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**EFFECT OF PRE-TREATMENTS AND AIR-FRYING, A NOVEL
TECHNOLOGY, ON ACRYLAMIDE GENERATION IN FRIED POTATOES**

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Short title: Acrylamide generation in air-frying

Section: Toxicology and Chemical Food Safety

14 **Abstract**

15 This paper investigated the effect of air-frying technology, in combination with a pre-
16 treatment based of soaking the samples in different chemical agent solutions (citric acid,
17 glycine, calcium lactate, sodium chloride or nicotinic acid (vitamin B3)), on the
18 generation of acrylamide in fried potatoes. The influence of reducing sugars on the
19 development of surface's color was also analyzed. The experiments were conducted at
20 180 °C by means of air-frying and deep oil-frying, as a reference technology. Based on
21 the evolution of colour crust with frying time, it could be concluded that the rate of
22 Maillard reaction decreased as the initial reducing sugars content increased in the raw
23 material, and was also lower for deep-oil frying than for air-frying regardless of pre-
24 treatments applied. Air-frying reduced acrylamide content by about 90% compared with
25 conventional deep-oil frying without being necessary the application of a pre-treatment.
26 However, deep-oil fried potatoes pre-treated with solutions of nicotinic acid, citric acid,
27 glycine at 1% and NaCl at 2% presented much lower acrylamide levels (up to 80-90%
28 reduction) than non-pretreated samples.

29

30 **Keywords:** acrylamide, air-frying, additives, reducing sugars, color

31

32 **Introduction**

33 In June 2010, the European Union published recommendations for the control of
34 acrylamide content in food (2010/307/UE). This publication was the result of the
35 adventitious discovery of the excessive incidence of acrylamide in different foods,
36 mainly submitted to frying and baking, by researchers of the Stockholm University and
37 the publication of analytical data of various foods on the web site of the Swedish
38 National Food Agency in 2002 (Rosén and Hellenäs 2002). According to
39 epidemiological studies, the toxicity of acrylamide, a probable carcinogenic compound
40 for humans classified in Group 2A (IARC), it is not only due to its consumption, but
41 also to its role as a precursor in the development of other compounds during hepatic
42 metabolism e.g. glycidamide (Blank 2005).

43 Acrylamide is an early product in Maillard reaction which is produced by the reaction
44 between the amino acid asparagine and reducing sugars (glucose, fructose etc.) when
45 food is subjected to temperatures above 120 ° C in processes such as frying or baking
46 (Yaylayan and others 2003). Acrylamide generation is especially relevant in processed
47 potato products due to the high presence of the precursors (asparagine and reducing
48 sugars) in the matrix. The presence of the precursors can vary depending on the variety,
49 soil conditions and postharvest storage conditions (Kumar and others 2004).

50 The influence of the main frying variables (oil temperature, frying time, potato:oil ratio,
51 surface:volume ratio, etc.) on acrylamide generation has been extensively studied over
52 the past ten years (Bråthen, and Knutsen 2005; Gökmen, and Palazoglu 2009), as well
53 as its relationship with potato properties such as pH, water activity, capillarity and
54 porosity (Low and others 2006). It has been reported that the higher the frying
55 temperature and the lower the moisture content in the potato, the higher the acrylamide
56 content in the fried product (Pedreschi and others 2005). Some alternatives such as low

57 pressure-frying (Troncoso and Pedreschi 2009; Dueik and others 2012), microwave
58 assisted frying or microwave pre-thawing (Ngadi and others 2009; Tuta and others
59 2010), and pre-drying (Pedreschi and Moyano, 2005) have been studied in order to
60 obtain healthier fried products in terms of fat and/or acrylamide content. Air-frying is a
61 technology which permits the reduction of fat content in fried products by 90% (Andrés
62 and others 2012). The technology consists of direct contact between the product and a
63 dispersion of oil droplets in hot air in a frying chamber. The product is constantly in
64 motion to promote homogeneous contact between both phases. In this way, the product
65 is dehydrated and the typical crust of fried products gradually appears (Andrés and
66 others 2012). To the authors' knowledge, no publication provides scientific information
67 about the impact of this technology on acrylamide generation.

68 The application of a treatment prior to frying for acrylamide reduction has raised
69 interest in recent years. Some pre-treatments based on the immersion of the product in
70 solutions with chemical agents in order to mitigate acrylamide generation during frying
71 could be grouped according to the following objectives: (I) to reduce the content of
72 asparagine and reducing sugars in the raw material e.g. by blanching (Pedreschi and
73 others 2005); (II) to enzymatically hydrolyze asparagine i.e. asparaginase (Zyzak and
74 others 2003, Ciesarova and others 2006, Pedreschi and others 2011); (III) to add amino
75 acids (e.g. glycine (Low and others 2006)) or cations (e.g. Ca^{2+} or Na^+ : (Gökmen and
76 Senyuva 2007) able to compete with asparagine in Maillard reaction and (IV) to
77 kinetically control the rate of acrylamide formation by pH reduction (Stadler and Scholz
78 2004) or by lactic acid fermentation (Anese and others 2009; Bartkiene and others
79 2012). The criteria taken into account to select the substances employed in this study to
80 perform pre-treatments were mainly two: (i) the number of studies in which the
81 substance was used and its effectiveness. This is the case of blanching, citric acid or

82 sodium chloride; (ii) the functionality of the substance. In this sense, the incorporation
83 of a bioactive compound such as glycine, nicotinic acid (vitamin B3) or calcium during
84 pre-treatment could contribute to the functionality of the final fried product.

85 Moreover, it is important to point out that the majority of published studies concerning
86 the effect of pre-treatments by immersion in different additives to reduce acrylamide
87 have been carried out in model systems, but not in real foods in which the structure
88 could play an important role. In this study, the advantages of the application of these
89 different pre-treatments could accurately be established because of all of them are
90 applied at the same frying conditions and raw material.

91 The main objective of this study was to evaluate the effect of air-frying, in combination
92 with different chemical pre-treatments, on acrylamide reduction in a real food matrix
93 such as fried potato.

94

95 **Materials and Methods**

96 Reagents

97 Citric acid, nicotinic acid and acrylamide standard ($\geq 99\%$) were obtained from Merck
98 (Darmstadt, Germany), calcium lactate and sodium chloride from Scharlab (Barcelona,
99 Spain). Glycine, L (+) potassium sodium tartrate tetrahydrate, formic acid (99-100%
100 purity) and magnesium sulfate were purchased from VWR (Fontenay-sous-Bois,
101 France). 3,5-dinitrosalicylic acid (DNS) and hexane were from Panreac (Barcelona,
102 Spain). Primary secondary amine (PSA) was obtained from Supelco (Bellefonte).
103 Double distilled water was prepared for chromatographic use (Milli-Q, Millipore Corp.,
104 Bedford, MA). All chemicals used were analytical grade, and those used for
105 chromatographic analysis were HPLC grade. The acrylamide standard was purchased
106 from Merck (Germany), the stock solution (1mg mL^{-1}) was prepared by dissolving 100

107 mg of the acrylamide in 100 mL of acetonitrile, and was kept at 4°C until use. All
108 working solutions were prepared daily by serial dilution in acetonitrile.

109 Raw material

110 The potatoes (*Solanum tuberosum* L., Frisia variety) were provided by a local supplier.
111 This variety was chosen because of its availability during all the year in Spain. This
112 variety is characterized by yellow skin, light yellow-white flesh and oval shape. The
113 potatoes were sorted, washed, peeled, cut into 7 mm thick slices and cored with a
114 stainless steel core borer (25 mm in diameter) to produce discs. The experiments were
115 performed in two different periods of time: March-April and May-June being necessary
116 to analyze the total reducing sugars content of potatoes used in each period due to the
117 relevance of these compounds in acrylamide generation as well as color changes along
118 frying.

119 Experimental Methodology

120 *Pre-treatments*

121 The potato discs were subjected to various treatments prior to frying. Specifically,
122 samples were blanched at 85 °C for 5 minutes or immersed in solutions of different
123 chemical agents: citric acid, glycine, calcium lactate, sodium chloride or nicotinic acid
124 at 1 and 2% for 60 minutes at room temperature. Likewise, a control consisting of
125 dipping the samples in distilled water at room temperature for 60 minutes was also
126 performed. In all cases, the potato:solution ratio was 1:5 (w/w).

127 The potato discs were removed from the solution and the excess liquid was soaked up
128 with absorbent paper for two minutes prior to frying.

129 *Frying Step*

130 The experiments were carried at a fixed frying temperature of 180 °C in a commercial
131 deep oil fryer (model: FM 6720 Ideal 2000 Professional, Solac with a nominal power of

132 2,000 W) and hot air frying equipment (model: AH-9000 Actifry, Tefal with a nominal
133 power of 1,400 W). For deep oil experiments, a potato-to-oil ratio of 1:20 (w/v) was
134 used. Therefore, a total amount of 2 L of oil was employed to deep-oil fry 100 g of
135 potatoes. Commercial sunflower oilseed was used in all frying experiments. This ratio
136 was large enough to avoid important changes in terms of product-to-oil ratio and
137 therefore in the oil composition and temperature. The oil was renewed every 2
138 experiments. For hot air frying experiments, 0.003 kg of oil per kilogram of potatoes
139 was added to the air chamber according to the specifications of the equipment, i.e. a
140 total amount of 0.3 g of oil per 100 g of potatoes. A constant frying temperature was
141 confirmed by means of two PT-100 temperature sensors (model: TF101K) located at the
142 top and the bottom of each fryer. Samples were immersed in the fryers when the initial
143 frying temperature of 180 °C was achieved. Each experiment was conducted in
144 triplicate.

145 The concentration of reducing sugars was determined in raw potatoes in order to
146 evaluate its influence on color development. Color development was based on the
147 CIEL*a*b* colorimetric characteristics of fried potatoes. A reference frying time was
148 complementary established based on a visual evaluation of color on the surface and the
149 external (crispy outer crust) and internal cooking level of the samples. For this purpose,
150 samples were removed from the fryers for color determination at different time intervals
151 depending on the frying technology and initial reducing sugars content. Concurrently,
152 samples with low reducing sugar content were removed every 2 minutes until 16
153 minutes and at minute 26 for deep oil-frying conditions, and every 2 minutes until 24
154 minutes and at minute 34 for air frying. Samples with high reducing sugar content were
155 removed every minute until minute 7 for deep oil-frying, and every 2 minutes until
156 minute 20 for air frying. The sampling time was chosen according to a previous study

157 on mass and heat transfer phenomena for both frying technologies (Andrés and others
158 2012). Before the analytical determinations, the excess oil was removed from the
159 samples with absorbent paper.

160 Determination of the acrylamide contents was only carried out at the reference frying
161 time.

162 Analytical Determinations

163 All determinations were carried out in triplicate.

164 *Optical properties*

165 The determination of the optical properties of the potato disks in each experiment and
166 frying time was carried out on the surface of the samples by means of a
167 spectrophotometer (MINOLTA, mod. CM-3600d). The color space coordinates
168 CIEL*a*b* were obtained from the absorption spectrum between 380 and 770 nm by
169 reflectance with the reference system: D65 illuminant and 10 ° observer, and a 7 mm
170 lens. Previously, samples were measured with both black and white calibration tiles in
171 order to study the possible translucency of the samples. Since the same spectrum was
172 obtained with the black and white tiles, the opacity of the samples was confirmed, and
173 only data corresponding to the black tile were analyzed.

174 *Reducing sugars content*

175 The reducing sugars content was determined by spectrophotometry (model V-630;
176 Jasco Inc., Tokyo (Japan) according to the protocol described by Miller (1959). This
177 colorimetric method is based on the reduction of acid 3,5-dinitrosalicylic by reducing
178 sugars. 90 g of Rochelle salt (L (+) potassium sodium tartrate tetrahydrate) were added
179 to 210 mL of dinitrosalicylic acid solution (2.3 % of NaOH and 1.4 % dinitrosalicylic
180 acid (w/w) in distilled water) and made up to 300 mL. In order to carry out the
181 determinations, 0.3 g of homogenized potato sample was mixed with 1 mL of distilled

182 water and 2 mL of dinitrosalicylic acid solution. A blank consisting of 1mL of distilled
183 water and 2 mL of dinitrosalicylic acid solution was also prepared. After that, the mix
184 was heated for 5 minutes in boiling water and then cooled at room temperature for 10
185 minutes. Finally, 1 mL of the mix was diluted in 4 mL of distilled water and the
186 absorbance determined at 546 nm. The quantification of reducing sugars content (g/100
187 g of potatoes) was calculated using equation 1:

$$188 \text{ Reducing sugars (g/100 g of potatoes)} = (\text{Absorbance} - 0.00385) \times 1.07893 \quad \text{eq. (1)}$$

189 *Analysis of acrylamide*

190 The acrylamide content determinations were carried out using the method of dispersive
191 solid phase extraction called QuEChERS, originally designed for pesticide analysis in
192 food and modified by Mastovska and Lehotay (2006) for the extraction of acrylamide,
193 following the method described by Al-Tasher (2011) (Agilent Technologies, Note of
194 Application) with some modifications. Three potato disks were ground in a blender and
195 a sub-sample (2 g) of potato was placed in a 50mL Falcon tube and 5 mL of n-hexane
196 was added. The tube was shaken in a vortex for 30 seconds and after that 10 ml
197 bidistilled water, 10 mL acetonitrile, 4 g MgSO₄ and 0.5 g NaCl were added and stirred
198 in the vortex for one minute. The suspension was then centrifuged at 2026 RCF
199 (Centronic BL II (Selecta, Spain)) for 5 minutes, the hexane layer (upper phase) was
200 discarded and 1mL of the acetonitrile phase, containing the acrylamide, was transferred
201 to a 2 mL polypropylene tube containing 50 mg PSA and 150 mg MgSO₄, and stirred
202 for 30 seconds. The homogenate was centrifuged at 2697 RCF (Labofuge 200 (Heraeus,
203 Germany)) for 1 minute and the supernatant was transferred to a vial for analysis by
204 LC/MS/MS. The vials were amber type with a 1.5 mL of capacity.

205 The chromatographic analysis was performed with an Agilent 1200 Series HPLC
206 system coupled to an Agilent 6410 triple quadrupole mass spectrometer (Agilent

207 Technologies Inc., CA, USA) with an electrospray type ionization source. The column
208 used in this study was a Zorbax Eclipse XDB C-18 (2.1mmx50mm, 1.8 μ m). The mobile
209 phase used consisted of 2.5% methanol and 97.5% aqueous formic acid (0.1%) (A) and
210 methanol (B). The elution gradient was as follows: 0-4 min 100% of A; 4.1-6 min 70%
211 A; 6.1 min 100% A, with 2 min post-time to equilibrate the column and the total run
212 time was 8 min. The column oven temperature was set at 30°C, the flow was maintained
213 at 0.2 mL/minute, and the injection volume was 10 μ L; after each injection a needle
214 wash was performed.

215 The electrospray operated in positive ion mode. The conditions used in the ionization
216 source were: 350°C at 12 L/min for the drying gas (N₂), a nebulizer pressure of 40 psi
217 and a capillary voltage of 4000 V. Identification and quantification of acrylamide in the
218 samples was performed using the multiple reaction monitoring (MRM) mode, and the
219 two most abundant ions, m/z 72 > 27 and m/z 72 > 55.2, were studied respectively.

220 Before analysis, the chromatography method was validated according to the directive
221 2002/657/CE. The validation included the determination of linearity, recovery,
222 precision (repeatability and reproducibility), LOD (limit of detection) and LOQ (limit of
223 quantification). The linearity and matrix effect of the analytical procedure were studied
224 using calibration standards prepared in neat solvent and in fortified samples submitted
225 to QuEChERS extraction. The five-point-calibration curves (10, 20, 50, 100, and 200 μ g
226 kg⁻¹) in solvent and in fortified raw and fried potatoes were constructed and compared.
227 The slope ratios (slope matrix/slope solvent) were < 1 in all cases. This indicated that a
228 matrix effect existed with a suppression of ionization (Cuadros-Rodriguez and others
229 2003). Therefore, acrylamide content (μ g kg⁻¹) in fried potatoes was quantified from the
230 “fried matrix curve”; whereas acrylamide content (μ g kg⁻¹) in pre-treated samples was
231 quantified from the “raw matrix curve”. The presence of Maillard reaction products in

232 fried potato matrices leads to a different MS/MS response than in raw potatoes samples
233 in which no Maillard reaction products exist (Zhang and others 2011).

234 The accuracy of both the fried and raw matrix was evaluated through recovery
235 experiments by fortifying an appropriate sample at three different levels (10, 50 and 100
236 $\mu\text{g kg}^{-1}$), with six replicates at each level (n=6).

237 Repeatability was calculated from the analysis of six blank samples fortified at each one
238 of the three specified levels of fortification (10, 50 and 100 $\mu\text{g kg}^{-1}$), and performed by
239 the same operator on the same day. To evaluate reproducibility the analyses were
240 repeated on three consecutive days by two different operators.

241 The LOD and LOQ of the method were obtained by fortifying samples with acrylamide
242 at different concentrations. The values were determined as the amount of analyte for
243 which signal-to-noise ratios (S/N) were higher than 3 and 10, respectively.

244 2.3.4. Statistical analysis

245 Statistical analysis of variance (ANOVA) was performed by Statgraphics Centurion to
246 estimate the effect of process variables (reducing sugars content in raw material, type of
247 pre-treatment and frying technology) on the obtained results. Evaluations were based on
248 a 95% significance level.

249

250 **Results and Discussion**

251 *Influence of frying technology, pre-treatment and initial reducing sugars content on*
252 *color changes during frying*

253 It is well known that browning reactions induce changes in the crust of fried products.
254 The kinetics of Maillard reaction depends on several factors such as precursor
255 concentrations in raw material, pre-treatments, frying technology and frying variables
256 (De Wilde and others 2005). Additionally, a correlation between acrylamide

257 development and reducing sugars content in raw material has been reported in previous
258 studies (Williams 2005; De Wilde and others 2005).

259 Statistical analysis showed (Table 1) that frying technology, initial reducing sugars in
260 raw potatoes, and their interaction, had a significant influence on the evolution of
261 CIEL*a*b* parameters with frying time. Moreover, frying technology was the factor
262 which most influenced the evolution of color parameters during frying (higher value of
263 F-ratio), and especially on a*. Pre-treatment was not found to have a significant effect
264 which means that it is possible to apply pre-treatments without affecting the color of
265 fried potatoes. Figures 1, 2 and 3 show the evolution of lightness (L*), a* and b* with
266 regard to air and deep-oil frying time, of untreated, control and blanched samples as
267 well as after dipping them in 1 and 2% of citric acid. The authors do not consider it
268 necessary to show the results corresponding to all pre-treatments in Figures 1-3. On one
269 hand, the results showed that the higher the initial reducing sugars content, the faster the
270 rate of browning reactions. Color changes in potato products are the direct consequence
271 of non-enzymatic browning reactions which reducing sugars (i.e. glucose and fructose)
272 participate in (Manzoco and others 2011). On the other hand, frying technology clearly
273 affected the kinetic of color changes. In fact, Maillard reaction was much faster for
274 deep-frying than air-frying. This is due to differences between both technologies in
275 terms of mass and heat transport kinetics, even if the temperature of the external
276 medium (air or oil) was 180 °C in both cases. The thermal properties of oil favor heat
277 transport by convection from the external medium to the potato surface, and lead to a
278 faster frying process (Andrés and others 2012).

279 For oil-frying, lightness (L*) increased initially but tended to decrease with time
280 resulting in a darkening of the fried potatoes; while it almost remained constant in
281 samples submitted to air-frying (Figure 1). Browning was directly reflected in a gradual

282 increase of the a* color parameter. As it can be observed in Figure 2, a* values
283 gradually increased from negative values to positive ones. Finally, the CIE b* parameter
284 (related to yellow tint when positive) experimented a notable increase during frying,
285 regardless of frying technology (Figure 3). Samples submitted to oil-frying reached
286 higher b* values than air-fried potatoes and the rate of changes of b* and a* was faster
287 in fried potatoes with high initial reducing sugars content. The frying time required to
288 achieve positive values of a* in fried potatoes with low initial reducing sugars content
289 was 26 and 34 min for oil and air-frying conditions; while the time at which samples
290 with high initial reducing sugars content was 7 and 20 min in oil and air-frying,
291 respectively. The reference frying times, those at which a* values were lower than 0,
292 were 16 and 24 minutes in samples with low reducing sugars content for oil and air-
293 frying, respectively; whereas 5 and 16 minutes were required in samples with high
294 reducing sugars content for oil and air-frying, respectively. It is important to point out
295 that the reference time varied depending on each technology and the initial reducing
296 sugars content, but not on the type of additives used for pre-treatment.

297 *Validation of the analytical method for acrylamide determination*

298 As there is no official methodology described for the analysis of acrylamide, validation
299 of the analytical procedure was carried out as first step in order to ensure the quality of
300 the obtained results. The calibration curves for raw and fried samples were linear in the
301 range 10-200 $\mu\text{g kg}^{-1}$, with a correlation coefficient (R^2) of 0.9926 and 0.9986,
302 respectively. The recoveries, performed by adding known quantities of acrylamide were
303 in the range 97 to 108 % for raw matrix and 98 to 116 % for the fried one. The relative
304 standard deviation for recovery data ranged from 4.4 to 15.5 % and 4.8 to 16.0 %,
305 respectively.

306 Repeatability and reproducibility, expressed as the relative standard deviation, were 0.5-
307 3.5 and 0.5-12.8 % for the raw matrix, whereas for the fried matrix they were 2.6-15.6
308 and 4.1-14.6 %, respectively. All the relative standard deviations are in agreement with
309 the 2002/657EC Directive, which permits up to 20 %. The limit of quantification of the
310 method assayed, in both cases, was $10 \mu\text{g kg}^{-1}$.

311 The results of validation demonstrate that the method used is accurate and precise, and
312 therefore it can be concluded that this analytical procedure guarantees the quantification
313 of acrylamide in the samples.

314 *Influence of frying technology and pre-treatment on reducing sugars content and* 315 *acrylamide generation*

316 The effectiveness of pre-treatment in initial reducing sugars lixiviation as well as the
317 influence of chemical agents involved in the pre-treatments and the frying technology
318 on acrylamide reduction at reference frying time were also studied. This analysis was
319 only conducted in samples with high initial reducing sugars (0.203 ± 0.005 %), as this
320 condition is the most disadvantageous for acrylamide formation.

321 Table 2 shows the reducing sugars content of the samples after pre-treatment. The
322 results have been expressed as g of reducing sugars per 100 g of fresh potatoes in order
323 to compare the effect of pre-treatment. For this purpose, the net mass variation
324 occurring under the different experimental conditions (data not shown) was taken into
325 account. The relative variation of initial reducing sugars content (%) taken place during
326 pre-treatment was also calculated.

327 On one hand, pre-treatment by immersion in different chemical agents, as well as
328 blanching, resulted in a statistically significant loss of the initial reducing sugars of the
329 raw potatoes. It is important to point out that placing the samples in water alone for 60
330 min (control) resulted only in a 6% reduction, therefore it could be considered that no

331 reduction took place under control conditions; while the presence of a solute in the
332 external immersion medium produced reductions of over 22%, excepting in 1% of
333 NaCl. In fact, the presence of a solute in the immersion medium generates a mild
334 chemical potential gradient acting as a driving force for mass fluxes, leading to an
335 exchange of soluble solutes between the potato tissue and the surrounding solution
336 (Pointing 1973). Concretely, there is an intake of the chemical agent, from the external
337 solution to the potato tissue, at the same time as a partial removal of native hydrosoluble
338 solutes, such as reducing sugars and asparagine, from the liquid phase of the potato to
339 the external solution (Wicklund and others 2006). However, an increase of solute
340 concentration in the external solution did not significantly affect the variation of
341 reducing sugars (Table 2); whereas some chemical agents such citric acid, glycine or
342 calcium lactate and sodium chloride at 2% improved the reducing sugars lixiviation.
343 The results also demonstrated the advantage of applying short high-temperature
344 treatments (blanching) instead of long low-temperature ones (control). In this way,
345 blanching (5 min at 85 °C) resulted in a 26% loss of reducing sugars instead of 6% for
346 control conditions (60 min at room temperature). Pedreschi and others (2004) reported
347 similar levels of reduction after blanching potatoes of *Tivoli* var.; whereas Mestdagh
348 and others (2008a) reported lower values of reduction. During blanching, the alteration
349 of amylaceous tissue takes place resulting in a higher migration of acrylamide
350 precursors (Pedreschi and others 2005).

351 Previous studies reported that lixiviation of reducing sugars in raw materials and
352 acrylamide generation on frying are linked, since reducing sugars content is one of the
353 main limiting acrylamide precursors in potato products (Amrein and others 2003; De
354 Wilde and others, 2005). In this study, the roles of different chemical agents and frying
355 technologies in inhibiting acrylamide formation were also studied. The acrylamide

356 content ($\mu\text{g Kg}^{-1}$) of pretreated deep-oil fried and air-fried potatoes at their respective
357 reference frying times (5 min and 16 min in deep-oil frying and air-frying, respectively)
358 is shown in Figure 4. Air-frying resulted in a 77% reduction in acrylamide formation for
359 unpretreated samples in comparison to deep-oil fried ones. Therefore, the application of
360 a pre-treatment with the aim of reducing this toxic compound is not necessary for air-
361 frying. The effectiveness of air-frying in limiting acrylamide formation could be related
362 to the higher relative humidity of the product/external medium interphase for air-frying
363 compared to deep-oil frying. Moreover, Gökmen and Palazoglu (2009) reported that a
364 certain amount of acrylamide formed in potatoes during frying is lost through
365 evaporation. In fact, the higher fluxes of water from the core of the product to the
366 surface and finally to the air, linked to the long frying time during air-frying (Andrés
367 and others 2012), prolonged the constant-rate frying step characterized by the saturation
368 of the product surface with water. On the other hand, deep-oil frying mainly occurs at a
369 falling-rate. It is well known that the presence of water on the potato surface prevents
370 acrylamide formation in fried products.

371 Surprisingly, samples pre-treated with 1% of NaCl or Calcium lactate presented higher
372 acrylamide content than the control and the unpretreated air-fried samples. The presence
373 of any of these mono or divalent cations, Ca^{2+} or Na^{+} , at the product surface could
374 diminish the water activity of the surface and favor acrylamide formation in the fried
375 product. This result revealed a strong interaction between the frying technology and
376 type of pre-treatment as these cations reduced acrylamide formation for deep-oil frying.
377 In addition, pre-treatments with blanching, nicotinic acid, citric acid, glycine at 1% and
378 NaCl at 2% limited acrylamide generation in deep-oil fried potatoes. Zeng and others
379 (2009) previously reported that nicotinic acid was the most effective water soluble
380 vitamin in the inhibition of acrylamide with up to 70% mitigation. Moreover, since no

381 undesirable flavor was found in fried potatoes after treatment, nicotinic acid could be a
382 promising inhibitor of acrylamide formation in food processing. The effect of citric acid
383 (77 and 91% reduction at 1 and 2% of solution concentration, respectively) could be
384 attributed to both an important pH drop and a spatial hindrance which hinder the
385 reaction between acrylamide precursors (Mestdagh and others 2008b). The ability of
386 glycine to reduce the acrylamide content in fried potatoes compared to other amino
387 acids was previously reported in model systems (Bråthen and Knutsen, 2005; Low and
388 others, 2006) and blanched potato crisps (Kim and others 2005). Unexpectedly, a
389 reduction of 80 and 55% was found in this study at 1 and 2% of glycine, respectively.
390 Alternatively, concentrations above 1% of glycine in the soaking medium, i.e. in the
391 pretreated potato, did not result in a higher acrylamide reduction. The effectiveness of
392 glycine, and other free amino acids or proteins, was attributed to the promotion of
393 competitive reactions with asparagine to react with reducing sugars and/or by covalently
394 binding the formed acrylamide through Michael type addition reactions (Mestdagh and
395 others 2008). The effect of mono and divalent cations Na^+ and Ca^{2+} at 2% gave as a
396 result significant acrylamide mitigation. A loss of 78 and 68% took place at this
397 concentration with NaCl and calcium lactate, respectively; whereas a reduction of $\approx 20\%$
398 occurred at 1% with both cations. Mestdagh and others (2008) reported a maximum
399 reduction of 10% and 49% by adding 50 $\mu\text{mol/g}$ mixture of NaCl and 100 $\mu\text{mol/g}$
400 mixture of calcium lactate to a model system. However, Gökmen and Senyuva (2007)
401 found a complete prevention of acrylamide formation by Ca^{2+} , whereas monovalent
402 cations, such as Na^+ , almost halved the acrylamide formed in the model system.
403 Calcium lactate and NaCl, as food additives, are widely used as firming and
404 preservative agents in commercial foods and they could be also used by the food
405 industry to control the formation of acrylamide.

406 Finally, a correlation between reducing sugars content at the beginning of frying (Table
407 2) and acrylamide content (Figure 4) at the reference frying time was separately
408 performed for both frying technologies (data not shown). Results showed that reducing
409 sugars content at initial frying time had lack of correlation with acrylamide content
410 ($R^2=0.648$ for deep-oil frying and 0.252 for air-frying). This fact again reflected that
411 acrylamide inhibition is a complex mechanism in which not only the amount of the
412 precursors (reducing sugars and asparagine) has a major role but also the interference of
413 additives in Maillard reaction and the kinetics and conditions of frying.

414

415 **Conclusions**

416 From the results obtained in this study, it could be concluded that the lower the initial
417 reducing sugars content the higher the time required for reaching the typical tonality of
418 fried potatoes. The rate of Maillard reaction during air-frying technology was much
419 lower than for deep oil-frying conditions leading to a drastically acrylamide reduction
420 (about $\approx 90\%$) with compared with conventional frying even in unpretreated samples.
421 This fact permits to affirm that air-frying technology is a promising technology for
422 obtaining healthy fried products.

423 The application of a pre-treatment became an important step for acrylamide mitigation
424 for deep-oil frying. In this sense, both the extraction efficiency of reducing sugars and
425 the penetration of additives during pre-treatment played a combining role for
426 acrylamide mitigation. Concretely, dipping the potatoes into a solution of nicotinic acid,
427 citric acid, glycine at 1% or NaCl at 2% might be a viable approach for the
428 minimization of acrylamide content. Nevertheless, the sensory repercussion of any
429 strategy to reduce acrylamide generation should be evaluated before application.

430

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434

435 **Author Contributions**

436 M. Sansano collected experimental data and drafted the manuscript

437 M. Juan-Borrás and I. Escriche performed method validation for LC/MS/MS
438 acrylamide analysis

439 A. Andrés planned the study and interprets the results

440 A. Heredia planned the study, interprets the results and draft the manuscript

441

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Table 1. Values of F-ratio and level of significance from ANOVA multifactor of the influence of the main effects (pre-treatment, frying technique and initial reducing sugars) on the CIEL*a*b* changes along frying.

Main effects	F-ratio		
	L*	a*	b*
A: Pre-treatment	1.77 NS	1.60NS	1.76NS
B: Frying Technique	19.61**	50.09**	192.47**
C: Initial Reducing sugars	4.29*	7.17**	81.36**
Interactions			
AB	1.42NS	1.64NS	3.99**
BC	3.80NS	10.45**	35.32**

** Statistical significance $\geq 99\%$ (p-value ≤ 0.01); * Statistical significance $\geq 95\%$ (p-value ≤ 0.05); NS (not statistical significance, p-value > 0.05)

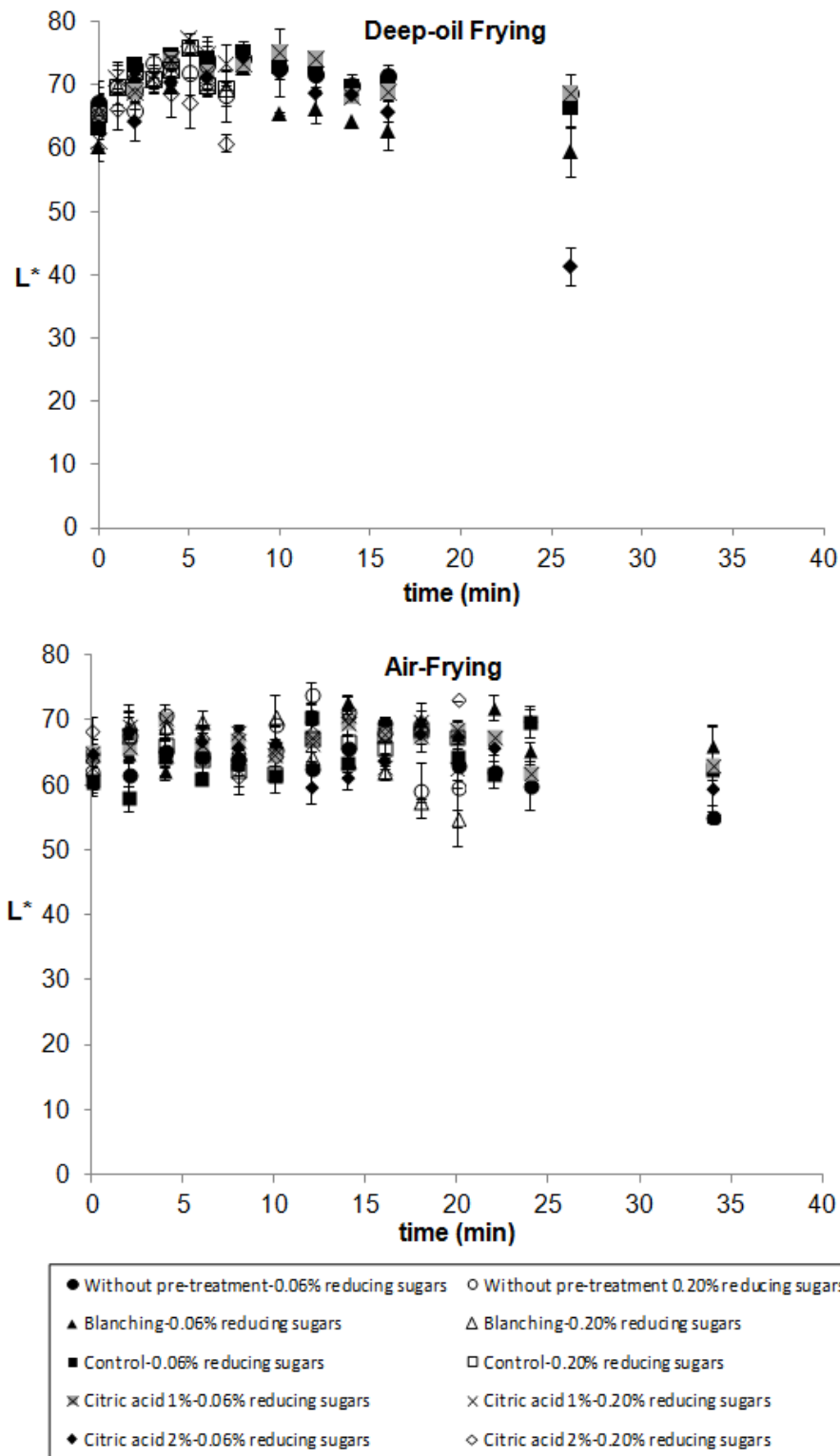
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Table 2. Reducing sugar content (g/ 100 g fresh potato) of the samples after pre-treatments and variation of initial reducing sugars (%) as a consequence of pre-treatments.

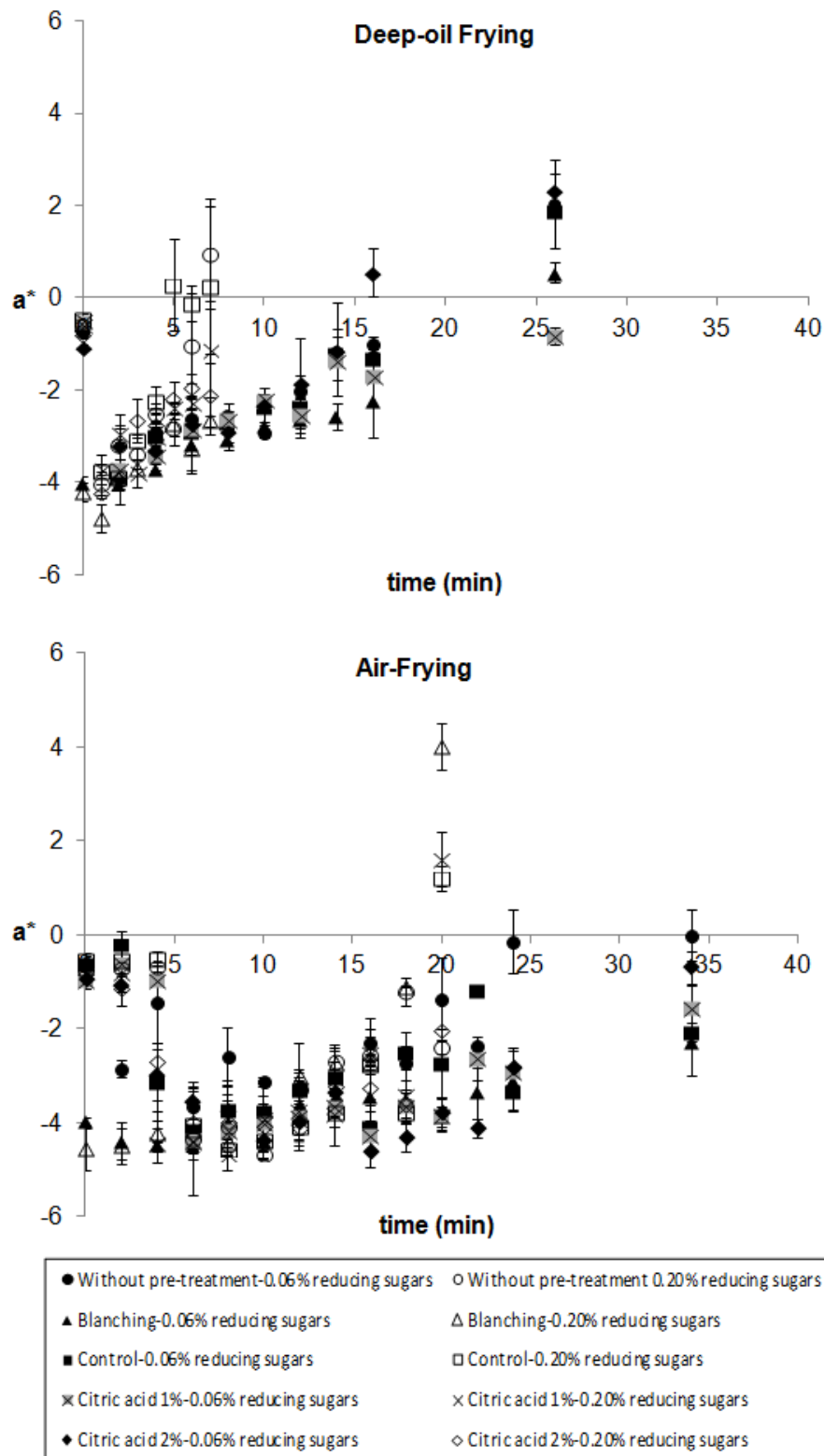
Pre-treatment	Reducing sugars content (g/100g fresh potato)	Variation of initial Reducing sugars (%)
Raw potatoes	0.203(0.005)d	
Control	0.20(0.03)d	-6.54(1.03)
Blanching	0.15(0.03)bc	-26(5)
Nicotinic acid 1%	0.146(0.016)bc	-31(6)
Nicotinic acid 2%	0.156(0.009)bc	-23(5)
Citric acid 1%	0.133(0.013)abc	-35(6)
Citric acid 2%	0.1237(0.0112)abc	-39(6)
Glycine 1%	0.13(0.02)bc	-39(5)
Glycine 2%	0.118(0.013)ab	-42(6)
NaCl 1%	0.21(0.02)d	1.3(1.4)
NaCl 2%	0.097(0.015)a	-55(5)
Calcium lactate 1%	0.158(0.006)c	-22(3)
Calcium lactate 2%	0.130(0.005)abc	-36(2)

^{abcd} different letters indicate differences between homogeneous groups at a 95% of significance level (p-value ≤ 0.05)
Mean (standard deviation)



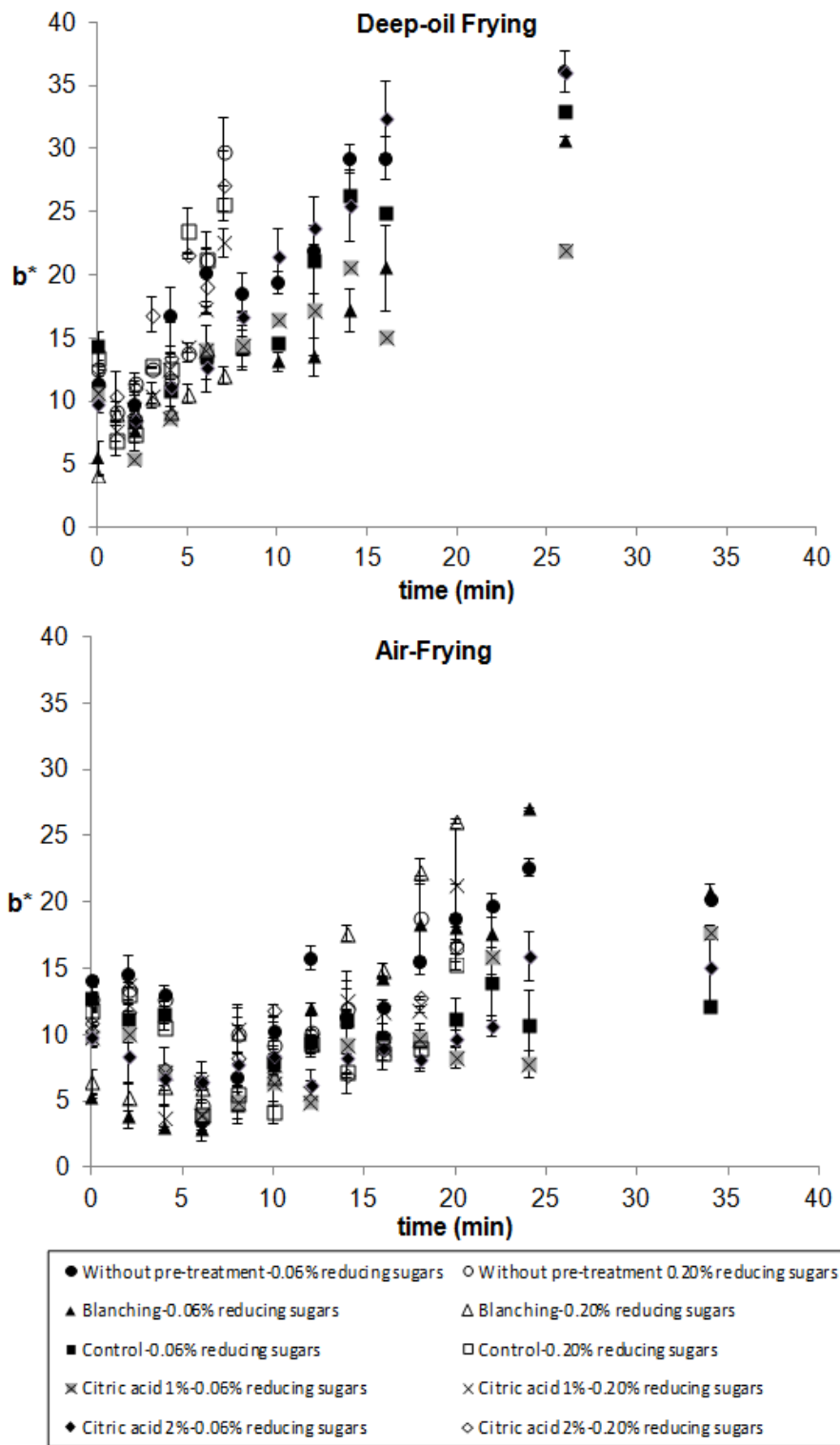
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586 **Figure 1.** Evolution of lightness (L*) along deep-oil and air-frying of unpretreated
 587 (without any pre-treatment), control (60 min at room temperature in distilled water),
 588 blanched and dipped in citric acid at 1 and 2% samples with low and high-medium
 589 initial reducing sugars content in raw potatoes.



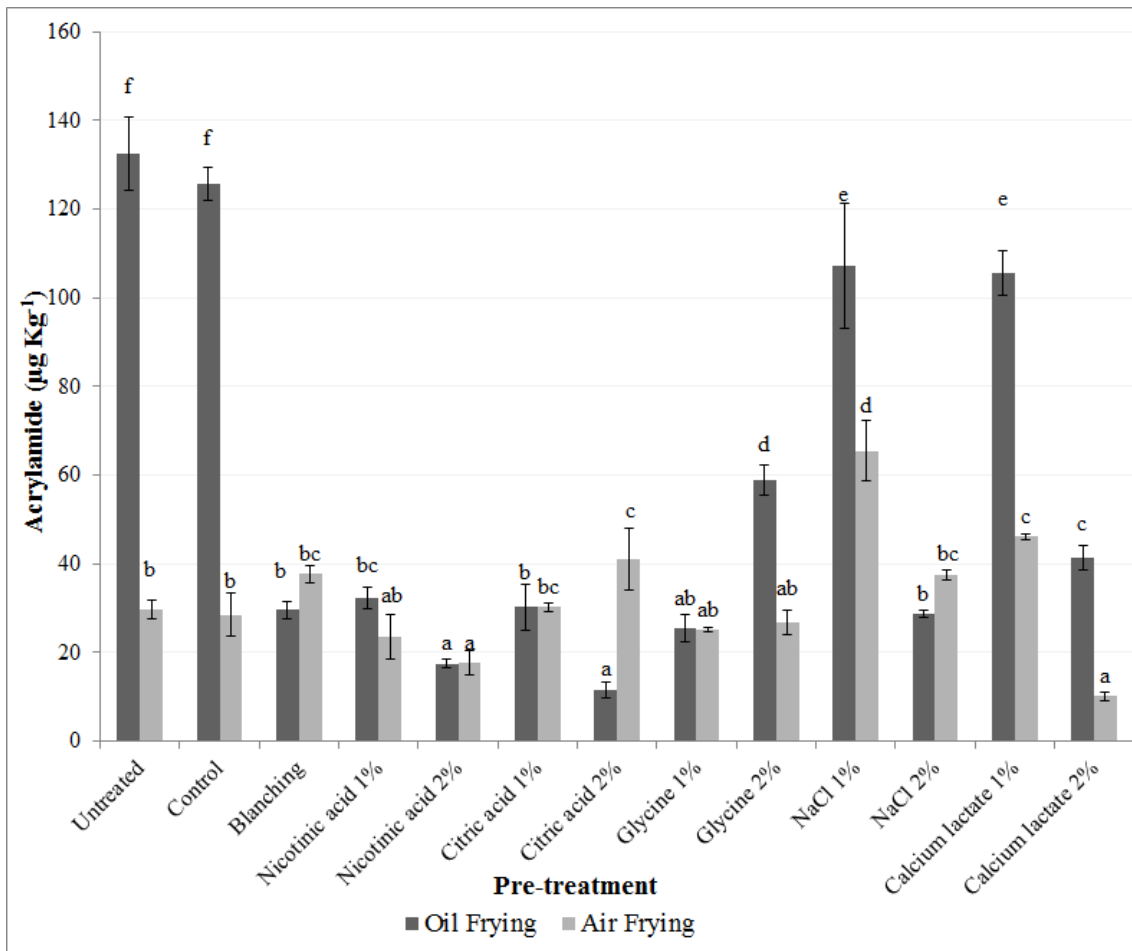
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591 **Figure 2.** Evolution of a^* colorimetric parameter along deep-oil and air-frying of
 592 unpretreated (without any pre-treatment), control (60 min at room temperature in
 593 distilled water), blanched and dipped in citric acid at 1 and 2% samples with low and
 594 medium-high initial reducing sugars content in raw potatoes.



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596 **Figure 3.** Evolution of b* colorimetric parameter along deep-oil and air-frying of
 597 unpretreated (without any pre-treatment), control (60 min at room temperature in
 598 distilled water), blanched and dipped in citric acid at 1 and 2% samples with low and
 599 medium-high initial reducing sugars content in raw potatoes.



601

602 **Figure 4.** Acrylamide content ($\mu\text{g kg}^{-1}$) of fried potatoes submitted to the different
 603 studied pre-treatments and at the reference frying time of deep-oil frying and air-frying.

604 These values correspond to fried potatoes with medium-high initial reducing sugars

605 content in raw potatoes.