the reinforced films can be influenced by several aspects, 499 such as the decrease of the films' hydrophilicity due to the presence of CNCs, or the increase of 500 the tortuosity of the diffusion path of water molecules due to the presence of highly crystalline 501 CNCs. In addition, the restricted polymer chain mobility and the higher matrix structure density 502 obtained due to the strong interactions of the functional groups of proteins and CNCs also 503 contribute to the reduction of the water vapour permeability of CNC-reinforced films. (Azevedo, 504 et al., 2020; Rafieian, Shahedi, Keramat, & Simonsen, 2014).



Fig. 5. a) OTR and b) WVTR of unreinforcing (control) FBP film and FBP film reinforcing

with different CNCs content.

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#### 510 **3.7** Colour and transparency properties

511 Fig. 6 shows the light transmission in the UV and visible ranges and the transparency of unreinforced FBP film and films reinforced with different CNCs content. As can be seen, all films 512 show a good ability to absorb UV light at wavelengths between 250 and 300 nm, while for 513 wavelengths between 300 and 400 nm, the CNCs addition to the FBP film reduces the UV light 514 515 absorption, which is of interest in the food preservation packaging sector. In the visible range 516 region (400-700 nm), the FBP film shows a high transmittance percentage, which is reduced after 517 the CNCs incorporation. As can be seen in Fig. 6a, the transmittance decreases as the CNC content 518 of the film increases.

519 The films' transparency is one of the most important properties to be considered for their use in 520 the food packaging industry. Fig. 6b shows that the control film has a high transparency, however, 521 after CNCs addition, this transparency is slightly reduced, obtaining a higher opacity as the CNCs 522 content in the film increases. The decrease in transparency of the reinforced films may be related 523 to light scattering caused by the dispersed CNCs in the matrix which leads to a reduction of light 524 passing through the film. The increased opacity of reinforced films may also be due to the 525 formation of CNCs aggregates during film fabrication, which cause a slight darkening and make 526 it difficult for light to pass through the matrix (Qazanfarzadeh, et al., 2016; Shabanpour, et al., 527 2018). This loss of transparency in CNCs-reinforced protein films has also been observed by 528 several authors such a Sukya et al. (Sukyai, et al., 2018), Han et al. (Han, et al., 2018) or Huang 529 et al. (Huang, Tao, Ismail, & Wang, 2020). In this case, FBP films are less transparent than other 530 CNCs-reinforced films, such as CNCs-reinforced whey protein films from sugar cane bagasse. 531 (Sukyai, et al., 2018). However, it is worth noting that the control film and the FBP film reinforced 532 with 1 wt% of CNCs show higher transparency than the commercial LDPE film used for food 533 packaging, 4.26 A/mm, being this very similar to that obtained in the FBP films reinforced with 534 3, 5 and 7 wt% of CNCs (Guerrero, Hanani, Kerry, & De La Caba, 2011).

535



Fig. 6. a) transmittance, b) transparency and c) macroscopic aspect of unreinforced (control) 538 539 FBP film and FBP film reinforced with different CNCs content.

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Table 4 shows the surface colour values of unreinforced film and films reinforced with different 541 542 CNCs content. As can be seen, the incorporation of CNCs into the FBP films results in a slight increase in  $L^*$  and a decrease in the  $a^*$  and  $b^*$  coordinate values, which is representative for a 543 544 slight increase of the brown colour in the samples, probably due to the formation of CNC 545 aggregates in the film. However, the  $\Delta E$  values of the CNC-reinforced films show that the 546 reinforcement addition into the FBP film hardly alters its colour properties.

548 **Table 4.** Colour parameters from CIELab space of unreinforced (control) FBP films and FBP

549

film reinforced with different CNCs content.

Parameter	FBP	FBP_CNC-1	FBP_CNC-3	FBP_CNC-5	FBP_CNC-7
$L^*$	$66.6\pm0.4$	$68.0\pm0.7$	$69.8 \pm 1.3$	$68.6 \pm 1.6$	$69.4 \pm 1.4$
$a^*$	$5.5\pm0.3$	$4.3\pm0.3$	$3.9\pm0.6$	$5.1\pm0.7$	$4.0\pm0.5$
$b^*$	$37.0\pm0.5$	$34.5\pm0.8$	$35.6\pm1.2$	$35.1 \pm 1.3$	$34.2\pm1.2$
$\Delta E  (Control)^{[a]}$	-	3.1	3.8	2.7	4.2

550 [a]  $\Delta E$  is calculated from the color difference between FBP film and each FBP/CNC film.

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#### 552 **4. CONCLUSIONS**

553 In the present work, the effect of the incorporation of different CNCs content (1, 3, 5 and 7 wt%) 554 synthesized from pine cone on the mechanical, thermal, barrier, water susceptibility, optical and morphological properties of faba bean protein (FBP) films has been evaluated. The CNCs addition 555 556 to the FBP film resulted in the formation of intramolecular interactions between the hydroxyl 557 groups of the CNCs and the amino and carboxyl groups of the proteins, mainly through hydrogen 558 bonds. This chemical affinity resulted in more compact films with lower chain mobility, which 559 was reflected in the mechanical properties of the reinforced films through an increase in tensile 560 strength and Young's modulus and a decrease in the elongation at break compared to the 561 unreinforced FBP film. In this case, the incorporation of 7 wt% CNC in the FBP film resulted in 562 an increase in tensile strength and Young's modulus of 50.5 and 76.1% respectively with respect 563 to the control film, as well as a decrease in elongation at break of about 53.1%. The thermal 564 stability of the FBP film was also affected by the presence of CNCs, increasing the onset 565 degradation temperature as the CNC content in the film increases. In this case, an increase of 15 566 °C was obtained in the onset degradation temperature of the film reinforced with 7 wt% of CNCs 567 compared to the control film. A slight increase in the maximum degradation temperature was also 568 observed in the reinforced films with respect to the FBP film. The interactions formed between 569 the CNCs and the proteins also reduced the amount of active sites able to react with the water 570 molecules in the proteins, which resulted in an increase in the hydrophobicity of the CNC-571 reinforced films, as evidenced by a reduction in moisture content and water solubility and an

572 increase in their contact angles. The CNCs addition to the FBP film also significantly improved 573 its barrier properties, reducing its WVTR and OTR as the CNC content in the matrix increased. 574 In this case, the film reinforced with 7 wt% CNC obtained 36.2 and 22.7% lower OTR and WVTR 575 values, respectively, compared to the unreinforced FBP film. Finally, it was observed that the 576 presence of CNCs in the FBP film led to an increase in its opacity as the CNCs content increases. 577 This work has shown that CNCs obtained from pine cone are of great interest to improve some 578 physical and mechanical properties of FBP films, obtaining biobased and biodegradable materials 579 that could present attracting potential for use in the food packaging sector as edible food 580 packaging material. These films could be a potential solution to reduce the amount of non-581 degradable food packaging waste of petrochemical origin.

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