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Ozuna López, C.; Puig Gómez, CA.; García Pérez, JV.; Carcel Carrión, JA. (2014). Ultrasonically enhanced desalting of cod (*Gadus morhua*). Mass transport kinetics and structural changes. *Food Science and Technology*. 59:130-137. doi:10.1016/j.lwt.2014.05.062.



The final publication is available at

<https://dx.doi.org/10.1016/j.lwt.2014.05.062>

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Additional Information

1 **Ultrasonically Enhanced Desalting of Cod (*Gadus morhua*). Mass**  
2 **Transport Kinetics and Structural Changes**

3  
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27 **ABSTRACT**

28 The search for an alternative means of reconstituting dried and salted products prior to  
29 consumption is of relevance for the food industry. New techniques should speed up the  
30 process while causing minimum impact on product quality. Thereby, the aim of this  
31 work was to evaluate both the effect of high-intensity ultrasound application on the  
32 desalting kinetics of cod, as well as the changes in its textural and microstructural  
33 properties. Moisture and NaCl transport was studied separately by taking the diffusion  
34 theory into account. The evolution in the swelling and hardness of cod during desalting  
35 was determined and modeled by assuming first-order kinetics. A microstructural  
36 analysis of raw salted and desalted cod was also carried out by means of light  
37 microscopy and SEM techniques. Ultrasound application significantly ( $p < 0.05$ ) affected  
38 both moisture and NaCl transport and the increase in both effective diffusivities (from  
39 24% to 103%) was linked to the acoustic pressure applied. The desalting process  
40 induced the swelling and the softening of the cod tissue, both of which are phenomena  
41 that are intensified by ultrasound application. From the microstructural observations, it  
42 was shown that the application of high-intensity ultrasound modified the cod structure,  
43 e.g. the increase in the fiber width.

44

45 *Keywords:* Ultrasound, Modeling, Diffusivity, Texture, Microscopy.

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## 48        **1. Introduction**

49        Salted cod (*Gadus morhua*) is one of the most widely-consumed heavy-salted products  
50        in southern European countries and Latin America (Martínez-Alvarez & Gómez-  
51        Guillén, 2013; Oliveira, Pedro, Nunes, Costa, & Vaz-Pires, 2012). Due to the  
52        unpalatably high salt concentration in the fish muscle, cod must be desalted before  
53        consumption. This process is traditionally carried out by immersing the product in  
54        stagnant water over a period of 24–48 h. Thus, desalting is accomplished not only  
55        through the loss of salt but also as a result of sample rehydration (Barat, Rodríguez-  
56        Barona, Andrés & Visquert, 2004). The combined effect of water gain and salt loss on  
57        the protein matrix (Thorarinsdottir et al., 2011) modifies the structure of product.  
58        Protein rehydration involves a loss of firmness (Barat et al., 2004). Moreover, the  
59        reduction in NaCl content improves the water holding capacity, which, in turn, increases  
60        the water absorption, thus contributing to the total weight gain (Oliveira et al., 2012).  
61        The main problems of industry-scale cod desalting are linked to the final product quality  
62        and long processing times. For this reason, recent research has focused on finding new  
63        cod desalting methods in order to improve the mass transfer process, like tumbling  
64        technology (Bjørkevoll, Olsen, & Olsen, 2004), vacuum pulses (Andrés, Rodríguez-  
65        Barona, and Barat, 2005) or high pressures (Salvador, Saraiva, Fidalgo and Delgadillo,  
66        2013).  
67        High-intensity ultrasound (US) is being used as a novel food process intensification  
68        technique (Chemat, Zill-e-Huma, & Khan, 2011; Pananun, Montalbo-Lomboy,  
69        Noomhorm, Grewell, & Lamsal, 2012). In liquid media, US enhances mass transfer  
70        mainly by inducing cavitation (Ozuna, Puig, García-Pérez, Mulet, & Cárcel, 2013). In  
71        addition, some other mechanical phenomena, such as the “sponge effect” (Cárcel,

72 Benedito, Rosselló, & Mulet, 2007a), the generation of microchannels in the solid or  
73 microstirring at the solid-liquid interfaces, could also affect both the external and  
74 internal mass transfer resistance (Fernandes, Gallão, & Rodrigues, 2008). Moreover, the  
75 mechanical stress that US causes in the product may modify both structural and textural  
76 properties (Gabaldón-Leyva et al., 2007; Stadnik, Dolatowski, & Baranowska, 2008).  
77 The effectiveness of applying US is directly linked to the actual acoustic energy  
78 introduced in the medium (Kulkarni & Rathod, 2014). This fact mainly depends on the  
79 emitter-liquid-product coupling, which should be experimentally determined in each  
80 specific application (Cárcel, Benedito, Bon, & Mulet, 2007b).  
81 US has been used to enhance mass transport in the treatments of solids in hypertonic  
82 solutions, such as meat or cheese brining (Ozuna et al., 2013; Siró et al., 2009; Cárcel et  
83 al., 2007b; Sanchez et al., 1999). However, as far as we are concerned, there is no  
84 previous literature on the use of US to improve the desalting of foodstuffs. Therefore,  
85 the aim of this work was to evaluate the feasibility of using US in cod desalting from  
86 the study of the mass transport kinetics, texture evolution and microstructure of desalted  
87 samples.

## 88 **2. Materials and Methods**

### 89 **2.1 Raw material and sample preparation**

90 Salted cod (*Gadus morhua*) pieces (1.50±0.25 kg) were provided by a local supplier  
91 (Carmen Cambra S. L., Spain) to ensure the homogeneity of the raw material.  
92 Parallelepiped-shaped samples (length 50 x width 30 x thickness 5 mm) were obtained  
93 from the central part of the cod loin using a sharp knife, wrapped in plastic waterproof  
94 film and kept refrigerated at 2±0.5 °C (maximum storage time 120 h) until the desalting  
95 experiments.

## 96 2.2 Ultrasonic equipment

97 The desalting treatments were carried out in an ultrasonic bath (71 L, 40 kHz; ATU  
98 Ultrasonidos, Spain), equipped with a cooling jacket (Fig. 1F). The equipment allows  
99 the applied ultrasonic power to be modulated (up to 1500 W) (Fig. 1L). For the  
100 temperature control, a glycol solution was pumped (1-38023 CLES, EBARA, Italy)  
101 (Fig. 1D) from a reservoir tank (Fig. 1E) into the cooling jacket. Afterwards, the glycol  
102 solution was cooled down in a plate heat exchanger connected to a chiller unit  
103 (Fig. 1A). The temperature in the bath was measured using seven type K thermocouples  
104 (Fig. 1H), placed in different positions and connected to a data logger (HP Data Logger  
105 34970 A, Hewlett-Packard S.A., Spain) (Fig. 1K).

## 106 2.3 Characterization of the acoustic field

107 Two different techniques were used to estimate the actual ultrasonic energy applied:  
108 calorimetry and acoustic pressure determination. The measurements were carried out at  
109 a depth of 100 mm from the water-air interface, where samples were placed during  
110 treatment. This depth corresponded to a maximum pressure plane detected through the  
111 erosion planes produced by cavitation on a piece of aluminum foil.

112 The calorimetric method consisted of recording the temperature at 7 different points  
113 (Fig. 1H) for the first 1 min of US application (Cárcel et al., 2007b). At least five  
114 replicates were carried out for each measurement. The slope of the temperature vs. time  
115 curve gave an estimation of the average ultrasonic power in the medium.

116 The acoustic pressure measurement was carried out with a hydrophone (TC4013, Reson  
117 A/S, Denmark) connected to a digital oscilloscope (Tektronix TDS 420 A, Tektronix  
118 Inc., USA). The output voltage level (average of 100 signal acquisitions) was converted  
119 into the equivalent acoustic pressure (bar) by using the known sensitivity of the

120 hydrophone. For the measurement, 2D scans were performed by moving the  
121 hydrophone in 30 mm increments in both X and Y-directions. In such a way, the surface  
122 of the ultrasonic bath (length 600 and width 300 mm) was mapped by measuring the  
123 acoustic pressure in a rectangular mesh of 171 nodes.

#### 124 2.4 Desalting experiments

125 Before each desalting experiment, samples were immersed in distilled water for 20 s to  
126 remove the superficially adhered salt. Then samples were blotted and weighed  
127 (PB3002-S/PH, J.P., Mettler Toledo, Spain). Afterwards, 28 samples were placed in a  
128 sample holder (Fig. 1G) and introduced into the ultrasonic bath containing 27 L of low-  
129 mineral content water (Cortes, S.A., Spain). Three sets of desalting experiments were  
130 carried out: without (CONTROL) and with US application at two different electric  
131 powers: 1500 W (US-1500) and 750 W (US-750). At least 3 replicates were made for  
132 each treatment tested.

133 At preset times (15, 30, 45, 60, 90, 120 and 180 minutes), four samples were randomly  
134 taken out of the bath, blotted, wrapped in plastic waterproof film and frozen ( $-18 \pm$   
135  $0.5$  °C) until moisture (method No. 950.46 AOAC, 1997) and NaCl content (Cárcel et  
136 al., 2007b) were determined; this was carried out in triplicate in each case.

137 The thickness of each sample was measured before ( $T_0$ ) and after ( $T$ ) desalting with a  
138 Vernier caliper and the thickness ratio (TR) (Eq. 1) was estimated as an index of  
139 swelling.

$$140 \quad TR = \frac{T}{T_0} \quad (1)$$

141 Hardness (H), characterized as the maximum penetration force, was also evaluated in  
142 desalted cod samples at different times using a Texture Analyzer (TAX-T2<sup>®</sup>, Stable

143 Micro System, United Kingdom). Penetration tests were conducted with a 2 mm flat  
 144 cylinder probe (SMS P/2N), at a crosshead speed of 1 mm/s and a strain of 70%  
 145 (penetration distance 3.5 mm). In each cod sample, a preset pattern was followed in  
 146 order to carry out penetration tests at, at least, 12 different points.

## 147 2.5 Mathematical modeling

### 148 2.5.1 Moisture and NaCl transport

149 A simple mathematical model based on Fick's 2nd law (Crank, 1975) was used as a first  
 150 approach (Carcel et al., 2007a; Garau, Simal, Femenia &, Rosselló, 2006) to describe  
 151 the evolution of moisture and NaCl content in the samples during desalting. Mass  
 152 transport was considered to be one-dimensional due to the fact that thickness (5 mm)  
 153 was 1/6 (30 mm) and 1/10 (50 mm) shorter than the other dimensions. The mass flow in  
 154 the thickness direction being much larger compared to the other ones due to the relative  
 155 mass transfer resistances (Singh & Heldman, 2001). Thus, approaching the slices as  
 156 infinite slabs sounds reasonable (Garau et al. 2006). Constant effective diffusivities  
 157 ( $D_{NaCl}$  and  $D_w$ ), negligible changes in temperature and sample volume, solid symmetry,  
 158 homogeneous initial NaCl and moisture contents and negligible external resistance were  
 159 assumed. Eqs. 2 and 3 show the solution of the diffusion model in terms of the average  
 160 moisture and NaCl content, respectively (Ozuna et al., 2013).

$$161 \quad W = W_{eq} + (W_0 - W_{eq}) \left[ 2 \sum_{n=0}^{\infty} \frac{1}{\lambda_n^2 L^2} e^{-D_w \lambda_n^2 t} \right] \quad (2)$$

$$162 \quad NaCl = NaCl_{eq} + (NaCl_0 - NaCl_{eq}) \left[ 2 \sum_{n=0}^{\infty} \frac{1}{\lambda_n^2 L^2} e^{-D_{NaCl} \lambda_n^2 t} \right] \quad (3)$$



163 where,  $\lambda_n$  are the eigenvalues that are calculated by  $\lambda_n L = (2n+1)\frac{\pi}{2}$ . In these  
 164 conditions and from a theoretical point of view, the effective diffusivity was considered  
 165 to be the only kinetic parameter that involved all the effects which influenced the  
 166 moisture or NaCl transport rate. However, in practice the effective diffusivity identified  
 167 by fitting the model to the experimental data included all unknown mechanisms that  
 168 affected the mass transfer kinetics (external resistance, sample swelling, etc.).

### 169 2.5.2 Evolution of hardness and swelling

170 First-order reaction models, widely used to describe the kinetics of structural changes in  
 171 processed foods (Blasco, Esteve, Frígola, & Rodrigo, 2004; Baik & Mittal, 2003), were  
 172 considered to study the evolution of the thickness ratio (Eq. 4) and hardness (Eq. 5)  
 173 during desalting.

$$174 \quad TR = TR_{eq} + (TR_0 - TR_{eq})e^{(-k_{TR} \cdot t)} \quad (4)$$

$$175 \quad H = H_{eq} + (H_0 - H_{eq})e^{(-k_H \cdot t)} \quad (5)$$

### 176 2.6 Model fitting

177 The parametric identification of the first order kinetics ( $k_{TR}$ ,  $TR_{eq}$ ,  $k_H$  and  $H_{eq}$ ) and the  
 178 diffusion models ( $D_{NaCl}$ ,  $D_W$ ) was carried out by using an optimization procedure  
 179 (Generalized Reduced Gradient) that minimized the sum of the squared differences  
 180 between the experimental and calculated data. The goodness of the fit was determined  
 181 by assessing the percentage of explained variance, VAR (%) (Eq. 6).

$$182 \quad VAR(\%) = \left[ 1 - \frac{S_{xy}^2}{S_y^2} \right] \cdot 100 \quad (6)$$

### 183 2.7 Microstructure

184 2.7.1 Scanning electron microscopy with combined dispersion X-ray analysis  
185 (SEM-EDX).

186 Cubic samples (side 3 mm) of salted and desalted cod (CONTROL, US-750 and US-  
187 1500 after 180 min of desalting) were immersed in liquid nitrogen and then freeze-dried  
188 at 1 Pa for 3 days (LIOALFA-6, Telstar, Spain). Then the samples were vacuum sealed  
189 in vials in the same freeze-drier so that they would remain stable (Hernando, Llorca,  
190 Puig, & Lluch, 2011). After that, they were individually placed on SEM slides with the  
191 aid of colloidal silver and then gold-coated with carbon (SCD005, Baltec, Germany) at  
192  $10^{-2}$  Pa and an ionization current of 40 mA. The samples were observed through a  
193 scanning electron microscope (JSM-5410, Jeol, Japan) equipped with an X-ray detector  
194 and LINK data-processing system (INCA 4.09, Oxford Instruments, England) at an  
195 acceleration voltage of 10-20 kV, which provides internal information about the  
196 standards of the energy dispersive X-ray spectra of the elements analyzed,  $\text{Na}^+$  and  $\text{Cl}^-$   
197 (Ozuna et al., 2013). For the EDX (energy-dispersive X-ray) analysis, the samples were  
198 carbon-coated (CEA035, Baltec, Germany). Mapping images of the  $\text{Cl}^-$  and  $\text{Na}^+$   
199 distribution in cod samples were taken using a voltage of 20 kV and at a working  
200 distance of 15 mm.

201 2.7.2 Light microscopy (LM)

202 For LM observation, cryostat sections (200  $\mu\text{m}$ ) were obtained from frozen desalted cod  
203 (CONTROL, US-750 and US-1500 after 180 min of desalting) using a CM1950  
204 microtome (Leica Biosystems, Germany). The sections were transferred to coated glass  
205 slides, which had previously been placed inside the cryo chamber to achieve an  
206 improved adherence of the tissue sections (Thorarinsdottir et al., 2011). The cryostat  
207 section samples were examined under a light microscope (Nikon Eclipse E800, Japan).

208 The fiber thickness was measured from micrographs obtained by both light microscopy  
209 and SEM-EDX using the ImageJ 1.44d software (Wayne Rasband, National Institute of  
210 Health, USA). All the measurements were assessed from at least six randomly acquired  
211 images.

### 212 **3. Results and Discussion**

#### 213 3.1 Acoustic field characterization

214 The acoustic field characterization was carried out at the two different levels of electric  
215 power used in the ultrasonically assisted desalting experiments: 750 W (US-750) and  
216 1500 W (US-1500).

217 From the calorimetric measurements, it could be observed that the average volumetric  
218 energy available in the medium was  $19\pm 5$  kW/m<sup>3</sup> and  $37.9\pm 5.2$  kW/m<sup>3</sup> for US-750 and  
219 US-1500, respectively. The calorimetric/electric yields were close to 70%.

220 The acoustic pressure measurements showed a very irregular acoustic field distribution  
221 in the bath (Figs. 2A and 2B): the acoustic field pattern for US-750 was different from  
222 that for US-1500. The reflections of ultrasonic waves at the moving air-water interface  
223 or the bath walls and the implosions of cavitation bubbles (Kulkarni & Rathod, 2014)  
224 can generate this heterogeneous ultrasonic field. As regards the integrated average  
225 acoustic pressure, it was  $0.5\pm 0.1$  bar (Fig. 2B; range 0.3-0.8 bar) for US-1500, but  
226  $0.4\pm 0.1$  bar (Fig. 2A; range 0.1-0.7 bar) for US-750.

227 As can be observed, doubling the electric energy supplied to the bath transducers  
228 doubles the energy measured by calorimetry but not that measured using the  
229 hydrophone.

230 The calorimetric method measured the thermal effects produced by US, while acoustic  
231 pressure is a measurement of only mechanical effects. The interaction between waves

232 produced by reflections or cavitation can partially neutralize them producing a  
233 conversion of acoustic energy into heat. Therefore, acoustic pressure is a more accurate  
234 measurement of the ultrasonic mechanical energy available to produce effects.

## 235 3.2 Mass transport

### 236 3.2.1 Experimental NaCl and water content

237 Salted cod loin showed an initial NaCl content of  $0.6\pm 0.2$  kg NaCl/kg SFS (salt-free  
238 solids). The immersion of samples in water produced a decrease in NaCl content, which  
239 was greater when US was applied (Fig. 3A). Thus, after 180 min of desalting, the NaCl  
240 content in US-1500 ( $0.114\pm 0.007$  kg NaCl/kg SFS) was 32% lower than that measured  
241 in CONTROL samples ( $0.17\pm 0.03$  kg NaCl/kg SFS). US-750 showed an intermediate  
242 NaCl content ( $0.13\pm 0.04$  kg NaCl/kg SFS). Therefore, the influence of US depended on  
243 the acoustic power applied (Fig. 3A).

244 As regards the moisture content, the cod loin's initial content of  $1.9\pm 0.2$  kg water/kg  
245 SFS increased during the experiments. The US application also significantly ( $p<0.05$ )  
246 accelerated the water transport (Fig. 3B). For example, the moisture content of  
247 CONTROL after 180 min of treatment ( $3.0\pm 0.4$  kg water/kg SFS) was achieved for US-  
248 1500 in only 63 min (65% reduction in treatment time). As in the case of NaCl  
249 transport, the water gain was dependent on the acoustic power applied.

### 250 3.2.2 Modeling transport kinetics

251 The fit of the proposed diffusion models to the experimental data provided percentages  
252 of explained variance ranging from 93 to 95% for NaCl and 95 to 97% for moisture  
253 kinetics (Table 1). These low values can be attributed to the great variability of the raw  
254 material (Oliveira et al., 2012; Barat et al., 2004). However, the trend between

255 calculated and experimental data was quite similar (Fig. 3), showing the feasibility of  
256 the model.

257 The  $D_{\text{NaCl}}$  values identified for CONTROL ( $4.57 \times 10^{-10} \text{ m}^2/\text{s}$ ) were in the same order of  
258 magnitude as others found in literature (Barat et al., 2006 and 2004) for cod desalting.

259 The application of US during desalting produced a significant ( $p < 0.05$ ) increase in  
260  $D_{\text{NaCl}}$ , which depended on the power applied. Thus,  $D_{\text{NaCl}}$  was 25% higher for US-750  
261 and 62% higher for US-1500 than the figure identified for CONTROL experiments

262 (Table 1). The relationship between the applied acoustic energy and the identified  $D_{\text{NaCl}}$   
263 was better described from acoustic pressure measurements than from calorimetric.

264 Between the US-750 and US-1500 experiments there was observed to be a 30%  
265 increase in acoustic pressure (from 0.39 bar to 0.51 bar, respectively) and this provided

266 a close fit to the 30% increase in the identified  $D_{\text{NaCl}}$  (Table 1), while using calorimetry,  
267 an effective diffusivity increase of 103% would be expected. During the desalting

268 experiments, the temperature was held at 4 °C and, as a result, only ultrasonic  
269 mechanical effects on mass transport took place, which are well quantified from the

270 acoustic pressure measurement. Calorimetric measurements, on the other hand, are  
271 more suitable for quantifying the thermal effects produced by ultrasound.

272 In the case of moisture transport, the increase in  $D_w$  brought about by US application  
273 was close to 41% for US-750 and 103% for US-1500 as compared to CONTROL

274 experiments. As in the case of  $D_{\text{NaCl}}$ , the increase in  $D_w$  was also well correlated with  
275 the increase in acoustic pressure.

276 The increase in solute and moisture transport when applying ultrasound has been  
277 previously described in other solid-liquid systems. Thus, Siró et al. (2009) found

278 increases of 96% in  $D_{\text{NaCl}}$  when US was applied during meat brining (5 °C, 4% NaCl),

279 and Gabaldón-Leyva et al. (2007) observed an improvement of 190% in the total solid  
280 diffusion coefficients during the brining of bell pepper (55 °C, 13.51% NaCl). Ozuna et  
281 al. (2013) reported increases in  $D_w$  of around 101% when US was applied in meat  
282 brining (5 °C, 28% NaCl) and Cárcel et al. (2007a) found increases of 117% in moisture  
283 transport during the osmotic dehydration of apple (30 °C, 30 °Brix).

284 It should be pointed out that, even the proposed theoretical model considers the  
285 moisture and NaCl transport only controlled by diffusion, the effective diffusivities  
286 ( $D_{NaCl}$  and  $D_w$ ) identified are just kinetic parameters, which also include other  
287 phenomena affecting the mass transport, such as external convective flow or volume  
288 sample changes (Mulet, 1994) like swelling. The influence of US on mass transport  
289 comes from the appearance of effects such as the so-called "sponge effect" or the  
290 generation of microchannels (Cárcel et al., 2010) that reduce the internal resistance and  
291 the generation of microstirring or the implosion of cavitation bubbles at interfaces that  
292 enhance the external mass transport (Chemat et al., 2011). As previously explained,  
293 temperature control avoids any improvement linked to thermal energy.

### 294 3.3 Evolution of swelling

295 The application of US during desalting increased the swelling, measured from the TR  
296 (Eq. 1), and this increase depended on the level of acoustic power applied (Fig. 4A).  
297 Thus, after 180 min desalting, while US-750 samples showed a TR value ( $2.8 \pm 0.1$ )  
298 which was 22% higher than that of CONTROL ( $2.3 \pm 0.4$ ), in the case of US-1500 there  
299 was a 50% increase ( $3.4 \pm 0.3$ ).

300 As for the experimental variability, the first order kinetic (Eq. 4) provided an adequate  
301 estimation of the TR (Table 2). The rate constant ( $k_{TR}$ ), which is related with the rate of  
302 swelling during desalting, was not significantly ( $p < 0.05$ ) increased by US application

303 (Table 2). As can be observed in Fig. 4B, there was no direct kinetic effect on the  
304 swelling itself produced by US, but the influence observed on the experimentally  
305 measured swelling can be simply linked to the fact that US application intensified water  
306 transport.

### 307 3.4 Evolution of hardness

308 The measurements of the maximum penetration force showed that the initial degree of  
309 hardness of the salted cod ( $3.6 \pm 0.4$  N) reduced as the desalting process progressed  
310 (Fig. 5). Major hardness changes took place in the first 45 min and an asymptotic value  
311 was found after approximately 60 min of desalting (Fig. 5A). US application  
312 accelerated this reduction in hardness (Fig. 5A), with the greatest differences between  
313 CONTROL and US experiments observed at the beginning of the desalting. Unlike  
314 what was observed in the case of swelling, the influence of US on the evolution of  
315 hardness can be attributed not only to the ultrasonic intensification of mass transport but  
316 also to some textural effects (Stadnik et al., 2008), since, at a similar water content, US  
317 samples exhibited a lower hardness value than CONTROL (Fig. 5B). No differences  
318 were observed between US-750 and US-1500.

319 The ability of US to induce structural effects has already been reported (Gabaldón-  
320 Leyva et al. 2007; Ozuna et al., 2013). The compressions and expansions produced by  
321 US induced mechanical stress on both the protein structure and the constituents of the  
322 connective tissue that can lead to softening.

323 The first-order kinetic model (Eq. 5) accurately described the hardness changes during  
324 desalting (Fig. 5A). Both model parameters ( $k_H$  and  $H_{eq}$ ) were significantly ( $p < 0.05$ )  
325 affected by US application. However, no significant differences were found between the  
326 two acoustic powers tested (Table 2). The rate constant ( $K_H$ ) significantly ( $p < 0.05$ )

327 increased when acoustic energy was applied as indicated by the more marked reduction  
328 in hardness when US was applied (Fig. 5B). The equilibrium hardness value ( $H_{eq}$ ) was  
329 lower for US treatments than CONTROL, which may be ascribed to the  
330 abovementioned structural effects of US.

### 331 3.5 Microstructure

332 SEM micrographs showed the microstructure of salted cod (Fig. 6). The cod fibers,  
333 covered by salt deposits (Fig. 6A), presented an intense dehydration and compaction  
334 with a total degradation of connective tissue. The high content of salt masked the  
335 underlying structures (Fig. 6B). During desalting, important changes take place as a  
336 consequence of cell rehydration. After 180 min of treatment, samples showed a  
337 significant swelling of muscle fibers (Figs. 7A, 7B and 7C) compared to salted cod  
338 (Fig. 6). Thus, the fibers' average width increased from  $64.4 \pm 9.2 \mu\text{m}$  in the case of  
339 salted cod fibers to  $84.1 \pm 11.7 \mu\text{m}$  in CONTROL samples. This increase was even  
340 greater when US was applied (Figs. 7B and 7C). In this sense, US-1500 fibers (Fig. 7C)  
341 were 27% wider ( $106.7 \pm 11.5 \mu\text{m}$ ) than CONTROL. This fiber width increase was also  
342 observed in LM micrographs. In this case, the measured fiber width increased from  
343  $90.4 \pm 4.4 \mu\text{m}$  in CONTROL (Fig. 7D) to  $108.8 \pm 13.2$  and  $139.4 \pm 11.7 \mu\text{m}$  in US-750  
344 (Fig. 7E) and US-1500 (Fig. 7F), respectively. The differences between LM and SEM  
345 measurements can be attributed to the differences in sample preparation.

346 The fiber width increase explains the macroscopic swelling already described in Fig. 4.  
347 In addition, the US application increased the interfibrillar spaces (IS) (Figs. 7B and 7C)  
348 that can also contribute to the softening of US samples (Fig. 5). The SEM-EDX  
349 technique confirmed that US application intensified the NaCl leakage (Figs. 7B and



350 7C), which was manifested by a lower number of dots in micrographs than in  
351 CONTROL (Fig. 7A).

352 The LM micrographs also showed the mechanical effects of acoustic energy. The US  
353 samples presented greater fiber and connective tissue degradation (Figs. 7E and 7F)  
354 than CONTROL (Fig. 7D). This fact can be attributed to violent microjets produced by  
355 the asymmetric implosion of bubbles near the solid surface. These results coincide with  
356 those found by other authors who related the application of US with the physical  
357 disruption of the myofibril structure (Ozuna et al., 2013), the degradation of collagen  
358 macromolecules, or the creation of micro-channels (Jayasooriya, Bhandari, Torley, &  
359 D'Arcy, 2004).

#### 360 **4. Conclusions**

361 The application of high-intensity ultrasound improved cod desalting, increasing both  
362 moisture and NaCl effective diffusivities by up to 103 and 62%, respectively. The  
363 desalting induced the tissue swelling and the decrease in sample hardness, both of  
364 which were intensified by ultrasound application. Microstructural analyses showed that  
365 cod fibers were significantly affected by ultrasound application. These facts not only  
366 coincide with the intensification of mass transport and the observed enhancement of  
367 swelling, but also with the decrease in hardness.

#### 368 **Acknowledgments**

369 The authors acknowledge the financial support of the Spanish Ministerio de Economía y  
370 Competitividad and FEDER (Ref. DPI2013-37466-C03-03). César Ozuna thanks  
371 Universitat Politècnica de València for an FPI grant (Ref. 2009-02). The author also  
372 wishes to thank Carmen Cambra S.L. for selecting the raw material.

373 **Nomenclature**

<b>Symbol</b>	<b>Reference</b>	<b>Unit</b>
SFS	Salt-free solids	
$W_0$	Initial moisture content,	kg water/kg SFS
$W_{eq}$	Equilibrium moisture content,	kg water/kg SFS
$NaCl_0$	Initial sodium chloride content,	kg NaCl/ kg SFS
$NaCl_{eq}$	Equilibrium sodium chloride content,	kg NaCl/kg SFS
t	Time,	s
$D_{NaCl}$	Effective NaCl diffusivity,	$m^2/s$
$D_W$	Effective moisture diffusivity,	$m^2/s$
$\Delta D_{NaCl}$	NaCl effective diffusivity increase by ultrasound	%
$\Delta D_W$	Effective moisture diffusivity increase by ultrasound	%
L	Half length,	m
$\lambda_n$	Eigenvalues	
$T_0$	Initial sample thickness,	m
T	Thickness after a certain time of treatment	m
$TR_0$	Initial thickness ratio	
$TR_{eq}$	Equilibrium thickness ratio	
$H_0$	Initial hardness,	N
$H_{eq}$	Equilibrium hardness,	N
$k_{TR}$	Rate constant of thickness ratio,	1/s
$k_H$	Rate constant of hardness,	N/s
$S_y$	Standard deviation of the sample	
$S_{yx}$	Standard deviation of the estimation	

374

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458

459 **Figure captions**

460 **Fig. 1** Desalting set-up. A) chiller B) valve C) heat exchanger D) pump E) cooling  
461 reservoir F) ultrasonic bath G) sample holder H) thermocouples I) cod samples J)  
462 ultrasonic transducers K) data logger L) ultrasonic generator M) Computer.

463 **Fig. 2.** Acoustic pressure (bar) distribution in the ultrasonic bath with an applied electric  
464 power of 750 W (A) and 1500 W (B) in a horizontal section at 100 mm from the  
465 water surface.

466 **Fig. 3** Experimental NaCl (A) and moisture (B) transport kinetics of salted cod slices  
467 (thickness 5 mm) desalted with (US, 40 kHz, 750 and 1500 W) and without  
468 ultrasound application (CONTROL). Average values (each point represents the  
469 average of 12 measurements; 4 independently desalted samples analyzed in  
470 triplicate) and LSD intervals ( $p < 0.05$ ).

471 **Fig 4.** Evolution of thickness ratio vs time (A) and vs moisture content (B) during  
472 desalting of salted cod with (US, 40 kHz, 750 and 1500 W) and without ultrasound  
473 application (CONTROL). Average values (each point represents the average of 8  
474 measurements) and LSD intervals ( $p < 0.05$ ).

475 **Fig 5.** Evolution of hardness vs time (A) and vs moisture content (B) during desalting of  
476 salted-cod with (US, 40 kHz, 750 and 1500 W) and without ultrasound application  
477 (CONTROL). Average values (each point represents the average of 30  
478 measurements) and LSD intervals ( $p < 0.05$ ).

479 **Fig 6.** Longitudinal (A) and transversal (B) section of salted cod (*Gadus morhua*) fibers  
480 observed by SEM (x500). S: NaCl deposits, IS: Interfibrillar space.

481 **Fig 7.** Longitudinal section observed by SEM-EDX (A, B and C; x500) and LM (D, E  
482 and F; x4) of cod loin desalted for 180 min without (CONTROL, A, D), and with US  
483 at 750 (B, E) and 1500 W (C, F). IS: Interfibrillar space.

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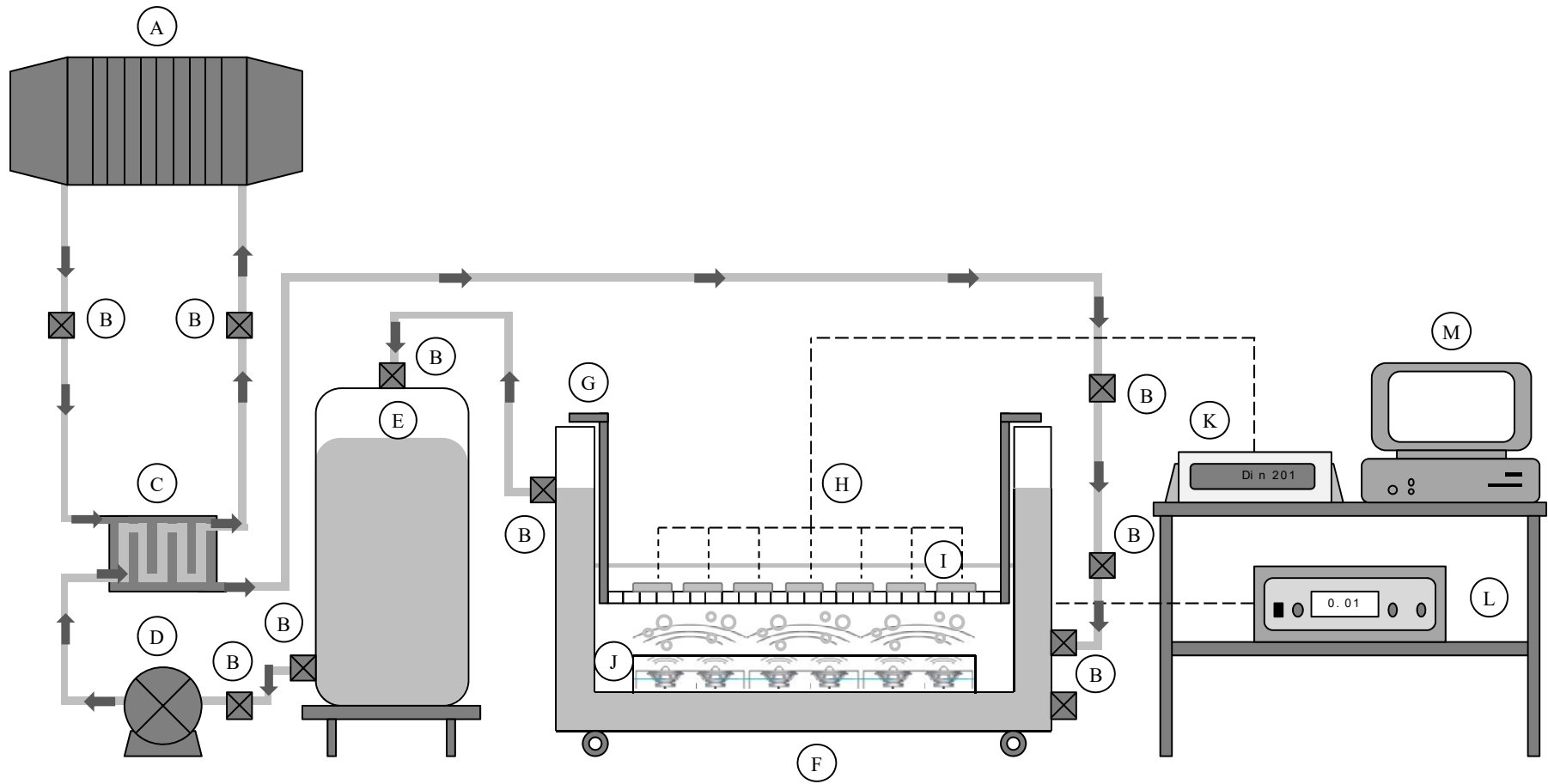
#### 485 **Table captions**

486 **Table 1.** Average values and confidence intervals (95%) of effective diffusivity of  
487 moisture ( $D_w$ ) and NaCl ( $D_{NaCl}$ ) identified from modeling of cod desalting ( $4 \pm 1$  °C)  
488 with (US, 40 kHz, 750 and 1500 W) and without US application (CONTROL) and  
489 percentage of explained variance by the model (VAR (%)).  $\Delta D_w$  and  $\Delta D_{NaCl}$  (%) are  
490 the increases in effective diffusivity produced by ultrasound application.

491 **Table 2.** First-order kinetic parameters for thickness ratio ( $TR_{eq}$ ,  $K_{TR}$ ) and evolution of  
492 hardness ( $H_{eq}$ ,  $K_H$ ) during cod desalting ( $4 \pm 1$  °C) with (US, 40 kHz, 750 and 1500  
493 W) and without US application (CONTROL). Average  $\pm$  confidence intervals of the  
494 estimation (95%) are shown.



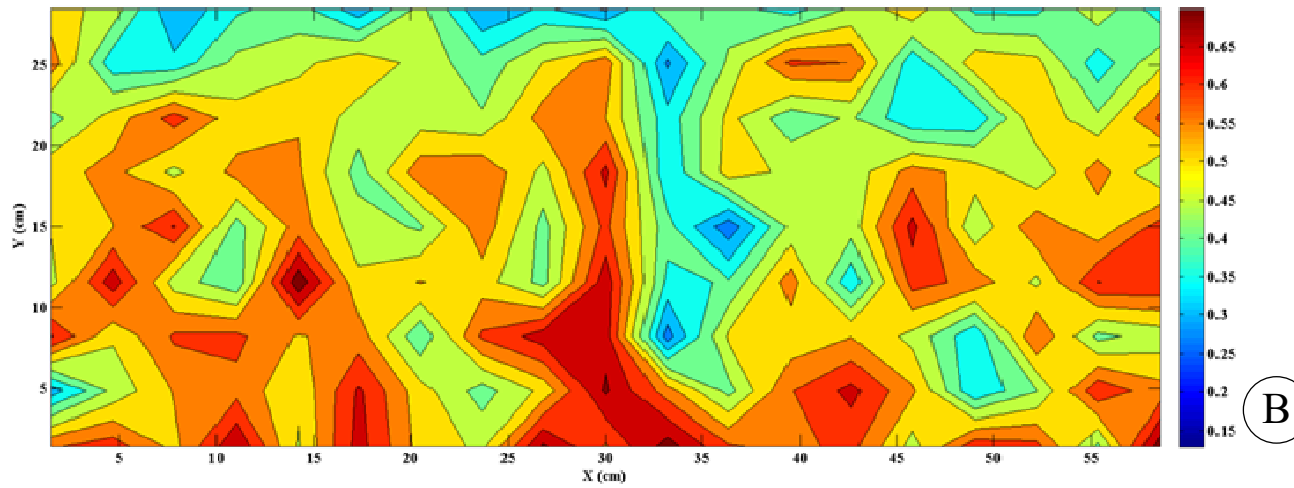
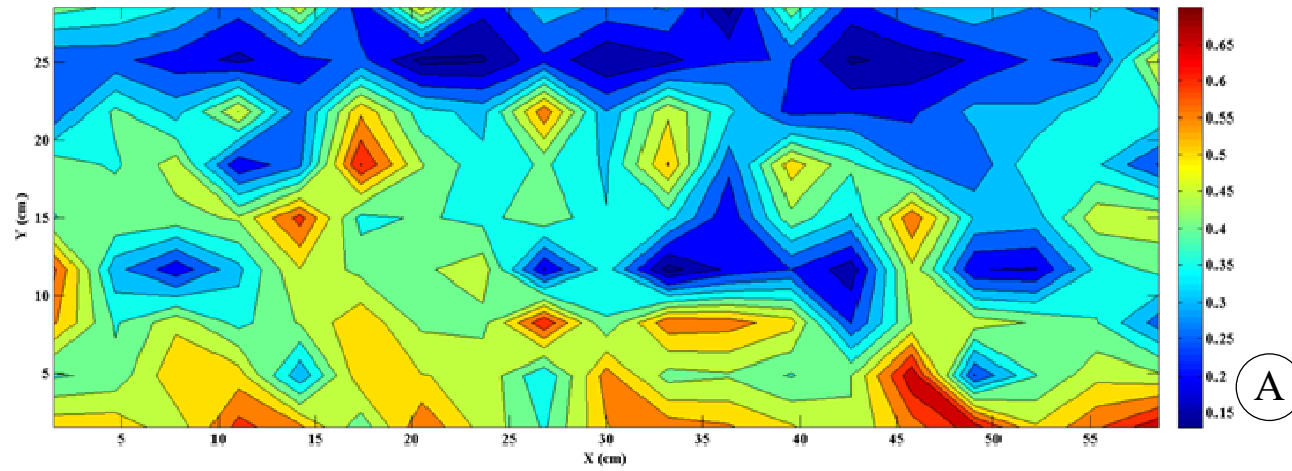
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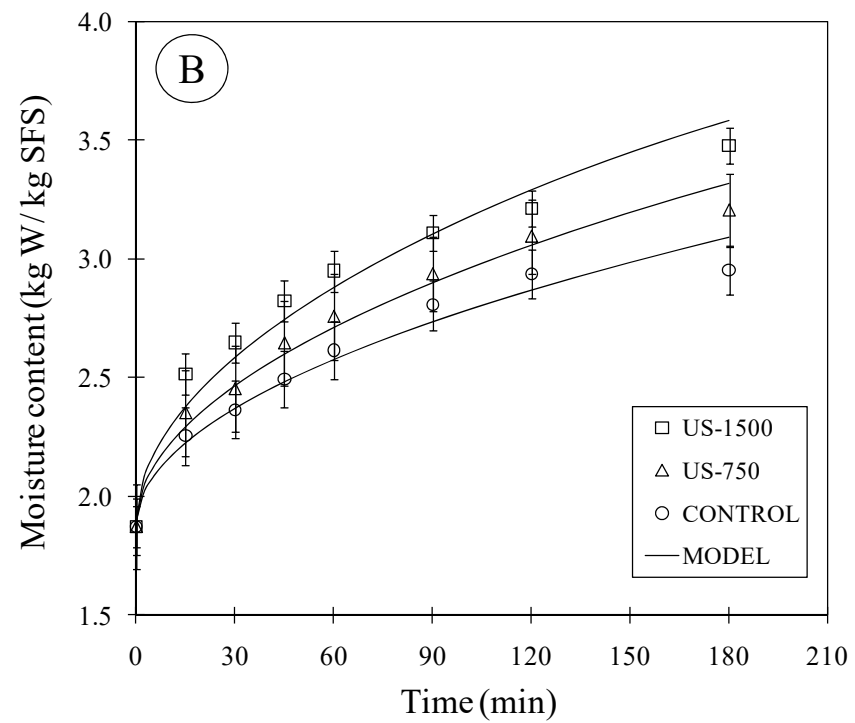
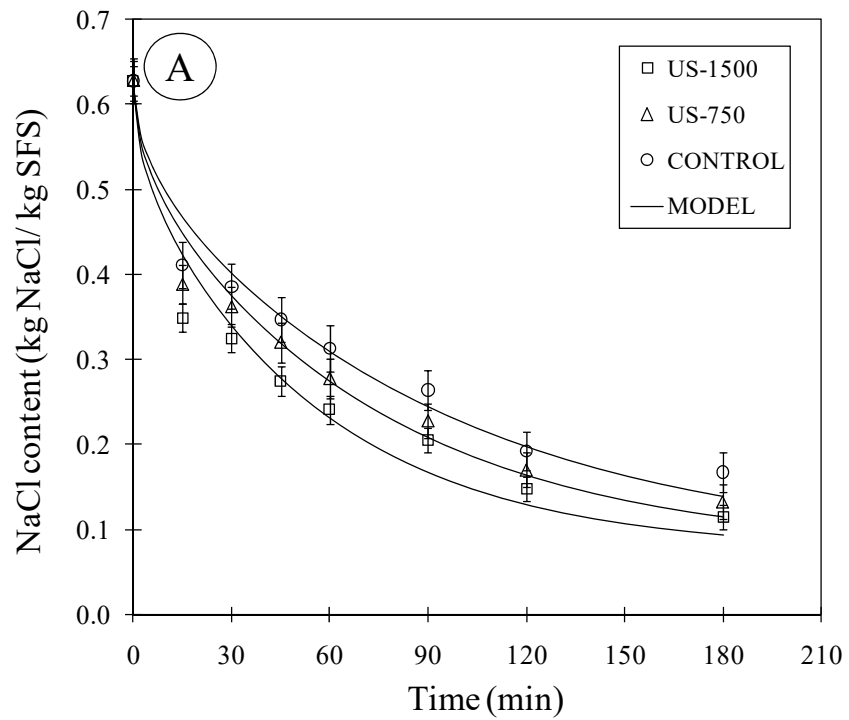
497 **Fig 1**

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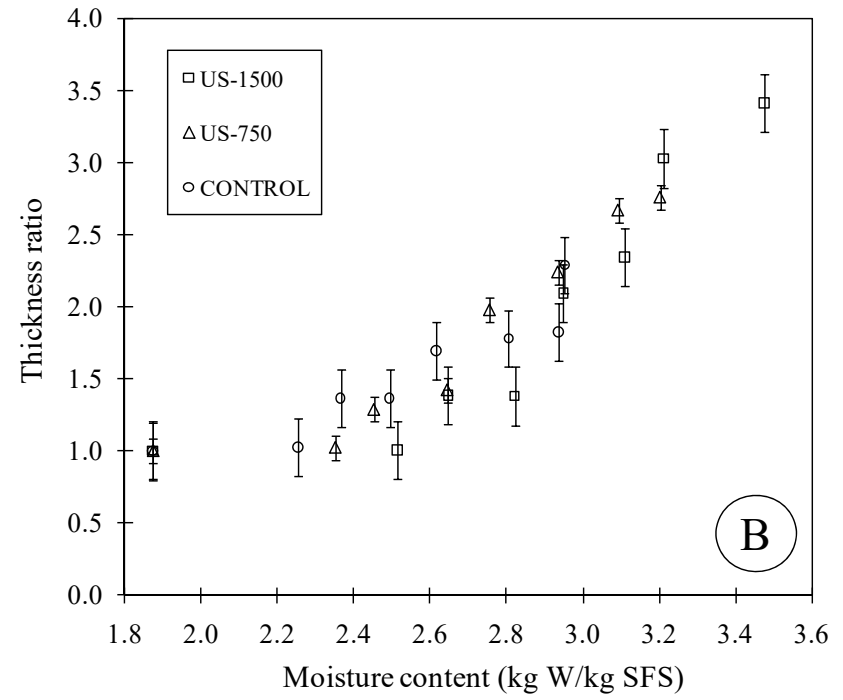
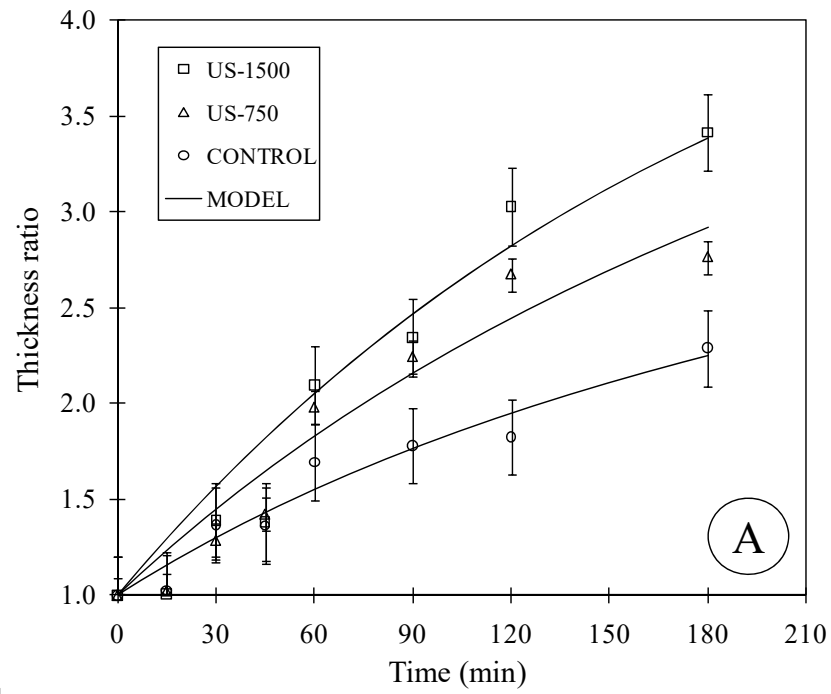
500 **Fig 2**



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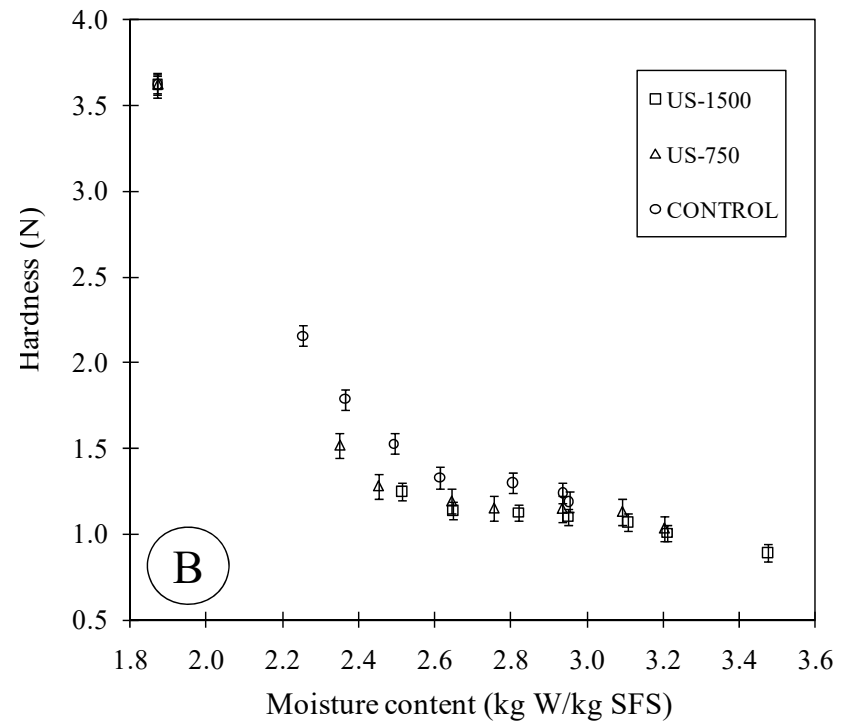
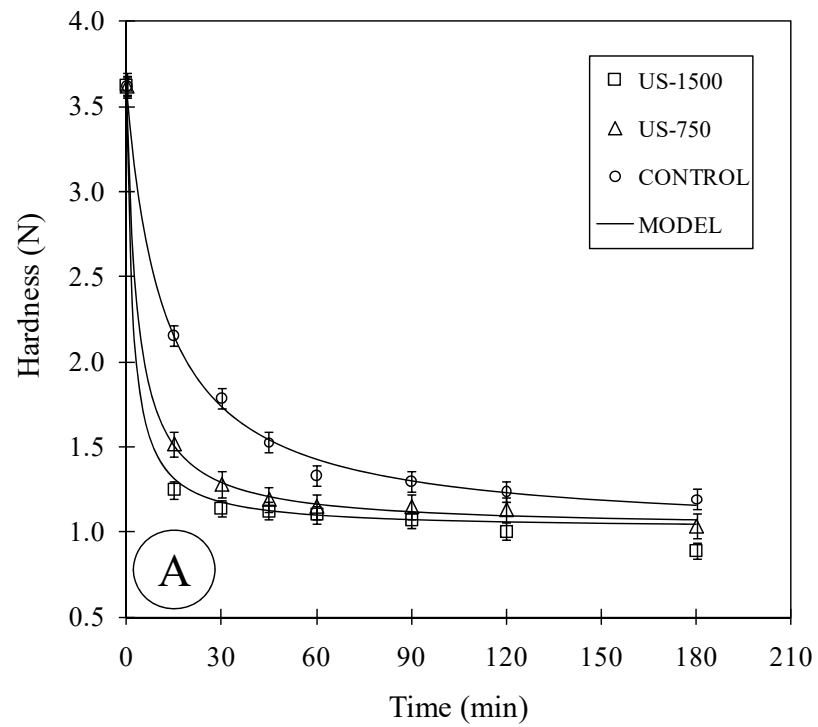
502 **Fig 3**

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504 Fig 4

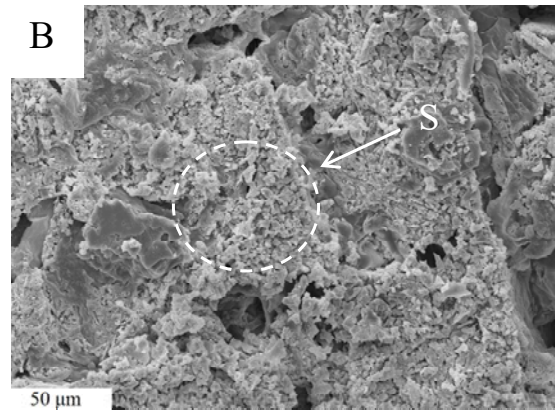
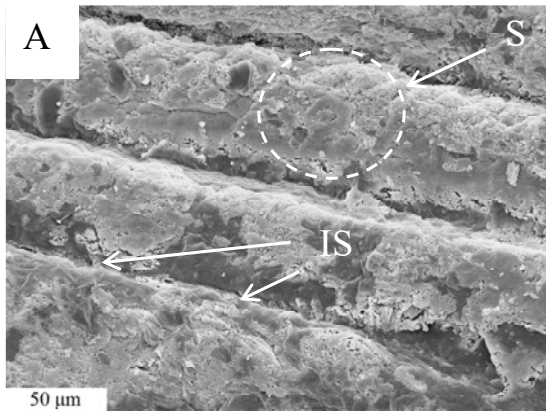
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507 **Fig 5**

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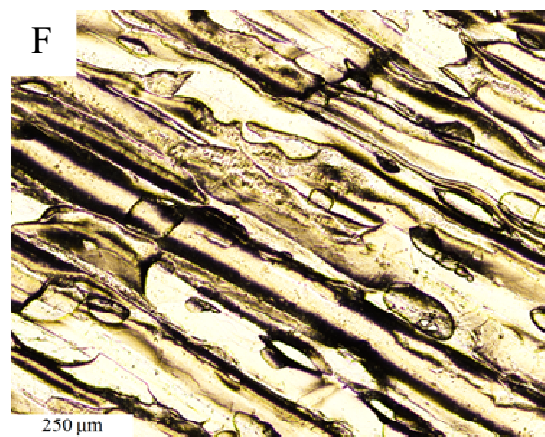
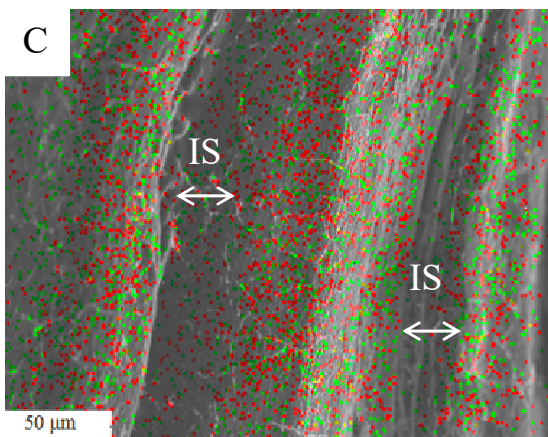
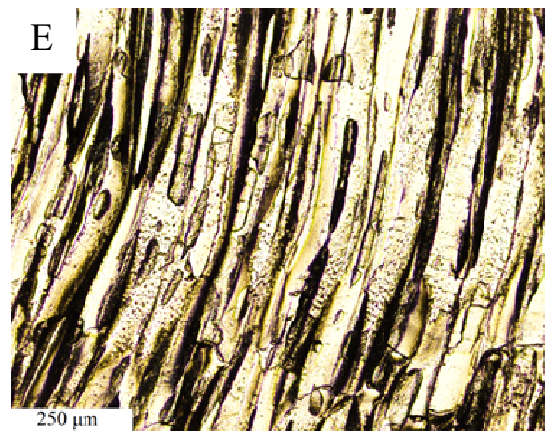
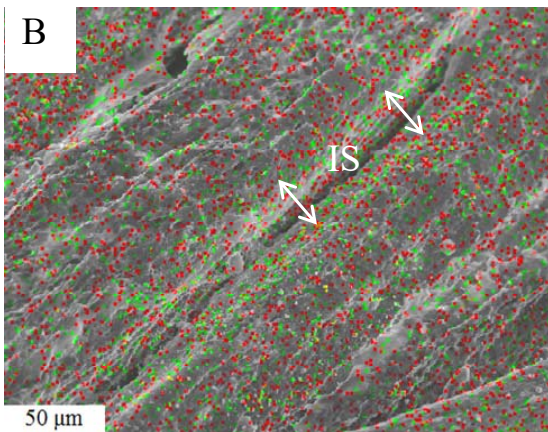
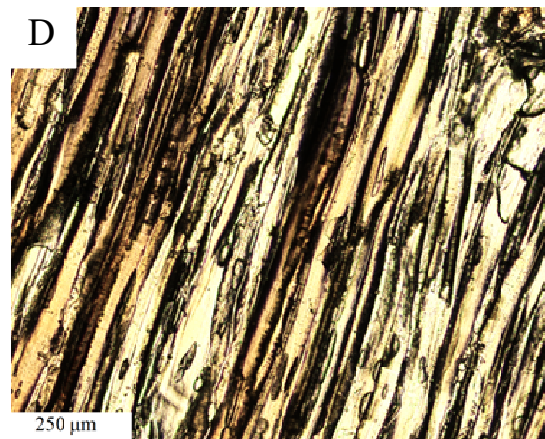
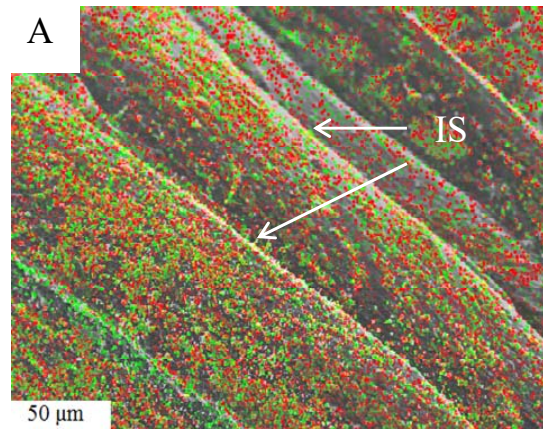
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510 **Fig 6**

511

SEM. X-ray

Light microscopy



512

513 **Fig. 7**

514 **Table 1**

	NaCl transport			Water transport		
	$D_{\text{NaCl}} (10^{-10}) [\text{m}^2/\text{s}]$	$\Delta D_{\text{NaCl}} (\%)$	VAR (%)	$D_{\text{w}} (10^{-10}) [\text{m}^2/\text{s}]$	$\Delta D_{\text{w}} (\%)$	VAR (%)
CONTROL	$4.6 \pm 0.8$	-	95.0	$1.1 \pm 0.2$	-	95.0
US-750	$5 \pm 1$	24.9	95.7	$1.5 \pm 0.2$	41.3	97.2
US-1500	$7.0 \pm 1.1$	61.9	93.3	$2.2 \pm 0.3$	102.8	95.0

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	Swelling evolution			Hardness evolution		
	TR <sub>eq</sub>	k <sub>TR</sub> (10 <sup>-3</sup> ) [min <sup>-1</sup> ]	VAR (%)	H <sub>eq</sub> (N)	k <sub>H</sub> [min <sup>-1</sup> ]	VAR (%)
CONTROL	3.1±0.2	5.1±2.1	94.4	1.26±0.08	0.06±0.02	99.1
US-750	4.1±0.4	5.2±1.0	92.9	1.14±0.06	0.12±0.03	99.6
US-1500	4.9±0.4	5±1	92.7	1.15±0.09	0.17±0.07	99.2