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Additional Information

1 2 3	ASSESSMENT OF VINE-SHOOTS IN A MODEL WINES AS ENOLOGICAL ADDITIVES
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13	ABSTRACT
14	For the first time, the possibility of using vine-shoots as enological additives, similar
15	to oak chips, is proposed. Their volatile and phenolic compounds contribution to a model
16	wine was studied. Variables such as vine-shoot vinifera (Airén and Cencibel), size (chip
17	and granule), toasting (toasted and non-toasted), dosage (4 and 12 g/L) and maceration
18	time (1, 7, 21, 35 and 120 days) were tested. Results showed an important transference of
19	high-value enological compounds after 35 days of maceration with 12 g/L of toasted vine-
20	shoots. Vanillin or guaiacol were found in concentrations always higher than their wine
21	odor thresholds, depending on the variety and vine-shoot formats used. Moreover, trans-
22	resveratrol and ellagic acid stand out for their high concentrations. This transference rate
23	shows the enological aptitude of vine-shoots with a possibly positive contribution in real
24	wines, both at sensory and functional level, although future studies in real wines must be
25	carried out.
26	
27	KEYWORDS: antioxidant activity, model wine, phenolic compounds, tannin, vine-
28	shoots, volatiles.

#### 1. INTRODUCTION

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Among the enological additives, the fragments of oak wood known as chips, stand out 31 since they contribute to the wine with aromatic characteristics similar to wines aged in 32 oak barrels (Bautista-Ortín, Lencina, Cano-López, Pardo-Mínguez, López-Roca, & 33 Gómez-Plaza, 2008; Cerdán, Rodríguez Mozaz, & Ancín Azpilicueta, 2002). As a result 34 of such addition to wine, the transference of several oak compounds takes place, 35 especially volatile and phenolic compounds (Martínez-Gil, del Alamo-Sanza, Gutiérrez-36 Gamboa, Moreno-Simunovic, & Nevares, 2018) which give to the wine a special highly 37 appreciated woody aroma. In this process, the size of the oak chips, dosage, seasoning, 38 39 toasting degree and time of maceration, are the main factors that determine the type and 40 quantity of compounds released from wood to wine (Schumacher, Alañón, Castro-Vázquez, Pérez-Coello, & Díaz-Maroto, 2013). 41 Vine-shoots are the principal waste in viticulture, In Spain, Castilla-La Mancha region 42 the vineyard surface is around 443.818,31 ha, which accounts for 13% of European Union 43 (EU) and almost 6% of the world area (OIV, 2017). If it is assumed an average of 1.3 kg 44 of vine-shoots/vine as weight for an annual average production, approximately around 45 2.107 tons of vine-shoots are harvested per year in the world. However, the current 46 47 interest for the revaluation of this wastes and the search for innovative uses to consider them as resource rather than a waste has demonstrated that vine-shoots are an important 48 source of high value compounds with biostimulant (Sánchez-Gómez, Zalacain, Pardo, 49 Alonso, & Salinas, 2016, 2017) and bioplaguicide effects (Sánchez-Gómez, Sánchez-50 Vioque, Santana-Méridas, Martín-Bejerano, Alonso, Salinas, et al., 2017). Recently, 51 vine-shoots have been characterized and prepared to be used during winemaking in the 52 same way that oak chips are used (Cebrián-Tarancón, Sánchez-Gómez, Gómez-Alonso, 53

Hermosín-Gutierrez, Mena-Morales, García-Romero, et al., 2018; Cebrián-Tarancón, 54 55 Sánchez-Gómez, Salinas, Alonso, Oliva, & Zalacain, 2018; Delgado de la Torre, 2015). Oak chips can be consider as the residues from barrel manufacturing, but their quantity 56 is insignificant in comparison with vine-shoots viticultural waste generated, but highly 57 studied due to their appreciated wine contribution. Oak chips are richer in certain furanics 58 and benzenoids volatile compounds followed by whiskey lactones and terpenes, 59 especially significant is the contribution of vanillin within the benzenoids groups, a key 60 compounds for wine organoleptic characteristics, all of them also present in toasted vine-61 shoots (Cebrián-Tarancón, et al., 2018; Cebrián-Tarancón, Sánchez-Gómez, Salinas, 62 Alonso, Oliva, & Zalacain, 2018). It is worth mentioning that vine-shoots are a valuable 63 64 source of stilbenes, especially trans-resveratrol, which is not present in oak wood. Another important difference between both woods is on the tannin fraction, being 65 ellagitannins present in oak whereas vine-shoots only have oligomeric proanthocyanidins 66 (Cebrián-Tarancón, et al., 2018). 67 However, to propose vine-shoots as a new enological additive, it is important to carry 68 out a preliminary study about the transference of compounds with enological interest from 69 70 vine-shoots to a model wine, which is the main objective of this work. For it, the size of 71 vine-shoot fragments, toasting effect, dosage and the maceration time within a model wine were the variables tested, using vine-shoots from Airén and Cencibel Vitis vinifera. 72

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#### 2. MATERIALS AND METHODS

#### 2.1. Vine-shoot samples

Two vine-shoot cultivars were selected, Airén (*VIVC: 157*) and Cencibel (*VIVC: 12350*), the last one also known as Tempranillo, which represent the third and first cultivars in the world respectively. Two 15 years old vineyards of O.D. Mancha (Castilla-

79	La Mancha,	Spain)	were	selected	for	the	experiment.	Their	plant	density	was	2000
80	vines/ha and	their su	rface	of each vi	ineya	ırd v	was 3 ha.					

Pruning of vine-shoots was established 90 days after the last day of harvest in each vineyard, in case of Cencibel was the end of August and in case of Airén by the middle of September 2017. Fifty vines of each vineyard were selected randomly and 0.5 kg pruned, making a total of 25 kg of vine-shoots collected for each variety and vineyard. The higher accumulation of volatile and non-volatile compounds in vine-shoots takes place at six months of post-pruning storage (Cebrián, Sánchez-Gómez, Salinas, Alonso, & Zalacain, 2017). For this reason, vine-shoots were stored intact in the dark and at a room temperature (18 ± 3 °C) with a relative humidity ranging between 60-66%, until their processing.

As the minimum size oak chips for winemaking is set up two millimeters (ECC, 2006), two different size according to their size were tested: similar to oak chips (called *chips*) with size of 2.5 cm - 3.5 cm long and a second one between 2 mm and 2 cm long (called *granules* onwards).

After this time, half of these vine-shoots were subjected to a toasting process in an air circulation oven (Heraeus T6, Hanau Deutschland) at 180°C and 45 min according to (Cebrián-Tarancón, Sánchez-Gómez, Salinas, Alonso, Oliva, & Zalacain, 2018). Final moisture was resulting 2.84%. Moreover, as grapevines are commonly treated with different fungicides, a study on the evolution of these residues in toasted vine-shoots was also carried out. The results confirm the disappearance of these fungicides when vine-shoots are submitted to a toasting process (data not shown).

#### 2.2. Model wine and treatments

The model wine was composed by an ethanol/water solution (12.5/87.5; v/v) and 5g/L of tartaric acid, adjusted to pH 3.5 with NaOH 1M (García-Estévez, Alcalde-Eon, Martínez-Gil, Rivas-Gonzalo, Escribano-Bailón, Nevares, et al., 2017). For each variety, Airén and Cencibel vine-shoots, were tested: both sizes (*chips* and *granule*), toasted and non-toasted, two different dosages (4 g/L and 12 g/L) and five maceration times (1, 7, 21, 35 and 120 days) starting the first day after vine-shoot samples were added in model wine.

Vine-shoot samples were introduced in infusion bags, in similar way as the oak chips are introduced in real winemaking, and keeping the bags submerged in the model solution with the help of inert glass balls. The maceration step was carried out in one litter amber glass bottles, standing, at dark and under temperature control conditions (16 ± 1°C). The studied conditions were choose taken in account the traditional use of oak chips in model wines (Bautista-Ortín, Lencina, Cano-López, Pardo-Mínguez, López-Roca, & Gómez-Plaza, 2008; Schumacher, Alañón, Castro-Vázquez, Pérez-Coello, & Díaz-Maroto, 2013). A total of 40 different wine model solutions of each variety were prepared in duplicate.

#### 2.3. Chemical composition analysis

*2.3.1. Volatile compounds* 

Volatiles were determined according to (Sánchez-Gómez, Zalacain, Alonso, & Salinas, 2014). Their extraction was carried out by means of Stir Bar Sorptive Extraction (SBSE) with a polydimethylsiloxane twister bar (10 mm length, 0.5 mm film thickness). Samples were stirred at 500 rpm during 60 minutes later analysis was performed using an automated thermal desorption unit (TDU, Gerstel, Mülheim and der Ruhr, Germany) mounted on an Agilent 7890A gas chromatograph system (GC) coupled to a quadrupole Agilent 5975C electron ionization mass spectrometric detector (MS, Agilent

Technologies, Palo Alto, CA, USA), equipped with a fused silica capillary column (BP21 129 130 stationary phase, 30-m length, 0.25 mm I.D. and 0.25 µm film thickness) (SGE, Ringwood, Australia). The carrier gas was helium with a constant column pressure of 131 20.75 psi. 132 The stir bars were thermally desorbed in a stream of helium carrier gas at a flow rate 133 of 75 mL/min with the TDU programmed from 40 to 295 °C (held 5 min) at a rate of 60 134 135 °C/min, splitless desorption mode. The analytes were focused in a programmed temperature vaporizing injector (PTV) (CIS-4, Gerstel), containing a packed liner (20 mg 136 tenax TA), held at - 40 °C with cryo cooling prior to injection. After desorption and 137 138 focusing, the CIS-4 was programmed from -40 to 260 °C (held for 5 min) at 12 °C/min to transfer the trapped volatiles onto the analytical column. The GC oven temperature was 139 programmed to 40 °C (held for 2 min), raised to 80 °C (5 °C/min, held for 2 min), raised 140 141 to 130 °C (10 °C/min, held for 5 min), raised to 150 °C (5 °C/min, held for 5 min), and then raised to 230 °C (10 °C/min, held for 5 min). The MS was operated in scan 142 acquisition (27-300 u) with an ionization energy of 70 eV. The temperature of the MS 143 transfer line was maintained at 230 °C. 144 145 MS data acquisition was carried out at positive scan mode, although to avoid matrix 146 interferences, the MS quantification was performed in the SIM mode using their characteristic m/z values. Compounds identification was performed using the NIST 147 library and confirmed by comparison with the mass spectra and retention time of their 148 149 pure standards. The standards used to identify and quantify volatile (GC-MS) were purchased in Sigma-Aldrich (Steinheim, Germany), the numbers in brackets indicates the 150 m/z used for quantification: benzaldehyde (m/z = 106), citronellol (m/z = 69), citronellal 151 (m/z = 69), ethyl hexanoate (m/z = 101), ethyl octanoate (m/z = 101), 4-ethylphenol (m/z = 101)152

= 107), eugenol (m/z = 164), farnesol (m/z = 69), furfural (m/z = 96), geranyl acetone (m/z = 107)

= 43), guaiacol (m/z = 109), 1-hexanol (m/z = 56), hexyl acetate (m/z = 43), 5-hydroxymethylfurfural (m/z = 97),  $\beta$ -ionone (m/z = 177),  $\beta$ -ionol (m/z = 205), linalool (m/z = 71), 5-methylfurfural (m/z = 110), nonanal (m/z = 57), 1-octen-3-ol (m/z = 57), syringol (m/z = 154), vanillin (m/z = 151), 4-vinylguaiacol (m/z = 151) and trans/cis-whiskey lactones (m/z = 99). As internal standard 3-methyl-1-pentanol (220  $\mu$ l/25 mL model wine) was used. Quantification was based on calibration curves of the respective standards in the model wine at five different concentrations (2  $\mu$ g/L-15 mg/L) (R<sup>2</sup> = 0.95-0.99). All analyzes were made in triplicate. 

### 2.3.2. Low molecular weight phenolic compounds (LMWPC)

The samples were filtered through a PVDF Durapore filter of 0.22 μm (Millipore, Bedford, MA, USA) and 20 μL were injected into an Agilent 1200 HPLC chromatograph (Palo Alto, California, USA) equipped with a Diode Array Detector (Agilent G1315D) and atmospheric pressure chemical ionization (MM-ESI/ APCI-MS) system, coupled to an Agilent ChemStation (version B.03.01) data-processing station. Separation was performed on a reverse phase ACE Excel 3 C18-PFP (4.6 mm × 150 mm, 3 μm particle size) and a precolumn ACE Excel HPLC Pre-colum Filter 1PK (0.5 μm particle size) at 30 °C. The HPLC proportion of solvents used was water/formic acid/acetonitrile (97.5:1.5:1 v/v/v) as solvent A and acetonitrile/formic acid/solvent A (78.5:1.5:20 v/v/v) as solvent B. The elution gradient was set up for solvent B as: 0 min, 5%; 8.40 min, 5%; 12.50 min, 10%; 19 min, 15%; 29 min, 16%; 30 min, 16.5%; 34.80 min, 18%; 37.20 min, 32%; 42 min, 62%; 52 min, 90%; 54 min, 100%; 56 min, 100%; 60 min, 5%; 65 min, 5%.

All compounds detection was carried out by means of DAD detector by comparison

with the corresponding UV-vis spectra and retention time of their pure standards (Sigma-

Aldrich, Steinheim, Germany). Although other LMWPC were looked for, the compounds quantified and identified at different wavelength were: (+)-catechin, (-)-epicatechin, gallic acid, protocatechuic acid at 280 nm; ellagic at 256 nm; conyferaldehyde and sinapaldehyde at 324 nm; *trans*-piceid (*t*-resveratrol-3-glucoside) and *trans*-resveratrol at 308 nm. Quantification was based on calibration curves of the respective pure standards in the model wine at five different concentrations achieved by UV-vis signal ( $0.40 - 260 \, \text{mg/L}$ ) ( $R^2 = 0.96-1.00$ ). All analyzes were made in triplicate.

#### 2.3.3. Tannin fraction

The tannin level was measured in samples according to the methylcellulose precipitation method by (Sarneckis, Dambergs, Jones, Mercurio, Herderich, & Smith, 2006). Tannin precipitation was performed using a 0.04% methylcellulose solution (w/v, in deionized water). The tannin concentration was determined in triplicate by the difference in absorption at 280 nm between the sample without the addition of methylcellulose and the methylcellulose precipitated. The tannin content was quantified as (-)-epicatechin.

## 2.3.4. Antioxidant capacity

The scavenging activity against the 1, 1-diphenyl-2-picrylhydrazyl (DPPH) free radical was determined according to (Braca, De Tommasi, Di Bari, Pizza, Politi, & Morelli, 2001). Samples (0.5 mL) were mixed with 2.5 mL of DPPH in methanol (0.04 mg/mL), incubated 1 hour at room temperature and the absorbance measured at 517 nm in a Perkin-Elmer Lambda 20 UV-visible Spectrophotometer. Distilled water and (+)-catechin standard were used as negative and positive control of the assay, respectively.

The reduction of DPPH was tested in triplicate for five randomly different volumes: 10, 20, 40, 80 and 100 µl of the wine model solutions.

#### 2.4. Statistical analysis

To guarantee quality of data, firstly a complete descriptive analysis was carried out. Special attention was paid to detect potential outliers. For that, firstly each group was analyzed of the three repetitions comparing their variability. Subsequently, numeric and graphic (box-whiskers) descriptive analysis were carried out. Finally, a detailed residual analysis was done in each one of the analysis of variance. Due to the experimental plan, the statistical analysis was done with the average values of each one of the experimental conditions. For each compound, the Tukey test was used to analyze significant differences among the selected treatments ( $\alpha < 0.05$ ). To achieve a general view and compare the effect of each treatment, afactor analysis of correlations was used. For that, variables selected were the more important aroma compounds related with the classical sensory profile of wood aged wines. As a method of extraction, the principal component method was applied with varimax rotation.

#### 3. RESULTS AND DISCUSSION

#### 3.1. Vine-shoot conditions selection

It is well known that the compounds extracted when wine is in contact with oak wood are of great importance for wine quality, since the aroma is markedly modified (Cerdán, Rodríguez Mozaz, & Ancín Azpilicueta, 2002; García-Estévez, et al., 2017). Oak wood is always submitted to a toasting process and volatile compounds such as vanillin, guaiacol or whiskey lactones are significantly generated. These volatiles have also been identified in toasted and non-toasted vine-shoots (Cebrián, Sánchez-Gómez, Salinas,

Alonso, & Zalacain, 2017; Cebrián-Tarancón, Sánchez-Gómez, Salinas, Alonso, Oliva, & Zalacain, 2018; Delgado De La Torre, Priego-Capote, & Luque De Castro, 2015; Sánchez-Gómez, Zalacain, Alonso, & Salinas, 2014, 2016a), contributing with a vanilla aroma, smoked and toasted notes or a characteristic coconut aroma respectively (Schranz, Lorber, Klos, Kerschbaumer, & Buettner, 2017). But there are other aroma which are not in common in both woods, as for example linalool, which has been only quantified in vine-shoots but not in oak chips (Cebrián-Tarancón, Sánchez-Gómez, Salinas, Alonso, Oliva, & Zalacain, 2018). It is a well-known vine compound, which can be accumulate in grapes and to be extracted along winemaking, contributing to the varietal aroma of wine and its typicality.

In order to determine the best transference conditions of such compounds (vanillin, guaiacol, *cis* and *trans* whiskey lactones and linalool) from vine-shoots to the model wine, a factor analysis was performed (Table 1) considering their size, dosage, toasting and maceration time in contact with the model wine.

The factorial analysis for both vine-shoot varieties (Airén and Cencibel) provided in both cases, two factors with a similar structure (Table 1). The first factor gathers the information of the variables *trans*-whiskey lactone, *cis*-whiskey lactone and linalool, while the second was generated by guaiacol and vanillin. Figures 1 and 2 shows the scores of Airén and Cencibel experiments carried out in the factors obtained, allowing a global view and where it is possible to compare the effects of the sizes, dosages, toasting and maceration time within the model wines. It is noteworthy that for the two varieties, the two underlying factors obtained were the same; however, the response to the treatments (dosages, toasting and maceration time) in each variety was different as it can be observed in Figures 1 and 2. All variables had a positive weight, which means that the positioning of the model wines on the right side of the first factor implies simultaneously greater

values of trans-whiskey lactone, cis-whiskey lactone and linalool. In the second factor, the positioning in the upper part implies greater values of guaiacol and vanillin. In this way, the treatments with a higher extractability capacity were the ones positioned in the upper right of the two graphs: in case of Airén variety (Figure 1), the treatment with *chips* was in the right part of factor 1, which means that it generally presents greater values of trans-whiskey lactone, cis-whiskey lactone and linalool. In the axis, it is observed that model wines placed in the upper and right part and therefore with greater values of vanillin and guaiacol, correspond to different combinations of dosages, toasting and maceration time, suggesting that interactions of higher order are produced between the tested experimental conditions. So, according to these results, the best treatments for this variety would be add 12 g/L of toasted *chips* and macerate during 21 and 35 days (model wines named as C-T-12-21 and C-T-12-35) (Figure 1). In case of Cencibel variety (Figure 2), a very different behaviour was observed between chips and granule formats, but with an opposite response to that of Airén. Here, the model wines in contact with granules have a greater value of trans-whiskey lactone, cis-whiskey lactone and linalool, due to the position of the scores in the factor 1. In factor 2, again, it was observed that the best treatments to achieve greater values of vanillin and guaiacol, correspond to the different combinations of treatments. The selection of the best conditions for the vine-shoots was carried out by analysing the positioning of the model wines on the factorial plans. In addition, for Cencibel variety, in contrast by Airén variety, it has been observed that the two factors present certain correlation. Looking at loading values (Table 1), it is concluded that the best treatments were the five located in the upper right, all of them with a dosage of 12 g/L and granule size. It is due that they were located on the main direction marked by almost all the variables and that was located on the quadrant bisector. Although the most efficiency treatments for Cencibel variety were 35 and 120 days of

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maceration time (Figure 2), the last one was not significantly different, so 35 days was select for Cencibel vine-shoots (model wine G-T-12-35). In addition, such results suggest that after 35 days of maceration, the kinetics of extraction of the studied compounds remains constant. Moreover, the maceration treatment for 21 days (model wine G-T-12-21) also will be studied in detail, since it was the second most important treatment and for to compare with the Airén vine-shoots.

# 3.2. Detailed composition of the selected model wine solutions

Table 2 shows a detailed analysis of the transferred compounds from the two varieties of vine-shoots, when 12 g/L of toasted pieces (*chips* and *granules*) were macerated during 21 and 35 days in the model wine. In case of pH (Table 2), it was observed lower values when higher maceration times were used, independently of the variety and vine-shoot formats, in relation to the control model wine. In case of Airén model wines macerated 21 days, an increment of 0.7 units were observed and about 0.48 units in case of 35 maceration days. In case of Cencibel model wines macerated 21 days, an increment of 0.73 units were observed, and about 0.55 units in case of 35 maceration days. Such pH increment should be taking into account when vine-shoots will be added to real wines. Compounds were grouped according to their chemical structure as volatiles and low molecular weight phenolic compounds.

The total volatile composition of the resulting model wines was clearly different for both varieties according to the formats and maceration time used. When Airén *chips* were used, in the case of the lowest maceration time (21 days), it was observed a significant higher level of compounds than in the rest of treatments. By contrast, the transfer of compounds from Cencibel *chips* to model wine was higher when they were macerated for

a longer period of time (35 days) and such contribution showed a similar content than in case of the *granule* format.

Furanics were the most abundant volatile compounds in wines, reaching value ranges of  $2891.68 - 5516.08 \,\mu\text{g/L}$  for Airén and  $2825.02 - 3625.62 \,\mu\text{g/L}$  for Cencibel ones. Such compounds result from the hemicellulose degradation during the toasting process (Chatonnet, Boidron, & Pons, 1989) and can contribute to wine aroma with caramel, toasted and honey like notes. The high presence of these compounds was expected by their high affinity to ethanol/water (model wine) (Alañón, Alarcón, Marchante, Díaz-Maroto, & Pérez-Coello, 2017). However, its content was lower than their corresponding wine odor thresholds (Table 2). It is accepted that compounds whose Odor Activity Value (ratio between the concentration of a compounds and its odor threshold, OAV) is  $\geq 1$  contribute individually to the matrix aroma (Guth, 1997), so such furanics do not participate in the perception of the aroma of these wine model solutions (Table 2).

Volatile phenols were the second most abundant group, where the most important compounds, from the point of view of its contribution to the wine aroma, were vanillin and guaiacol. Both compounds come up from lignin degradation while wood toasting takes place (Cebrián-Tarancón, Sánchez-Gómez, Salinas, Alonso, Oliva, & Zalacain, 2018). In case of vanillin, similar trend was found for both varieties, the highest concentrations were observed when *granule* format was used. The most significant increase with maceration time was observed in Cencibel wines from *granule* format, where vanillin concentration reached 227.40 µg/L and 261.25 µg/L after 21 and 35 days, respectively. The OAV of vanillin in all wine model solutions was > 1 (Table 2). However, while in Airén model wines from *granule* format the OAV values were 2.73 and 2.84 for 21 and 35 maceration days respectively, in Cencibel model wines with similar format, the OAV values were 3.79 and 4.35, respectively. These results suggest

that *granule* toasted vine-shoots from both varieties, could increase wine vanillin content with an important vanilla aromatic note. Such possibility also was presented when *chip* format was used, but in a less extension. Moreover, vanillin concentration in all model wines studied were comparable to the levels reported for others works when oak barrels and fragments were tested (Navarro, Kontoudakis, Gómez-Alonso, García-Romero, Canals, Hermosín-Gutíerrez, et al., 2018). According with previous works, the size of wood fragment may have a relevant effect on the formation of vanillin during the toasting of the pieces of vine-shoots (Campbell, Sykes, Sefton, & Pollnitz, 2005).

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Guaiacol showed different behavior regardless of the vine-shoots variety, but not significant differences were observed between the concentrations achieved by the best treatments. For model wines in contact with Cencibel vine-shoots, the higher concentration of guaiacol was found when granule size was used, 15.21 µg/L and 16.30 µg/L for 21 and 35 days of maceration, respectively. In contrast, *chip* size contributed with a greater quantity of guaiacol to the wine, 17.98 μg/L and 18.25 μg/L, in case of Airén variety (Table 2) for 21 and 35 days of maceration. Very low concentrations of this compound when oak chips were used in other previous works (Bautista-Ortín, Lencina, Cano-López, Pardo-Mínguez, López-Roca, & Gómez-Plaza, 2008). Moreover, in the same way that vanillin, the guaiacol OAV values were  $\geq 1$ , so vine-shoots used in wines in the conditions of this study, could be contribute with its "smoked" or "toasted" notes to the wine aroma (Schranz, Lorber, Klos, Kerschbaumer, & Buettner, 2017), as oak barrels and chips (Chira & Teissedre, 2013). With similar aroma descriptors, syringol and 4-vinylguaiacol also were detected in model wine solutions, but their OAV values were < 1. However, synergies effects may exist between volatile compounds, especially in those with similar aromatic nuances (Ferreira & Cacho, 2009). Eugenol, a volatile phenol with a characteristic clove aroma (Ferreira, López, & Cacho, 2000), had a different

behavior according to the vine-shoot varieties and formats, but in all cases the concentration was lower than its odor threshold. The highest and lowest content were in model wines from Cencibel and Airén *granule* sizes, respectively. Very low concentration of 4-ethylphenol was detected in the different model wines, ranging between of 2.43 -  $4.01~\mu g/L$ , fall far below its odor threshold, so its leather and horse note should not be detected (Bautista-Ortín, Lencina, Cano-López, Pardo-Mínguez, López-Roca, & Gómez-Plaza, 2008).

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Terpenoids, a group of compounds that are associated with the grape variety, with roses, citrusy or lilac notes, also showed the highest extraction when the granule format was used. No significant differences were found between both times of maceration in Airén model wines for total terpenoids, but in case of Cencibel, the highest concentration, 127.46 µg/L according to Table 2, was reached after 35 days of maceration (Table 2). It is important to highlight the presence of  $\beta$ -ionol, which was not described previously in toasted vine-shoots (Cebrián-Tarancón, Sánchez-Gómez, Salinas, Alonso, Oliva, & Zalacain, 2018), which showed up five times content when granule format was used. This suggest that a minor size of particle favors its extraction. Compounds such as citronellal, linalool, farnesol, among others terpenoids, were also detected in all model wines, being linalool the most abundant when the granule format of Cencibel was used. At individual level, the OAV values were < 1 for all of them, although the information about the odor threshold for β-ionol was not found in the literature (Table 2). However, the synergist effect that may exist between them, could enhance the wine varietal aroma if vine-shoots were used as enological additives. On the other hand, if the  $\beta$ -ionone precursor, β-ionol, is considered as a powerful odorant, these results could suggest an important contribution of its positive floral aroma to the wines, since the contents found were much higher than the odor threshold of the  $\beta$ -ionone.  $C_6$  compounds, also known for their herbaceous aroma contribution to wine aroma, are the products of the enzymatic breakdown of unsaturated fatty acids (Aubert & Chalot, 2018). In this case, 1-hexanol was the only one representing this chemical family and showed a different behavior depending of the vine-shoot varieties used. Model wine in contact with Airén vine-shoots showed the highest concentration after 21 days of maceration, independently of the size. However, wines in contact with Cencibel vine-shoots showed the highest concentration when the *chip* format was used, independently of the maceration time.

The influence of vine-shoots in the total esters and "others" group compounds content (Table 2) was not relevant.

At sensory level, the lactones (*cis* and *trans*- whiskey lactone isomers) found in the model wines were not relevant, since their content were much lower than their corresponding odor thresholds (Table 2). Nevertheless, it is worth discussing them, because until recently, they were considered exclusive as oak compounds, but according to (Sánchez-Gómez, Zalacain, Alonso, & Salinas, 2016b), they were also present in toasted vine-shoots. Such compounds may contribute with a woody coconut aroma, which is positively correlated with the *cis*-isomer (Chatonnet, Boidron, & Pons, 1990) or an herbaceous and also woody aroma (Bautista-Ortín, Lencina, Cano-López, Pardo-Mínguez, López-Roca, & Gómez-Plaza, 2008), which is attributed to the *trans*-isomer. In our work, both isomers were quantified when the *chip* format was used, being necessary a long time of maceration in case of Cencibel vine-shoots. Higher concentrations were found in the case of Cencibel cultivar, 8.89 μg/L, when the *granule* format was tested.

Summarising, in an attempt to extrapolate the actual results to real wines and considering the volatiles with OAV >1, we suggest that: 1) If wines with vanilla notes are desired, then vine-shoots with a *granule* format must be used independently of the variety,

macerating them during 35 days; 2) If wines with toasted notes (guaiacol) are desired, Airén vine-shoots must be used in *chips* format, macerating during 21 days; 3) If wines with both aromas (vanilla and toasted) are desired, Cencibel vine-shoot must be used in *granule* formats, macerating during 35 days.

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The low molecular weight phenolic compounds identified in model wines from the selected treatments were also shown in Table 2, grouped according to their chemical phenolic families such as flavanols, phenolic acids, stilbenes and aldehydes. All quantified compounds were also found in non-toasted vine-shoots (Cebrián, Sánchez-Gómez, Salinas, Alonso, & Zalacain, 2017) although their content decreased with toasting, except for trans-resveratrol compound which increased (Cebrián-Tarancón, Sánchez-Gómez, Salinas, Alonso, Oliva, & Zalacain, 2018). (+)-Catechin and (-)epicatechin were the flavanols detected and only quantified when granule format was used for both varieties, reaching concentrations of 18.24 mg/L and 7.34 mg/L in Airén and Cencibel, respectively. These contents were in the same range than in real wines which have not been in contact with any wood (Gordillo, Baca-Bocanegra, Rodriguez-Pulído, González-Miret, García Estévez, Quijada-Morín, et al., 2016; Sánchez-Gómez, Zalacain, Pardo, Alonso, & Salinas, 2017). Ellagic acid was the most abundant phenolic compound transferred from vine-shoots to model wines, whereas gallic and protocatechuic acids were also transferred but in very small quantities. Such acids were previously detected in vine-shoots, toasted and non-toasted, being ellagic acid the most significant (Cebrián-Tarancón, Sánchez-Gómez, Salinas, Alonso, Oliva, & Zalacain, 2018). The mean content of ellagic acid in all model wines was approximately 33 mg/L and not relevant differences were observed among format, varieties and maceration time. The content of ellagic acid in real wines was lower than in the studied model wines. However, it has recently been shown that the addition of ellagic acid in pre-fermentative stage of winemaking, at concentrations of 300 mg/L, markedly improves the color of red wines, acting as an important cofactor of anthocyanins, but also stabilizing the phenolic profile (Zhang, He, Zhang, Reeves, Liu, Zhao, et al., 2018). This result suggests the interest of using vine-shoots as enological additives, both before and after the alcoholic fermentation of wines.

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In the stilbenes group, trans-resveratrol was found in all model wines, while its glucoside, trans-piceid, was only detected when granule formats and 35 days of maceration were used. Both stilbenes were present in high concentration in vine-shoots, non-toasted and toasted, but the toasting process increase them significantly, especially trans-resveratrol (Cebrián-Tarancón, Sánchez-Gómez, Salinas, Alonso, Oliva, & Zalacain, 2018). The concentration of *trans*-resveratrol in the model wines reached levels of 4.05 mg/L and 2.64 mg/L for Airén and Cencibel varieties respectively, when the granule format was used during 35 days of maceration (Table 2). The presence of transresveratrol in white wines according to the bibliography may range between 0.80 to 4.61 mg/L (Fabjanowicz, Płotka-Wasylka, & Namieśnik, 2018) depending if there is any contact with skins, seeds and grape juice during winemaking. This stilbene is considered the major functional compound in red wine (Sumby, Grbin, & Jiranek, 2010). Due to the fact that there was higher trans-resveratrol content in model wines in contact with vineshoots than in real wines, vine-shoots could be used as an enological additive to increase trans-resveratrol and therefore, the healthy benefits of wines, being Airén granule size in contact during 35 days, the one that may have the highest contribution in real wines. This possibility does not present when oak chips are used, since this stilbene does not form part of this wood.

Phenolic aldehydes such as coniferaldehyde and sinapaldehyde were only quantified in model wines from Airén, when *chips* were macerated during 21 days, and in model

wines from Cencibel when the *granule* format was macerated during 35 days, the highest concentrations being in this last case (Table 2).

In the pool of phenolic compounds, the tannin content was also determined. Model wines with Cencibel, independently of the used format, had always higher tannin content than when Airén was used. In case of Cencibel, *chips* contributed with 26.96 mg/L of (-)-epicatechin while with granule the content was 22.53 mg/L of (-)-epicatechin. In case of Airén chips, the tannin content within the model wines was 11.92 mg/L of (-)-epicatechin versus the 18.55 mg/L of (-)-epicatechin for granule format. Then depending on the variety, the contribution of tannins to the model wine solutions in terms of formats was different.

The tannin fraction of vine-shoots is constituted only by proanthocyanins, of which procyanidins are the majority but it is noteworthy the high amount of prodelphinidins content in relation with other grape derivates. The mean value of the tannin fraction in white wines is lower than in red wines, in the latest between 0.5 and 3 g/L (Sarneckis, Dambergs, Jones, Mercurio, Herderich, & Smith, 2006). Therefore, if vine-shoots would be used as enological additives, their contribution to the tannin sensorial properties of wines, it does not seem to be so relevant.

#### 3.3. Antioxidant capacity

Previously the antioxidant capacity of wines, seeds, grape pomaces or skins has been evaluated. Such capacity is attributed to certain wine components, since they can prevent reactions of oxidative degradation, and therefore they have a positive effect on health (Bartolomé, Nuñez, Monagas, & Gómez-Cordovés, 2004).

The antioxidant capacity of toasted vine-shoots as a pool has not been analyzed before, although their individual chemical characterization (Cebrián-Tarancón, Sánchez-

Gómez, Salinas, Alonso, Oliva, & Zalacain, 2018; Sánchez-Gómez, Zalacain, Alonso, & Salinas, 2016a) suggest their antioxidant potential. In Figure 3 it is shown the percentage of DPPH radical reduction of model wines solutions macerated with toasted Airén and Cencibel vine-shoots as *chip* and *granule* formats during 35 days, 100 µl of these were used. Independently of the variety used, the highest reduction was observed for the *granule* size, reaching a 50% and 37% of DPPH radical inhibition for Airén and Cencibel respectively. These results suggest that the vine-shoot formats had a direct influence in the antioxidant capacity, which increase when a low particle size was used and also, they were in agreement with the higher content of phenolic compounds in these model wine solutions, especially stilbenes and flavanols.

#### 4. CONCLUSIONS

The transference of chemical compounds with an enological interest from vine-shoots to a model wine was studied for first time, in order to propose them as new enological additives for wines. Such contribution depends on the vine-shoot variety and format used. In all cases significant quantities of vanillin and guaiacol, above their odor thresholds, were transferred from vine-shoots. *trans*-Resveratrol, the highest functional compound in wines, was found in all samples in larger quantities than those found in most real wines. Moreover, the ellagic acid content found make possible its participation on wine copigmentation and color stabilization phenomena. Taking into account all compounds and vine-shoot formats and varieties, the ones with better overall contribution was Cencibel granule format macerated at 35 days.

Such preliminary results suggest that vine-shoots can contribute positively to modify the aroma, color and healthy effect of wines, by adding to the wine their vine own compounds. Future works should be addressed in real wines.

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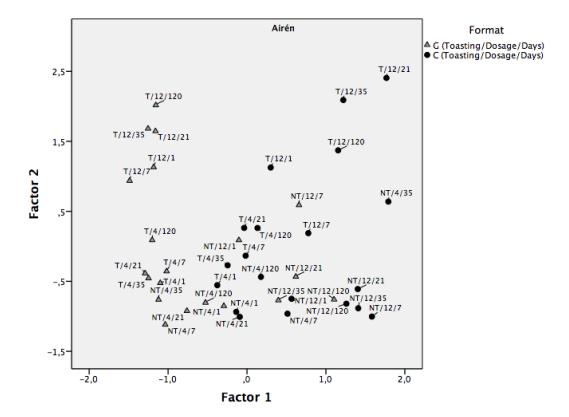
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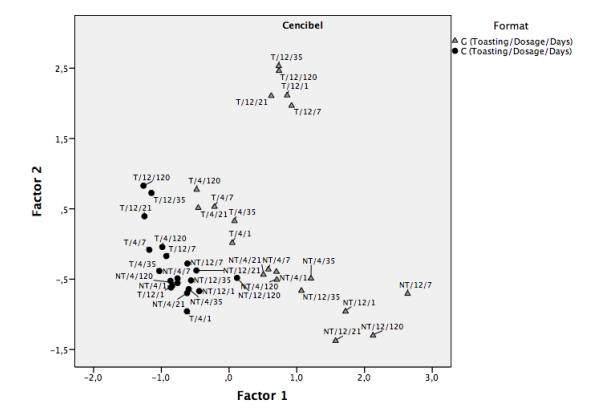
#### FIGURE CAPTIONS

- Figure 1. Scores of the forty model wines in contact with Airén vine-shoot fragments
- obtained after the application of the Factor Analysis. Format (C: chips, G: granules);
- toasted (T: toasted, NT: non-toasted); dosages (4: 4g/L, 12: 12g/L); time of sampling (1,
- 712 7, 21, 35 and 120 days).

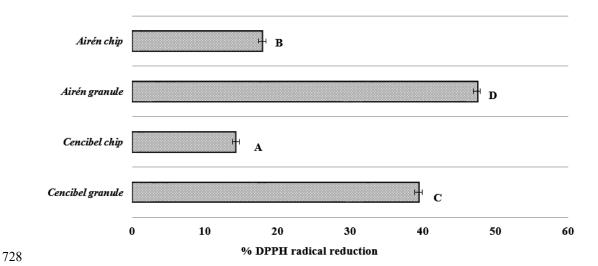




**Figure 2**. Scores of the forty model wines in contact with Cencibel vine-shoot fragments obtained after the application of the Factor Analysis. Format (C: *chips*, G: *granules*); toasted (T: toasted, NT: non-toasted); dosages (4: 4g/L, 12: 12g/L); time of sampling (1, 7, 21, 35 and 120 days).



**Figure 3.** Percentages of DPPH radical inhibition (antioxidant capacity of model wines in contact with 12 g/L of toasted Airén and Cencibel vine-shoots (*chip* and *granule* formats) during 35 days (n=3). 100 μl of model wines.



# **TABLES**

# **Table 1**. Factor analysis of the model wine solutions in all of the tested conditions.

	Airén (n=40) Cencibel (n=40)			
	Factor 1	Factor 2	Factor 1	Factor 2
% Of explained variance	51.9 %	31.9 %	81.9 %	11.4 %
trans-Whiskey lactone	0.914	0.000	0.767	0.616
cis-Whiskey lactone	0.896	-0.131	0.708	0.646
Linalool	0.891	0.097	0.950	0.218
Guaiacol	0.262	0.893	0.545	0.771
Vainillin	-0.300	0.879	0.220	0.944

Table 2. Volatile and phenolic content (μg/L and mg/L, respectively) of model wines in contact with 12 g/L of toasted Airén and Cencibel vine-shoots (*chips* and *granules* formats) during 35 and 21 maceration days.

			AIRÉ	EN .		CENCIBEL				
	<b>dor threshold</b> (µg/L or mg/L)	Chi	ps	Gran	nules	Ci	hips	Granules		
Volatile Compounds	µg/L 01 mg/L)	21 days, 12 g/L	35 days, 12 g/L	21 days, 12 g/L	35 days, 12 g/L	21 days, 12 g/L	35 days, 12 g/L	21 days, 12 g/L	35 days, 12 g/L	
C6 compounds (µg/L)										
1-Hexanol	8000°	$28.17 \pm 0.01 \text{ b}, \epsilon$ (< 0.02)	$15.79 \pm 0.39 \text{ a}, \alpha\beta$ (< 0.02)	$26.12 \pm 4.72 \text{ b}, \delta \epsilon$ (< 0.02)	$21.92 \pm 1.05 \text{ ab}, \gamma \delta$ (< 0.02)	$19.65 \pm 1.25 \text{ B,}\beta\gamma$ (< 0.02)	$19.27 \pm 1.09 \text{ B}, \beta \gamma$ (< 0.02)	$13.46 \pm 1.64 \text{ A}, \alpha$ (< 0.02)	$14.37 \pm 1.80 \text{ A}, \alpha\beta$ (< 0.02)	
	Total	$28.17 \pm 0.01$ b, $\varepsilon$	$15.79\pm0.39~a,\alpha\beta$	$26.12 \pm 4.72$ b, $\delta\epsilon$	$21.92 \pm 1.05$ ab, $\gamma \delta$	$19.65 \pm 1.25 \text{ B,} \beta \gamma$	$19.27\pm1.09~B,\beta\gamma$	$13.46 \pm 1.64 \text{ A}, \alpha$	$14.37 \pm 1.80 \text{ A}, \alpha\beta$	
Esters (µg/L)										
Hexyl acetate	1500 <sup>f</sup>	$0.19 \pm 0.01 \text{ b, } \beta$ (< 0.02)	$0.10 \pm 0.00 \text{ a}, \alpha$ (< 0.02)	$0.10 \pm 0.02 \text{ a, } \alpha$ (< 0.02)	$0.07 \pm 0.01 \text{ a, } \alpha$ (< 0.02)	$0.08 \pm 0.01 \text{ A}, \alpha$ (< 0.02)	$0.09 \pm 0.02 \text{ A}, \alpha$ (< 0.02)	$0.10 \pm 0.01 \text{ A}, \alpha$ (< 0.02)	$0.10 \pm 0.01 \text{ A}, \alpha$ (< 0.02)	
Ethyl hexanoate	14°	$0.65 \pm 0.01$ b, $\alpha\beta\gamma$ $(0.05)$	$\begin{array}{c} 0.61 \pm 0.04 \text{ ab, } \alpha\beta \\ (0.04) \end{array}$	$0.50 \pm 0.06 \text{ a}, \alpha$ (0.04)	$0.83 \pm 0.05 \text{ c}, \beta \gamma$ (0.06)	$0.64 \pm 0.11 \text{ B,aby}$ $(0.05)$	$0.92 \pm 0.07 \text{ D, } \delta$ (0.07)	$0.41 \pm 0.03 \text{ A}, \alpha$ (0.03)	$0.65 \pm 0.09 \text{ C},  \alpha\beta\gamma$ (0.05)	
Ethyl octanoate	5 °	$0.10 \pm 0.01 \text{ a, } \alpha\beta$ (0.02)	$0.08 \pm 0.00 \text{ a, } \alpha\beta$ (0.02)	$0.12 \pm 0.05 \text{ a, } \alpha\beta$ (0.02)	$0.14 \pm 0.01 \text{ a, } \alpha\beta$ (0.03)	$0.07 \pm 0.01 \text{ A}, \alpha\beta$ (< 0.02)	$0.08 \pm 0.00 \text{ A}, \alpha\beta$ 0.02	$0.04 \pm 0.00 \text{ A}, \alpha$ (< 0.02)	$0.27 \pm 0.18 \text{ A, } \beta$ (0.05)	
	Total	$0.94 \pm 0.03$ b, $\beta \gamma \delta$	$0.79 \pm 0.04$ a, $\alpha\beta\gamma$	$0.71 \pm 0.09$ a, $\alpha\beta$	$1.04 \pm 0.04$ b, $\gamma \delta$	$0.79 \pm 0.12 \text{ B}, \alpha\beta$	$1.09 \pm 0.07 \text{ D}, \delta$	$0.55 \pm 0.04 \text{ A}, \alpha$	$1.02 \pm 0.18 \text{ C}, \gamma \delta$	
Furanics (mg/L)										
Furfural	14°	$2.63 \pm 0.00 \text{ c}, \gamma$ (0.26)	$2.20 \pm 0.013 \text{ b}, \beta$ (0.16)	$1.72 \pm 0.42 \text{ ab}, \alpha$ (0.12)	$1.31 \pm 0.045 \text{ a, } \alpha$ (0.09)	$1.30 \pm 0.083 \text{ A}, \alpha$ (0.09)	$1.52 \pm 0.02 \text{ A}, \alpha$ (0.11)	$1.55 \pm 0.19 \text{ A}, \alpha$ (0.11)	$1.39 \pm 0.05 \text{ A}, \alpha$ (0.10)	
5-Methylfurfural	$20^{b}$	$0.38 \pm 0.00 \text{ c, } \epsilon$ $(0.02)$	$0.26 \pm 0.006 \text{ b}, \delta$ (< 0.02)	$0.15 \pm 0.04 \text{ a}, \alpha\beta$ (< 0.02)	$0.116 \pm 8.99 \text{ a}, \alpha$ (< 0.02)	$0.16 \pm 0.008 \text{ A}, \alpha\beta\gamma$ (< 0.02)	0.18 ±0.003 AB, βγ (< 0.02)	$0.18 \pm 0.009 \text{ AB}, \beta \gamma$ (< 0.02)	$0.19 \pm 20.91 \text{ B}, \beta \gamma$ (< 0.02)	
5- Hydroxymethylfurfura		$2.51 \pm 0.00 \text{ b}, \beta$ (0.04)	$1.80 \pm 0.14 \text{ a}, \alpha$ (0.02)	$1.68 \pm 0.30 \text{ a}, \alpha$ (0.02)	$1.47 \pm 0.09 \text{ a}, \alpha$ (< 0.02)	$1.37 \pm 0.13 \text{ A}, \alpha$ (< 0.02)	$1.90 \pm 0.19 \text{ A}, \alpha$ (0.02)	$1.89 \pm 0.25 \text{ A}, \alpha$ (0.02)	$1.73 \pm 0.25 \text{ A}, \alpha$ (0.02)	
T ( (T)	Total	$5.52\pm0.00$ c, $\gamma$	$4.25 \pm 0.14 \text{ b}, \beta$	$3.54 \pm 0.76 \text{ ab}, \alpha\beta$	$2.89 \pm 0.14 \text{ a}, \alpha$	$2.83 \pm 0.20 \text{ A}, \alpha$	$3.60 \pm 0.19 B$ , $\alpha\beta$	$3.62 \pm 0.34 \text{ B}, \alpha\beta$	$3.32 \pm 0.20$ AB, $\alpha$	
Lactones (µg/L)		0.50 . 0.01 . 0	0.01 . 0.001 . 0				0.07 . 0.00 .	2.22 . 0.22 D	251 . 2215	
cis- whiskey lactone	54 <sup>h</sup>	$0.78 \pm 0.01 \text{ a, } \beta$ (< 0.02)	$0.81 \pm 0.00 \text{ b}, \beta$ (< 0.02)	n.d.	n.d.	n.q.	$0.27 \pm 0.08 \text{ A}, \alpha$ (< 0.02)	$3.33 \pm 0.22 \text{ B}, \gamma$ (0.06)	$3.71 \pm 0.24 \text{ B}, \gamma$ (0.07)	
trans- whiskey lactone	370 <sup>h</sup>	$0.69 \pm 0.01 \text{ a, } \beta$ (< 0.02)	$1.04 \pm 0.01 \text{ b}, \gamma$ (< 0.02)	n.d.	n.d.	n.q.	$0.39 \pm 0.12 \text{ A}, \alpha$ (< 0.02)	$4.65 \pm 0.14 \text{ B}, \delta$ (< 0.02)	$5.17 \pm 0.08 \text{ C}, \epsilon$ (< 0.02)	
	Total	$1.47\pm0.02~\beta$	$1.84 \pm 0.01~\beta$	n.d.	n.d	n.q.	$0.66 \pm 0.16 \text{ A}, \alpha$	$7.98 \pm 0.31 \; \mathrm{B}, \gamma$	$8.89 \pm 0.29 \text{ C}, \delta$	
Terpenoids (µg/L)										
Citronellal	references not found	$0.62 \pm 0.01$ b, $\alpha$	$0.42 \pm 0.03$ a, $\alpha$	$0.41\pm0.13$ a, $\alpha$	$0.36 \pm 0.02$ a, $\alpha$	$0.48 \pm 0.08$ A, $\alpha$	$0.47 \pm 0.01 \text{ A}, \alpha$	$0.36 \pm 0.03$ A, $\alpha$	$0.53 \pm 0.23 \text{ A}, \alpha$	
Citronellol	100 <sup>b</sup>	n.d.	$1.89 \pm 0.12 \text{ a}, \gamma$ (0.02)	n.d.	n.d.	n.d.	$1.20 \pm 0.05 \text{ A}, \alpha$ (< 0.02)	n.d. (< 0.02)	$1.39 \pm 0.11 \text{ B}, \beta$ (< 0.02)	
Farnesol	$1000^{\rm g}$	$0.69 \pm 0.01 \text{ a, } \alpha\beta$ (< 0.02)	$2.25 \pm 0.01 \text{ b}, \delta$ (< 0.02)	$0.70 \pm 0.30$ a, $\alpha\beta$ (< 0.02)	$0.60 \pm 0.03$ a, $\alpha\beta$ (< 0.02)	$0.29 \pm 0.02 \text{ A}, \alpha$ (< 0.02)	$1.01 \pm 0.09 \text{ B}, \beta$ (< 0.02)	$0.38 \pm 0.03 \text{ A}, \alpha$ (< 0.02)	$1.66 \pm 0.40 \text{ C}, \gamma$ (< 0.02)	
Geranyl Acetone	60 <sup>a</sup>	$0.17 \pm 0.01 \text{ a, } \gamma \ (< 0.02)$	$0.10 \pm 0.01 \text{ a, } \alpha\beta$ (< 0.02)	$0.10 \pm 0.06 \text{ a},  \alpha\beta$ (< 0.02)	$0.11 \pm 0.02 \text{ a, } \alpha\beta\gamma \ (< 0.02)$	$0.10 \pm 0.01 \text{ B},  \alpha\beta$ (< 0.02)	$0.06 \pm 0.02 \text{ A}, \alpha$ (< 0.02)	$0.08 \pm 0.01 \text{ AB, } \alpha \ (< 0.02)$	$0.15 \pm 0.01 \text{ C},  \beta \gamma$ (< 0.02)	
β-ionol	references not found	$61.34 \pm 0.01~a,\alpha$	$24.42\pm0.07~a,\alpha$	$138.63 \pm 44.36 \ b,  \beta$	$136.30 \pm 5.11 \ b,  \beta$	$38.95 \pm 2.97 \text{ B}, \alpha$	$24.56\pm2.00~A,\alpha$	$110.68\pm6.76~C,\beta$	$121.20 \pm 6.69 \text{ C}, \beta$	
β-ionone	0.09e	$0.02 \pm 0.01 \text{ a, } \alpha$ (0.20)	$0.02 \pm 0.01 \text{ a, } \alpha$ (0.20)	$0.01 \pm 0.00 \text{ a, } \alpha$ (0.15)	$0.01 \pm 0.00 \text{ a, } \alpha$ (0.15)	$0.01 \pm 0.00 \text{ A}, \alpha$ (0.15)	$0.01 \pm 0.00 \text{ A}, \alpha$ (0.15)	$0.01 \pm 0.00 \text{ A}, \alpha$ (0.15)	$0.04 \pm 0.00 \; B,  \beta$ (0.43)	

Linalool	25°	$1.67 \pm 0.01 \text{ c}, \delta$ (0.07)	$0.70 \pm 0.05 \text{ b}, \gamma$ (0.03)	$0.23 \pm 0.11 \text{ a, } \alpha\beta$ (< 0.02)	$0.15 \pm 0.02$ a, $\alpha$ (< 0.02)	$0.33 \pm 0.04 \text{ A}, \alpha\beta$ (< 0.02)	$0.43 \pm 0.09 \text{ A}, \beta$ (0.02)	$2.22 \pm 0.02 \text{ B}, \epsilon$ (0.09)	$2.48 \pm 0.15 \text{ C}, \zeta$ (0.10)
	Total	$64.50 \pm 0.06 \text{ a}, \alpha$	$29.80 \pm 0.11 \text{ a, } \alpha$	$140.09 \pm 44.96 \text{ b}, \beta$	$137.54 \pm 5.15$ b, $\beta$	$40.16 \pm 3.10 \text{ A}, \alpha$	$27.73 \pm 2.07 \text{ A}, \alpha$	$113.73 \pm 6.71 \text{ B}, \beta$	$127.46 \pm 6.52 \text{ C}, \beta$
Volatile phenols (µg/L)									
1 "0"		$4.01 \pm 0.01$ a, $\alpha$	$2.46 \pm 0.82 \text{ a, } \alpha$	$2.97 \pm 0.01 \text{ a. } \alpha$	$2.43 \pm 0.48 \text{ a. } \alpha$	$2.98 \pm 0.45 \text{ AB, } \alpha$	$2.60 \pm 0.17 \text{ A}, \alpha$	$3.10 \pm 0.53$ AB, $\alpha$	$3.89 \pm 0.12 \text{ B}, \alpha$
4-Ethylphenol	440e	(< 0.02)	(< 0.02)	(< 0.02)	(< 0.02)	(< 0.02)	(< 0.02)	(< 0.02)	(< 0.02)
Eugenol	6°	$0.80 \pm 0.01~b,\beta$	$0.61 \pm 0.03$ ab, $\alpha\beta$	$0.58 \pm 0.17$ a, $\alpha\beta$	$0.47 \pm 0.00$ a, $\alpha$	$0.63 \pm 0.13$ A, $\alpha\beta$	$0.85\pm0.07~AB,\beta\gamma$	$1.13\pm0.02~B,\gamma$	$2.03 \pm 0.19$ C, $\delta$
Lugenor	Ü	(0.13)	(0.10)	(0.10)	(0.08)	(0.10)	(0.14)	(0.19)	(0.34)
Guaiacol	9.5°	$17.98 \pm 0.01 \text{ b}, \beta$ (1.89)	$18.25 \pm 0.73 \text{ b}, \beta$ (1.92)	$11.51 \pm 2.00 \text{ a, } \alpha$ (1.21)	$11.40 \pm 0.53 \text{ a, } \alpha$ (1.20)	$9.23 \pm 0.78 \text{ A}, \alpha$ (1.00)	$11.48 \pm 2.07 \text{ A}, \alpha$ (1.21)	$15.21 \pm 0.67 \text{ B}, \beta$ (1.60)	$16.30 \pm 1.50 \text{ B}, \beta$ (1.72)
		$14.75 \pm 0.01 \text{ c, } \delta \epsilon$	$9.02 \pm 1.27 \text{ a, } \alpha\beta$	(1.21) 15.47 ± 0.00 c, $\varepsilon$	$11.33 \pm 0.81 \text{ b}, \beta \gamma$	$8.01 \pm 1.64 \text{ A}, \alpha$	$9.08 \pm 0.88 \text{ A}, \alpha\beta$	$12.51 \pm 1.17 \text{ B}, \gamma \delta$	$12.27 \pm 0.48 \text{ B}, \gamma \delta$
Syringol	570e	(0.03)	(0.02)	(0.03)	(0.02)	(< 0.02)	(0.02)	(0.02)	(0.02)
Vanillin	$60^{\rm f}$	$118.77 \pm 0.01 \text{ a, } \beta$	$104.25 \pm 0.71 \text{ a, } \alpha\beta$	$164.07 \pm 0.00 \text{ b}, \gamma$	$170.62 \pm 17.47 \text{ b}, \gamma$	$86.54 \pm 1.69 \text{ A}, \alpha$	$95.79 \pm 8.41 \text{ A}, \alpha\beta$	$227.40 \pm 8.07 \text{ B}, \delta$	261.25 ±15.99 C, ε
v ammin	00	(1.98)	(1.74)	(2.73)	(2.84)	(1.44)	(1.60)	(3.79)	(4.35)
4-Vinylguaiacol	40 °	$1.09 \pm 0.01 \text{ a, } \alpha\beta$	$0.56 \pm 0.05 \text{ a}, \alpha$	$1.31 \pm 1.56 \text{ a}, \alpha\beta$	$0.97 \pm 0.11 \text{ a, } \alpha\beta$	$0.87 \pm 0.13 \text{ A}, \alpha\beta$	$0.67 \pm 0.30 \text{ A}, \alpha$	$2.16 \pm 0.26 \mathrm{C}, \beta$	$1.57 \pm 0.14 \text{ B}, \alpha\beta$
	Total	(0.03) $157.40 \pm 0.06 \text{ a, } \beta$	(< 0.02) 135.14 ± 0.45 a, $\alpha\beta$	(0.03) 195.91 ± 5.02 b, $\gamma$	$(0.02)$ 197.77 ± 16.41 b, $\gamma$	$(0.02)$ $108.26 \pm 1.34$ , $\alpha$	$(0.02)$ $120.48 \pm 10.07 \text{ A, } \alpha$	(0.05) $261.50 \pm 8.32 \text{ B}, \lambda$	(0.04) 297.32 ± 16.88 C, $\delta$
Others (µg/L)	10111	137.40 ± 0.00 a, p	155.14 ± 0.45 a, up	175.71 ± 5.02 0, γ	177.77 ± 10.41 0, γ	100.20 ± 1.54 ,0	120.40 ± 10.07 A, a	201.30 ± 6.32 B, K	277.32 ± 10.00 €,0
Benzaldehyde	2000 <sup>d</sup>	$8.86 \pm 0.01$ b, $\gamma$	$4.93\pm0.10~a,\alpha$	$7.10 \pm 2.39$ ab, $\beta \gamma$	$5.48\pm0.38~a,\alpha\beta$	$4.98 \pm 0.14 \text{ A}, \alpha$	$5.29 \pm 0.06~A,\alpha\beta$	$6.46\pm0.26~B,\alpha\beta$	$6.15\pm0.17~B,\alpha\beta$
Benzaidenyde	2000"	(< 0.02)	(< 0.02)	(< 0.02)	(< 0.02)	(< 0.02)	(< 0.02)	(< 0.02)	(< 0.02)
Nonanal	15 <sup>f</sup>	$1.70 \pm 0.01 \text{ b}, \beta$	$0.80 \pm 0.02 \text{ a}, \alpha$	$1.11 \pm 0.33 \text{ a}, \alpha$	$0.79 \pm 0.05 \text{ a}, \alpha$	$0.94 \pm 0.14 \text{ A}, \alpha$	$0.84 \pm 0.19 \text{ A}, \alpha$	$0.96 \pm 0.24 \ 0.06 \ A, \alpha$	$1.18 \pm 0.04 \text{ A}, \alpha$
		(0.11)	(0.05)	(0.07)	(0.05)	(0.06)	(0.06)	(0.06)	(0.08)
1-Octen-3-ol	3°	$1.43 \pm 0.01 \text{ c}, \gamma$ (0.48)	$1.10 \pm 0.13 \text{ b}, \beta \gamma$ (0.37)	$0.41 \pm 0.13 \text{ a, } \alpha$ (0.14)	$0.85 \pm 0.06 \text{ b}, \beta$ (0.28)	$0.71 \pm 0.04 \text{ A}, \alpha\beta$ (0.24)	$0.99 \pm 0.07 \text{ A}, \beta$ (0.33)	$0.94 \pm 0.01 \text{ A}, \beta$ (0.31)	$0.94 \pm 0.32 \text{ A}, \beta$ (0.31)
	Total	$11.98 \pm 0.03$ b, $\beta$	$6.83 \pm 0.21 \text{ a, } \alpha$	$8.62 \pm 2.85 \text{ ab, } \alpha$	$7.12 \pm 0.45 \text{ a, } \alpha$	$6.63 \pm 0.17 \text{ A}, \alpha$	$7.13 \pm 0.13 \text{ A}, \alpha$	$8.36 \pm 0.41 \text{ B}, \alpha$	$8.26 \pm 0.11 \text{ B}, \alpha$
Total volatile compounds (µ	g/L)	5876.63±0.28b, δ	4713.64±178.57a, γ	4234.96±907.27a,βγ	3511.36±19.27a, αβ	3222.50±205.9A, α	4028.73±199.36Β,αβγ	4262.31±334.0Β,αβγ	4060.44±177.33Β,αβγ
Phenolic compounds									
Flavanols (mg/L)									
(+) -Catechin		n.d.	n.d.	n.d.	$4.09 \pm 0.12 \text{ a, } \beta$	n.d.	n.d.	n.d.	$1.12 \pm 0.01 \text{ A}, \alpha$
(-) -Epicatechin		n.d.	n.d.	n.d.	$14.14 \pm 0.95 \text{ a}, \beta$	n.d.	n.d.	n.d.	$6.23 \pm 0.21$ A, $\alpha$
	Total	n.d.	n.d.	n.d.	$18.24 \pm 0.83$ a, $\beta$	n.d.	n.d.	n.d.	$7.34 \pm 0.21$ A, $\alpha$
Phenolic acids (mg/L)		22 10 : 0.011	22.00 + 0.07.1	22.02 + 0.00	22.52 . 0.02 . 0	22.22 . 0.02 . 0	22.54 + 0.01 + 0	22.22 . 0.21 . 0	22.52 . 0.04 . 0
Ellagic acid Gallic acid		$33.18 \pm 0.01 \text{ b}, \alpha$ $0.62 \pm 0.03 \text{ c}, \gamma$	$33.98 \pm 0.07 \text{ d}, \gamma$ $0.23 \pm 0.01 \text{ a}, \beta$	$33.03 \pm 0.00 \text{ a, } \alpha$ n.d.	$33.53 \pm 0.02 \text{ c}, \beta$ $0.55 \pm 0.01 \text{ b}, \gamma$	$33.22 \pm 0.03 \text{ A}, \alpha\beta$ n.d.	$33.54 \pm 0.01 \text{ A}, \beta$	$33.33 \pm 0.21$ A, $\alpha\beta$ n.d.	$33.53 \pm 0.04 \text{ A}, \beta$ $0.68 \pm 0.02 \text{ B}, \delta$
Protocatechuic acid		$0.02 \pm 0.03 \text{ c}, \gamma$ $0.25 \pm 0.00 \text{ b}, \gamma$	$0.23 \pm 0.01 \text{ a, p}$ $0.21 \pm 0.00 \text{ a, \beta}$	n.d.	$0.35 \pm 0.01 \text{ b}, \gamma$ $0.25 \pm 0.02 \text{ b} \gamma$	n.d.	$0.16 \pm 0.00 \text{ A}, \alpha$	n.d.	$0.08 \pm 0.02 \text{ B}, \delta$ $0.34 \pm 0.01 \text{ B}, \delta$
Trotocutectiate deta	Total	$34.05 \pm 0.03$ b, $\beta \gamma$	$34.42 \pm 0.08 \text{ c}, \delta$	$33.00 \pm 0.00 \text{ a, } \alpha$	$34.33 \pm 0.03 \text{ c}, \gamma \delta$	$33.22 \pm 0.03 \text{ A, } \alpha$	$33.69 \pm 0.03 \text{ B}, \beta$	$33.33 \pm 0.21 \text{ AB}, \alpha$	$34.54 \pm 0.06 \text{ C}, \delta$
Stilbenes (mg/L)		.,,	· ·	,	· •	,	· ,	,	,
trans-Piceid		n.d.	n.d.	n.d.	$0.68 \pm 0.00$ a, $\alpha$	n.d.	n.d.	n.d.	$0.71\pm0.00~A,\beta$
trans-Resveratrol	T . 1	$1.16 \pm 0.03 \text{ c}, \gamma$	$0.74 \pm 0.00 \text{ b}, \beta$	$0.64 \pm 0.00$ a, $\beta$	$4.05 \pm 0.02 \text{ d, } \varepsilon$	$0.48 \pm 0.01 \text{ A}, \alpha$	$0.63 \pm 0.00 \text{ A}, \beta$	$0.48 \pm 0.00 \text{ A}, \alpha$	$2.64 \pm 0.09 \text{ B}, \delta$
Aldahudas (mc/I)	Total	$1.16 \pm 0.03 \text{ c}, \gamma$	$0.74 \pm 0.00$ b, $\beta$	$0.64 \pm 0.00$ a, $\beta$	$4.73 \pm 0.02$ d, $\varepsilon$	$0.48 \pm 0.01 \text{ A}, \alpha$	$0.63 \pm 0.00 \text{ A}, \beta$	$0.48 \pm 0.00 \text{ A}, \alpha$	$3.35 \pm 0.09 \text{ B}, \delta$
Aldehydes (mg/L) Coniferaldehyde		$0.63 \pm 0.05$ b, $\alpha$	n a	n a	n a	n a	n a	n a	$0.73 \pm 0.00 \text{ A, } \beta$
Sinapaldehyde		$0.03 \pm 0.03$ b, $\alpha$ $0.20 \pm 0.03$ a, $\alpha$	n.q n.q	n.q n.q	n.q. n.q	n.q n.q	n.q n.q	n.q. n.q	$0.73 \pm 0.00 \text{ A}, \beta$ $0.40 \pm 0.00 \text{ A}, \beta$
paraerij de	Total	$0.83 \pm 0.08 \text{ b}, \beta$	n.q.	n.q.	n.q.	n.q.	n.q.	n.q.	$1.13 \pm 0.04 \text{ A}, \lambda$
Total phenolic compound		$36.03 \pm 0.14 \text{ b}, \gamma$	$35.15 \pm 0.08 \text{ ab}, \beta \gamma$	$33.64 \pm 0.00a$ , $\alpha$	$57.30 \pm 0.84 \text{ c, } \epsilon$	$33.71 \pm 0.04$ A, $\alpha$	$34.32 \pm 0.03$ B, $\alpha\beta$		

For each compound, different small letters indicate significant differences among all treatments carried out with Airén variety, capital letters indicate significant differences among all treatments carried out with Cencibel variety. Different Greek letters indicate significant differences among all treatments, independently of the variety. Tuckey test (α < 0.05) was used for statistical analysis. The mean values (n= 3) are shown with their standard deviation. n.d. not detected. n.q. not quantified. Mean pH of model wines: Airén *chips* 21 days (4.19), Airén *chips* 35 days (3.96), Airén *granule* 21 days (4.22), Airén *granule* 35 days (4.05), Cencibel *chips* 21 days (4.04), Cencibel *chips* 35 days (3.80), Cencibel *granule* 21 days (4.28).

a) (Guth, 1997)

- b)
- c)

- (Etiévant, 1991)
  (Ferreira, López, & Cacho, 2000)
  (Khairallah, Reynolds, & Bowen, 2016)
  (López, Aznar, Cacho, & Ferreira, 2002)
  (Culleré, Escudero, Cacho, & Ferreira, 2004)
  (Franco, Peinado, Medina, & Moreno, 2004)
  (Boidron, Chatonnet, & Pons, 1988)