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Additional Information

1 **Drying-rewetting cycles in ordinary Portland cement mortars**
2 **investigated by electrical impedance spectroscopy**

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20 ABSTRACT

21 Changes caused in the porous microstructure of ordinary Portland cement (OPC)
22 mortars were studied using electrical impedance spectroscopy (EIS) and equivalent
23 circuit (EqC). Two successive processes, at 20 °C and 50 °C, consisting of several
24 drying-rewetting cycles, were applied to the mortars. After each cycle, the electrical
25 impedance and the amount of water absorbed were measured. The EIS-EqC
26 methodology allowed to find two distributed impedance relaxations, associated to
27 capillary and gel-C-S-H porosities, respectively. At room temperature any
28 microstructural change was not detected. Nevertheless, at 50 °C two microstructural
29 changes were inferred: 1) the volume of accessible porosity increased (pore
30 coarsening) and 2) the surface of the conductive path through C-S-H gel became more
31 conductive (surface smoothing).

32

33 **Key words:** mortar, gel-porosity, capillary-porosity, drying-rewetting, electrical-
34 impedance-spectroscopy.

35

36 1. INTRODUCTION

37 The physical properties of the pore network in hardened cement-based materials
38 (HCB), such as pastes, mortars and concretes, determine their fundamental
39 engineering properties, such as mechanical strength and durability. The pore network
40 can be characterized by a complex function of its pore size, pore shape, pore surface
41 area, volume fraction of pores, connectivity between pores and water saturation level.
42 In mature HCB materials, the pores that contain non-bound liquid water are classified
43 into three classes of porosity, with decreasing size and with different pore shape: i)
44 capillary porosity (> 8 nm in diameter) that includes inter-hydrate spaces 8-20 nm, ii)
45 gel porosity (large pores ≈ 8-4 nm in diameter and small pores ≈ 4-2 nm in diameter)

46 and iii) interlayer porosity (< 2nm in width). Henceforth, gel porosity (GeP) will be the
47 volume of pores associated with the C-S-H gel, whose size is less than 8 nm, and
48 capillary porosity (CaP) will refer to the pores greater than 8 nm.

49 The transition between percolation and depercolation of CaP (communicated or
50 interrupted by the gel C-S-H, respectively) has implications on water curing, transport
51 properties and the durability response of HCB structures. The relationship between the
52 depercolated state and CaP volume has been studied in several ways [1]. The
53 investigation of removing water from HCB materials through drying-rewetting
54 processes is a method for assessing the durability, and is also a useful approach for
55 characterizing different parameters of the pore structure [2].

56 The intensity of drying depends on the temperature, ambient relative humidity (RH%)
57 and duration of the process. Each drying intensity affects the porosity down to a certain
58 pore size, but the movement of water in these pores alters the redistribution of water in
59 the pores of smaller sizes [3]. Some important features about water displacement and
60 changes in the porosity of HCB materials, subjected to drying processes, are reported
61 in the literature. The most remarkable results at different temperatures are:

- 62 1) Drying at 60 °C:
 - 63 a. the cement paste exhibited a coarsening of capillary porosity (increasing the
64 mean size pore) and a collapse of low-density C-S-H gel [4-6].
 - 65 b. for 14 days, the mortar lost the water of capillary and gel porosity, but not that
66 of the interlayer porosity [3].
 - 67 c. the mortar removed a large fraction of water from the interlayer porosity, but
68 did not at 40 °C [7].
- 69 2) Oven-drying of HCB materials at 105 °C for 24 hours was used as a reference
70 method for removing completely the non-bound water [8,9].
- 71 3) Drying at room temperature and relative humidity RH > 25% was shown to be a
72 reversible process regards to the water content, because the water in the

73 interlayer pores did not move and the water in gel and capillary pores could
74 return after saturation [10]. Drying at room temperature and RH = 0%, and
75 subsequent re-saturating, made the gel particles closer [11].

76 Several methods have been used to measure changes in the porosity of HCB materials
77 after drying treatments:

- 78 1) H-NMR (nuclear magnetic resonance) can evaluate quantitatively the
79 percentages of remaining water [8]. It is also possible to calculate the ratio
80 surface/volume of pores [12] and the gel pore size [3,7,13].
- 81 2) SANS (small-angle neutron scattering) gives a direct measurement of i) total
82 internal surface area accessed by mobile water, ii) volume fractal and iii)
83 surface fractal. The first fractal parameter is associated to the packing of C-S-H
84 gel particles, and the second one can be associated with the roughness of the
85 cement grains [2,11,14].
- 86 3) WVSI (water vapour sorption isotherms) allows to relate the mass water content
87 to the RH% conditions and the minimum size of saturated pores [6,9,10,15].
- 88 4) MIP (mercury intrusion porosimetry) gives a pore size distribution after a drying
89 process [16-18].
- 90 5) EIS (electrical impedance spectroscopy) has been used to evaluate the effect of
91 drying on microstructural changes in HCB materials, but only in a few articles
92 [19,20]. The pore coarsening due to drying treatment was related to the size of
93 the frequency arc in the impedance plot. The presence of an intermediate arc in
94 the impedance plot was associated with the formation of denser phases and
95 new interfacial regions between collapsing C-S-H surfaces. However,
96 quantitative analysis was not performed.

97 The main advantages of the EIS method are twofold: samples can be measured
98 without previous treatments, and the measurement process is non-destructive and non-
99 invasive. The usefulness of EIS (up to 1 MHz), when applied to saturated HCB

100 materials, is that their response shows the ionic conductivity in the pores, distinguishing
101 between the bulk space and the surface.

102 During the last 20 years, EIS in the range of 1 Hz to 1 MHz has been used in HCB
103 materials in order to relate the electrical macroscopic properties to the microstructure
104 [20-38]. In the following paragraphs a brief review related to the EIS technique and the
105 equivalent electrical circuit (EqC) method, applied to impedance data performed in a
106 two-electrode conductive cell, is presented.

107 The EqC method is based on the configuration of an electrical circuit with passive
108 electrical elements such as resistance R, capacitor C, and constant phase element Q.
109 These elements are connected in series, in parallel or in other arrangement, in order to
110 fit experimental impedance data to theoretical impedance of the circuit, in the
111 frequency range of the experiment.

112 The admittance of Q is frequency dependent: $Y(Q) = Y_0 (j \cdot 2\pi \cdot f)^n$, characterized by two
113 parameters Y_0 and n (where f is the frequency and j the complex imaginary unit). This
114 complex admittance has two components:

$$115 \quad \text{Re}Y(Q) = Y_0 \omega^n \cdot \cos\left(n \frac{\pi}{2}\right) \quad (1a)$$

$$116 \quad \text{Im}Y(Q) = Y_0 \omega^n \cdot \text{sen}\left(n \frac{\pi}{2}\right) \quad (1b)$$

117 Circuits with three branches in parallel have been proposed [39-42], being represented
118 as $(R_1 Q_1 [R_2 Q_2])$ following the circuit description code (CDC) [43]. This parallel circuit
119 represents three main phases of the material with different types of conductivity (with
120 respect to the applied voltage): resistive (R_1 , in phase), capacitive (Q_1 , out of phase
121 $n_1 \cdot \pi/2$ radians) and resistive-capacitive in series ($R_2 Q_2$, lagged $< n_2 \cdot \pi/2$ radians). This
122 three-branch circuit has the advantage that allows to identify three different phases of
123 the material and to monitor their evolution. However, its weakness is that other

124 alternative circuits with different number of branches in parallel also allow to explain the
125 bulk electrical conductivity.

126 Recently, a circuit with two Randles in series: $R_s (C_1 [R_1 W_1]) (C_2 [R_2 W_2])$ has been
127 applied [21,37,44,45], being W a Warburg element equivalent to a Q element with
128 $n=1/2$.

129 A general circuit for fitting the experimental data, without the need of a priori
130 assumptions, is a Voigt circuit with a certain number of pairs elements in parallel (RC)
131 connected in series. Any set of impedance data can be fitted to a circuit with sufficient
132 number of (RC) [46-49]. This circuit also serves to: i) check if the experimental data
133 fulfil the Kramers-Kronig relations, ii) obtain the time constant of each (RC) ($\tau = RC$)
134 and iii) estimate continuous distributions with some approximate results [46].

135 If the material system has continuous impedance relaxations, as it is the case of HCB
136 materials, a series circuit with different pairs of (RQ) elements can be used. Each (RQ)
137 represents a different distributed relaxation of the impedance [47-49]. The
138 characteristic time constant T of the distributed relaxation is defined as [22]:

$$139 \quad T = (R \cdot Y_0)^{(1/n)} \quad (2)$$

140 Some researchers found a single relaxation (RQ) in OPC mortars [26,50], and also in
141 cement pastes [51]. Other authors presented two relaxations, $(R_1 Q_1)(R_2 Q_2)$ in OPC
142 mortars [52,53], and even a single relaxation in series with a resistance has been
143 proposed, $R_1(R_2 Q_2)$ [54].

144 A common feature in these investigations with saturated mortars is the presence of a
145 distributed relaxation (RQ) with an exponent n of Q close to 0.80. This exponent n
146 represents the width of the distributed relaxation in the frequency dimension. The value
147 $n = 1$ corresponds to a narrow Debye relaxation and a decreasing value of n (from 1 to
148 0.5) means an increase in the width of the relaxation. The exponent $n = 0.80$ has been
149 related to: i) the fractal dimension of the C-S-H gel [26,50,51], ii) the ratio of

150 dielectric/conductive components in the admixture that constitutes the C-S-H gel
151 [42,55] and iii) the water confined in nanometer size pores [56]. Whatever its meaning
152 is, a change of n depicts a change in the structure of the gel C-S-H in HCB materials.

153 The main objective of this paper is to establish the methodology EIS-EqC to assess the
154 changes in microstructure of OPC mortars subjected to several cycles of drying at low
155 intensity conditions (20 °C and 50 °C) and subsequent resaturating.

156 Given the controversy about the number of relaxations in saturated mortars, and in
157 order to facilitate comparability of future studies, one specific objective of this work
158 aims to demonstrate that electrical conductivity of saturated OPC mortars can be
159 characterised by two impedance relaxations of type (RQ). This physical model is
160 validated with the applicability of the same EqC to all saturated states, the original and
161 the resultant states after the successive drying-rewetting cycles throughout the
162 experiment. The relation between the two relaxations and the main two porosities, CaP
163 and GeP, is discussed by analysing the amount of water removed-absorbed in the
164 drying-rewetting cycles.

165

166 2. EXPERIMENTAL

167

168 2.1 Materials

169 Mortar samples were prepared using ordinary Portland cement (OPC) of the type CEM
170 I 52.5R according to the composition, specifications and conformity criteria standard
171 UNE-EN 197-1:2011 [57].

172 To obtain mortars of different porosities, three different w/c ratios were used: 0.40, 0.50
173 and 0.60 labelled as m040, m050 and m060, respectively. The aggregate-to-cement

174 ratio (a/c) was 3/1 (62% in volume). Siliceous sand with a fineness modulus of 4.1 was
175 used as the aggregate.

176 Mortar specimens were prepared according to European standard UNE-EN 196-1:2005
177 [58]. Fresh mortar was cast into prismatic moulds measuring 4 x 4 x 16 cm³.
178 Specimens were cured at temperatures of $t = 20 \pm 2$ °C and a relative humidity RH >
179 99% for 24 hours. After demoulding, samples were immersed in a saturated lime
180 solution and were cured in this environment at 20 °C for 270 days. Three samples of
181 each mortar were prepared for impedance measurements.

182

183 2.2 Methods

184 2.2.1 Impedance measurements

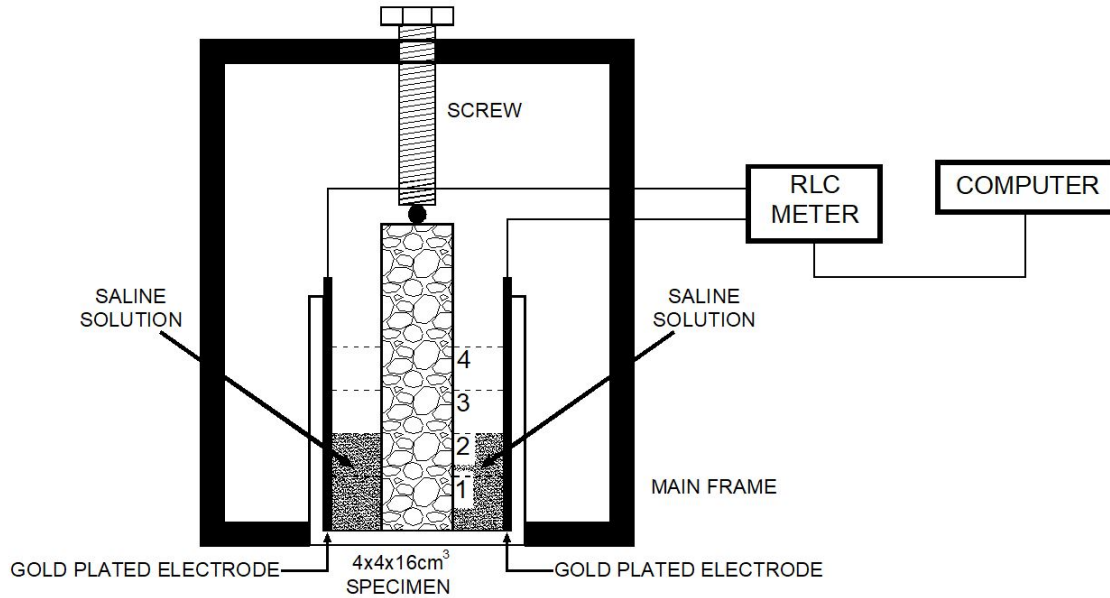
185 Mature mortars, 270 days old, were measured by EIS before and after each drying-
186 rewetting cycle, always in saturated state.

187 The electrical impedance measurements were performed with the impedance meter
188 HP-4284A (impedance measurement range 0.01 mΩ - 100 MΩ, basic accuracy 0.05%)
189 at a constant intensity of 100 μA, in the frequency range 0.1-1000 kHz. In this range,
190 59 measurements equidistant in logarithmic scale were taken.

191 A conductive measuring cell with two electrodes was used for impedance
192 measurements. Fig. 1 shows the schematic for the measurement method as it was
193 described in a previous paper [42].

194 Impedance measurements were performed in saturated samples. The prismatic
195 samples were settled vertically, the spaces between electrodes and two opposite faces
196 of mortar were filled with saline solution. The other two faces and the bottom of the
197 sample were sealed with silicone. The solution that was in equilibrium with the mortars
198 was placed as a conductive contact between the sample and the electrodes.

199 Four impedance measurements were taken at 1.5 cm, 3 cm, 4.5 cm and 6 cm in
 200 height. The admittance data (inverse of impedance) were adjusted linearly with the
 201 height of measurement to separate the parasitic admittance from the intrinsic
 202 admittance of the mortar.



203

204 **Fig. 1.** Schematic view of the two-electrode cell for EIS measurement. Four measuring heights
 205 1.5 cm, 3 cm, 4.5 cm and 6 cm are shown as 1, 2, 3, 4, respectively.

206

207 The resulting experimental admittance Y_{ex} was obtained per unit length. The
 208 impedance of the mortar Z is in series with the impedance due to the interface
 209 electrode-solution-mortar Z_{es} : $Z_{ex} = Z + Z_{es}$. Both impedances were separated by
 210 means of an EqC using LEVM software which applies a complex non-linear least
 211 squared fitting (CNLS) [60]. The value of the mortar impedance, Z , matches with the
 212 electrical resistivity because the geometrical factor resulting for Y_{ex} is the unit (1 m^{-1}).

213 2.2.2 Drying-rewetting procedure

214 Two successive processes, at room temperature ($20 \text{ }^\circ\text{C}$) and at $50 \text{ }^\circ\text{C}$, consisting of six
 215 drying-rewetting cycles, were applied to the mortars.

216 The first and second drying-rewetting cycles, lasting 7 and 14 days, respectively,
 217 involved an ambient-drying (AD) process, temperature $t = 20 \pm 2 \text{ }^\circ\text{C}$ and relative
 218 humidity $\text{RH} = (55 \pm 3)\%$, and a subsequent vacuum saturation.

219 The other four drying-rewetting cycles involved an oven-drying (OD) process at $50 \text{ }^\circ\text{C}$
 220 for circa 7 days, until constant weight, and subsequent vacuum saturation.

221 The vacuum saturation process was performed in two stages: a vacuum extraction
 222 during 30 minutes under pressure of 20 mmHg (2.7 kPa) and 1 hour of slow wetting
 223 until saturation with distilled water at the same pressure (UNE-EN 1936:2007) [59].

224 After the saturation process, mortar samples were immersed in distilled water for at
 225 least 24 hours before taking the impedance measurements, to ensure the equilibrium
 226 between the mortar and saline solution. This solution was used in the impedance
 227 measurement.

228 The nomenclature of the experiment is as follows: S_j , being $j=0,1,2,..6$ for each of the
 229 seven saturated states. The first saturated state, S_0 , corresponds to the original mature
 230 mortar of 270 days, vacuum saturated. The dried states are referenced as: XD_j , being j
 231 $= 1, 2,..6$ the number of drying-rewetting cycles, and $X = A, O$ depending on whether
 232 they result from drying at AD or OD conditions, respectively. Table 1 shows the
 233 nomenclature used during the experiment.

234

Process State	Ambient Drying (AD)			Oven Drying (OD)			
	Dried	AD1	AD2	OD3	OD4	OD5	OD6
Saturated	S0	S1	S2	S3	S4	S5	S6

235

236 **Table 1** Nomenclature of drying-rewetting states of the mortars and processes.

237

238 The water absorbed after saturation in each cycle (XD_j-S_j) was measured as a
239 percentage of sample volume W_a . The impedance measurements were taken in every
240 saturated state S0, S1, S2, S3, S4, S5 and S6.

241

242 3. RESULTS AND DISCUSSION

243

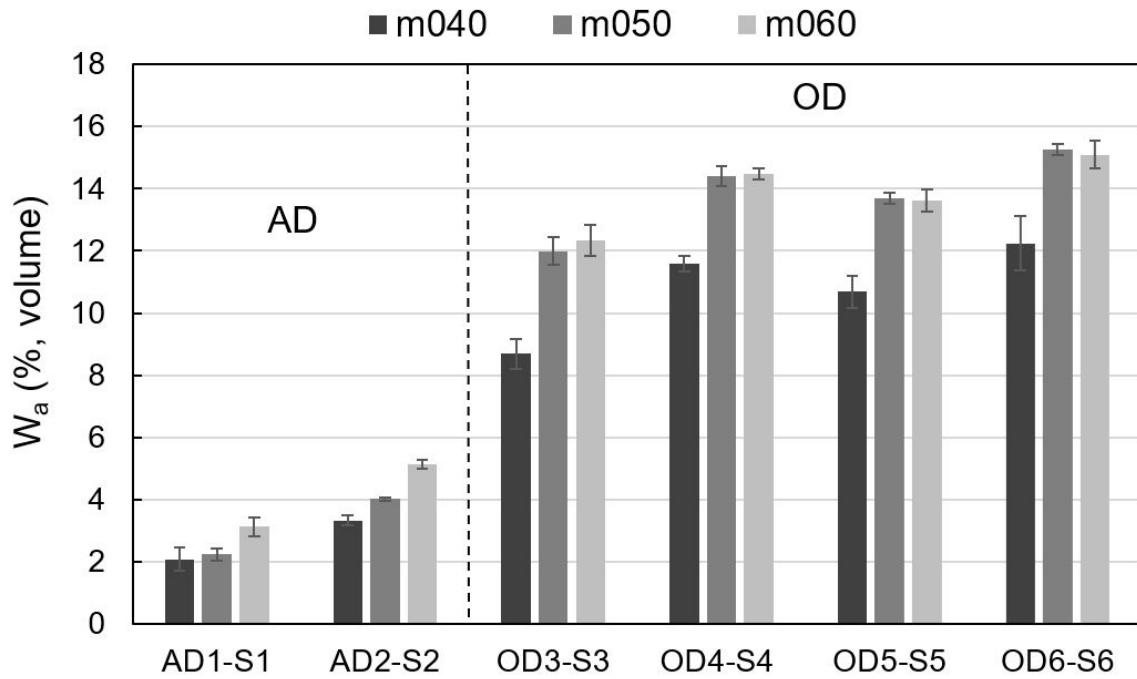
244 3.1 Water removed during the drying-rewetting cycles

245 Fig. 2 shows the amount of water absorbed W_a at each cycle of the process between
246 the dry state (XD_j) and the subsequent saturated state (S_j). In each cycle, XD_j-S_j ($j=1-6$
247 and $X=A,O$), the same drying strength for all mortars was applied, and all water inside
248 the pores above a minimum size was removed. The minimum accessible pore size
249 reached at OD conditions was smaller than at AD conditions.

250 The mass of the samples was approximately the same in all saturated states. In all
251 cycles, the same amount of water that was removed in the drying was recovered in the
252 re-saturating W_a , highlighting the good repeatability of the experiment. The mass in the
253 dry states decreased for successive cycles, and therefore W_a increased from AD1-S1
254 to OD6-S6 continuously. The continuous increase indicates that in each new cycle
255 water was absorbed into new porosity and that this new accessible porosity appeared
256 after the drying process.

257 In the two AD cycles, AD1-S1 and AD2-S2, W_a did not exceed 5% in any mortar,
258 although the repetition of the cycle with double time of drying (14 days) caused an
259 increase in W_a from 2.1%, 2.2%, and 3.1% up to 3.3%, 4.0%, and 5.1% for mortars
260 m040, m050, and m060, respectively (Table 2).

261



262

263 **Fig. 2.** Water absorbed W_a (percentage in volume) for OPC mortars with w/c = 0.40, 0.50 and
 264 0.60 in ambient drying (AD) cycles (AD1-S1, AD2-S2) and oven drying (OD) cycles (OD3-S3,
 265 OD4-S4, OD5-S5, OD6-S6).

266

267 By applying the Powers model [10] and assuming that: i) the degree of hydration of
 268 cement is 0.85, ii) the intrinsic porosity of the gel C-S-H is 26% and iii) the mortars
 269 have 42% of cement paste, the calculated capillary porosity (CaP) is 4%, 7%, and 9%
 270 (in volume) for mortars m040, m050, and m060, respectively (Table 2). Therefore,
 271 water of CaP was not completely removed during the AD process, only 83%, 57% and
 272 57% of Cap, for m040, m050 and m060, respectively, was involved in AD process.
 273 These values show a higher percentage of depercolated CaP in mortars m050 and
 274 m060 than in m040, because all water in accessible CaP (>8 nm) is removed at AD
 275 conditions [10].

276

277

278

Mortars	W_a (%)						CaP (%)	CaP+GeP (%)
	AD1-S1	AD2-S2	OD3-S3	OD4-S4	OD5-S5	OD6-S6		
m040	2.1	3.3	8.7	11.6	10.7	12.2	4	13
m050	2.2	4.0	12.0	14.4	13.7	15.2	7	17
m060	3.1	5.1	12.3	14.5	13.6	15.1	9	20

279

280 **Table 2** Water absorbed W_a (%) shown in Fig. 2, in ambient drying (AD) cycles (AD1-S1, AD2-
281 S2) and oven drying (OD) cycles (OD3-S3, OD4-S4, OD5-S5, OD6-S6). Calculated capillary
282 porosity CaP (%) and total porosity CaP+GeP (%) (Powers model) for the three mortars.
283 Percentages in volume. The uncertainties of W_a are shown in Fig. 2.

284

285 According to Jennings et al. [10], by applying the WWSI technique, the large gel pores
286 (4 to 8 nm in diameter) are emptied at $RH \approx 50\%$ and temperature of 24 °C. Therefore,
287 it is possible that some water of GeP, specifically the water of GeP located on the
288 surface of the sample, might have been removed during the AD cycles.

289 In the third cycle of the experiment (OD3-S3), which corresponds to the first cycle at 50
290 °C of OD process, W_a increased more than double, and reached up to 8.7% for m040,
291 12.0% for m050 and 12.3% for m060.

292 Returning to the Powers model, the calculated total porosity CaP+GeP was 13%, 17%,
293 and 20% (in volume) for mortars m040, m050, and m060, respectively (Table 2). These
294 values of estimated total porosity agree with data for $w/c=0.50$ mortars, whose loss of
295 water during OD treatment at 60 °C was 16% [3]. Therefore, in the first OD cycle the
296 water of GeP was not completely removed. Furthermore, it is possible that the isolated
297 capillary pores pertaining to depercolated CaP remain filled of water.

298 During the second OD cycle, OD4-S4, W_a increased with respect to the first, OD3-S3,
299 although the same intensity of drying was applied. This result indicates that the
300 repetition of the drying process induced rearrangements in the microstructure that
301 increased the accessible porosity for water [2,4,5,10,17].

302 A significant decrease in W_a between cycles OD4-S4 and OD5-S5 is observed in all
303 mortars. This result is meaningful because is directly related to a decrease in the
304 electrical conductivity G_2 , associated to GeP that will be discussed below in the section
305 about electrical properties. This decrease of W_a probably is due to the fact that drying
306 time did not last enough, and only a transient constant weight was reached.

307 A remarkable feature is that W_a shows the same value for m050 and m060 throughout
308 the OD treatment, in the range 12%-15%, well above the range 9%-12% for m040 (Table
309 2). Furthermore, at the end of the experiment, W_a reached values of 94%, 89% and 76%
310 of their total porosity, in increasing order of w/c: 0.40, 0.50 and 0.60. These results
311 confirm that mortars with higher w/c ratio have more volume of inaccessible small pores
312 in GeP, and this larger quantity of small pores increases the depercolated CaP;
313 therefore, the OD treatment could not remove the inaccessible water contained in both
314 spaces. Mature m060 has more porosity inaccessible than m050 but the accessible
315 porosity throughout OD process at 50 °C is the same.

316 The increase of W_a in the AD process was not due to a change in the microstructure, as
317 will be seen in the section on electrical properties, but to a greater emptying of the CaP
318 due to an increase in drying time.

319 The increase of W_a from OD2-S2 to OD3-S3 was due to the increase in the intensity of
320 drying from 20 °C to 50 °C and the consequent decrease in the minimum size accessible
321 pore.

322 The successive increase in W_a from the first OD3-S3 to the last OD6-S6, was due to
323 the repetition of the cycle: drying (50 °C) - rewetting, that caused the successive
324 shrinkage of the solid part of the mortar (small pores of gel C-S-H) and the increasing
325 in size of neighbour pores (largest pores of GeP and smallest pores of CaP). These
326 new pores that exceeded the minimum accessible pore size at 50 °C, were generated
327 in a drying and were filled in the following rewetting. Therefore, W_a increased with the

328 number of cycles, showing the effect known as pores coarsening [2,4,17]. The
329 successive OD cycles at 50 °C modified the microstructure of the gel C-S-H as it will be
330 demonstrated by the analysis of electrical properties.

331

332 3.2 Impedance data and proposed EqC

333 Fig. 3(a) shows, as example of data set, the Z-plot (Nyquist plot) of the experimental
334 impedance Z_{ex} for mortar m060 in the seven saturated states (S0-S6) of the
335 experiment.

336 Every plot shows a small tail on the right, representing the impedance of interface
337 between the electrodes and the sample at low frequencies, in the range 100 Hz - 5
338 kHz. The rest of the graph, in the range of frequency from 5 kHz to 1 MHz, displays two
339 sub-arcs representing one complete relaxation and the beginning of a second one. An
340 increase in both $\text{Re}(Z_{ex})$ and $|\text{Im}(Z_{ex})|$ is observed from S0 to S6. Furthermore, the
341 minimum value of $|\text{Im}(Z_{ex})|$ increases with the number of cycles. This feature makes
342 essential a fitting to an EqC in order to separate the electrode effects and identify the
343 real part of Z at zero frequency or dc conditions.

344 Fig. 3(b) shows the series EqC: $R_{sol} Q_e (R_1 Q_1) (R_2 Q_2)$ for fitting the experimental
345 impedance data. The right part of the circuit $R_{sol} Q_e$ corresponds to the impedance at
346 the electrode-solution interface Z_{es} , simulating the tail at low frequencies (Fig. 3(a)).
347 The left part of EqC represents the impedance of the mortar in the frequency range of
348 the experiment (100 Hz - 1 MHz) with two distributed relaxations of type (RQ)
349 composed by two electrical resistances, R_1 and R_2 , and two constant phase elements,
350 Q_1 and Q_2 . Each Q is characterized by two electrical parameters: the frequency
351 exponent, n , and the admittance factor, Y_0 .

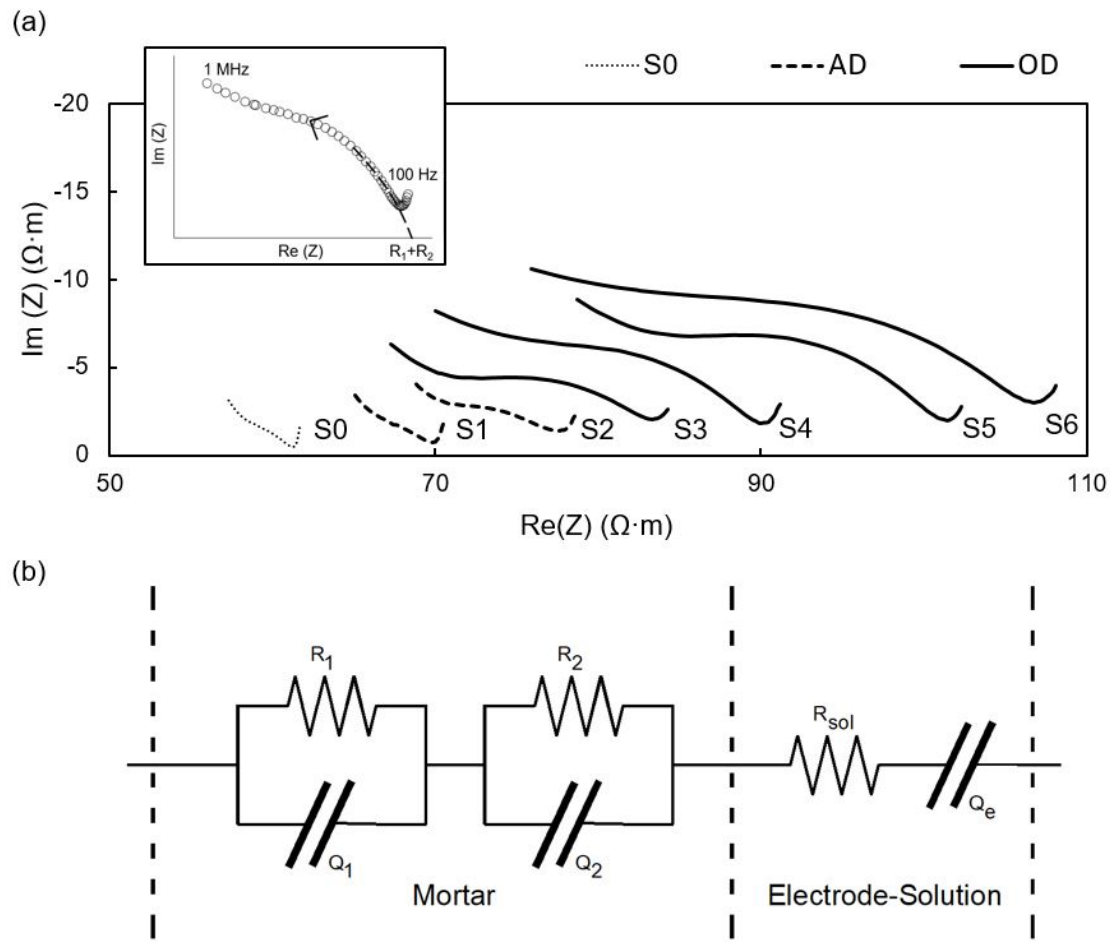
352 Kramers-Kronig requirements were tested by fitting impedance data to a Voigt circuit
353 with seven pairs (RC) in series, prior to adjusting the impedance data to EqC. Only
354 those data that exhibited a random scatter of residuals were selected for the analysis.

355 The calculated statistical parameters of fitting for both, Voigt circuit and two-distributed-
356 relaxations EqC (Fig. 3(b)), for every data set in the experiment were:

357 1) root mean square values of the estimated relative standard deviation of the fit
358 residuals (PDRMS) that was less than 0.021, 2) residuals with respect to the model
359 (Res/Mod) that were less than 5%, and 3) relative standard deviation of the fit residuals
360 (SF) that was less than 0.016.

361 These good statistical parameters indicate that the proposed electrical model, with
362 (R_1Q_1) and (R_2Q_2) in series, represents appropriately the electrical impedance of the
363 saturated mortar, both in its original mature state and after the drying-rewetting cycles.

364



365

366 **Fig. 3.** Impedance data and equivalent circuit (EqC): (a) Nyquist plot of Z_{ex} ($Re(Z_{ex})$ vs $Im(Z_{ex})$)
 367 for m060 in their original saturated state S0 and in their saturated states: S1, S2 (ambient
 368 drying, AD), and S3, S4, S5, S6 (oven drying, OD), (b) EqC composed of two pairs (R_1Q_1),
 369 (R_2Q_2) in series with two elements R_{sol} Q_e . Inset: Any curve of Nyquist plot. Bulk resistance
 370 R_1+R_2 is the intersection of the mortar curve at low frequency with the $Re(Z_{ex})$ axis. Arrow
 371 points towards increasing frequency.

372

373 A key point of IES studies is the microstructural interpretation of the electrical results.

374 The bulk electrical resistance of the mortar is R_1+R_2 (Inset of Fig. 3(a)). In Nyquist plot
 375 $Re(Z_{ex})$ decreases from R_1+R_2 to $Re(Z_{ex}(1 \text{ MHz}))$ with frequency. This decrease
 376 means an impedance relaxation. $Im(Z_{ex})$ increases and decreases with frequency. The
 377 maxima of $|Im(Z_{ex})|$ show the characteristic frequencies F around which relaxations
 378 spread. Only if these characteristic frequencies are significantly different they can be
 379 identified by adjusting the EqC.

380 The impedance and its inverse, the admittance, are complex magnitudes that
381 henceforth are equivalent to resistivity and conductivity, respectively, because the
382 geometric factor of the sample that represents Z_{ex} is the unit.

383 The two relaxations (R_1Q_1) and (R_2Q_2) have their characteristic frequency F_1 and F_2 ,
384 respectively, around which different ionic conduction phenomena occur. The
385 frequencies F_1 and F_2 are related to the characteristic time constants T_1 and T_2 ,
386 defined in Eq (2), with the relationship $2\pi \cdot T_i \cdot F_i = 1$.

387 Each relaxation is characterized by 3 parameters: R , Y_0 and n . According to the
388 literature on EIS applied to HCB materials, the only electrical parameter that
389 characterizes the impedance relaxation of gel C-S-H is the exponent n with values
390 around 0.80 [26,50,51].

391 Furthermore, the admittance of each relaxation (RQ) is defined as the sum of two
392 admittances in parallel $Y(R)$ and $Y(Q)$. The admittance $Y(Q)$ in turn is the sum of its
393 real and imaginary parts defined in Eq (1). Therefore, the ionic conductivity associated
394 with each relaxation has three additive components: 1) $G (= 1 / R)$ is in phase with the
395 applied voltage and independent of the frequency of the applied voltage, 2) $Re(Y(Q))$ of
396 Eq (1a), is in phase with the voltage but dependent on the frequency, and 3) $Im(Y(Q))$
397 of Eq (1b), is out of phase $\pi / 2$ rad with the voltage and also dependent on the
398 frequency.

399 These properties of the conductivities allow to associate them to different spaces of the
400 microstructure. Conductivity G is related to the bulk solution of the pore space, and
401 $Re(Y(Q))$ and $Im(Y(Q))$ are related to the surface of the pores.

402 According to the literature, the dispersive elements Q_1 and Q_2 represent the conduction
403 in the surface of the pore, which is due to diffusion and polarization in the
404 electrochemical double layer close to the solid wall of the pores [26,61-63]

405 Changes in these electrical parameters will allow to relate them to changes in the
 406 microstructure.

407

408 3.3 Equivalent circuit parameters in the original saturated state S0

409 Table 3 shows the parameters of EqC (Fig. 3(b)) obtained in the state S0 for the three
 410 mortars (mean value of 3 samples). The parameters (R_1 , Y_{01} , n_1) and (R_2 , Y_{02} , n_2)
 411 characterize the two mortar relaxations. Conductivities G_1 , G_2 and bulk dc conductivity
 412 of the mortar G_{eq} ($= 1/(R_1+R_2)$) are also shown in Table 3.

413 Being $n_2 = 0.80-0.83$ and $G_2 \ll G_1$, relaxation (R_2Q_2) can be associated to GeP and
 414 therefore (R_1Q_1) to CaP.

415 Moreover, G_{eq} is in the range of 10-17 mS/m for the three mortars, with their original
 416 pore solution. This value is much lower than 150 mS/m, accepted limit value below
 417 which the capillary porosity is depercolated [1]; and therefore it can be assumed that
 418 the two porosities, CaP and GeP, were arranged in series throughout the path of the
 419 ionic species.

420

S0	Q ₁			Q ₂			G ₁ (mS/m)	G ₂ (mS/m)	G _{eq} (mS/m)
	R ₁ (Ω·m)	Y ₀₁ (Ω ⁻¹ ·s ^{n₁})	n ₁ (-)	R ₂ (Ω·m)	Y ₀₂ (Ω ⁻¹ ·s ^{n₂})	n ₂ (-)			
m040	0.96	4.18·10 ⁻³	0.496	93.24	5.79·10 ⁻⁹	0.803	1042	10.7	10.6
m050	3.14	2.63·10 ⁻³	0.463	67.61	5.23·10 ⁻⁹	0.825	318	14.8	14.1
m060	4.13	2.96·10 ⁻⁴	0.555	57.08	6.38·10 ⁻⁹	0.823	242	17.5	16.3

421

422 **Table 3** Mean values (three samples for each mortar) of the EqC parameters: R_1 , Y_{01} , n_1 (of
 423 Q_1), R_2 , Y_{02} , n_2 (of Q_2) obtained for every mortar at the beginning of the experiment S0 (original
 424 saturated 270-day mortar). Calculated dc conductivities: G_1 ($=1/R_1$), G_2 ($=1/R_2$) and bulk
 425 conductivity $G_{eq} = (R_1+R_2)^{-1}$. The uncertainties of the parameters were about 3% for Y_{02} , 1% for
 426 R_1 and Y_{01} , 0.2% for n_1 and n_2 , and 0.05% for R_2 .

427

428 Resistances R_2 and R_1 are associated to GeP and CaP, respectively. The resistance
429 R_2 were 15, 20 and 90 times higher than R_1 for m060, m050, and m040, respectively.
430 These resistances represent the ionic conduction within the volume of the pore and
431 they are related to: 1) the shape and dimensions of the pore, 2) the volume of mortar
432 occupied by the pores and 3) the dc conductivity of the solution that saturates the
433 pores. The relationship between the resistances (R_1 , R_2) and the pores size that they
434 represent can not be established only with electrical measurements. It would be
435 necessary to know the dc conductivity of their respective solutions.

436 The parameter Y_{02} was 5-6 orders of magnitude smaller than Y_{01} . They represent the
437 amplitude of conductivity at the interface for $\omega^n \sim 1$ (about 1 Hz for both, Q_1 and Q_2).
438 Exponent n_2 was about 0.80 and n_1 was about 0.50. With de values obtained for Y_0
439 and n , in state S0, the conductivity at the pore surface for all mortars ($Y_0 \cdot \omega^n$) was about
440 10^3 times greater in the pores represented by Q_1 than in those represented by Q_2 , in
441 the frequency range from 100 Hz to 1 MHz.

442 The large difference in conductivity associated with the surface of the pores reaffirms
443 the idea that there are two types of conductive pores with very different surfaces.
444 Element Q_2 corresponds to GeP associated to gel C-S-H and Q_1 is related to CaP.

445 In HCB materials, an impedance relaxation with an exponent n in the range 0.80-0.82
446 has been reported [26,50,51]. It has been related to the fractal surface dimension d_f of
447 the C-S-H gel by means of the equation $d_f = 2 \cdot n_2 + 1$ [64]. For the three mortars in the
448 original state S0, with $n_2 = 0.80-0.83$, the calculated fractal parameters were $d_f = 2.60-$
449 2.66 that are in agreement with those measured by other techniques for the C-S-H gel
450 [5,18,65].

451 The characteristic time constant T_i of the ($R_i Q_i$) relaxation, displayed in Eq (2),
452 indicates the characteristic frequency $F_i = (T_i \cdot 2\pi)^{-1}$ at which the imaginary part of the
453 conductivity reaches approximately the same value as the real part.

454 Applying Eq (2) for (R_1Q_1) and (R_2Q_2) data, the characteristic time constants were $T_1 \approx$
455 10^{-5} s and $T_2 \approx 10^{-8}$ s. The estimated characteristic frequencies were $F_1 \approx 16$ kHz and
456 $F_2 \approx 16$ MHz. The frequency F_2 determined by the parameters of the relaxation (R_2Q_2)
457 although exceeded the highest frequency applied in the experiment and despite
458 representing an incomplete relaxation, agrees with data presented in other articles
459 [54,66].

460

461 3.4 Evolution of EqC parameters

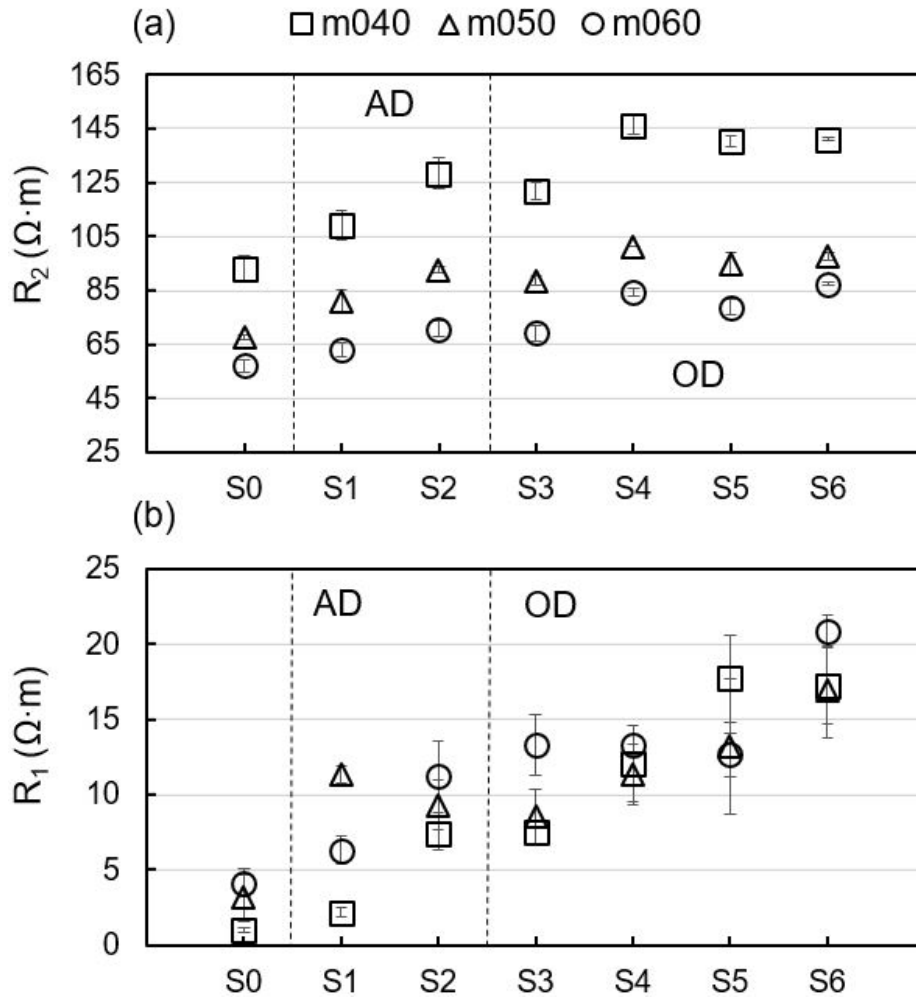
462 In this section, the evolution of the EqC parameters between the saturated states S0
463 and S6 are analysed for all mortars.

464 Fig. 4 depicts the values of R_2 and R_1 , the resistances associated to GeP and CaP,
465 respectively. The resistance R_2 exhibited significant differences between mortars, in
466 increasing order: m060, m050 and m040. Therefore, R_2 behaves like a parameter that
467 characterizes the w/c ratio. Resistance R_1 had greater uncertainty than R_2 and did not
468 show significant differences between mortars throughout the experiment.

469 In the AD process, from the original state S0 to S2, both R_1 and R_2 increased. There is
470 a linear relationship between them with determination coefficients higher than 0.85
471 (graphs not shown), indicating that their increase was due to the same effect.

472 Both resistances increased in the AD process because the rewetting process was
473 performed with distilled water. The conductivity of the new solution was lower than the
474 original one because some of the precipitated salts did not dissolve. If the saturation
475 had been done with the original saline solution, the resistances would have decreased
476 [20].

477



478

479 **Fig. 4.** Resistance R_2 (a) and R_1 (b) for all studied OPC mortars ($w/c=0.40-0.50-0.60$) at the
 480 initial state of saturation, S0, and after successive drying-rewetting cycles. Saturated states S1,
 481 S2 (ambient drying, AD) and S3, S4, S5, S6 (oven drying, OD). The units of R_1 and R_2 are $\Omega \cdot m$
 482 because the geometrical factor of experimental impedance Z_{ex} is $1 m^{-1}$

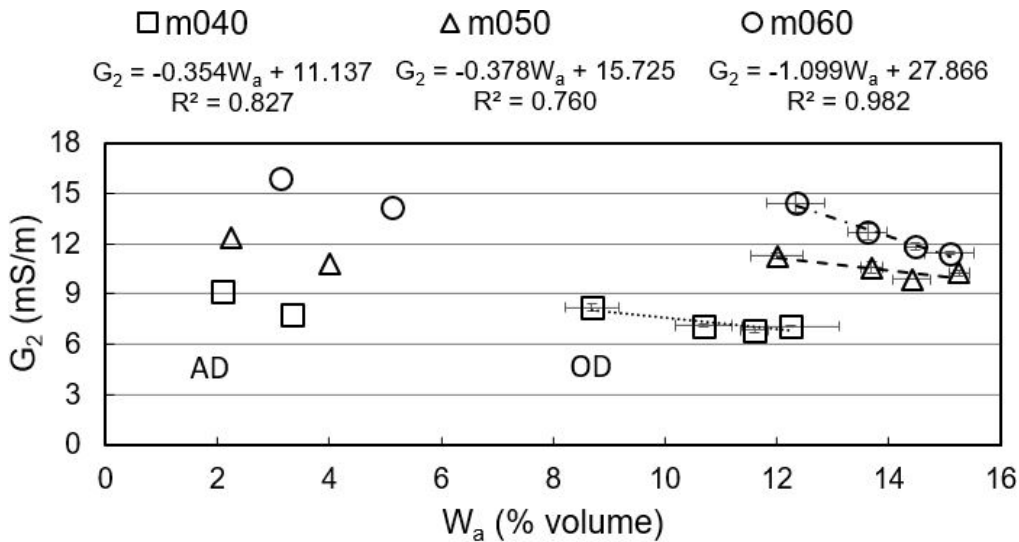
483

484 The resistance R_2 , associated to GeP, also increased in the AD process, despite none
 485 or very few water of GeP was removed in the AD cycles ($W_a\% < CaP\%$ in AD
 486 process). Nevertheless, this increase of R_2 indicates that the presence of distilled water
 487 in CaP affected the solution in pores of GeP after the vacuum-saturation process.

488 Between the states S2 and S3 there was no significant variation of R_2 , although in S3 a
 489 considerable increase in W_a was observed (Table 2). Resistance R_2 peaked in S4 for
 490 every mortar, and it hardly exhibited any change in the subsequent states, while R_1
 491 increased continuously.

492 No relationship was observed between R_1 or G_1 and W_a , but it was found between G_2
 493 and W_a . Fig. 5 shows this relationship between the conductivity G_2 associated to GeP
 494 and water absorbed W_a .

495



496

497 **Fig. 5.** Relationship between conductivity G_2 and water absorbed W_a for all mortar in saturated
 498 states S1, S2 (ambient drying, AD) S3, S4, S5 and S6 (oven drying, OD). Linear equations G_2 -
 499 W_a and the determination coefficients R^2 for each mortar in the four OD cycles.

500

501 In the two processes, AD ($W_a < 6\%$) and OD ($W_a > 8\%$), the conductivity associated to
 502 GeP decreased with W_a with a similar behaviour. This suggests that the electrical
 503 conductivity of the pore solution decreased in both processes. In the OD cycles, the
 504 conductivity of mortar decreased linearly with W_a , as shown in the linear fittings (Fig.
 505 5).

506 However, between the last AD cycle and the first OD cycle, there was a large increase
 507 in W_a , 7% for m060, 8% for m050 and 5% for m040, without a decrease in the
 508 conductivity. This confirms the idea that in the first OD cycle at 50 °C, some structural
 509 changes were produced in the gel C-S-H, compensating the decrease in the
 510 conductivity of the solution.

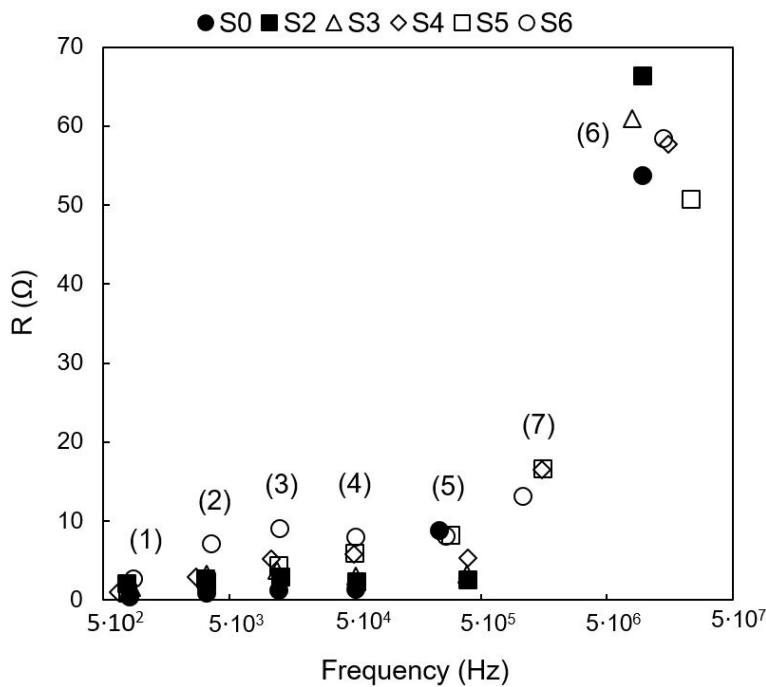
511 This behaviour agrees with the coarsening of pores involved in the conductivity [4-6,
 512 17]. The smallest pores got closer due to the shrinkage of the C-S-H gel and neighbour
 513 pores became larger, which increased the accessible porosity. These largest pores
 514 provided a new path for the ionic current.

515 From S3 to S6 the coarsening continued but did not compensate the decrease in
 516 conductivity of pore solution.

517 The absence of relationship between R_1 and W_a may be due to the fact that R_1 can be
 518 related to different spaces of the mortar matrix, among them the zone that surrounds
 519 the aggregates (interfacial transition zone, ITZ), which presents a wide variety in size
 520 and shape.

521 Fig. 6 shows the resistance R of the seven (RC) of the Voigt circuit as a function of the
 522 characteristic frequency of each single relaxation (RC). The values correspond to
 523 mortar m060 in six saturated states. The circuit was applied to check Kramers-Kronig
 524 conditions.

525



526

527 **Fig. 6.** Resistance R of seven (RC) of Voigt circuit (1) - (7) versus characteristic frequency of
528 each (RC). Values for m060 in six saturated states. Full symbols for S0 and ambient drying (AD)
529 cycle (S2) and open symbols for oven drying (OD) cycles (S3, S4, S5, S6). The (RC) number
530 (7) appears only in the last states S4, S5 and S6.

531

532 The largest resistances were obtained for (RC) number (6), with values between 50 Ω
533 and 70 Ω . This range of values is the same as that of the resistance R_2 of (R_2Q_2),
534 which was related to GeP (60-85 Ω in Fig. 4 for m060). These values of R correspond
535 with single relaxations whose characteristic frequency is between 10 MHz and 20 MHz,
536 around $F_2 \approx 16$ MHz which was assigned to the distributed relaxation (R_2Q_2). This
537 result confirms the existence of a frequency of relaxation above the frequency range of
538 measurements that characterizes a homogeneous phase, the gel C-S-H.

539 Conversely, in the frequency range $5 \cdot 10^2$ - $5 \cdot 10^5$ Hz, five Debye relaxations around the
540 characteristic frequency $F_1 \approx 16$ kHz of (R_1Q_1) were found. If their five resistances
541 determine R_1 of distributed relaxation (R_1Q_1), which was associated to CaP, it can be
542 state that this CaP represents a heterogeneous phase. This may be the reason
543 because R_1 is not directly related to the W_a , unlike it is R_2 .

544 In Fig. 7 the exponents n_2 and n_1 of Q_2 and Q_1 , associated to GeP and CaP,
545 respectively, are shown.

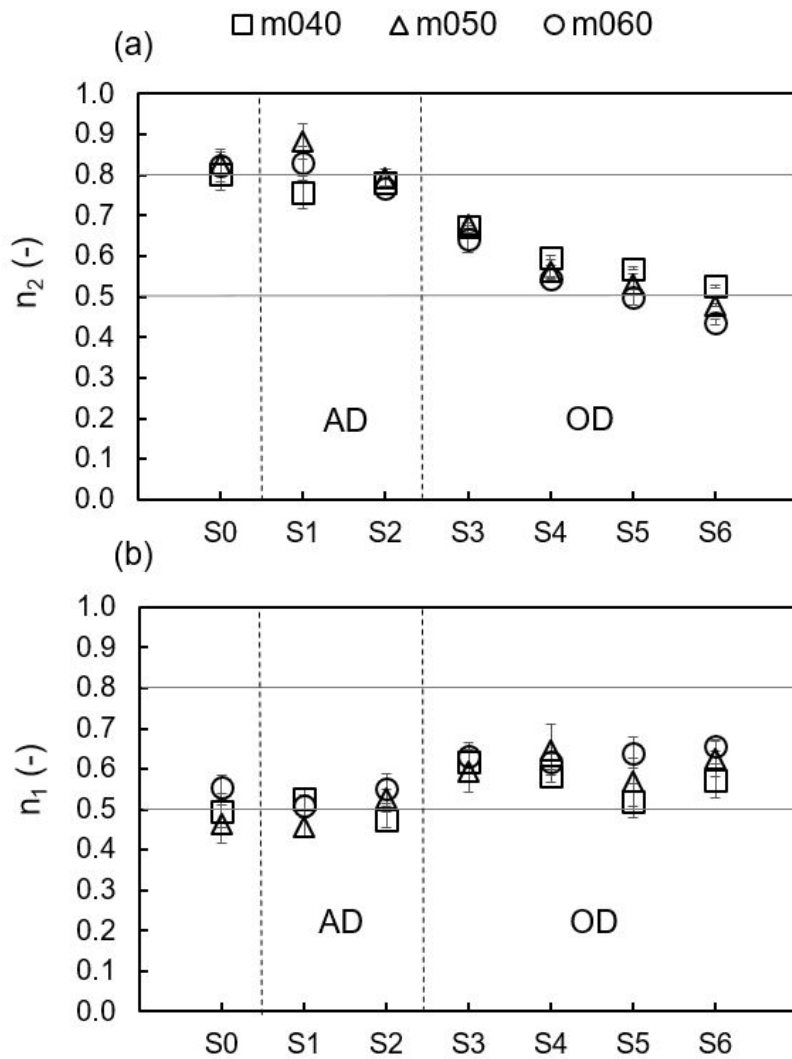
546 Exponent n_2 shows a similar value for all mortars throughout the experiment, with a
547 small uncertainty. This exponent changed in the same way for all mortars, indicating
548 that the surface of GeP pores suffered the same change in the OD process, regardless
549 of the w/c ratio.

550 In the two states of AD process, S1 and S2, the exponent n_2 was around 0.80, the
551 same value as in the original state, S0. This result indicates that the surface of the
552 pores of the C-S-H gel did not change during the AD cycles. This behaviour is in
553 accordance with the fact that the volume of water removed in the AD process (3.3%,
554 4.0% and 5.1% for m040, m050 and m060, respectively), was smaller than the free

555 water of CaP (4%, 7% and 9%, respectively), and that the AD rewetting process did not
556 affect the surface of the GeP pores.

557 During the OD process, n_2 decreased monotonously from 0.80 to a value around 0.50.
558 This indicates, from the electrical point of view, that the width of the relaxation (R_2Q_2)
559 increased. From the physical point of view, the conductivity in the ionic layer close to
560 the wall of the pores depends to a lesser extent on the frequency (Eq (1)). The fractal
561 parameter d_f associated to n_2 decreased from 2.60 to 2.00. This decrease is in
562 agreement with the variation of the number of SiO_4 tetrahedral chains linked to the
563 CaO sheets of the C-S-H gel [65] and it is related to the polymerization of the gel that is
564 a consequence of thermal treatments [67]. In other studies where a decrease of n_2
565 from 0.80 to 0.60 was observed, the pore solution was replaced by NaCl solutions, and
566 likely the effect of Na^+ on Ca-sheets of gel C-S-H was similar to the one observed in
567 the OD process [41].

568



569

570 **Fig. 7.** Evolution of Q_1 and Q_2 exponents (a) n_2 , (b) n_1 during the successive drying-rewetting
 571 cycles for every type of studied OPC mortar ($w/c = 0.40-0.50-0.60$). Initial state S0 and
 572 successive states S1, S2 (ambient drying, AD) and S3, S4, S5, S6 (oven drying, OD).

573

574 Exponent n_1 also shows a value similar to all mortars throughout the experiment.

575 However, this behaviour is very different from n_2 . In the first states S0, S1 and S2 (AD),

576 n_1 showed a value around 0.50. The constant value of n_1 in AD process indicates no

577 effects or reversible effects on the surface of the CaP. In the first cycle of the OD

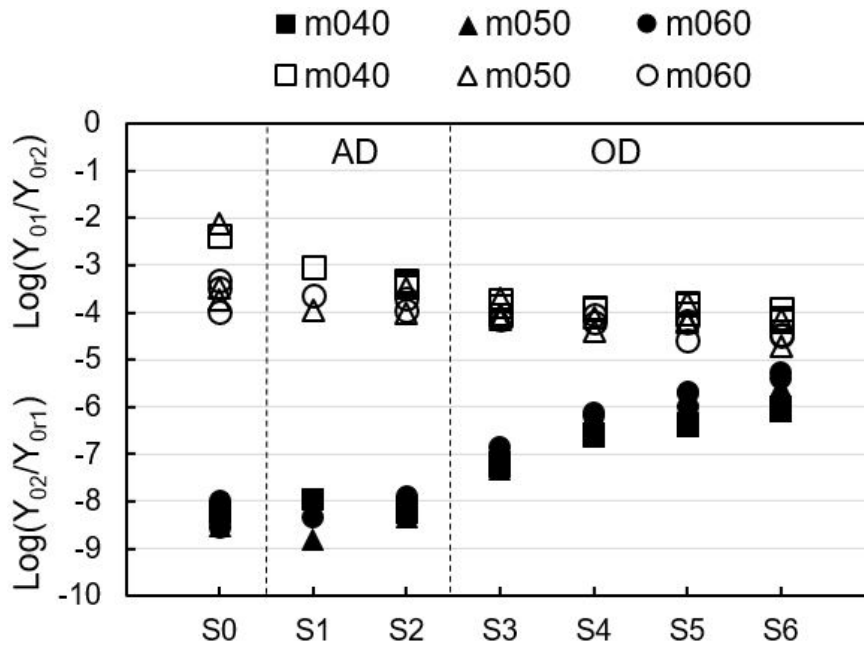
578 process, n_1 increased to values around 0.60 and remained constant until the end.

579 The effect of the drying-rewetting process was different for the two relaxations. The

580 repetition of the OD cycles did not affect significantly the n_1 exponent associated to

581 CaP, unlike the n_2 exponent associated to GeP, which decreased continuously.

582 The values of Y_{02} and Y_{01} , corresponding to Q_2 and Q_1 , associated to GeP and CaP,
 583 respectively, are depicted in Fig. 8.



584

585 **Fig. 8.** Evolution of $\log(Y_{01}/Y_{0r})$ and $\log(Y_{02}/Y_{0r})$ during successive drying-rewetting cycles for
 586 every studied OPC mortar ($w/c = 0.40-0.50-0.60$). $Y_{0r1} = Y_{0r2} = 1$ ($\Omega^{-1}\cdot s^n$) is a reference value
 587 (for units $n = n_1, n_2$, respectively). Original saturated state S0, S1 S2 (ambient drying, AD), and
 588 S3, S4, S5, S6 (oven drying, OD). Full symbols for Y_{02} and open symbols for Y_{01} .

589

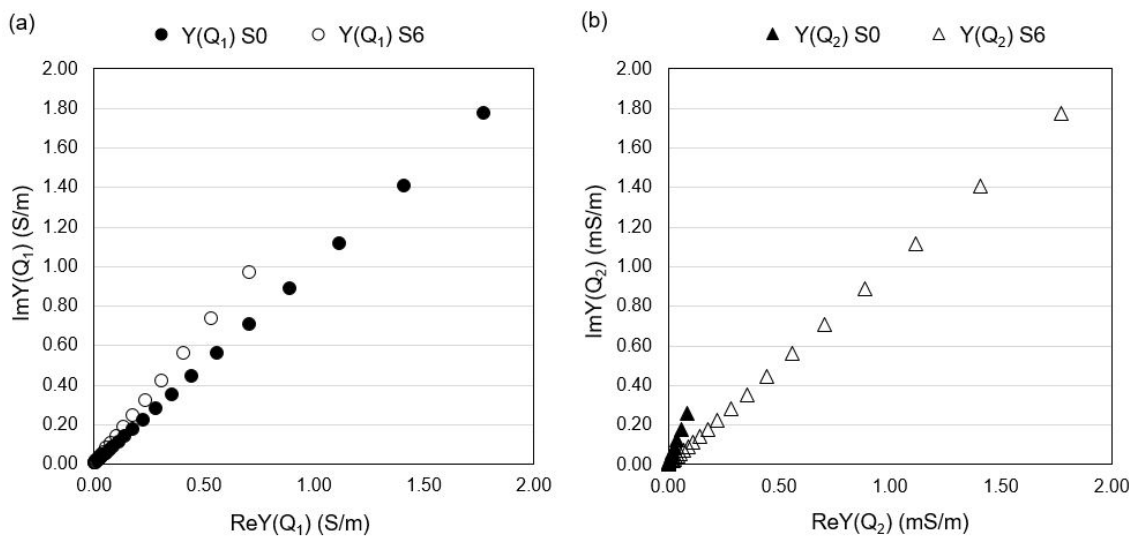
590 In the AD process (states S1, S2), the values of Y_{02} and Y_{01} hardly changed, having a
 591 difference about 4-5 orders of magnitude between them. In the OD process, Y_{01}
 592 decreased slightly, around one order of magnitude, whereas Y_{02} increased 2-3 orders
 593 of magnitude.

594 The evolution of surface conductivities, between the original state S0 and the last state
 595 of the experiment S6, is summarized in Fig. 9. Calculated values for m050 are shown
 596 as an example.

597 Both, real and imaginary part of admittance (conductivity) are shown, between 100 Hz
 598 and 1 MHz. The conductivity represented by Q_1 is 1000 times greater than Q_2 (units in
 599 S and mS, respectively).

600 The decrease in Y_{01} and increase in n_1 result in a decrease of the surface conductivity
 601 associated to CaP by approximately 50% (Fig. 9 (a)). The increase in Y_{02} and the
 602 decrease in n_2 result in a rise of the surface conductivity associated to GeP
 603 approximately 10 times from the initial S0 to last state S6 (Fig. 9 (b)). This increase in
 604 surface conductivity is the consequence of a smoothing of the pores surface in GeP.
 605 These pores are the result of the coarsening effect.

606



607

608 **Fig. 9.** Imaginary part of admittance $\text{Im}(Y(Q))$ versus real part $\text{Re}(Y(Q))$ (or conductivity) for Q_1
 609 (a) and Q_2 (b), in the range from 100 Hz (lower values) to 1 MHz (higher values), in the first
 610 saturated state S0 (full symbols) and the last saturated state S6 (open symbols). Calculated
 611 values with m050 data.

612

613 The continuous evolution of the electrical parameters R_1 , R_2 , n_1 , n_2 , Y_{01} and Y_{02}
 614 demonstrates the suitability of the electrical model used to explain the electrical
 615 conductivity of OPC mortars. In addition, the physical meaning of the parameters that
 616 define the two relaxations is conserved, and the changes that occurred during the
 617 drying-rewetting processes detected by EIS-EqC methodology agree with the changes
 618 of the microstructure deduced from W_a measurements.

619

620 4. CONCLUSIONS

621 The main conclusions obtained from the results previously discussed in this work are:

622 1) The electrical impedance spectroscopy (EIS), in the range from 100 Hz to 1 MHz,
623 enabled detection of changes in the porosity of the mortars subjected to drying-
624 rewetting cycles and distinguished between mortars with different w/c ratios.

625 2) The physical model that explains the electrical conductivity of saturated mortars with
626 two distributed relaxations of type (RQ) is consistent because it explains the
627 conductivity of mortars in the original state, and after they were subjected to successive
628 drying-rewetting cycles, at two drying levels: 20 °C (room temperature) and 50 °C.

629 3) Although the two relaxations can not be directly related to pores size, because the
630 conductivities of the respective solutions are unknown, electrical parameters of the
631 proposed equivalent circuit allowed to identify two different conductivities associated to
632 gel and capillary porosities.

633 4) The drying at ambient temperature produced reversible changes in the mortars,
634 while the drying effects at 50 °C were irreversible. The changes in electrical parameters
635 measured with EIS-EqC methodology agree with the amount of water absorbed and
636 with the microstructural changes reported by other methods.

637 5) In the drying-rewetting process at room temperature, only the conductivity of the
638 pores solution changed. In the drying-rewetting cycles at 50 °C, the volume of pores
639 detected at 50 °C increased monotonously (pore coarsening) and the surface of C-S-H
640 gel became continuously more conductive (surface smoothing).

641

642

643

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647

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