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Biodegradability of bio-based materials produced with DHT-modified cassava starch

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Abstract

Among the different types of starch modification that increase their performance, dry heating treatment (DHT) is a promising green technique that results in plastic materials with enhanced characteristics, mainly in their mechanical and barrier properties. The effect of this type of modification on its biodegradability and disintegration behavior has yet to be described in the literature. Therefore, this work aims to produce bio-based plastic with cassava starch previously modified by dry heating treatment (DHT) and study its disintegration and biodegradation in compost media. The main results showed that all materials showed up to 80 % and 90 % biodegradation and disintegration, respectively, for all materials in 45 days. Moreover, higher rates of disintegration were visualized until day 7, as well as microbial growth and disappearance of functional groups, which is attributed to the highly hydrophilic nature of starch that facilitates the hydrolysis of the polymer in the early stages. In particular, the sheet submitted to DHT for 4 h showed higher rates.

Keywords: extrusion, biodegradable plastics, compostable materials.

1. Introduction

The amount of materials used for packaging is constantly growing. If current consumption models and waste management practices do not improve by 2050, there will be more than 12 billion tons of plastic litter [1]. The inappropriate disposal of plastic materials has led to their dispersion and accumulation in the marine environment in smaller pieces called “microplastics” [2]. The finding of plastic degrading into microparticles in oceanic environments is no longer shocking news [3]. In fact, during the current COVID-19 pandemic, the worldwide production and disposal of face masks, plastic laboratories and medical materials have increased drastically, adding microplastic waste to the environment [3].

Therefore, the development of biopolymer-based packaging is one such alternative to reduce the use of conventional plastics made from petrochemicals. Regarding starch, in 2017, the global production of starch-blend materials corresponded to 43.8 % of the total biodegradable plastic production [2]. However, it is well-known that native starch performance is reduced, and the properties of their plastics are not competitive against conventional plastics. Among the numerous varieties of starch modifications to improve the performance of native starch, dry heating treatment (DHT) is an interesting alternative with potential use in different industries, resulting in materials with increased mechanical and barrier properties [8,9]. According to the literature, DHT results in partial replacement of hydroxyl groups by carbonyl groups, resulting in new intermolecular associations. Moreover, DHT led to a reduction of molecular size, in particular the molecules of intermediate size, forming two well-defined groups with larger and smaller molecules, similar molecule sizes are more suitable for reassociation favoring short- and long-term retrogradation [12]. Literature also reported that polymer

structure, i.e., polymer chain length, and functional groups influence the biodegradation process [10].

Biodegradation studies of polymers in compost media started to be studied for some bio-based packaging, i.e., wheat gluten-based [4,5]; Polylactic acid (PLA) - Poly(3-hydroxybutyrate-co-3-hydroxyvalerate), PHBV [6] and starch [7], among others. However, the effect of the DHT modification process on the biodegradation and disintegration of plastic materials is not yet described in the literature. Obtaining materials with increased performance is a good achievement; however, these materials must also result in proper rates of biodegradation and disintegration, otherwise, the grave problem of plastic waste contamination in the environment will not be solved at all.

Therefore, this work aims to produce bio-based plastic with cassava starch previously modified by dry heating treatment (DHT) and study its disintegration and biodegradation in compost media.

2. Material and Methods

2.1. Materials and dry heating process (DHT)

Native cassava starch (Amilogill 1500) was supplied by Cargill Agrícola - Brazil. Subsequently, it was divided into 50 g portions, then allocated into aluminum envelopes (30 x 30) cm² and submitted to dry heating treatment (DHT) process in a convective oven (FANEM, model 515, Brazil), for 2 h (DH2T) and 4 h (DH4T) at 130 °C, following the modification process described in La Fuente [14]. The modified starch was then cooled, grounded, sieved (250 mm), packaged, and stored until further use. Glycerol grade P.A. (Sigma-Aldrich, Brazil) and Milli Q water were used as plasticizers.

2.2. Extrusion process

The mixes of starch, glycerol (43.1 g/100 g of starch) and water (25.9 g/100 g of starch) were mixed in a blender (Kitchen Aid, 525, Brazil). These mixes were fed to the twin-screw extruder (Thermo Fisher Scientific, Process 11, Germany) L/D 40, and 11-mm screw diameter from a double screw feeder (Brabender, Volumetric Minitwin Process 11, Germany) at 0.05 kg/h and 80 rpm to produce pellets (2 mm). The temperature profile from the feed zone (1) to the extrudate outlet (zone 8) was 60, 75, 90, 100, 110, 110, 105, and 95 °C. Subsequently, from a conical-type feeder (Brabender, Kulturstrasse 55-73, Germany), those pellets were supplied to the same corotating twin-screw extruder with a feed flow of 0.3 kg/h. The plastic materials were produced using the same temperature profile of the pellets at 95 °C and 80 rpm with a die of (3 x 0.1) cm², as previously described in La Fuente et al. [14]. Those materials were conditioned for at least 7 days in aquariums (75 % RH) at ~25 °C before the disintegration/biodegradation experiments.

2.3. Water Content and Solubility

The water content of the biobased materials was determined in triplicate in an oven (Marconi, M030/2057, Brazil) at 105 °C, according to the methodology described by Gontard et al. [15].

The solubility was evaluated in quintuplicate with disks (diameter = 20 mm) in Milli Q water at 25 °C and 58 °C, stirred at 120 rpm in a shaker (SOLAB, mod. SL-223, Brazil) for 24 h. After that, water was filtered and the retained material was weighed and then dried to constant mass at 105 °C according to Gontard et al. [15]. Solubility (*S*) and water uptake capacity (*WUC*) were calculated according to Eq. (1) and Eq. (2), respectively.

$$S(\%) = \frac{m_f - m_0}{m_0} \cdot 100 \quad (1)$$

$$WUC(\%) = \frac{m_w - m_f}{m_f} \cdot 100 \quad (2)$$

wherein: m_o , m_w , and m_f are the initial, the wet, and the final mass of the discs (g), respectively.

2.4. Synthetic Solid Residue (SSR)

A synthetic solid residue (SSR) was prepared according to the EN ISO 20200 standard [12]. The water content was corrected at 55 g/100 g at the beginning of the composting period. At time zero and the end of the assay, the SSR was characterized by the dry (DS) and volatile solids (VS), in triplicate, according to Eq. (3) and Eq. (4).

$$DS(\%) = \frac{m_d^{105}}{m_w^0} \cdot 100 \quad (3)$$

$$VS(\%) = \frac{m_d^{105} - m_d^{550}}{m_d^{105}} \cdot 100 \quad (4)$$

wherein: m_w^0 is the initial mass of the sample, m_d^{105} is the mass after drying at 105 °C and m_d^{550} is the mass of the ashes dried at 550 °C.

The assay was validated at the end of the composting time according to Eq. (5).

$$R = \frac{[m_o \cdot (DS)_o \cdot (VS)_o] - [m_f \cdot (DS)_f \cdot (VS)_f]}{[m_o \cdot (DS)_o \cdot (VS)_o]} \cdot 100 \quad (5)$$

wherein: m_o is the initial mass, DS_o is the initial dry solids, and VS_o is the initial volatile solids, while m_f , DS_f and VS_f are their respective values at the end of the composting time.

2.5. Disintegration assay

The film disintegration assay was performed on a laboratory scale according to the EN ISO 20200 [12] methodology, with some modifications, as described below.

The amount of 5 g of samples cut into (25 × 25) mm² was weighed using an analytical balance (±0.1 mg) and put into mesh bags, later placed into plastic reactors containing 1 kg of SSR, prepared as described previously. Before the experimental setting, plastic reactors were perforated with a 5 mm diameter hole on one side of the reactor to allow gas exchange between the inside and the outside. Three reactors per film formulation and then placed in the oven (FABBE-PRIMAR, Brazil) at 58 ± 5 °C to ensure thermophilic conditions. Throughout the testing period, the reactors were weighed and, if necessary, Milli-Q water was added to keep the compost wet and stirred as specified in ISO 20200 [12]. The disintegration percentage (D_t) after 45 days was determined according to Eq. (6).

$$D_t(\%) = \frac{m_0 - m_t}{m_0} \cdot 100 \quad (6)$$

wherein: m_0 is the initial mass and m_t is the mass after day 45 dried at 40 °C under vacuum in an oven (Marconi, M030, Brazil), until a constant mass.

Additionally, samples of materials were removed from the reactor at different control times on days 2, 7, 10, 14, 21, 28, 35, and 45 to monitor sample weight loss, morphological and mechanical changes, and FTIR spectra. Before these analyses, the mesh bags (holding the pieces of material) were cleaned with a brush to eliminate the adhered residues and then dried at 40 °C, under a vacuum in an oven (Marconi, M030, Brazil), until constant mass.

2.5.1. Morphology

The morphological changes of the biobased materials throughout the disintegration experiments were photographed with a digital camera and with an image capture system consisting of a light microscope and a digital camera (LEICA, model S6D, Germany) at 80x magnification, immediately after being collected from the reactors. Pieces of the materials, previously dried at 40 °C, were visualized by a scanning electron microscope

(SEM) (FEI, model Quanta 600FEG, Netherlands), coated with 10-nm platinum, and observed at 15 kV and 5000x.

2.5.2. Mechanical Properties

Plastic material thickness was determined at five arbitrary positions using a digital micrometer (± 0.001 mm) (MITUTOYO, Japan). Tensile strength (TS), elongation at break (ϵ), and Young's modulus (E) were estimated throughout the uniaxial tensile assay according to the ASTM D882-12 standard method [13] on a texture analyzer (Stable Micro Systems, TAXT2i, UK), using the probe A/TGT self-tightening grips at a speed of $1 \text{ mm} \cdot \text{s}^{-1}$ and grip separation of 50 mm; ten strips (100×25) mm^2 per formulation were prepared.

Strips of materials were carefully taken out of the reactors and tested on day 0, day 2, and day 4. The materials from days 2 and 4 were previously dried under vacuum in an oven (Marconi, M030, Brazil) at 40°C until a constant mass was reached to eliminate excess water. Afterward, they were conditioned for at least 2 days in aquariums (75 % RH) at $\sim 25^\circ\text{C}$ before the mechanical test. However, for both DHT sheets (DH2T and DH4T), it was impossible to remove the material because it was already broken or broken during the collection procedure after 2 days. From day 7, it was impossible to collect materials from any formulation.

2.5.3. Fourier-transform infrared (FTIR) assay

Plastic material spectra were evaluated by Fourier Transform Infrared spectroscopy (Shimadzu, IR Prestige-21, Japan) in the Attenuated Total Reflectance (ATR) mode from 4000 to 500 cm^{-1} , with a resolution of 2 cm^{-1} and 20 scans. The data were then processed using Origin 2020 software (OriginLab Corporation, Massachusetts, USA).

2.6. Biodegradation assay

Biodegradation was evaluated throughout the laboratory setup methodology for 45 days at 58 ± 2 °C simulating aerobic composting following ISO 14855-1 [14] with modifications. Before the test, the C, N, and H contents were determined by an elementary analyzer (PERKIN ELMER, model 2400 Serie II, USA). The reactors (glass containers of 2 L) containing two polypropylene flasks were then prepared, one with 3 g of synthetic solid residue (SSR) with 1 g of vermiculite and approximately 100 mg of material, while the other flask had distilled water to ensure 100 % RH. Reactors were prepared in triplicate per formulation; reactors containing cellulose microcrystalline (CMC) as the reference material and reactors with only compost as blank were also prepared in triplicate.

The CO₂ generated inside the reactors throughout the biodegradation process was directly measured using a CO₂ analyzer (MOCON, PAC CHECK model 650, U.S.A), at intervals of 2 days. The theoretical concentration of CO₂ generated from the plastic material (CO₂Th_s) was calculated by Eq. (7).

$$CO_2^{Th}_s = m_s \cdot C_s \cdot \frac{M_{wCO_2}}{M_{wC}} \quad (7)$$

wherein: m_s is the dry mass of the sample (g), C_s is the concentration of carbon in the dry material determined by the elementary analysis, M_{wCO_2} and M_{wC} are the molecular weight of CO₂ and C, respectively.

The percentage of biodegradation was determined by Eq. (8), assuming that all the carbon present in the material was converted to CO₂.

$$B(\%) = \frac{\sum(CO_2)_t - \sum(CO_2)_{bt}}{CO_2^{Th}} \cdot 100 \quad (8)$$

wherein: $\sum CO_{2t}$ is the cumulative amount of CO₂ produced in the sample reactor and $\sum CO_{2bt}$ is the cumulative amount of CO₂ produced in the blank reactor.

The kinetics of biodegradation was adjusted to Hill's equation [19, 20].

$$B(\%) = B(\%)_{\max} \cdot \frac{t^n}{k^n + t^n} \quad (9)$$

wherein: $B(\%)_{\max}$ is the percentage of biodegradation at infinite time, t is the time (days), k is the time at which $B(\%) = 0.5 B(\%)_{\max}$ (days), and n is the curve radius of the sigmoid function.

2.7. Statistical analyses

All data were evaluated by the Analysis of Variance Multifactor (ANOVA), applying Tukey's test ($p < 0.05$) using the Statgraphics Centurion XV software (StatPoint, Inc., USA).

3. Results and Discussion

3.1. Solubility, water uptake capacity, water content, thickness, and carbon content

Solubility and water uptake capacity was determined at 58 °C because the disintegration/biodegradation in compost media was held at this temperature, and these properties could infer or predict the behavior of disintegration/biodegradation since one of the main components of the compost is water (55 g/100 g). Table 1 presents the solubility and water uptake capacity at 25 °C and 58 °C of the native and DH2T and DH4T materials. Similar values of solubility have been reported in the literature for cassava starch materials produced by extrusion [21]. At 25 °C, the solubility of all the materials was statistically similar; however, the effect of the temperature is observed at 58 °C. As shown in this table, the DHT modification process increased the solubility of biobased materials probably caused by the depolymerization that occurred during the modification. Indeed, Guo et al. [22] showed a decrease from 1.75×10^{20} (native starch) to 1.89×10^{19} and 5.61×10^{16} (g/mol) after 2 h or 4 h at 130 °C of the DHT process. Lei et al. [23] also showed a dramatic decrease in DHT-modified starch granules size from 100 to 10 nm in the case of maize starch.

Moreover, Shah et al. [24] reported that this molecular polymerization increases the solubilization due to the fragmentation of starch into low molecular weight chains and La Fuente et al. [21] showed that the ozone-modified cassava starch sheets produced by extrusion also resulted in a higher solubility than the native cassava starch materials. However, the modification process is different (ozone modified *versus* DHT); depolymerization of molecules occurs in both modification processes, confirming that polymerization facilitates water penetration and dissolves polymeric fractions more easily, thus increasing the solubility of the DHT films.

At 25 °C, a decrease in water uptake capacity (WUC) can be observed due to the DHT-time, while at 58 °C, a decrease in this property was observed only for the DH4T cassava starch sheet. According to Edhirej et al. [25] the presence of hydroxyl groups in water molecules facilitates the formation of hydrogen bonds with starch, resulting in the highest tendency of WUC in the plasticized polymer material. As exposed previously, DHT promotes the oxidation of hydroxyl to carbonyl groups; therefore, the lowest hydroxyl groups available to form hydrogen bonds and lower WUC, are in agreement with the authors' hypothesis, which probably describes the results herein. In sum, the total solubility of the DHT materials (both 2 h and 4 h) is greater than the native one, which implies that DHT-modified materials could probably degrade at higher rates than the native ones.

Table 1 also presents the values of moisture, thickness, and carbon content of the materials studied. Similar values of moisture and thickness were also reported in the literature for cassava materials produced by extrusion [9,15]. Similar carbon content values were reported for starch films [5,7]. No statistical differences were observed for the thickness and the C content, which is a positive result, as both properties influence the disintegration and biodegradation (CO₂ production) processes. According to ASTM

D882-18 [13], the plastics obtained from extrusion are ‘sheets’ since their thickness is greater than 0.250 mm.

3.2. Compost characteristic

Table 2 presents the compost characterization in dried solids (*DS*) and volatile solids (*VS*), as well as the reduction in the volatile content (*R*) of the compost after the composting period. The initial values (time = zero composting) were *DS* of 33.75 ± 0.63 g/100 g and *VS* of 92.07 ± 2.00 g/100 g. Table 2 allows observing the decrease in the *VS* as regards the initial value, which is an expected behavior already described in the literature [7,18]. After 45 days, no statistical differences were observed in the percentage of disintegration for the native, DH2T, and DH4T sheets; similar values of disintegration were reported in the literature for native pea starch after 73 days [19] and corn starch after 84 days [7]. Since the *R* values were greater than 30 % and the standard deviations of the disintegration were less than 10 %, the test could be validated according to EN ISO 20200 [16].

3.3. Morphological changes

The visual appearance of the sheets at different composting times is presented in Figure 1. Initially, at time = zero, all materials presented the same morphological characteristic without difference between native and DHT sheets (2 h and 4 h). On the second day, all the sheets are verified to have absorbed water; however, while the native sheet remained with a joint structure, the DHT materials started to separate into pieces, with the visible presence of cracks. On day 7 the disintegration of all the sheets could be visualized; the zoom images show that DHT materials, especially DH4T sheets, resulted in more and smaller pieces, implying a higher disintegration rate. From day 14 to the end of the assay (day 21 - 28 images are not shown), the disintegration continues to occur. Finally,

on day 45, all the materials are observed to practically completely disintegrate, with few particles bound in the mesh bags.

The morphological changes shown by SEM at different composting times are presented in Figure 2. According to Oliveira et al. [20] the SEM allows the detection of microbial colonization and changes in the physical aspect of the polymer surface, which therefore implies surface degradation. At time zero, a smooth surface could be seen for the native sheet, and in the case of DHT materials (2 h and 4 h), ungelatinized starch granules could be seen, which is probably caused by two reasons. Chandanasree et al. [28] reported that during the modification process starch granules grew agglomerated due to the leaching of amylose as the decreased granule interspacing. Moreover, Lei et al. [29] point out that the low moisture content of DHT-starches (due to the heating) reduces the available water acting as a plasticizer during gelatinization. In fact, ungelatinized granules of DHT-modified starch were also seen for materials produced by extrusion [13] and by melt blending and subsequent compression molding [10].

On day 2, a smoother surface remained for the native sheet, while cracks are visualized in the DHT sheets, more visible in the DH4T one, which agrees with the images shown in Figure 1. Czaja-Jagielska [32] also visualized a porous structure with cracks and cavities in the disintegration of corn and potato starch films on the initial days of degradation. On day 7, it is possible to observe the microorganisms attacking the starch granules, again more markedly for the DH4T sheet. From day 14 to the end of the assay, microbial colonization is clearly visualized, with significant increases day after day.

Balaguer et al. [11] observed fungal growth in wheat gluten films with the formation of a mycelium network, similar to Trujillo-Hernández et al. [33] in the degradation of *jicama* starch films, in which the presence of fungal spores and mycelia invading the

surface were visualized after 30 days. In our study, the mycelium network was observed on day 35, markedly in the DH2T sheet.

3.4. Weight loss curve

Figure 3 presents the weight loss of biobased sheets during the composting period, whereby the behavior of the curve is similar to those described in the literature [5,7,18] for biopolymer materials degradation under similar conditions. Higher weight loss rates are observed till day 7 when they start decreasing considerably. In fact, high rates of disintegration were also observed on the initial days of the assay for starch materials, which was related to the highly hydrophilic nature of starch, facilitating the degradation of the polymer backbone and promoting faster surface erosion [19]. On day 45, weight loss reached values from 80 to 90 % for all the materials; moreover, during experimentation time, the DH4T sheet showed a slightly higher weight loss rate during composting time, especially at the end of the assay (day 45). This behavior could probably be attributed to the modification process in which molecular depolymerization occurs. Maniglia et al. [8] reported that DHT promoted the formation of molecules of lower size, while Sheel & Pant [9] said that polymer depolymerization increases the available polymer sections for microorganism attack, which probably explains the results here, similar to those exposed by Chiu & Solarek [34] who reported that starch oxidation (DHT is a type of oxidation) produced more biodegradable materials; however, this is an initial hypothesis and additional analysis may be needed in the future to prove this.

As pointed out before, Sheel & Pant [9] reported that solubility is directly related to biodegradability, which partially explains the results herein, since higher solubility was observed for DHT sheets. The fact that DH2T resulted in higher values of solubility (Table 1) could imply that not only hydrophilic characteristics (available hydroxyl

groups) interfere with biodegradability behavior but also does polymer structure, since higher depolymerization was visualized in the DH4T sheet, as described by Maniglia [8].

3.5. Mechanical properties

Figure 4 shows the mechanical properties of the native, DH2T and DH4T sheets during the composting time. At time zero, similar values were reported for tensile strength, elongation, and Young's modulus for cassava sheets produced by extrusion [9,15]. As previously exposed, materials were also collected on day 2 and day 4. However, for the DHT sheet (both 2 h and 4 h), the collection of material was impossible because the materials were already broken into pieces or broken during the collection procedure, as can be seen in Figures 1 and 2. For the native sheet, a decrease in the mechanical properties is directly related to the experimentation time. Tosin [35] also observed a decrease in mechanical properties during the biodegradation period of *Mater-Bi* films (a biodegradable plastic produced by Novamont), which according to Oliveira [20] are indications of biodegradation caused by microbial attack on polymers. Sheel & Pant [9] also attributed this decrease in the mechanical properties to the reduced polymer chain length and their weakness due to chain scissions. Therefore, according to these results, the biodegradation process of all biobased sheets had already started on day 2.

3.6. Fourier-transform infrared (FTIR) spectra

Figure 5 shows the FTIR spectra of the sheets obtained from native cassava starch and modified by DHT (2 h and 4 h), collected during the disintegration assay. FTIR spectra provide information on the polymer chemical functional groups, which, according to Oliveira [20], give information on the chemical changes during biodegradation, either through their consumption or by the production of functional groups due to microbial activity.

At time zero, the spectra exhibited a conventional behavior for native cassava starch plastics plasticized with glycerol [26,27] and also for sheets produced with cassava starch previously modified by DHT [9]. No differences were observed among the different materials (native, DH2T, and DH4T) as previously discussed in the literature (7). The band at 3300 cm^{-1} is associated with the free hydroxyl groups (O—H), while at 2925 cm^{-1} related to C—H stretch [26,27], near to 1645 cm^{-1} to the bending vibration of H—O—H, water molecules absorbed in the amorphous region and, at 1000 cm^{-1} the band is related to the C—O stretching [28,29].

Note that the same profile is maintained for native and DHT- modified materials until day 2, while from day 7 to the end of the assay (day 45), the presence of peaks is not verified, showing that all functional groups disappeared during the disintegration.

According to Ostadi et al. [40] the peak declines near 3404 cm^{-1} indicating that glycerol has percolated from the material, which, according to Mittal [41] is due to the breakage of the hydroxyl groups bonds during the degradation. The decrease in the peak near $2700\text{--}2900\text{ cm}^{-1}$ is due to the breakage of methylene groups [31] and near 1666 cm^{-1} due to the duction of carbonyl groups [32]. Ostadi et al. [40] also reported that the decrease in the peak near 1027 cm^{-1} is related to the breakage of the α (1→4) bond of starch, suggesting the degradation of the polymer by α -amylase produced by the microorganisms. The fact that this change in the FTIR path was observed on day 7, is consistent with the images shown in Figure 2, in which microbial growth can be seen, and also with the higher rate of weight loss observed in Figure 3.

3.7. Biodegradation

Figure 6 presents biodegradation during 45 days of the composting period of native, DHT (2h and 4h) materials and the reference material, in which the symbols are experimental data in comparison with the adjusted Hill's model. First, it can be seen

that the biodegradation of the reference material (CMC) reached 70 % after 45 days; the assay could thus be validated according to the ISO 14855-1 guidelines [14]. In general, the model describes the experimental data well, and the biodegradation curves showed a sigmoidal shape, characteristic of bio-based polymer biodegradation [4]. Similar values of biodegradation were reported in the literature for commercial potato starch films [5] and wheat gluten [4].

In general, the polymer biodegradation process can be divided into *i)* biodeterioration, *ii)* depolymerization, *iii)* bio assimilation, and *iv)* mineralization [9]. In the first step, the formation of a microbial biofilm leads to superficial degradation in which the polymeric material is fragmented into smaller particles. During depolymerization, the biofilm microorganisms secrete extracellular enzymes, which catalyze the polymer chain into oligomers, dimers, or monomers [10]. During this step, a lag phase can occasionally be detected in the sigmoid curves, corresponding to the time needed for water to penetrate the matrix and enzymes to cleave the polymer chains [5]. Uptake of the small molecules produced in the microbial cell and the subsequent production of primary and secondary metabolites is called bioassimilation. Lastly, these metabolites are mineralized and products such as CO₂, CH₄ and H₂O are formed and released into the environment [10]. The materials produced with cassava starch (native or DHT) reached at least 80 % of their biodegradation after 45 days (Figure 6A). A higher market value was reached by DH4T, in comparison to the native and the DH2T material. Analogous to the weight loss curve rate (Figure 3), the microbial growth (Figure 2), and the losses of functional groups (Figure 5), the biodegradation effects were observed mainly on day 7, when the maximum value of the biodegradation rate was observed, again for the DH4T sheet, as shown in Figure 6B.

Note that biodegradation values of the DH4T sheets reach values higher than 100 % (considering the standard deviation), which is due to the ‘priming’ effect, which occurs when the compost inoculum in the reactors produces more CO₂ than the blank sample [7]. This is attributed to the stimulation of organic matter mineralization as a consequence of its interaction with microorganisms [19].

Table 3 presents the parameters of Hill’s model for the native, DH2T, and DH4T bio-based sheets. It can be observed that the DH4T sheet presented the highest value of $B(\%)_{\max}$, followed by the native, the DH2T, and the CMC materials; similar values are found in the literature for starch films [5,7,19]. The time in days to reach 0.5 $B(\%)_{\max}$ (k value) was similar for all the materials and also similar to those reported previously [5,7,19].

According to the literature, the higher degree of crosslinking in starch resulted in a more resistant matrix (due to the new linkages created); lower swelling capacity (measured as water uptake capacity), thus resulting in lower biodegradation rates due to the new bonds being cleaved by microorganisms and the lower water vapor diffusion coefficient [5]. Moreover, Wang et al. [46] stated that both high crystallinity and low hydrophilicity retard penetration and attack by water, resulting in a slow hydrolysis process. Those statements oppose the results reported herein. The DHT modification process results in new functional groups (more resistant matrix due to the new linkages created) and also reduces the water uptake capacity as observed for DH4T (Table 1). The biodegradation of DHT materials was expected to be slower; however, DH4T resulted in higher disintegration (Figure 3) and biodegradation (Figure 6A and 6B) rates. The ‘key’ factor could be the depolymerization, as previously exposed, which results in shortening the polymer chain and makes it more susceptible to microorganism attack. Indeed, Wang et al. [46] showed that even with the same structure, the size of the

polymer chain can severely affect the rate of hydrolysis during biodegradation. Moreover, Sheel & Pant [9] reported that the introduction of the carbonyl groups such as C=O increases the number of chromophores, increases the sites for photon absorption, and thus the rate of photodegradation. The results presented here are a great achievement because the literature reported that DHT modification enhances the starch performance for several applications, i.e., 3D impression [23] and especially in the packaging field [8,9]; from this study, depending on the conditions of modification, the biodegradation process can be assumed to also be improved. For example, compared to the native sheet, with the modification of starch for 2 h (DH2T), the behavior of the disintegration/biodegradation patterns is not altered. However, when the modification is carried out for 4 hours (DH4T), the mechanical performance of the sheets at time zero (Figure 4), as well as the disintegration/biodegradation rates are improved (Figure 3 and Figure 6A and 6B).

4. Final Remarks and Conclusions

Disintegration and biodegradation of native and DHT sheets were determined by the ISO 20200 and ISO 15855-1 methodology, respectively, and both assays were validated according to the criteria described in both standards. In general, the disintegration and biodegradation of all the materials resulted in up to 90 % and 80 %, respectively, after 45 days. In both assays, maximum rates were achieved on day 7, which is attributed to the highly hydrophilic nature of starch that facilitates the hydrolysis of the polymer in the early stages.

Complementary assays, morphological and mechanical changes, and FTIR spectra helped to describe the disintegration/biodegradation process, in a specific way. The microbial colonization was visualized on day 7 by SEM images, concurrently with the disappearance of functional groups in the FTIR spectra. On day 2, a decrease in

mechanical performance was observed; in fact, it was not possible to measure it in DHT sheets due to the higher hydrolyzed structure, as a result of the DHT modification process, which results in the shortened polymer chain, and probably this fact facilitates the microorganism attack.

The solubility characteristic partially explains the disintegration/biodegradation behavior. According to the literature, higher values of molecular depolymerization were reported in the literature for DH4T (in comparison to native and DH2T starch), which increases solubilization in water. In this study, higher solubility was observed in the DH2T sheets; conversely, higher disintegration/biodegradation was observed for the DH4T sheets. This could imply that the greater polymeric depolymerization of this material and the presence of the microorganism population in the compost are crucial elements promoting increases in the rates of disintegration/biodegradation.

These results are an advance from the knowledge related in the literature today because DHT modification showed to enhance starch performance resulting in improved plastic material performance and, from the results reported herein, it is possible to assume that depending on the conditions of modification, the biodegradation process could also be improved, which helps to reduce the world's waste accumulation problem. Therefore, impacting positively the packaging sector.

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Figure Caption

Figure 1. Morphological changes of native, DH2T and DH4T bio-based sheets collected at different times during the composting period. **Note.** Zoom images were obtained at 80x.

Figure 2. Scanning electron microscopy images of native, DH2T, and DH4T biobased sheets collected at different times during the composting period obtained at 5000x.

Figure 3. Weight loss of native, DH2T and DH4T biobased sheets collected at different times during the composting period at 58 °C.

Figure 4. Mechanical properties of native, DH2T and DH4T bio-based sheets collected at different times during the composting period.

Figure 5. FTIR spectra of native, DH2T, and DH4T bio-based sheets collected at different times during the composting period.

Figure 6. Biodegradation (**A**) and biodegradation rate (**B**) of native, DH2T, and DH4T biobased sheets during the composting period. Symbols are experimental data points, solid lines correspond to the degradation adjusted curves with Hill's equation, and error bars represent standard deviation.

655 **Table 1.**

656 Solubility (%) and water uptake capacity (WUC) (%) at (25 and 58) °C, water Content (*X*), thickness, and carbon content (*C*) of the native,
 657 DH2T, DH4T bio-based materials.

Material	Solubility (%)		WUC (%)		<i>X</i> (g/100 g d.b.)	Thickness (mm)	<i>C</i> (g/100g)
	25 °C	58 °C	25 °C	58 °C			
Native	27.7 ± 2.7 ^a	36.1 ± 3.0 ^c	468.2 ± 19.0 ^a	1069.1 ± 54.0 ^a	18.73 ± 0.43 ^a	0.67 ± 0.17 ^a	39.82 ± 0.11 ^a
DH2T	30.8 ± 2.0 ^a	54.4 ± 2.6 ^a	380.2 ± 16.0 ^b	1040.8 ± 55.0 ^a	18.81 ± 0.38 ^a	0.79 ± 0.08 ^a	39.65 ± 0.01 ^a
DH4T	28.3 ± 2.1 ^a	44.1 ± 3.2 ^b	290.0 ± 11.4 ^c	724.6 ± 15.0 ^b	20.02 ± 0.54 ^b	0.82 ± 0.12 ^a	39.55 ± 0.12 ^a

658 Means with different letters in the column are significantly different (p<0.05).

Table 2.

Dried solids (*DS*) g/100 g, volatile solids (*VS*) g/100 g, the reduction of volatile solids (*R*) %, and the disintegration after 45 days (*D*₄₅) of composting of native, DH2T, and DH4T bio-based sheets.

Material	Post-Composting		<i>R</i> (%)	Disintegration <i>D</i>₄₅ (%)
	<i>DS</i> (g/100 g)	<i>VS</i> (g/ 100 g)		
Native	78.21 ± 8.61	86.38 ± 5.95	39.55 ± 3.27	89.8 ± 2.4 ^a
DH2T	49.40 ± 12.63	91.28 ± 2.21	45.20 ± 8.05	90.4 ± 2.5 ^a
DH4T	97.98 ± 1.20	89.96 ± 1.20	42.77 ± 4.30	90.7 ± 3.4 ^a

Means with different letters in the column are significantly different (p<0.05).

Table 3. Parameters of the Hill adjusted model $B(\%)_{\max}$, k and, n is the curve radius of the sigmoid function of native, DH2T and DH4T bio-based sheets collected at different times during the composting period.

Material	$B(\%)_{\max}$ (days)	K (days)	n (-)	r^2
Native	94.80	10.20	2.47	0.99
DH2T	81.60	10.73	2.96	0.99
DH4T	98.94	10.09	2.65	0.99
CMC	80.74	11.28	2.71	0.99

Note. $B(\%)_{\max}$ is the percentage of biodegradation at infinite time, k is the time (days) at which $B(\%) = 0.5 B(\%)_{\max}$ (days), and n is the curve radius of the sigmoid function.