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Additional Information

1	Elastic and thermodynamic properties of α -Bi ₂ O ₃ at high pressures:
2	study of mechanical and dynamical stability
3	
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Abstract

31 The elastic and thermodynamic properties of the monoclinic polymorph of bismuth oxide (α -Bi₂O₃); aka mineral bismite, have been theoretically investigated both at room 32 pressure and under hydrostatic compression by means of first principles calculations 33 based on density functional theory. In this work, the elastic stiffness coefficients, elastic 34 35 moduli, Poisson's ratio, B/G ratio, elastic anisotropy indexes (A_B , A_G , A_1 , A_2 , A_3 , A_U) and directional dependence of Young modulus and linear compressibility have been 36 37 obtained. Vickers hardness, and sound wave velocities have been calculated. Our simulations show that bismite has a high elastic anisotropy. α -Bi₂O₃ is a ductile material 38 whose elastic anisotropy increases under compression and presents a stronger ability to 39 resist volume compression than shear deformation at all pressures. Besides, it has a very 40 small minimum thermal conductivity, which is well suited for thermoelectric 41 42 applications. Finally, the mechanical and dynamical stability of bismite at high pressure has been studied and it has been found that α -Bi₂O₃ becomes mechanically unstable at 43 pressures beyond 19.3 GPa and dynamically unstable above 11.5 GPa. These 44 45 instabilities could be responsible for the amorphization of bismite observed 46 experimentally between 15 and 20 GPa.

47

48

49 Keywords:

50 oxides

51 *ab initio* calculations

52 high pressure

53 elastic properties

54 thermodynamic properties

55 **1. Introduction**

56 Bismuth trioxide has been extensively studied due to its high polymorphism and technological importance [1]. The stable phase of Bi_2O_3 at room conditions is the α 57 phase, which occurs in nature as mineral bismite. α -Bi₂O₃ crystallizes in a monoclinic 58 59 crystalline structure (space group: $P2_1/c$ No. 14, Z=4) and this phase presents the lowest symmetry of all the known polymorphs of bismuth trioxide [1, 2]. The structure of α -60 Bi₂O₃ can be described as layers of bismuth atoms parallel to the (100) plane of the 61 monoclinic cell, separated by layers of oxygen atoms (see Fig. 1) [1]. Noteworthy, 62 recent studies on α -Bi₂O₃ have focused on its use as a photocatalyst, a gas sensor and a 63 64 supercapacitor [3-7].

The structural, vibrational, optical, electric and magnetic properties of α -Bi₂O₃ have 65 66 been studied under different conditions of temperature or pressure [1, 8-23]. These studies report that this material has unusual magnetic and electrical properties such as 67 the existence of internal magnetic fields and a longitudinal magnetoelectric effect [9, 68 10]. In addition, shock compression experiments have demonstrated that bismite can be 69 70 used as thermite mixtures for high pressure (HP) applications [8]. In this context, powder X-ray diffraction and Raman scattering experiments have shown that α -Bi₂O₃ 71 undergoes an amorphization at HP between 15 and 20 GPa depending on the hydrostatic 72 73 conditions [16, 22, 23].

Despite the technological applicability of α -Bi₂O₃, its elastic properties are barely known and, to the best of our knowledge, only its axial compressibilities, bulk modulus at zero pressure, and sound velocity are recently known [8, 16]. The present work reports a theoretical study of the elastic and thermodynamic properties of bismite both at room pressure (0.0001 GPa) and at HP up to ~20 GPa. This study has allowed us to discuss the mechanical stability of this sesquioxide at HP. Besides, we report lattice dynamics *ab initio* calculations at different pressures. In particular, the knowledge of the phonon dispersion curves (PDCs) at different pressures has allowed us to study the dynamical stability of bismite at HP. We will show that the monoclinic structure of Bi₂O₃ becomes dynamically unstable prior to become mechanically unstable. These instabilities likely cause the amorphization experimentally observed in α -Bi₂O₃ between 15 and 20 GPa.

86

87 **2.** Calculation methods

Ab initio simulations of α -Bi₂O₃ under compression were carried out with the 88 density functional theory (DFT) [24] employing the plane-wave pseudopotential method 89 implemented in the VASP code (Vienna Ab initio Simulation Package) [25]. The use of 90 the projector-augmented wave scheme (PAW) [26] has allowed us to describe the full 91 92 nodal character of the all-electron charge density in the core region taking into account 15 valence electrons $(5d^{10}6s^24p^3)$ for bismuth and 6 valence electrons $(2s^22p^4)$ for 93 oxygen. The exchange-correlation energy was computed within the generalized gradient 94 95 approximation (GGA) using the Perdew-Burke-Ernzenhof functional for solids (PBEsol) [27]. A 6 x 4 x 4 Monkhorst-Pack grid of special k-points was used for 96 integrations over the Brillouin zone (BZ). In order to obtain accurate results for a 97 precise description of the electronic properties, the basis set of plane waves was 98 developed up to an energy cutoff of 520 eV. In this way, a high convergence in total 99 energy (better than 1 meV per formula unit) was obtained. Our calculations do not take 100 into account the spin-orbit interaction because those calculations are computationally 101 very demanding and it has been previously shown that spin-orbit interaction is small in 102 103 α -Bi₂O₃ [23].

DFT calculations are a well-tested method that describes properly the properties 104 of semiconductors under compression [28]. For that purpose, the structure of bismite 105 106 was fully relaxed to its optimized configuration at several chosen volumes (or 107 equivalently pressures) through the calculation of forces on atoms and stress tensors. In this way, the structural parameters were obtained, as well as a set of related energies, 108 volumes, and pressures. For each optimized structure, Hellmann-Feynman forces on 109 atoms were smaller than 0.006 eV/Å, and the deviation of the stress tensor from the 110 diagonal hydrostatic form was lower than 0.1 GPa. The simulations also provided the 111 formation energy and cohesive energy of α -Bi₂O₃, -6.531 eV and 2.947 eV, 112 respectively; in good agreement with experimental results [29]. 113

The elastic constants were evaluated by computing the macroscopic stress for a 114 small strain applying the stress theorem [30, 31] as implemented in the VASP code. 115 116 Taking into account their symmetry, the ground state and the optimized structures are strained in different directions [32]. A Taylor expansion of the total energy with respect 117 to the applied strain is used to evaluate with high precision the small total-energy 118 differences between different strained states [33]. During the simulations some tests 119 120 were performed using small strains in order to ensure the harmonic approximation from the strain-stress relationship. We have also studied the mechanical properties of α -Bi₂O₃ 121 using the elastic moduli obtained from the calculated elastic constants. 122

Finally, the dynamical stability of α -Bi₂O₃ was studied with *ab initio* DFT-based lattice-dynamics simulations at the zone centre (Γ point) and along high symmetry directions of the BZ by using the PHONON code [34]. To calculate the dynamical matrix using the harmonic approximation with the direct force constant approach, highly converged results on forces are required [34, 35]. The PDCs were obtained with a supercell of size 2 x 2 x 4 (320 atoms) that gave a good description of the phonon
branches.

130

131 **3. Results and discussion**

3.1. Elastic properties and mechanical stability of bismite

 α -Bi₂O₃ has 13 independent second-order elastic constants (C_{ii}) which, in the 133 134 Voigt notation, are: C₁₁, C₁₂, C₁₃, C₁₅, C₂₂, C₂₃, C₂₅, C₃₃, C₃₅, C₄₄, C₄₆, C₅₅ and C₆₆ [36]. When the crystal is loaded under external hydrostatic compression, the elastic stiffness 135 coefficients, B_{ij} , must be employed instead of C_{ij} elastic constants. The B_{ij} coefficients 136 137 for a monoclinic crystal, in the Voigt notation, under an external hydrostatic pressure, P, are: $B_{11} = C_{11} - P$, $B_{12} = C_{12} + P$, $B_{13} = C_{13} + P$, $B_{15} = C_{15}$, $B_{22} = C_{22} - P$, $B_{23} = C_{23} + P$, 138 $B_{25} = C_{25}, B_{33} = C_{33} - P, B_{35} = C_{35}, B_{44} = C_{44} - P, B_{46} = C_{46}, B_{55} = C_{55} - P \text{ and } B_{66} = C_{66}$ 139 -P [37]. Note that the values of B_{ij} and C_{ij} are equal when P = 0 GPa (value close to 140 room pressure). 141

142

a) At room pressure

143 Table 1 lists the calculated C_{ij} elastic constants for α -Bi₂O₃ at 0 GPa. Elastic constants are referred to an orthonormal coordinate system (x, y, z) [36]. These 144 145 orthonormal axes, following the IRE (Institute of Radio Engineers) convention, are related to the monoclinic crystal lattice vectors (a, b, c) in the following way: y-axis is 146 parallel to *b*-axis, *z*-axis is parallel to *c*-axis, and *x*-axis is perpendicular to (*yz*) plane. 147 C_{11} , C_{22} and C_{33} elastic constants indicate the resistance of the material to be elongated 148 or compressed along x, y and z directions, respectively. For α -Bi₂O₃, $C_{33} > C_{11} > C_{22}$ 149 indicates that unidirectional compressibility increases in the following sequence of 150 directions: [001] < [100] < [010]. On the other hand, elastic constants C_{44} , C_{55} and C_{66} 151 are related to the resistance to shear deformation. In our particular case, $C_{66} > C_{55} > C_{44}$ 152

denotes that the [010](001) shear is easier than the [001](100) shear which is easier than the [100](010) shear. C_{11} , C_{22} and C_{33} are greater than C_{44} , C_{55} and C_{66} in bismite. This indicates that this material has a greater resistance to unidirectional compression than to shear deformation. Finally, elastic constants with mixed compression/shear coupling $(C_{15}, C_{25} \text{ and } C_{35})$ along with elastic constant C_{46} related to shear deformation have the smallest values in bismite.

159 The calculated elastic constants for α -Bi₂O₃ can be compared to those experimentally obtained for glass Bi_2O_3 using the pulse echo overlap technique [38]. 160 The value of C_{11} in the glass ($C_{11} = 86.5$ GPa) is 26% smaller than the theoretical 161 average $(C_{11} + C_{22} + C_{33})/3$ in the crystal. The value of C_{44} in the glass $(C_{44} = 26.5 \text{ GPa})$ 162 is 30% smaller than the theoretical average $(C_{44} + C_{55} + C_{66})/3$ in the crystal. These 163 164 results clearly suggest that the crystalline phase is stiffer than the glass, as observed in a 165 number of compounds [39], and give confidence to our calculated results to continue exploring the mechanical properties of bismite from a theoretical ground. 166

The knowledge of the elastic constants at 0 GPa allows us to study the mechanical stability of the monoclinic structure. In this context, the mechanical stability of a crystal at zero pressure requires that the whole set of elastic constants, C_{ij} , satisfies the Born stability criteria [40]. These stability criteria for monoclinic crystals with 13 independent elastic constants, taking into account that C_{ij} matrix must be positivedefinite, are given by the following conditions:

173
$$m_1 = C_{11} > 0$$
 (1)

174
$$m_2 = C_{11}C_{22} - C_{12}^2 > 0 \tag{2}$$

175
$$m_3 = (C_{22}C_{33} - C_{23}^2)C_{11} - C_{33}C_{12}^2 + 2C_{23}C_{12}C_{13} - C_{22}C_{13}^2 > 0$$
(3)

176
$$m_4 = C_{44} > 0$$
 (4)

$$m_{5} = C_{12}^{2}C_{35}^{2} - C_{33}C_{55}C_{12}^{2} + 2C_{55}C_{12}C_{13}C_{23} - 2C_{12}C_{13}C_{25}C_{35}$$
$$-2C_{12}C_{15}C_{23}C_{35} + 2C_{33}C_{12}C_{15}C_{25} + C_{13}^{2}C_{25}^{2} - C_{22}C_{55}C_{13}^{2}$$
$$-2C_{13}C_{15}C_{23}C_{25} + 2C_{22}C_{13}C_{15}C_{35} + C_{15}^{2}C_{23}^{2} - C_{22}C_{33}C_{15}^{2}$$
$$-C_{11}C_{55}C_{23}^{2} + 2C_{11}C_{23}C_{25}C_{35} - C_{11}C_{33}C_{25}^{2} - C_{11}C_{22}C_{35}^{2}$$
$$+C_{11}C_{22}C_{33}C_{55} > 0$$
(5)

178
$$m_6 = C_{44}C_{66} - C_{46}^2 > 0 \tag{6}$$

¹⁸⁰ Values of m_i (i = 1-6) at 0 GPa are given in **Table 2**. As it can be observed, all ¹⁸¹ the stability criteria are satisfied at 0 GPa; thus, the monoclinic structure of bismite is ¹⁸² mechanically stable at room pressure. This result is in good agreement with the fact that ¹⁸³ mineral bismite is found at room conditions.

The elastic moduli can be obtained from the elastic stiffness coefficients B_{ij} . In particular, the bulk modulus, B, and shear modulus, G, have been calculated with the expressions [41]:

187
$$B_{V} = \frac{B_{11} + B_{22} + B_{33} + 2(B_{12} + B_{23} + B_{13})}{9}$$
(7)

188
$$\frac{1}{B_R} = S_{11} + S_{22} + S_{33} + 2(S_{12} + S_{23} + S_{13})$$
(8)

$$B_H = \frac{B_V + B_R}{2} \tag{9}$$

190

189

191
$$G_V = \frac{B_{11} + B_{22} + B_{33} - (B_{12} + B_{23} + B_{13}) + 3(B_{44} + B_{55} + B_{66})}{15}$$
(10)

192
$$\frac{1}{G_R} = \frac{4(S_{11} + S_{22} + S_{33}) - 4(S_{12} + S_{23} + S_{13}) + 3(S_{44} + S_{55} + S_{66})}{15}$$
(11)

$$G_H = \frac{G_V + G_R}{2} \tag{12}$$

194 The elastic moduli are derived in the Voigt [42], Reuss [43], and Hill [44] 195 approximations labeled with subscripts *V*, *R*, and *H*, respectively; in Eqs. 7-12. 196 The Young modulus, *E*, has been obtained from the bulk and shear moduli with 197 the expression [45]:

$$E_X = \frac{9B_X G_X}{G_X + 3B_X} \tag{13}$$

199

198

200 The subscript X in Eq. 13 refers to the symbols V, R, and H.

201 Elastic moduli in α -Bi₂O₃ at 0 GPa are tabulated in **Table 1**. It must be stressed 202 that the Hill bulk modulus, $B_{\rm H} = 68.3$ GPa is in rather good agreement with the theoretical value ($B_0 = 72.8$ GPa) obtained from our structural data calculated with 203 204 VASP. This agreement (there is only a 6% difference) suggests the consistency of our 205 calculations. On the other hand, our Hill bulk modulus is 20% smaller than the experimental value of $B_0 = 85.4$ GPa previously reported [16]. It must be stressed that 206 the last two values for B_0 were obtained from a 3rd order Birch-Murnaghan equation of 207 state [16]. Besides, our calculated value for $B_{\rm H}$ is 36% smaller than the bulk modulus 208 209 obtained from shock compression experiments ($B_0 = 106$ GPa) [8]. In this context, it should be commented that the small value of the first derivative of the bulk modulus, 210 B_0 '= 1.28 found in the shock compression work results in a greater value for B_0 due to 211 the strong correlation between B_0 and B_0 ' [46]. The fact that the bulk modulus is greater 212 than the shear modulus indicates that α -Bi₂O₃ presents a stronger ability to resist 213 volume compression than shear deformation at 0 GPa. 214

The Poisson's ratio, v, is connected with the way structural elements are packed and provides information about the characteristics of the bonding forces and chemical bonding [47]. The v ratio has been calculated with the expression [45]:

218
$$v_{X} = \frac{1}{2} \left(\frac{3B_{X} - 2G_{X}}{3B_{X} + G_{X}} \right)$$
(14)

The subscript X in Eq. 14 refers to the symbols V, R, and H. The Poisson's ratio in α -Bi₂O₃, with the Hill approximation, is v = 0.29 at 0 GPa (see Table 1). Since v > 0.25the interatomic bonding forces are predominantly central and the ionic bonding prevails against covalent bonding in bismite at room pressure [48, 49].

The *B/G* ratio was proposed by Pugh as a relationship which relates empirically the plastic properties of a material with its elastic moduli **[50]**. This relationship quantifies the ratio between the resistance to fracture and the resistance to plastic deformation. A value of the *B/G* ratio higher (lower) than 1.75 is associated with a ductile (brittle) material. In our calculations, we have found a value of B/G = 2.04 for α -Bi₂O₃ at 0 GPa (see **Table 1**) which indicates that bismite is ductile at room pressure.

It is well known that microcracks may be induced in materials due to elastic 230 231 anisotropy [51]. We have evaluated this anisotropy using different anisotropy indexes. 232 Firstly, we have calculated the percentage of anisotropy in both compressibility and shear moduli with expressions $A_B = (B_V - B_R)/(B_V + B_R)$ and $A_G = (G_V - G_R)/(G_V + G_R)$, 233 234 respectively [52]. A value of 0% indicates no anisotropy, while a value of 100% is 235 associated with the largest anisotropy. In α -Bi₂O₃, $A_B = 4.35\%$ and $A_G = 7.73\%$ at 0 GPa (see **Table 1**). These results indicate a larger anisotropy in shear than in compression in 236 bismite at room pressure. Secondly, we have calculated the shear anisotropic factors that 237 provide a measure of the degree of anisotropy in the bonding between atoms in different 238 planes [47]. The formulas that define the shear anisotropy factors are the following: 239

240
$$A_{1} = \frac{4B_{44}}{B_{11} + B_{33} - 2B_{13}}$$
(15)

241
$$A_2 = \frac{4B_{55}}{B_{22} + B_{33} - 2B_{23}}$$
(16)

242
$$A_3 = \frac{4B_{66}}{B_{11} + B_{22} - 2B_{12}}$$
(17)

243 where A_1 is the factor for the {100} shear planes between the (011) and (010) directions, A_2 is for the {010} shear planes between the $\langle 101 \rangle$ and $\langle 001 \rangle$ directions, and A_3 is for the 244 $\{001\}$ shear planes between the $\langle 110 \rangle$ and $\langle 010 \rangle$ directions [47]. For an isotropic crystal 245 the shear anisotropy factors must be one. A value of A_1 , A_2 or A_3 smaller or greater than 246 one is a measure of the elastic anisotropy of the crystal. In α -Bi₂O₃, $A_1 = 0.80$, $A_2 = 0.84$ 247 and $A_3 = 2.32$ at 0 GPa (see **Table 1**). These results indicate that bismite exhibits a 248 249 larger anisotropy in A_3 and shows more distinct performance in {001} planes than in {100} and {010} planes. Thirdly, we have also calculated the anisotropy of the material 250 251 by using the universal elastic anisotropy index A_U defined by: $A_U = 5(G_V/G_R) + (B_V/B_R) - 6$ 252 [53]. For a locally isotropic crystal, A_U must be zero, so the deviation of A_U from zero is a measure of the degree of elastic anisotropy for the crystalline structure. We have 253 found that α -Bi₂O₃ is elastically anisotropic because it has an A_U value above zero at 0 254 255 GPa (see Table 1).

The anisotropy in the monoclinic cell of bismite can also be estimated by the axial compressibilities κ_a , κ_b and κ_c . These compressibilities have been obtained by using the expressions [36]:

- 259 $\kappa_a = S_{11} + S_{12} + S_{13} \tag{18}$
- 260

261 $\kappa_b = S_{12} + S_{22} + S_{23} \tag{19}$

- 262
- 263 $\kappa_c = S_{13} + S_{23} + S_{33} \tag{20}$
- 264

where S_{ij} are the components of the elastic compliances tensor. **Table 3** includes the values for κ_a , κ_b and κ_c , obtained at 0 GPa using **Eqs. 18, 19** and **20**. It is found that $\kappa_b >$ $\kappa_c > \kappa_a$; i.e., the *b*-axis is the most compressible one and the *a*-axis the less compressible one. Our theoretical results (see **Table 3**) are in good agreement with those given in **Ref. 16** obtained from equation of state fits. Therefore, this result again gives us confidence about the correctness of our simulations. The elastic anisotropy of bismite given by the different anisotropy indexes is in agreement with the anisotropy observed in the axial compressibilities, since $\kappa_c/\kappa_a > 1$ and $\kappa_b/\kappa_c > 1$.

To conclude with the study of the anisotropy, we have obtained the directional 273 dependence of both the Young modulus, E, and the linear compressibility, κ , for bismite 274 at 0 GPa using the ELATE program [54]. The results are shown in Fig. 2 with a 3D 275 parametric surface. For an absolutely isotropic medium this surface must be a sphere. It 276 is clearly evidenced that α -Bi₂O₃ is highly anisotropic. The direction of minimum E, in 277 Cartesian coordinates (x, y, z), is (0, 1, 0) [see **Table 4**]. This indicates that the direction 278 with the smallest stiffness is along the *b*-axis. The stiffest direction is (0.30, 0, 0.95). 279 This direction is in the (010) plane at an angle of 17.5° to the z-axis. The directions for 280 281 maximum and minimum values of E (E_{max} and E_{min}) are correlated with the fact that C_{33} $> C_{11} > C_{22}$. The high anisotropy is also evidenced quantitatively by the ratio $E_{\text{max}} / E_{\text{min}}$ 282 = 2.4. On the other hand, the linear compressibility takes maximum and minimum 283 values of $\kappa_{\text{max}} = 8.8 \text{ TPa}^{-1}$ and $\kappa_{\text{min}} = 1.7 \text{ TPa}^{-1}$ (see **Table 4**). The anisotropy ratio in 284 linear compressibility is also quite high, $\kappa_{max} / \kappa_{min} = 5.1$. The direction of maximum 285 compressibility is (0, 1, 0), that is, the *b*-axis. We note that κ_{max} has the same value as 286 the axial compressibility $\kappa_b = 8.79 \cdot 10^{-3} \text{ GPa}^{-1}$ (see **Table 3**). The direction of minimum 287 compressibility is (0.89, 0, 0.45). This direction is in the (010) plane at an angle of 27.0° 288 to the x-axis and 49.3° to the *a*-axis. 289

Another common mechanical magnitude to characterize materials is hardness. It is known that hardness is a complex variable which involves elasticity and plasticity in a material. It has been recently proposed that the Vickers hardness, H_{ν} , can be obtained with the expression of Tian *et al.* [55]:

$$H_{\nu} = 0.92(G/B)^{1.137} G^{0.708}$$
⁽²¹⁾

296

295

The values of H_v in α -Bi₂O₃ at 0 GPa in the Voigt, Reuss and Hill approximations are included in **Table 1**. Bismite has a value of $H_v = 4.90$ GPa at 0 GPa in the Hill approximation. Consequently, this sequioxide can be considered as a relatively soft material because H_v is smaller than 10 GPa. The soft behavior of α -Bi₂O₃ is related with its ductile behavior at 0 GPa, as previously shown.

Finally, we have calculated the average sound velocity, v_m , in α -Bi₂O₃ from the known elastic moduli [56]. This velocity is a quantity which provides information on the elastic moduli, elastic anisotropy, and other thermodynamic properties. In polycrystalline materials v_m can be obtained by the expression [57]:

306
$$v_m = \left[\frac{1}{3}\left(\frac{2}{v_{trans}^3} + \frac{1}{v_{lon}^3}\right)\right]^{-1/3}$$
(22)

307 In Eq. 22, v_{trans} and v_{lon} are the transverse and longitudinal elastic wave velocities of the 308 polycrystalline material:

309
$$v_{lon} = \left(\frac{B + \frac{4}{3}G}{\rho}\right)^{1/2}$$
(23)

310
$$v_{trans} = \left(\frac{G}{\rho}\right)^{1/2}$$
(24)

where ρ is the density of the material and *B* and *G* are the elastic moduli. **Table 5** shows the density and wave velocities v_m , v_{lon} and v_{trans} in α -Bi₂O₃ at 0 GPa. The obtained value of v_m in the Hill approximation (2088.8 m/s) at 0 GPa is similar to that obtained for porous α -Bi₂O₃ ($v_m = 2111\pm432$ m/s) in shock compression experiments [8], and 315 39% smaller than that obtained by the calculated Hugoniot for bulk α -Bi₂O₃ ($v_m = 3432$ 316 m/s) in dynamic experiments [8].

317

b) At high pressure

The HP evolution of the elastic constants, C_{ij} , and of the elastic stiffness 318 coefficients, B_{ij} , in α -Bi₂O₃ are given in Figs. 3 and 4, respectively. We report the HP 319 dependence of B_{ij} and C_{ij} , although only B_{ij} are valid at any pressure, because B_{ij} are 320 321 obtained from the C_{ij} . As it can be observed, B_{11} , B_{22} , B_{33} , B_{12} , B_{13} , B_{15} and B_{23} (B_{55} and B_{35}) increase (decrease) with pressure in the whole pressure range studied. However, 322 B_{44} , B_{66} , B_{25} and B_{46} increase with pressure up to 4.5, 15.5, 11.2 and 10.3 GPa, 323 324 respectively, and then decrease as pressure increases. It must be noted that B_{35} changes its sign at 14.0 GPa. 325

326 From the HP dependence of the elastic stiffness coefficients, the HP evolution of 327 the B, G, and E elastic moduli, v Poisson's ratio, and B/G ratio for α -Bi₂O₃ is calculated (Fig. 5). For the sake of simplicity, only the HP dependence of the elastic moduli in the 328 329 average Hill approximation will be commented. As it can be seen, B_H increases under 330 compression, reaches a maximum value of 118.3 GPa at 12.6 GPa, and then it decreases 331 under pressure. Similarly, E_H increases with pressure achieving a maximum value of 87.3 at 1.6 GPa and decreases above that pressure. In turn, G_H decreases with pressure 332 in the whole pressure range studied. Finally, ν also increases with pressure, reaching a 333 value of 0.42 at 18.1 GPa. It must be stressed that the HP increase of v indicates that the 334 ductility and the metallic behavior increase under compression. This metallization can 335 be understood as a decrease of the ionic character of the material. In this sense, the 336 effect of pressure on bismite is the increase of atomic coordination and progressive 337 338 reduction of interatomic bond directionality. We note that bond directionality decreases in the sequence covalent-ionic-metallic of bond character. 339

It can also be observed that the B/G ratio increases at HP in α -Bi₂O₃, reaching a value of 6.1 at 18.1 GPa. Since the B/G ratio is related to the Poisson's ratio [49], the increase in B/G also indicates an increment of the ductility under compression.

Figure 6 shows the evolution with pressure of the different anisotropy indexes previously defined for bismite. The compression and shear anisotropy factors (A_B , A_G), A_3 shear anisotropy factor, and A_U universal anisotropy factor increase significantly under compression. This denotes a considerable increment of the elastic anisotropy in α -Bi₂O₃ with pressure. On the other hand, A_1 and A_2 shear anisotropy factors have a small variation with pressure.

Figure 7 presents the HP dependence of the Vickers hardness in α -Bi₂O₃. Clearly, H_{ν} decreases as pressure increases because the *G* elastic modulus and, consequently, the *G/B* ratio decrease with pressure. Therefore, bismite becomes softer as pressure increases; a result which is related to the increase of its ductility (*B/G* ratio), as previously commented.

The HP dependence of the sound velocities for α -Bi₂O₃ is given in **Figure 8**. The calculated v_{lon} , obtained with elastic moduli in the Hill approximation, increases with pressure reaching a maximum value of 3761.7 m/s at 9.7 GPa and decreases above that pressure. Moreover, the calculated velocities v_{trans} and v_m in all approximations decrease as pressure increases, thus reflecting the decrease of *G*.

It must be stressed that the crystalline structure of a compound in a given structure is stable if it is mechanically and dynamically stable **[58-60]**. Taking into account that we have evaluated the mechanical stability of bismite at room pressure in section 3.1.a) through the Born stability criteria, and that bismite is also dynamically stable at room pressure **[23]**, now we evaluate in this section the mechanical stability of bismite at HP on the basis of the theoretically calculated elastic constants at HP.

When a non-zero stress is applied to the crystal, a generalization of the Born stability criteria is required [61, 62]. The generalized Born stability criteria for monoclinic crystals are obtained by replacing in Eqs. 1-6 the C_{ij} elastic constants by the B_{ij} elastic stiffness coefficients. These criteria are given by:

369
$$M_1 = B_{11} > 0$$
 (25)

370
$$M_2 = B_{11}B_{22} - B_{12}^2 > 0 \tag{26}$$

371
$$M_3 = (B_{22}B_{33} - B_{23}^2)B_{11} - B_{33}B_{12}^2 + 2B_{23}B_{12}B_{13} - B_{22}B_{13}^2 > 0$$
(27)

372
$$M_4 = B_{44} > 0$$
 (28)

$$M_{5} = B_{12}^{2}B_{35}^{2} - B_{33}B_{55}B_{12}^{2} + 2B_{55}B_{12}B_{13}B_{23} - 2B_{12}B_{13}B_{25}B_{35}$$

$$- 2B_{12}B_{15}B_{23}B_{35} + 2B_{33}B_{12}B_{15}B_{25} + B_{13}^{2}B_{25}^{2} - B_{22}B_{55}B_{13}^{2}$$

$$- 2B_{13}B_{15}B_{23}B_{25} + 2B_{22}B_{13}B_{15}B_{35} + B_{15}^{2}B_{23}^{2} - B_{22}B_{33}B_{15}^{2}$$

$$- B_{11}B_{55}B_{23}^{2} + 2B_{11}B_{23}B_{25}B_{35} - B_{11}B_{33}B_{25}^{2} - B_{11}B_{22}B_{35}^{2}$$

$$+ B_{11}B_{22}B_{33}B_{55} > 0$$

(29)

 $M_6 = B_{44} B_{66} - B_{46}^2 > 0 \tag{30}$

375

Figure 9 reports the evolution of the generalized stability criteria in α -Bi₂O₃ under compression. At HP up to 23 GPa, all stability criteria are satisfied except M_5 (Eq. 29), which is violated at 19.3 GPa. Therefore, our simulations show that α -Bi₂O₃ becomes mechanically unstable above 19.3 GPa. This value is consistent with the quick increase of the A_B , A_G and A_U anisotropy factors above 16 GPa in α -Bi₂O₃ (see Figure 6), which is typically observed when a compound approaches the mechanical instability [63].

383

In this section, we will focus on the study of the dynamical stability of the 385 monoclinic structure of α -Bi₂O₃ at HP. For that purpose, we have performed lattice 386 387 dynamics calculations and have evaluated the PDC at different pressures (see Figure 388 10). It can be observed that our calculations do not show any phonon branch with imaginary frequencies at any point of the BZ in α -Bi₂O₃ at 0 GPa in good agreement 389 with a previous report [23]. Therefore, bismite is dynamically stable at room pressure as 390 391 previously commented. However, imaginary frequencies (below 0 in the ordinate scale of Figure 10) are developed above 11.5 GPa. Therefore, our theoretical calculations 392 predict that bismite becomes dynamically unstable at HP (above 11.5 GPa) before 393 394 becoming mechanically unstable (above 19.3 GPa).

To conclude this section, we want to comment that our results on the mechanical 395 396 and dynamical stability of bismite at HP allow us to understand previous works that 397 have studied the behavior of Bi₂O₃ under compression [16, 22, 23, 64, 65]. In this context, a phase transition was predicted to occur around 5.5 GPa from the monoclinic 398 399 α -Bi₂O₃ phase to the hexagonal HPC-Bi₂O₃ phase [64]. However, this phase transition 400 was not observed experimentally at room temperature, probably due to the hindrance of 401 the pressure-driven phase transition at room temperature. Additionally, a pressure-402 induced amorphization of α -Bi₂O₃ was observed between 15 and 20 GPa depending on the hydrostatic conditions [16, 22, 23]. Noteworthy, the experimental amorphization of 403 α -Bi₂O₃ under compression is consistent with the results of the present theoretical study 404 of the mechanical and dynamical stability of α -Bi₂O₃ which suggest that amorphization 405 406 could be caused by the dynamical and mechanical instabilities theoretically predicted to occur on α -Bi₂O₃ under purely hydrostatic pressure above 11.5 and 19.3 GPa, 407 respectively. The difference between experimental and theoretical pressure values for 408 409 the onset of the pressure-induced amorphization could be ascribed to the non-purely

410 hydrostatic compression of α -Bi₂O₃ in the reported experiments, which could favour a 411 larger stability of the monoclinic phase [16, 22, 23].

- 412
- 413

3.3. Thermodynamic properties of bismite

To conclude this work, we want to calculate some thermodynamic properties of bismite, such as the Debye temperature and the minimum thermal conductivity, from the knowledge of the average sound velocity, $v_{\rm m}$. The Debye temperature, $\theta_{\rm D}$, corresponds to the upper limit of phonon frequency in a crystal lattice and it can be calculated with the semi-empirical formula [57]:

419
$$\theta_D = \frac{h}{k_B} \left[\frac{3n}{4\pi} \left(\frac{N_A \rho}{M} \right) \right]^{1/3} v_m$$
(31)

In Eq. 31 *h* is the Planck's constant, $k_{\rm B}$ is the Boltzmann's constant, *n* is the number of atoms in the molecule, $N_{\rm A}$ is the Avogadro's number, ρ is the density, *M* is the molecular weight, and $v_{\rm m}$ is the average sound velocity. The value of $\theta_{\rm D} = 245.7$ K in the Hill approximation, for α -Bi₂O₃ at 0 GPa (see **Table 5**), is 22% smaller than the estimated value of $\theta_{\rm D} = 316$ K, obtained from shock compression data using Lindemman's melting law [8]. A decrease of $\theta_{\rm D}$ with pressure, clearly following the decrease of $v_{\rm m}$ with pressure, is observed in bismite (see Fig. 11(a)).

427 On the other hand, the thermal conductivity is a property that allows modeling 428 the heat transfer in solids. The theoretical minimum of the thermal conductivity has 429 been estimated by using the expression given by Clarke [66]:

430
$$\kappa_{\min} = k_B v_m \left(\frac{M}{n\rho N_A}\right)^{-2/3}$$
(32)

and the expression given in Long's model [67]:

432
$$\kappa_{\min} = \left\{ \frac{1}{3} \left[2(2+2\nu)^{\frac{3}{2}} + \left(1 - \frac{2\nu^2}{1-\nu}\right)^{\frac{3}{2}} \right] \right\}^{-\frac{1}{3}} \cdot k_B m^{\frac{2}{3}} \cdot \left(\frac{E}{\rho}\right)^{\frac{1}{2}}$$
(33)

433 where *m* is the number of atoms per volume. Both expressions give the same result for κ_{\min} and take into account the anisotropic elasticity of monoclinic α -Bi₂O₃. The value of 434 κ_{\min} in the Hill approximation, for α -Bi₂O₃ at 0 GPa, is 0.45 W m⁻¹ K⁻¹ (see **Table 5**). 435 Therefore, this sesquioxide can be considered as a low κ material interesting for 436 thermoelectric applications [68]. As shown in Figure 11(b), κ_{min} in the Hill 437 438 approximation increases slightly as pressure increases from 0 to 1.6 GPa and then decreases as pressure increases. It should be stressed that the calculated value of κ_{\min} = 439 0.45 W m⁻¹ K⁻¹ is smaller than the thermal conductivity measured at room temperature 440 for α -Bi₂O₃ which turns out to be 1.3 W m⁻¹ K⁻¹ [69]. Therefore, we consider that there 441 442 is room for improvement in the decrease of the thermal conductivity of bismite both at room pressure and at HP conditions for thermoelectric applications [70]. 443

444

445 **4. Conclusions**

In the present work, the elastic and thermodynamic behavior of α -Bi₂O₃ has been studied theoretically both at room pressure and under hydrostatic compression. The values at room pressure and the HP dependence of the elastic moduli, Poisson's ratio, *B/G* ratio and elastic anisotropy indexes of bismite are reported as well as a study of the directional dependence of Young modulus and linear compressibility at 0 GPa.

451 Our results show that this sesquioxide: i) is ductile at all pressures and has a 452 larger resistance to volume compression than to shear deformation; ii) shows an 453 intrinsic elastic anisotropy which increases with pressure; iii) is relatively soft at room 454 pressure and its hardness decreases under compression. Besides, its average sound 455 velocity and Debye temperature decrease with increasing pressure. Moreover, we have 456 found that α -Bi₂O₃ is a material with a very low thermal conductivity that could be 457 useful for thermoelectric applications both at low and high pressures.

Finally, the study of the mechanical and dynamical stability of the monoclinic structure of bismuth oxide at high pressure shows that this structure becomes mechanically and dynamically unstable above 19.3 and 11.5 GPa, respectively. This result indicates that amorphization of α -Bi₂O₃, found experimentally between 15 and 20 GPa, could be due to the dynamical and mechanical instabilities of this monoclinic phase. We hope that the current theoretical work on bismite will stimulate experimental studies of its elastic and thermodynamic properties.

465

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585

- 587
- 588

589	Table 1. C_{ij} elastic constants (in GPa) for α -Bi ₂ O ₃ . Elastic moduli <i>B</i> , <i>G</i> , and <i>E</i> (in GPa),
590	Possion's ratio (v), B/G ratio, and Vickers hardness (H_v in GPa) are obtained in the
591	Voigt, Reuss and Hill approximations, labeled respectively with subscripts V , R , and H .
592	The percentage of anisotropy in both compression (A_B) and shear (A_G) , as well as the
593	shear anisotropic factors (A_1, A_2, A_3) and the universal anisotropy index (A_U) are also
594	included. Data are calculated at 0 GPa.

	α -Bi ₂ O ₃ ^a
<i>C</i> ₁₁	114.0
<i>C</i> ₁₂	59.1
<i>C</i> ₁₃	57.1
<i>C</i> ₁₅	10.2
C ₂₂	84.3
C ₂₃	30.4
C ₂₅	2.6
C ₃₃	150.3
C ₃₅	10.5
C ₄₄	30.0
C ₄₆	8.3
C ₅₅	36.4
C ₆₆	46.4
B_V, B_R, B_H	71.3, 65.3, 68.3
G_{V}, G_{R}, G_{H}	36.0, 30.9, 33.4
E_V, E_R, E_H	92.5, 80.0, 86.3
v_V , v_R , v_H	0.28, 0.30, 0.29
$B_V/G_V, B_R/G_R, B_H/G_H$	1.98, 2.12, 2.04
$A_{\rm B}$ (%), $A_{\rm G}$ (%), $A_{\rm 1}$, $A_{\rm 2}$, $A_{\rm 3}$, $A_{\rm U}$	4.35, 7.73, 0.80, 0.84, 2.32, 0.93
$H_{v}\left(V,R,H\right)$	5.36, 4.44, 4.90

^a Our *ab initio* calculations.

Table 2. Born stability conditions in α -Bi₂O₃ at 0 GPa.

<i>m</i> ₁ (GPa)	114.0
m_2 (GPa ²)	$61.1 \cdot 10^2$
m_3 (GPa ³)	$74.4 \cdot 10^4$
<i>m</i> ₄ (GPa)	30.0
$m_5 (\mathrm{GPa}^4)$	$261.4 \cdot 10^5$
$m_6 (\mathrm{GPa}^2)$	$13.2 \cdot 10^2$

- **Table 3.** Axial compressibilities, κ_a , κ_b and κ_c , in α -Bi₂O₃ obtained from the elastic 601 constants at 0 GPa. The axial compressibilities reported in **Ref 16**, obtained from a
- 602 Murnaghan equation of state fit of data, are also given for comparison.

$\kappa_{\rm a} (10^{-3}{\rm GPa}^{-1})$	$\kappa_{\rm b} (10^{-3} {\rm GPa}^{-1})$	$\kappa_{\rm c} (10^{-3} {\rm GPa}^{-1})$	
2.36	8.79	4.15	This work
1.53(1)	7.84(2)	2.50(1)	Ref 16
2.07(1)	6.64(1)	4.41(1)	Ref 16

Table 4. Maximum and minimum values for the Young modulus, *E*, and linear compressibility, κ . The anisotropy ratio (maximum value/minimum value) and direction of minimum and maximum value for *E* and κ are also included. Data are calculated at 0 GPa with the ELATE software [54].

610

	Young modulus (GPa)		Linear compressibility (TPa ⁻¹)	
	E _{min}	E _{max}	K _{min}	K _{max}
value	53.4	127.0	1.7	8.8
anisotropy ratio		2.4	5	.1
direction of	(0, 1, 0)	(0.30, 0, 0.95)	(0.89, 0, 0.45)	(0, 1, 0)
minimum and				
maximum value ^a				

611

612 ^a in Cartesian coordinates (x, y, z).

Table 5. Density (ρ in g/cm³), longitudinal (v_{lon}), transverse (v_{trans}) and averaged (v_m) elastic wave velocity (in m/s), Debye temperature (θ_D in K), and minimum thermal conductivity (κ_{min} in W m⁻¹ K⁻¹) in α -Bi₂O₃ at 0 GPa. The labels *V*, *R*, and *H* refer to the Voigt, Reuss and Hill approximations, respectively.

618

ρ	9.539
$v_{\rm lon}(V,R,H)$	3537.0, 3341.3, 3440.5
$v_{\text{trans}}(V, R, H)$	1943.6, 1798.7, 1872.5
$v_{\rm m}(V,R,H)$	2166.5, 2008.1, 2088.8
$\theta_{\rm D}(V, R, H)$	254.8, 236.2, 245.7
$\kappa_{\min}(V, R, H)$	0.47, 0.43, 0.45

619

- 621 Figure captions
- 622

Figure 1. (Color online) Crystalline structure for α -Bi₂O₃. Large (blue) spheres represent Bi atoms and small (red) spheres O atoms.

625 Figure 2. (Color online) (left) Directional dependence of the Young modulus, in units

of GPa, and (right) spatial dependence of the linear compressibility, in units of TPa⁻¹, in
bismite at 0 GPa. The Young modulus and linear compressibility in a given direction

are shown as a green surface. Data plotted with the ELATE software [54].

Figure 3. (Color online) Theoretical pressure evolution of the elastic constants, C_{ij} , in *α*-Bi₂O₃. Solid lines connecting the calculated data points are guides to the eyes.

Figure 4. (Color online) Theoretical pressure evolution of the elastic stiffness coefficients, B_{ij} , in α -Bi₂O₃. Solid lines connecting the calculated data points are guides to the eyes.

Figure 5. (Color online) Pressure evolution of the elastic moduli (*B*, *G*, and *E*), Poisson's ratio (ν), and *B/G* ratio in α -Bi₂O₃. The results are given in the Voigt (squares), Reuss (circles), and Hill (triangles) approximations. Solid lines connecting the calculated data points are guides to the eyes.

Figure 6. (Color online) Pressure evolution of the percentage elastic anisotropy in both compression (A_B) and shear (A_G), shear anisotropic factors (A_1 , A_2 , A_3) and universal anisotropy index (A_U). Solid lines connecting the calculated data points are guides to the eyes. **Figure 7. (Color online)** Pressure dependence of the Vickers hardness (H_V) in α-Bi₂O₃. The results are given in the Voigt (squares), Reuss (circles), and Hill (triangles) approximations. Solid lines connecting the calculated data points are guides to the eyes.

Figure 8. (Color online) Pressure dependence of the longitudinal (v_{lon}) , transverse (v_{trans}) , and average (v_m) elastic wave velocity in α -Bi₂O₃. The results are given in the Voigt (squares), Reuss (circles), and Hill (triangles) approximations. Solid lines connecting the calculated data points are guides to the eyes.

Figure 9. (Color online) General Born stability criteria in α -Bi₂O₃. P_{mu} reports the pressure at which α -Bi₂O₃ becomes mechanically unstable. Solid lines connecting the calculated data points are guides to the eyes.

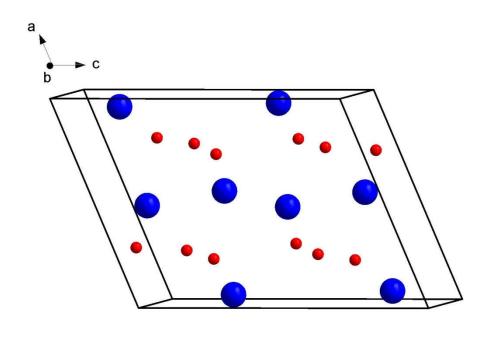
Figure 10. (Color online) Phonon dispersion curves in α -Bi₂O₃ at (a) 0 GPa and (b) 11.5 GPa.

Figure 11. (Color online) Evolution with pressure of (a) the Debye temperature (θ D),

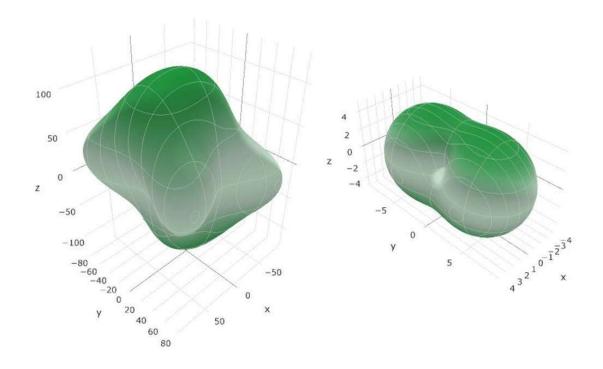
and (b) the minimum thermal conductivity (κ_{min}) in α -Bi₂O₃. The results are given in the Voigt (squares), Reuss (circles), and Hill (triangles) approximations. Solid lines

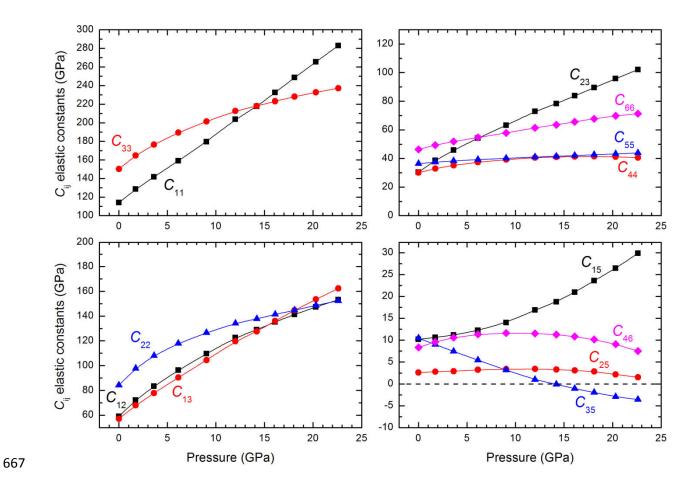
657 connecting the calculated data points are guides to the eyes.

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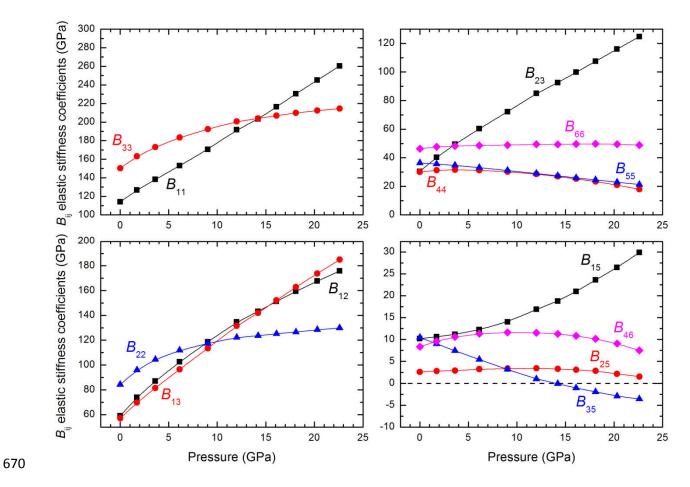
663 Figure 2

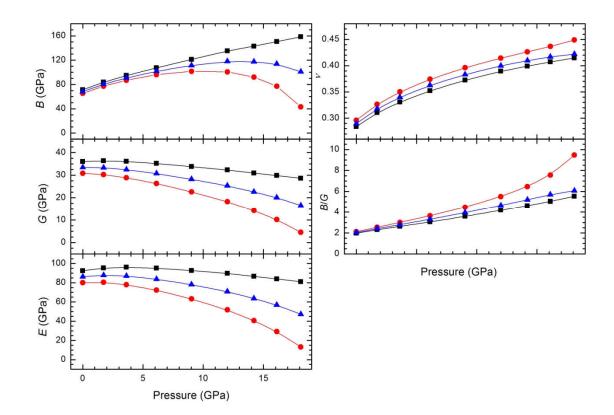


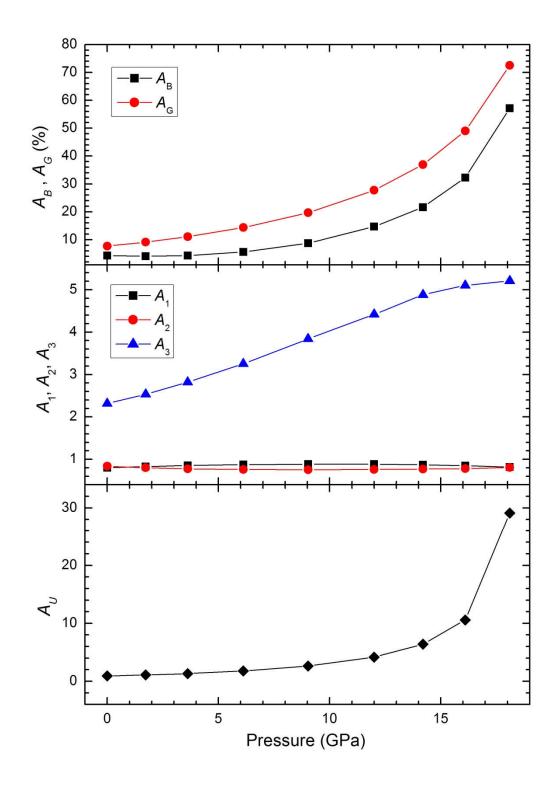












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