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Additional Information

24 1. Introduction

25 Osteochondral defects are focal areas where cartilage damage and injury of the adjacent
26 subchondral bone takes place, that can be treated using different strategies, such as (i)
27 osteochondral autograft [1], (ii) autologous chondrocytes [2] or (iii) matrix-induced autologous
28 chondrocyte implantation [3]. Nowadays, no successful method for complete regeneration of
29 osteochondral defects exists [4]. A graft designed for treating large osteochondral defects should
30 be a tissue-engineered osteochondral (bone–cartilage) composite (with a predefined size and shape)
31 characterised by mechanical stability and an appropriate postoperative functionality under
32 physiological conditions [5], to achieve simultaneous regeneration of both cartilage and
33 subchondral bone. Bilayered scaffolds are proposed for repairing osteochondral defects, in order
34 to allow the preparation of optimized different layers able to mimic the native extracellular matrix
35 for each tissue type (bone and cartilage), tuning the physico-chemical, structural, and mechanical
36 properties in a single structure [6]. Bilayered scaffolds have been classified into three types: (i)
37 “Cartilage tissue on bone scaffold” in which chondrocytes or neocartilage tissue are seeded directly
38 onto a bone scaffold, (ii) “Assembled bilayered scaffolds” in which two distinct and different
39 cartilage and bone scaffolds are assembled together before or during surgical implantation, and (iii)
40 “Integrated bilayered scaffolds”, consisting of two different structures that are joined together
41 through the integration of a material contained in both layers [7].

42 The potential advantages of glass/polymer composite scaffolds for regenerative medicine have
43 been widely emphasized in the recent literature [8], [9], [10] In this work, innovative bilayered
44 sponge-like scaffolds, based on a bioresorbable phosphate glass and a
45 glycidoxypropyltrimethoxysilane-crosslinked network of gelatin (G) were studied in order to

46 investigate their potential for osteochondral tissue regeneration. The proposed matrices represent a
47 new category of bilayered scaffolds, that could be easily obtained by a single step procedure.

48 **2. Materials and methods**

49 **2.1. Scaffolds preparation**

50 G (type A from porcine skin) and GPTMS were supplied from Sigma-Aldrich, Milan. Powders
51 (particle size <30 μm) of resorbable phosphate glass (I-CEL2; molar composition:45% P₂O₅, 3%
52 SiO₂, 26% CaO, 7% MgO, 15% Na₂O, 4% K₂O) were prepared as reported elsewhere [11], [12],
53 [13]. The scaffolds were prepared according to the following procedures: G was dissolved in
54 demineralised water at 50 °C to obtain a 2.5%(w/v) solution. I-CEL2 was added to the gelatin
55 solution to obtain I-CEL2/G composites with various weight ratios between the components:
56 0/100;30/70;70/30 (%w/w). The composites were coded as follows: I-CEL2/G 0/100;30/70;70/30.
57 Then, GPTMS was added to the G solution as previously described [14]. The solutions were poured
58 into polystyrene 24-multiwell containers for 24 h to complete the crosslinking reaction, and then
59 freeze-dried (Scanvac-CoolSafe) at -20 °C for 48 h. Uncrosslinked sponges were prepared as
60 control.

61 **2.2. Scaffold characterization**

62 The swelling behaviour was evaluated at 37 °C using a phosphate buffered saline at pH 7.4 (Sigma-
63 Aldrich). The swelling degree was measured after 3, 6 and 24 h and calculated as:
64 $\Delta W_s(\%) = ((W_s - W_0) / W_0) \times 100$, where W₀ and W_s are the sample weights before and after
65 swelling, respectively.

66 Morphological (SEM, Philips 525 M) and compositional analysis (energy-dispersive spectroscopy,
67 EDS; Philips EDAX 9100) were performed on fractured specimen sections. The samples were
68 sputter coated with carbon prior to the examination.

69 Pore dimension and distribution was quantified by micro-computed tomography (μ -CT,
70 MicroXCT-200 series, XRADIA). No contrasting agent was added and the samples had a minimum
71 size of $1 \times 1 \times 0.5$ mm³. The scanner was set at a voltage of 40 kV; the samples were scanned at 0.597
72 μ m pixel resolution by 1000 slices covering the sample height.

73 Compressive stress–strain curves were measured using MTS QTest/10 device and a load cell of 50
74 N. Test specimens (n=3) were cylinder-shaped sponges (1.2 cm diameter and an average height of
75 1.2 cm). The cross-head speed was set at 0.01 mm s⁻¹ and the load was applied until the specimen
76 was compressed to 70% of its original length. Young's modulus (E), collapse strength and strain
77 (σ^* and ε^*) and collapse modulus (E*) were measured from the stress-strain curves. E is the linear
78 elastic regime slope, E* is the collapse regime slope, σ^* and ε^* are, respectively, the stress and
79 strain of transition from linear to collapse regime [15].

80 Calorimetric measurements were performed using TA-INSTRUMENTS DSC/Q20. The samples
81 (6–8 mg) were hermetically sealed in aluminium pans. Heating was carried out at 10 °C min⁻¹ in
82 the 30–130 °C temperature range. Denaturation temperature (Td) and enthalpy (ΔH_d) were
83 calculated as the temperature of the maximum value of the denaturation endotherm and the peak
84 area. Denaturation enthalpy was normalised with respect to G content.

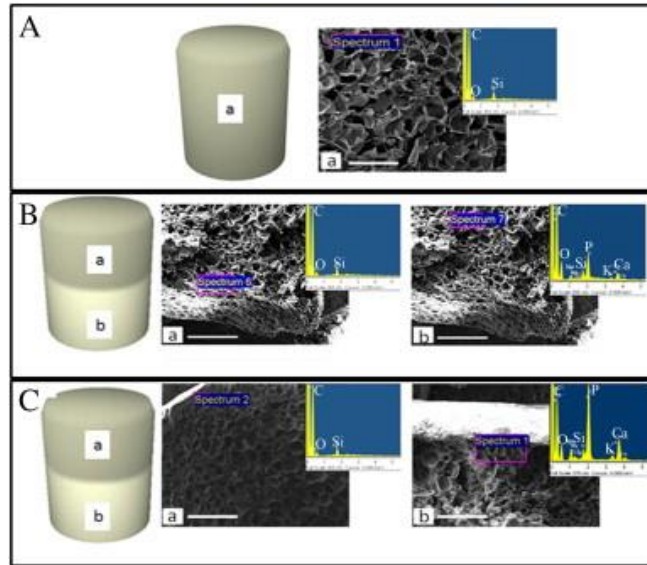
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87

88 3. Results and discussion

89 SEM analysis was performed to evaluate the effect of composition on scaffold morphology. Fig.
90 1(A) shows the typical foam-like morphology with interconnected pores of I-CEL2/G 0/100
91 scaffold. EDS spectra of pure G scaffold (insert in Fig. 1(A)) indicates the presence of the
92 characteristic elements contained in gelatin: carbon(C), nitrogen(N) and oxygen(O). Fig. 1(B) and
93 (C) show I-CEL2/G 30/70 and I-CEL2/G 70/30 scaffolds: EDS analysis demonstrated that these
94 samples were characterized by a typical bilayered structure, consisting of a top gelatin layer and a
95 bottom layer mainly constituted of I-CEL2 and a very low gelatin amount. EDS spectra of the
96 bottom layers showed the characteristic elements of I-CEL2, namely silicon(Si), potassium(K),
97 sodium(Na), magnesium(Mg), calcium(Ca) and phosphorus(P) (insert in Fig. 1(B) and (C)) where
98 EDS spectra of the top layers showed the characteristic elements of gelatin. The porosity degree
99 was different in the top and in the bottom layers, and dependent on the concentration of I-CEL2.
100 In particular, the gelatin top layer exhibited a total porosity of 86.2 vol%; on the other hand, the
101 porosity degree of the bottom layer was found to vary from 67.1 to 86.2 vol% with decreasing of
102 the concentration of I-CEL2. The dependence of porosity degree on I-CEL2 amount has to be
103 ascribed to the deposition of the bioresorbable glass particles on the pore walls as confirmed
104 subsequently by SEM examination (Fig. 1(B) and (C)). The mean pore size of I-CEL2/G 0/100 and
105 of the top layers of I-CEL2/G 30/70 and I-CEL2/G 70/30 scaffolds was found to be around 149.2
106 μm ; on the other hand, the mean pore size of the bottom layers was found to vary from 135.7 μm
107 for I-CEL2/G 30/70 to 126.5 μm for I-CEL2/G 70/30 composites, which demonstrates that the
108 average pore size decreased with increasing I-CEL2 content. All samples showed a high
109 interconnected network of pores (95%) with higher size than 95.1 μm , as assessed by μ -CT
110 analysis.

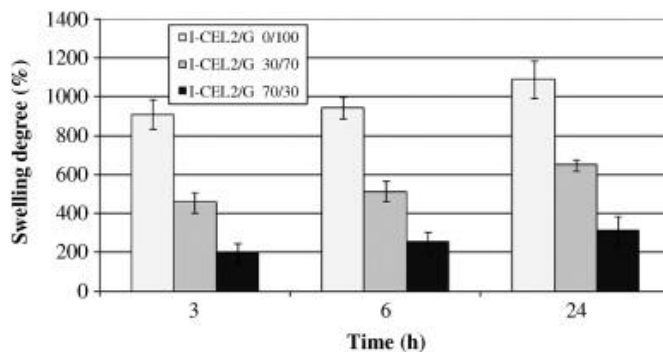


111
 112 Fig. 1. SEM micrographs and EDS spectra of I-CEL2/G composite scaffolds: (A) I-CEL2/G
 113 0/100, (B) ICEL2/G 30/70, (C) ICEL2/G 70/30 (bar: 100 μm). For each SEM micrographs (a)
 114 indicates the top side, and (b) the bottom side.

115 Different methods were reported in literature to prepare bilayered scaffolds [16], [17], [18], [19],
 116 generally based on two consecutive different procedures (e.g., sintering and freeze-drying). In our
 117 case, bilayered scaffolds could be easily obtained by casting I-CEL2/G mixture solutions: the
 118 higher density of I-CEL2 (2.6 g/cm^3) as compared to the G phase caused the progressive
 119 precipitation of I-CEL2 at the bottom of multiwell containers for gravity when the solution was
 120 poured into polystyrene 24-multiwell containers for 24 h to complete the crosslinking reaction.

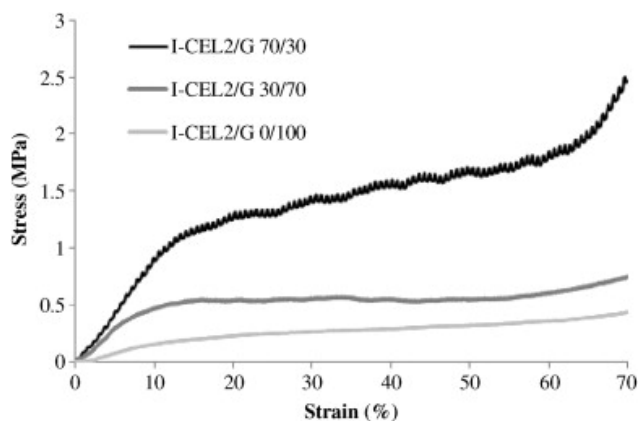
121 The increase in swelling also allows the scaffold to avail nutrients from culture media more
 122 effectively [20]. Fig. 2 reports the swelling degree as a function of time for composite porous
 123 matrices with different compositions. All composites showed a similar swelling behaviour: the
 124 swelling degree slightly increased over time from 3 to 24 h. I-CEL2/G 0/100 scaffolds displayed
 125 the highest swelling at each time interval (from $909 \pm 52\%$ at 3 h to $1088 \pm 60\%$ at 24 h). For I-
 126 CEL2/G 30/70 samples, at 3 h the swelling degree was about $457 \pm 50\%$. At 12 h the swelling degree

127 did not increased significantly, while at 24 h the swelling ratio was about $650\pm 45\%$. Moreover, for
 128 I-CEL2/G 70/30 composites, swelling ratio was about $197\pm 53\%$ after 3 h, while at 6 and 24 h, the
 129 swelling degree did not increase significantly. At each time, swelling degree was found to decrease
 130 with increasing I-CEL2 amount, because of the lower hydrophilicity of the inorganic phase as
 131 compared to the polymeric matrix causing a decrease of the water sorption as suggested in previous
 132 works [21], [22].



133
 134 Fig. 2. Swelling behavior of scaffolds as a function of time.

135 Fig. 3 shows the stress-strain curves obtained for the matrices by compression tests. A significant
 136 increase of compression Young's modulus was obtained by adding I-CEL2 into the gelatin matrix,
 137 due to the superior stiffness of the inorganic phase. compression behaviour of the glass as compared
 138 to G phase (1.9 ± 0.2 MPa for I-CEL2/G 0/100 up to 7.6 ± 0.5 MPa for I-CEL2/G 70/30). As shown
 139 in Table 1, the collapse strength and collapse strain were characterized by a different trend as a
 140 function of the I-CEL2 amount. In particular, the increase of the inorganic phase caused a
 141 progressive, slight decrease in the deformability of the composite scaffold and an increase of the
 142 collapse strength and modulus.



143

144 Fig. 3. Stress–strain curves of the porous scaffolds compressed at a strain of (0–70%).

145

146 Table 1. Elastic modulus, collapse strength and strain, and collapse modulus calculated from the
147 corresponding stress–strain curves.

I-CEL2/G sample	E (MPa)	σ^* (MPa)	ε^* (%)	$\Delta\sigma/\Delta\varepsilon$ (kPa)
0/100	1.9±0.2	0.27±0.04	14.4±1.5	0.09±0.01
30/70	5.6±0.4	0.56±0.02	10.0±0.3	0.13±0.03
70/30	7.6±0.5	0.63±0.13	8.3±0.6	1.26±0.02

148

149 DSC analysis was performed to analyse the thermal behaviour of scaffolds as a function of
150 composition together with the influence of I-CEL2 on gelatin thermal properties. Crosslinking
151 increased the thermal stability of gelatin helices as shown by the shift of the Td to higher values
152 (95.1 °C for I-CEL2/G 0/100) as compared to uncrosslinked gelatin scaffolds (92.3 °C) [19].
153 Crosslinking generally induced a decrease in the denaturation enthalpy, which was ascribed both
154 to a reduction of hydrogen bonds, and to a simultaneous increase in the extent of covalent crosslinks
155 [23] (30.2 J g⁻¹ for uncrosslinked gelatin and 26.0 J g⁻¹ for I-CEL2/G 0/100). The denaturation
156 temperature of I-CEL2/G composites with different weight ratios between inorganic and organic
157 phase slightly increased with respect to pure crosslinked gelatin film (97.0 °C for I-CEL2/G 30/70

158 and 98.5 °C for I-CEL2/G 70/30). It is worth noting that the composites showed low denaturation
159 enthalpy values (7.1 J g⁻¹ for I-CEL2/G 30/70 and 5.2 J g⁻¹ for I-CEL2/G 70/30), probably due to
160 a reduction of the helical structure as a consequence of the strong interactions between
161 bioresorbable glass and gelatin.

162

163 **4. Conclusions**

164 A new category of bilayered scaffolds were successfully and easily prepared by a single step
165 procedure for osteochondral tissue regeneration. The obtained scaffolds showed an interconnected
166 network of macropores with 100–150 µm average size as shown by SEM and µ-CT analysis.
167 Moreover, scaffolds containing I-CEL2 were particularly interesting due to their (i) increased
168 stability in aqueous solution as evidenced by swelling tests, (ii) increased compressive Young's
169 modulus with respect to the pure G, and (iii) interactions between the phases as suggested by the
170 slight increase in the denaturation temperature. The obtained composites represent promising
171 candidates for future trials in the field of osteochondral regeneration. Additional work is in
172 progress, with the aim to investigate the biocompatibility of these composite scaffolds, in vitro and
173 in vivo.

174 **Acknowledgements**

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176

177 **References**

178 [1] Marcacci M, Kon E, Delcogliano M, Filardo G, Busacca M, Zaffagnini S. *Am J Sports Med*
179 2007;35:2014–21.

180 [2] Lee CY, Liu X, Hsu HC, Wang DY, Luo ZP. *Connect Tissue Res* 2005;46:93–9.

181 [3] Ait Si, Selmi T, Neyret P, Verdonk PCM, Barnouin L. *Tech Knee Surg* 2007;6:253–8.

- 182 [4] Swieszkowski W, Tuan BH, Kurzydowski KJ, Hutmacher DW. *Biomol Eng* 2007;24(5):489-
183 95.
- 184 [5] Schaefer D, Martin I, Jundt G, Seidel J, Heberer M, Grodzinsky A, et al. *Arthritis Rheum*
185 2002;46:2524–34.
- 186 [6] Malafaya PB, Reis RL. *Acta Biomater* 2009;5:644–50.
- 187 [7] O’Shea TM, Miao X. *Tissue Eng B* 2008;14:447–64.
- 188 [8] Rezwani K, Chen QZ, Blaker JJ, Boccaccini AR. *Biomaterials* 2006;27:3413–31.
- 189 [9] Yunus DM, Bretcanu O, Boccaccini AR. *J Mater Sci* 2008;43:4433–42.
- 190 [10] Ahmed I, Jones IA, Parsons AJ, Bernard J, Farmer J, Scotchford CA, et al. *J Mater Sci: Mater*
191 *Med* 2011;22:1825–34.
- 192 [11] Leonardi E, Ciapetti G, Baldini N, Novajra G, Verne’ E, Bains F, et al. *Acta Biomater*
193 2010;6:598–606.
- 194 [12] Vitale-Brovarone C, Bains F, Bretcanu O, Verne’ E. *J Mater Sci: Mater Med* 2009;20:2197–
195 205.
- 196 [13] Vitale-Brovarone C, Ciapetti G, Leonardi E, Baldini N, Bretcanu O, Verne’ E, et al. *J Biomater*
197 *Appl* 2011;26:465–89.
- 198 [14] Tonda-Turo C, Gentile P, Saracino S, Chiono V, Nandagiri VK, Muzio G, et al. *Int J Biol*
199 *Macromol* 2011;49:700–6.
- 200 [15] Kanungo BP, Silva E, Van Vliet K, Gibson LJ, et al. *Acta Biomater* 2008;4:490–503.
- 201 [16] Olivera JM, Rodrigues MT, Silva SS, Malafaya PB, Gomes ME, Viegas CA, et al.
202 *Biomaterials* 2006;7:6123–37.
- 203 [17] Chen J, Chen H, Li P, Diao H, Zhu S, Dong L, et al. *Biomaterials* 2011;32:4793–805.
- 204 [18] Tampieri A, Sandri M, Landi E, Pressato D, Francioli S, Quarto R, et al. *Biomaterials*
205 2008;29:3539–46.

- 206 [19] Keeney M, Pandit A. *Tissue Eng B* 2009;15:55–73.
- 207 [20] Janaki K, Elamathi S, Sangeetha D, et al. *Trends Biomater Artif Organs* 2008;22:169–78.
- 208 [21] Ciardelli G, Gentile P, Chiono V, Mattioli-Belmonte M, Vozzi G, Barbani N, et al. *J Biomed*
209 *Mater Res Part A* 2010;92:137–51.
- 210 [22] Gentile P, Chiono V, Boccafoschi F, Baino F, Vitale-Brovarone C, Verne E, et al. *J Biomater*
211 *Sci, Polym Ed* 2010;21:1207–26.
- 212 [23] Bigi A, Cojazzi G, Panzavolta S, Rubini K, Roveri N. *Biomaterials*.