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Additional Information

1 Structural, vibrational and electronic properties of α' -Ga₂S₃ 2 under compression

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25 Abstract

26 We report a joint experimental and theoretical study of the low-pressure phase of α' -Ga₂S₃
27 under compression. Theoretical *ab initio* calculations have been compared to X-ray diffraction
28 and Raman scattering measurements under high pressure carried out up to 17.5 and 16.1 GPa,
29 respectively. In addition, we report Raman scattering measurements of α' -Ga₂S₃ at high
30 temperature that have allowed us to study its anharmonic properties. To understand better the
31 compression of this compound, we have evaluated the topological properties of the electron
32 density, the electron localization function, and the electronic properties as a function of
33 pressure. As a result, we shed light on the role of the Ga-S bonds, the van der Waals interactions
34 inside the channels of the crystalline structure, and the single and double lone electron pairs of
35 the sulphur atoms in the anisotropic compression of α' -Ga₂S₃. We found that the structural
36 channels are responsible for the anisotropic properties of α' -Ga₂S₃ and the A'(6) phonon, known
37 as the breathing mode and associated with these channels, exhibits the highest anharmonic
38 behaviour. Finally, we report calculations of the electronic band structure of α' -Ga₂S₃ at different
39 pressures and find a nonlinear pressure behaviour of the direct band gap and a pressure-induced
40 direct-to-indirect band gap crossover that is similar to the behaviour previously reported in
41 other ordered-vacancy compounds, including β -Ga₂Se₃. The importance of the single and, more
42 specially, the double lone electron pairs of sulphur in the pressure dependence of the topmost
43 valence band of α' -Ga₂S₃ is stressed.

45 Introduction

46 Research on Ga_2S_3 chalcogenide has increased in recent years due to its promising properties in
47 different applications. For instance, the high specific capacity of Ga_2S_3 (theoretically estimated
48 to be $682\text{--}1591 \text{ mAh g}^{-1}$)¹⁻⁴ makes it a promising anode material candidate in the field of Na-
49 ion batteries, an alternative to the well-known commercial Li-ion batteries. In this context,
50 recently synthesized Ga_2S_3 rods mixed with graphene have shown a reversible specific capacity
51 of 476 mAh g^{-1} (100 cycles, current density 0.4 A g^{-1}).⁵ On the other hand, many GaS_4 -
52 tetrahedron-based compounds, such as AgGaGeS_4 ,⁶ LiGaS_2 ⁷ and $\text{Li}_2\text{Ga}_2\text{GeS}_6$,⁸ have exhibited
53 large second-harmonic generation (SHG) efficiency and high laser-induced damage threshold
54 (LIDT), necessary for nonlinear optical (NLO) devices. It has been demonstrated that the major
55 contribution to the high birefringence, high SHG and high LIDT in chalcopyrites AgGaX_2 and
56 LiGaX_2 ($X=\text{S, Se, Te}$) comes from the GaX_4 tetrahedra.^{9,10} These results have motivated a
57 systematic research on Ga_2S_3 itself¹¹ that has proved that both α' (monoclinic, space group (S.G.)
58 Cc , No. 9, $Z=4$) and γ (cubic “sphalerite”, S.G. $F-43m$, No. 216) phases have a good optical
59 transparency in the IR region and large NLO effects, with SHGs and LIDTs competitive with
60 commercial NLO materials, like KTIPOPO_4 and AgGaS_2 . In a wider perspective of the
61 optoelectronic industry, Ga_2S_3 is a wide-band semiconductor ($2.5\text{--}3.4 \text{ eV}$)^{12, 13} suitable for a
62 wealth of applications, such as emitters from near IR to blue region,¹⁴⁻¹⁷ UV optical absorbers,¹⁸
63 Terahertz receivers,¹⁹⁻²¹ photovoltaic devices,^{22, 23} gas sensors,^{24, 25} micro-tunable lasers,²⁶ and
64 fluorescent probe materials.²⁷

65 All the above applications reveal the great versatility of Ga_2S_3 and open exciting prospects for
66 inexpensive, non-toxic, and abundant-element-based devices. Therefore, understanding the
67 behavior of the different phases of Ga_2S_3 is necessary. According to the Ga-S phase diagram,^{28,}
68 ²⁹ the α' phase is the stable polymorph at room conditions and melts congruently at 1300 K. At
69 high temperatures (HTs) and non-stoichiometric compositions, three phases are found: γ phase
70 between 1130-1180 K; α (hexagonal, S.G. $P6_1$, No. 161) and β (hexagonal “wurtzite”, S.G. $P6_3mc$,
71 No. 186) phases between 1190 and 1300 K depending on the excess of Ga.³⁰ Due to the tendency
72 of Al, Ga and In atoms to show four-fold coordination when linked to S, Se and Te atoms, such
73 as in zincblende-like binary AX compounds, like GaS, GaSe and GaTe or zincblende-related
74 ternary ABX_2 compounds, like chalcopyrite AgGaS_2 , AgGaSe_2 and AgGaTe_2 , 1/3 of cations sites
75 are empty in the four existing phases with Ga_2S_3 stoichiometry due to the mismatch between
76 the number of cations and anions. The same occurs in zincblende-related defect chalcopyrite
77 CdGa_2S_4 , CdGa_2Se_4 and CdGa_2Te_4 and other chalcogen-based ordered-vacancy compounds
78 (OVCs).³¹⁻³³ However, only the α' phase exhibits an ordered array of these cation vacancies. The
79 S atoms are distributed in an almost hexagonal close-packed arrangement in α' , α and β phases
80 while they are in a cubic close-packed fashion in the γ phase. In fact, α' and α phases are
81 superstructures of the β phase, which means that Ga sites determine the resulting structure.³⁴
82 In fact, if vacancies of the α' phase were filled with cations, we would obtain a distorted wurtzite
83 lattice,^{32, 34} i.e. we would reproduce the β phase of Ga_2S_3 but without disordered vacancy arrays.

84 **Figure 1 a)** shows the monoclinic unit cell of the α' phase. This phase has 5 independent atoms,
85 two Ga (Ga1 and Ga2) and three S (S1, S2 and S3), all occupying $4a$ Wyckoff positions. Vacancies
86 (V) are also represented in **Fig. 1 a)**. Both Ga cations are four-fold coordinated, with Ga1 atoms
87 linked to two S1, one S2 and one S3 atoms and Ga2 atoms linked to one S1, two S2 and one S3
88 atoms. In contrast, the three S atoms are coordinated differently. S1 and S2 are three-fold
89 coordinated, S1 is linked to two Ga1 and one Ga2 atom and S2 is linked to one Ga1 and two Ga2
90 atoms. In turn, S3 is two-fold coordinated to one Ga1 and one Ga2 atoms. Based on the

91 coordination of Ga and S atoms observed in **Fig. 1 a)** and according to the Lewis diagrams (see
92 **Fig. 1 b)**), one of the Ga-S bonds in the GaS₄ tetrahedra should be a dative bond. Furthermore,
93 another feature (not emphasized in the literature) must be highlighted: S1 and S2 atoms have
94 one lone electron pair (LEP), while S3 atoms have two LEPS. The greater electrostatic repulsion
95 of these LEPS in comparison to bonded electron pairs³⁵ explain the decrease in symmetry and
96 the distortion of the S arrangement with respect to the hexagonal wurtzite lattice. In particular,
97 the LEPS give rise to empty channels lying along the *c* axis and pointing towards the ordered
98 cation vacancies (see **Figs. 1 c)** and **d)**).^{36, 37}

99 Several efforts have been made to explore the temperature-composition phase diagram of the
100 Ga-S system at room pressure; however, little information is available about the Ga-S system at
101 high pressure (HP). To our knowledge, only two studies have addressed the behaviour of Ga₂S₃
102 at HP.³⁸ Using *in situ* X-ray diffraction (XRD) and X-ray near edge structure (XANES)
103 measurements supported by theoretical simulations, a pressure-induced phase transition from
104 the monoclinic α' phase to the rhombohedral tetradymite-type structure (S.G. *R-3m*) was
105 observed at 16 GPa and 1300 K.³⁸ Curiously, the pressure-induced stabilization of the
106 tetradymite-like phase under pressure was also observed in other A₂X₃ compounds, as α (R)-
107 In₂Se₃.³⁹ Moreover, the tetradymite-like phase is the typical phase at room conditions of heavy
108 group-15 sesquichalcogenides (β -As₂Te₃, Sb₂Te₃, Bi₂Se₃ and Bi₂Te₃) with exceptional
109 thermoelectric properties and is strongly related to their recently discovered topological
110 insulating properties.⁴⁰ A most recent publication has reported Raman scattering (RS) and
111 electrical conductivity measurements under HP, showing that the α' -Ga₂S₃ undergoes to a
112 semiconductor-metal transformation at 11.3 (17.2) GPa under hydrostatic (non-hydrostatic)
113 condition at room temperature.⁴¹ Such phase at those pressures was associated to the *R-3m*
114 phase, according to the previous Lai *et al.*'s results.³⁸ On top of that, two HP transitions have
115 been observed under decompression, at around 8.0 and 3.0 GPa for both hydrostatic and non-
116 hydrostatic conditions.⁴¹ However, the lack of XRD measurements on downstroke hamper a
117 clear identification of such polymorphs.

118 Cationic LEPS present on oxides and chalcogenides of groups 14 (Ge, Sn and Pb) and 15 (As, Sb
119 and Bi) have been the subject of multiple studies under HP.⁴²⁻⁴⁵ Stereochemically-active cationic
120 LEPS (formed from the cation *s* valence electron pair) occur when the empty *p* orbital of the
121 cation hybridizes with the antibonding cation *s*-anion *p* orbital. This stabilization is promoted by
122 the small energy separation between the cationic *s* and anionic *p* orbitals.⁴⁶ This usually occurs
123 when cations are linked to light chalcogen anions, such as O and S. For Se and Te, with a higher
124 energy of anionic *p* orbitals, the *s*-*p* mixing is smaller and cationic inert LEPS are preferred. This
125 seems to be the reason why the rhombohedral layered tetradymite-like structure is present in
126 Sb₂Te₃, Bi₂Se₃ and Bi₂Te₃, as already commented. The behaviour of anionic LEPS at HP has not
127 been studied in these layered chalcogenides and in other chalcogenides with single LEPS, like
128 ordered vacancy compounds (OVCs).^{31, 47} For example, in molecular solids like S₈⁴⁸ and
129 thioarsenide molecular crystals As₄S_n (*n* = 3, 4 and 5)⁴⁸, their HP behaviour has been studied,⁴⁹⁻
130 ⁵³ but the role played by the double LEPS present in such structures under compression has not
131 been clarified. In this context, we consider that a detailed study of the properties of α' -Ga₂S₃
132 under compression has not been done. More specifically, the role of the channel-like structure
133 of the vacancies formed by the single and double LEPS of S atoms in the anisotropic compression
134 of the α' phase has not been studied. Therefore, the presence of single and double LEPS of S
135 atoms in α' -Ga₂S₃ makes this compound an ideal system to evaluate the influence of these two
136 different stereochemically-active anionic LEPS on the pressure behaviour of α' -Ga₂S₃ and could
137 shed light on the behaviour of anionic LEPS in other solids under compression.

138 The present work is devoted to the study of the structural, vibrational and electronic properties
139 of the α' phase of Ga_2S_3 at HP and addresses the following points: 1) the study of the structural
140 and vibrational properties by means of HP-XRD and HP-RS measurements, complemented with
141 *ab initio* simulations; 2) the study of anharmonic properties arising from the joint study of the
142 HP-RS and HT-RS measurements; 3) the characterization of the different chemical bonds and
143 interactions and the relevance of the single and double LEs at HP via the Quantum Theory of
144 Atoms In Molecules (QTAIM) and electron localization function (ELF) analysis at different
145 pressures; and finally 4) the understanding of the electronic properties under compression via
146 *ab initio* simulations.

147 **Experimental details**

148 Commercial powders of α' - Ga_2S_3 were purchased from Alfa Aesar, with a high purity (99.99%).
149 For HP measurements, powders were loaded in a 150- μm diameter hole made in an Inconel
150 gasket and inserted in a membrane-type diamond-anvil cell with 350 μm diamond-culet
151 diameter. A 4:1 methanol-ethanol mixture was used as a pressure-transmitting medium (PTM).⁵⁴

152 Room-temperature powder angle-dispersive HP-XRD measurements up to 17.5 GPa, with
153 cooper as a pressure sensor,⁵⁵ were carried out in the BL04-MSPD beamline at ALBA synchrotron
154 facility.⁵⁶ The beam was focused by Kirkpatrick-Baez mirrors and images were collected using a
155 Rayonix SX165 CCD detector, with a diameter active area of 165 mm, located at 240 mm from
156 the sample. XRD measurements were performed with a wavelength of 0.4246 Å. The 2D x-ray
157 diffraction area detector data and the calibration with standard LaB_6 were performed with the
158 Python-based DIOPTAS software.⁵⁷ Le Bail method was employed in our refinements with the
159 software GSAS-II.⁵⁸

160 Unpolarized HP-RS measurements up to 16.1 GPa were carried out with a LabRAM HR UV Raman
161 microspectrometer with a thermoelectrically cooled CCD camera and a spectral resolution
162 better than 2 cm^{-1} . Excitation of Raman signal was performed using the He:Ne line (632.8 nm)
163 with a power below 1 mW. The applied pressure was determined by the ruby luminescence
164 method.⁵⁹ Raman peaks have been fitted to Voigt profiles, where the spectrometer resolution
165 is taken as the fixed Gaussian width.

166 **Theoretical details**

167 Density functional theory (DFT)⁶⁰ calculations of α' - Ga_2S_3 have been carried out with the Vienna
168 Ab-initio Simulation Package (VASP).⁶¹ The projected augmented wave (PAW) potentials^{62, 63}
169 have been used to describe valence electrons of Ga ($4s^23d^{10}4p^1$) and S ($3s^23p^4$) atoms, taking
170 into account the full nodal character of the all-electron charge density in the core region but
171 with an affordable basis-set. Calculations were performed with the generalized gradient
172 approximation (GGA) of Perdew-Burke-Ernzenhof revised for solids (PBEsol).⁶⁴ A plane-wave
173 kinetic-energy cutoff of 380 eV was defined in order to achieve highly converged results. For
174 each relaxed structure, calculations were performed with the automatic k-point generation
175 method included in the VASP package using Γ -30, that provides a dense special k-mesh of $6 \times 6 \times 5$,
176 with Hellman-Feynman forces smaller than 0.006 eV/Å per atom and deviations of the stress
177 tensor from the diagonal hydrostatic form smaller than 0.1 GPa.

178 Lattice-dynamic calculations of transversal optical (TO) modes were performed at the zone
179 center (Γ -point) of the Brillouin zone. The direct force-constant approach (or supercell method)
180 with the primitive cell was employed for the calculation of the dynamical matrix at the Γ -point.⁶¹

181 ⁶⁵ In order to obtain the phonon density of states, a $4 \times 4 \times 4$ supercell was used.

182 The DFT charge densities were analysed with the Critic2 program.⁶⁶ The Yu-Trinkle method,⁶⁷
183 implemented in Critic2, was employed to calculate Bader atomic charges and atomic volumes
184 using VASP output files (CHGCAR and AECCAR files). In order to guarantee a smooth pressure
185 behaviour of the topological properties at the bond critical points (BCPs), VASP-optimized
186 structures were recalculated using the linearized augmented plane-waves (LAPW) method
187 implemented in the Elk program, version 6.3.02.⁶⁸ For the ELK calculations, we used a $2 \times 2 \times 2$
188 uniform k-point grid, $R_{min} \times \max \{|G + k|\}$ equal to 9 and $R^{MT} = 2.4$ a.u. and 2.2 a.u. for Ga and
189 S atoms, respectively. At each pressure, the number of critical points fulfils the Morse zero-sum
190 condition. The kinetic, potential and total energy densities were calculated using the Thomas-
191 Fermi approximation, with the semiclassical gradient correction proposed by Kirzhnits.^{69, 70}
192 Finally, the positions of the ELF maxima were located using Critic2 and ELF isosurfaces were
193 obtained with VESTA software⁷¹ by reading the cube files written by Critic2 from ELF3D.OUT files
194 of Elk.

195 **Structural properties under compression**

196 **Figure 2** shows selected HP-XRD patterns of the α' phase at HP up to 17.5 GPa. At 0.1 GPa, we
197 can distinguish several diffraction peaks at angles below 12 degrees that belong only to the α'
198 phase (at around 4.6, 5.2, 7.0, 7.6, 8.1, 8.9, 11.1 and 11.9 degrees) in good agreement with a
199 previous HP study.³⁸ Above 12 degrees, our diffraction peaks from the α' phase are overlapped
200 with those from an impurity that was also observed in a previous HP work using commercial
201 powder.³⁸ Such impurity has been identified as Ga_2O_3 in the commercial sample. For further
202 details, the authors are referred to the Supporting Information (SI) (see **Fig. S1** in the SI). To
203 avoid misunderstanding, we include in **Fig. 2** tick marks for the phases of Ga_2S_3 and Ga_2O_3 and
204 also Cu reflections at certain pressures. It can be observed that emerging diffraction peaks
205 observed at HP come from the splitting of those peaks overlapped from Ga_2O_3 and the α' phase.

206 Le Bail refinement of the pattern at 0.1 GPa (**Fig. S1**) yields the following lattice parameters of
207 α' - Ga_2S_3 : $a=11.121(6)$ Å, $b=6.4093(3)$ Å, $c=7.012(2)$ Å and $\beta=121.276(7)$ degrees with a unit-cell
208 volume of $427.16(2)$ Å³. These values are in good agreement with those reported in earlier
209 works.^{36, 37} At HP, the shortening of the different interplanar distances in the α' phase shift the
210 diffraction peaks at higher angles, and the same happens with the impurity and the Cu peaks.
211 Above 3.3 GPa, several overexposed spots start to emerge in the diffraction rings, making
212 difficult to mask such zones. That forces us to perform Le Bail refinements on regions richer in
213 Cu, leading to a decrease of reflection intensities of α' phase and Ga_2O_3 , especially above 5.7
214 GPa.

215 At 16.1 GPa, several peak intensities of the α' phase start to decrease notably (marked with
216 vertical arrows in **Fig. 2**). Those peaks disappear completely at 17.5 GPa, which does not allow
217 us to refine the α' phase anymore. At this pressure, a pressure-induced phase transition is
218 observed, in agreement with previous works.^{38, 41} New peaks are not observed for the HP phase,
219 maybe due to the lack of heating to overcome a large kinetic barrier, as was suggested in **Ref.**
220 **38**. Curiously, heating has not been required for such HP transition in a recent work.⁴¹ In any
221 case, as previously mentioned, in this work we are only concerned with the behaviour of the α'
222 phase at HP, so the nature and behaviour of the HP phase is out of the scope of the present
223 paper.

224 As can be seen in **Fig. 3 a**), our experimental unit-cell volumes below 2 GPa agree reasonably
225 with those of run-2 in **Ref. 38**. At higher pressures, our experimental unit-cell volume shows a
226 larger compression than that of **Ref. 38**. This mismatch could stem from the less hydrostatic

227 conditions achieved in **Ref. 38** (which used LiF as a PTM) than in this work (we have used 4:1
228 methanol:ethanol mixture as a PTM). We must stress that the decrease of our experimental
229 unit-cell volume is in good agreement with the behaviour observed in both LDA and GGA-PBEsol
230 theoretical calculations shown in **Fig. 3 a**). Note that our GGA-PBEsol calculations show a more
231 pronounced compression than LDA calculations up to 6 GPa,³⁸ and that both calculations show
232 a similar compression rate at higher pressures.

233 Concerning the pressure behaviour of the α' phase, Lai *et al.*³⁸ employed a 2nd-order Birch-
234 Murnaghan equation of state (BM2-EoS) to evaluate the zero-pressure volume, V_0 , and bulk
235 modulus, B_0 .³⁸ This could be reasonable for their run-1 since fixing the pressure derivative, B'_0 ,
236 to 4 allowed them to fit experimental data despite not having enough points. In order to
237 properly determine the order of the BM-EoS for our experimental data and those of run-2 of Lai
238 *et al.*³⁸, we have plotted the normalised stress, F_E , vs finite strain, f_E , (**Figure S2**) by employing
239 EosFit-7c software.⁷² It can be observed that both our experimental data (**Fig. S2 a**) and that
240 from Lai *et al.* (**Fig. S2 b**) fit better to a BM3-EoS than to a BM2-EoS (see **Table 1**) since B'_0 values
241 of both sets of experimental data, which are in agreement between them, are far from 4. In
242 addition, both experimental data sets show a similar B_0 and V_0 .

243 Importantly, the experimental B_0 values are between those of GGA-PBEsol (**Fig. S2 c**) and LDA
244 (**Fig. S2 d**) calculations. The former is better fitted with a BM3-EoS while the latter is better
245 fitted with a BM2-EoS. For the sake of comparison, we have also fitted the LDA data to a BM3-
246 EoS (see **Table 1**). GGA-PBEsol data predict a softer behaviour of α' -Ga₂S₃ at HP than previously
247 reported LDA data.³⁸ On the other hand, the B_0 and B'_0 values for α' -Ga₂S₃ are in agreement with
248 those values previously reported for isostructural β -Ga₂Se₃ (see **Table 1**).^{73, 74} Channels are
249 expected to shrink on compression, thus leading to a low B_0 value and a values of B'_0 much higher
250 than 4, as occurs in other compounds with channels caused by LEPs.^{42, 75} The obtained B'_0 values
251 for α' -Ga₂S₃ are only slightly larger than 4 for both experimental and theoretical data. A deeper
252 study of the pressure behaviour of the LEPs forming the channels will be later provided when
253 the evolution of the charge density at HP is analyzed.

254 The pressure dependence of the lattice parameters and monoclinic β angle of α' -Ga₂S₃ is shown
255 in **Fig. 3 b**) and its inset. In general, there is a good agreement between our experimental and
256 theoretical data on compression. We can observe that GGA-PBEsol data are closer to the
257 experimental lattice parameters than LDA, particularly below 4 GPa. The most striking feature
258 of the α' phase on compression is the change in behavior of the β angle at HP. The experimental
259 (theoretical) β angle shows a negligible pressure dependence below 4 GPa (8 GPa) and a
260 decrease above this pressure. Therefore, both experimental and theoretical data show a
261 symmetrisation of the monoclinic structure at HP.

262 Regarding the axial bulk modulus of the α' phase, Lai *et al.* reported a quasi-isotropic behaviour
263 based on the modified BM2-EoS.³⁸ In order to revise this assessment, we have first tabulated
264 the zero-pressure axial compressibilities, defined as $\kappa_x = -\frac{1}{x} \frac{\partial x}{\partial P}$. As **Table 2** shows, the b axis is
265 the most compressible for our experimental data and those from run-2 in **Ref. 38**. However,
266 theoretical calculations predict the a axis to be the most compressible. Our experimental and
267 theoretical data reflect a higher axial compressibility than the results from **Ref. 38**. Indeed, a
268 and b axes are the most compressible axes of the α' phase. In view of **Table 2**, the α' phase does
269 not have a quasi-isotropic behaviour. In any case, it must be stressed that the conclusions
270 reached in the paragraph as a result of fitting each lattice parameter of monoclinic α' -Ga₂S₃ to a
271 BM3-EoS or a Murnaghan-EoS cannot totally describe the change of the crystal at HP because

272 the β angle also changes with pressure. Consequently, a more elaborate analysis should be done
273 of the axial bulk moduli in monoclinic and triclinic systems.

274 For monoclinic systems, the non-orthogonality in the three crystallographic axes implies non-
275 zero shear stress when each axis is compressed. Therefore, a good description of the axial
276 compressibility requires the study of the isothermal compressibility tensor. For this purpose, we
277 have used the software PASCAL⁷⁶ with the Eulerian approximation⁷⁷ to obtain both orthogonal
278 principal axis ev_i and their compressibilities λ_i by diagonalizing the isothermal compressibility
279 tensor (see **Table 2**). These parameters have been tabulated in a range of pressures from 0 to
280 14 GPa for our experimental and theoretical data and that from Lai *et al.*³⁸ It can be observed
281 that ev_2 coincides with the b axis and shows an intermediate compressibility. On the other hand,
282 ev_1 and ev_3 , contained in the a - c plane, are the directions of maximum and minimum
283 compressibilities, respectively. We find that the direction of maximum compressibility (ψ_{max} ,
284 measured to the c axis, from c to a) for run-2 in **Ref. 38** differs almost 27 degrees from our
285 experimental data and theoretical calculations. We must stress that the compressibilities λ_1 and
286 λ_2 imply between the 70-80 % of the total compression. Concerning ev_3 , the direction of
287 minimum compression (ψ_{min} , measured to the c axis, from c to a), is quite similar in both our
288 results and those from **Ref. 38**, with an angle between 154 and 168 degrees.

289 **Figure 4** helps to visualize how the principal axes are related to the channels in the crystalline
290 structure. It can be seen that ev_2 is perpendicular to the channels, along the direction
291 connecting S1 atoms (**Fig. 4 a**). On the other hand, ev_1 involves the small section of the
292 channels. Specifically, the direction of ev_1 is perpendicular to the array of S2 and S3 atoms along
293 the c axis (**Fig. 4 b**). Therefore, we can say that the empty channels, concretely along their cross-
294 sections, are responsible for most of the compression of the α' phase. The minimum direction
295 of compression, ev_3 , implies that the stiffest direction is almost perpendicular to the layers
296 formed by coplanar GaS₄ polyhedra along the a - b plane (**Fig. 4 b**). We can conclude, in view of
297 the principal axes and their compressibilities, that the α' phase shows a high anisotropy due not
298 only to its monoclinic symmetry but also to the presence of channels along the c -axis of the
299 structure.

300 We turn now to evaluate the role of sulfur LEPs in the compression of α' -Ga₂S₃. First, we studied
301 the change in the GaS₄ polyhedra at HP. For that purpose, the pressure dependence of the
302 relative volume, V/V_0 , of the two different GaS₄ tetrahedra was compared to that the unit-cell
303 volume (see **Fig. S3**). A fit to a BM3-EoS yields $B_0 = 123.0$ (99.1) GPa and $B'_0 = 3.9$ (6.1) for the
304 Ga1S₄ (Ga2S₄) tetrahedron; i.e. both tetrahedra show a different pressure behaviour despite
305 their similar environment. Clearly, the Ga polyhedral bulk moduli are more than double than
306 that of the unit-cell volume (see **Table 1**), thus reflecting that the major contribution to the
307 compression of the α' structure comes from the compression of the empty channels of the
308 structure generated by the sulfur LEPs.

309 To study their similarities and differences, we have analysed the influence of pressure on both
310 GaS₄ polyhedra by examining the calculated Ga-S bond lengths, their compressibilities, effective
311 coordination number (ECoN), distortion index, bond angle variation and quadratic elongation
312 with the software VESTA.⁷¹ **Figure 5 a**) and **b**) plot the change in the Ga-S bond lengths of Ga1S₄
313 and Ga2S₄ tetrahedra, respectively. In both tetrahedra, the three longer Ga-S bonds involve the
314 S1 and S2 atoms (those with only one LEP), meanwhile the shortest Ga-S bond refers to the S3
315 atoms (with double LEPs) (**Fig. 1 b**). Connecting with the Lewis diagram depicted in **Fig. 1 b**), it
316 is clear that Ga-S3 bonds can not be dative due to the double LEPs of S3 atoms. Therefore, in

317 each GaS₄ tetrahedron the dative bond must be associated with either the Ga-S1 or the Ga-S2
318 bond. The dative bond in one of these two GaS₄ tetrahedra leads to different Ga-S bond lengths
319 in both tetrahedra, and therefore, their different pressure dependence.

320 **Figure 5 c) and d)** show the compressibility of each Ga-S bond for both tetrahedra when fitted
321 to a Murnaghan-EoS. As expected, we found that the three longer Ga-S bonds in both tetrahedra
322 are more compressible than the shortest bond. Additionally, the Ga2 tetrahedron shows two
323 bonds with a high compressibility (above $4 \cdot 10^{-3} \text{ GPa}^{-1}$) at low pressures while the Ga1
324 tetrahedron shows only one bond with a high compressibility. This difference explains the softer
325 behaviour of the Ga2 tetrahedron than the Ga1 tetrahedron at low pressure. On the other hand,
326 the longest Ga1-S1 bond in the Ga1S₄ tetrahedron, named Ga1-S1*, behaves in an anomalous
327 way: its compressibility increases slightly up to 4 GPa, and then becomes almost constant up to
328 16 GPa, showing the largest bond compressibility above 11 GPa. A careful examination of the
329 structure shows that bonds Ga1-S1* and Ga2-S2* (the longest distance between Ga2-S2 atoms)
330 are perpendicular to the layers (**Fig. 1 c)**); i.e. they are almost in the same direction of the ev_3 .
331 This result means that the direction of minimum compressibility is in fact related to how the
332 bonds Ga1-S1* and Ga2-S2* are compressed. In fact, we could tentatively identify the Ga1-S1*
333 and Ga2-S2* bonds as dative if we attend to the anomalous pressure dependence of their
334 compressibilities and their relation with the direction of minimum compressibility.

335 The ECoN and distortion index (see **Figure S4 a)**) are equal for both GaS₄ tetrahedra at 0 GPa;
336 however, the Ga2S₄ tetrahedron has a slightly higher ECoN than the Ga1S₄ tetrahedron at HP
337 and, consequently, a smaller distortion index. Largest differences are observed in the bond angle
338 variance and quadratic elongation (**Fig. S4 b)**). These parameters are larger for the Ga1S₄
339 tetrahedron over the entire pressure range. Curiously, both tetrahedra reach almost the same
340 value of these parameters near 16 GPa, thus suggesting that Ga atoms are increasingly displaced
341 from their centroid in both polyhedra as pressure increases and reach similar values just prior
342 to the phase transition. The increase of both parameters is more remarkable in the Ga2S₄
343 tetrahedron, where the Ga2 atom is strongly shifted towards the S1-S2-S3 plane (inset of **Figure**
344 **S4 b)**). Concretely, this increasing distortion under compression can be rooted in the different
345 compressibilities of the Ga1-S1* and Ga2-S2* bonds, tentatively proposed as dative, compared
346 to the other Ga-S bonds. Indeed, the increasing distortions in both polyhedra likely cause a
347 structural instability leading to the phase transition that occurs near 16 GPa on α' phase, as we
348 have observed (see **Fig. 2)** and was already reported in **Ref. 38 and 41**.

349 Finally, in order to understand the changes observed in the Ga-S bonds we have followed the
350 relative change of the free theoretical atomic coordinates (x,y,z) of the unequivalent Ga and S
351 atoms at HP, as can be seen in **Figs. S5, S6 and S7** in the SI and schematized in **Figs. S8 and S9**
352 in the SI. Among all the features observed, included in the SI, we must stress the symmetrisation
353 of the structure with increasing pressure, as clearly shown by the z coordinate of the Ga2 atom
354 (see **Fig. S7 b)**). This coordinate tends to 0 close to 16 GPa thus suggesting a phase transition
355 above that pressure, in agreement with our HP-XRD measurements and previous works.^{38,41}

356 To summarize the structural part, we have found a good agreement between our experimental
357 and theoretical results in the pressure dependence of the lattice parameters and the bulk
358 modulus. Concerning the bulk modulus, thanks to the data of **Ref. 38** and our own results, we
359 have demonstrated a more compressible pressure behaviour of the α' phase that was previously
360 reported and in agreement with the pressure behaviour β -Ga₂Se₃.⁷⁹ On the other hand, our
361 results for lattice parameters match quite well with those published in **Ref. 38**; however, we
362 have performed a detailed analysis of the pressure dependence of the axial compressibility in a

363 monoclinic symmetry, like that of the α' phase, by obtaining the pressure dependence of the
364 principal axes given by the isothermal compressibility tensor. As a result, we have obtained that
365 the compression of α' -Ga₂S₃ is rather anisotropic and that the principal axes with the maximum
366 and intermediate compressibilities are related with the cross-sections of the channels.
367 Moreover, the study of the Ga polyhedral volumes have revealed that the high bulk modulus of
368 both GaS₄ tetrahedra and their low contribution to the relative volume shows the dominant role
369 of the channels in the compression of the α' phase. Curiously, despite the high bulk moduli of
370 both GaS₄ tetrahedra, they experiment a high distortion at HP and show a different compression
371 due to the existence of a dative Ga-S bond in one of the two polyhedra. In particular, the bond
372 angle variance and quadratic elongation show the largest values (almost equal for both
373 tetrahedra) at 16 GPa. Indeed, this increase of the distortion of both GaS₄ tetrahedra and the
374 symmetrisation of the Ga₂ atoms points out to a phase transition above 16 GPa towards a more
375 symmetric HP phase, as previously reported.^{38,41}

376

377 **Vibrational properties under compression and thermal expansion. Anharmonic properties.**

378 **A) Vibrational properties at HP**

379 Now we turn to our HP- and HT-RS measurements on α' -Ga₂S₃. Group theory predicts for the
380 non-centrosymmetric α' phase 27 IR- and Raman-active optical modes at $\Gamma = 13 A' + 14 A''$, plus
381 three acoustic modes ($2 A' + 1 A''$). We have to note that all A' and A'' modes are both Raman-
382 and IR-active; therefore, can show transversal optic (TO) and longitudinal optic (LO)
383 components; i.e. two peaks can eventually be observed for each mode. This means that up to
384 54 vibrational modes can be observed in the Raman and IR spectrum of α' -Ga₂S₃. In this work,
385 experimental modes with TO and LO counterparts are indicated in the Tables and Figures;
386 otherwise only TO counterpart is referred.

387 **Figure 6** shows normalized HP-RS spectra up to 16.1 GPa. All the observed Raman modes have
388 been attributed to the α' phase with no mode corresponding to the Ga₂O₃ impurity. We have
389 observed changes in the Raman spectrum at 15.1 GPa, with broad peaks appearing at 150, 210,
390 410 and 460 cm⁻¹. The new broad peaks increase their intensities at higher pressures, being that
391 located at 410 cm⁻¹ the most intense. These features present at 15.1 GPa, and even more
392 remarkable at 16.1 GPa, show the onset of the pressure-induced phase transition observed by
393 our HP-XRD measurements and in previous works.^{38,41} In fact, recent HP-RS measurement locate
394 the onset of the phase transition at 11.3 and 17.2 GPa under hydrostatic (helium as PTM) and
395 non-hydrostatic (no PTM) conditions, respectively.⁴¹ Therefore, the slight differences in this
396 transition pressure could come from the different hydrosticity conditions achieved in both the
397 present work and the previous ones.^{38,41} The RS spectra of **Figure 6** are in agreement with those
398 published recently in **Ref. 41**, only differing in the transition pressure, as already commented,
399 and the lack of measurements below 100 cm⁻¹ in **Ref. 41**. Unfortunately, we can merely make a
400 visual comparison between our HP-RS measurements and those from **Ref. 41** due to the lack of
401 assignment of the symmetry and pressure coefficients of the observed Raman modes.

402 The RS spectrum of the α' phase can be divided into two regions: the low-frequency region
403 (below 200 cm⁻¹) and the high-frequency region (above 200 cm⁻¹). According to Lucazeau *et al.*,
404 modes at the low (high)-frequency regions correspond to bending (stretching) vibrations of the
405 GaS₄ tetrahedra.⁷⁸ We have observed and followed the pressure dependence of 26 Raman-
406 active modes. As observed in previous studies,^{20, 22, 25, 26, 78-82} the RS spectrum at 0.5 GPa is
407 dominated by the breathing mode near 230 cm⁻¹, which is assigned to the A' symmetry. This

408 peak is related with the symmetric stretching vibration of S atoms around the vacancies in the S
409 atom-vacancy direction, and it derives from the symmetric Ga-S stretching vibrations of GaS₄
410 units.^{20, 26, 78, 79} It is worth mentioning that the breathing mode is also found in adamantite OVCs
411 of AGa₂S₄ stoichiometry,^{31, 83, 84} which are synthesized from Ga₂S₃ and several binary AS (A= Zn,
412 Cd and Hg) compounds. Their crystalline structures are strongly dominated by the Ga₂S₃
413 compound and have ordered vacancies, therefore explaining the presence of the breathing
414 mode.⁸⁰⁻⁸² Moreover, the symmetric stretching mode of S anions around Ga cations has been
415 also observed in GaS near 190 cm⁻¹,⁸⁵ i.e. at a smaller frequency than the breathing mode in
416 Ga₂S₃, likely due to the influence of the Ga-Ga bond in the vibration of the Ga₂S₆ units in GaS.
417 Finally, we can mention that RS spectra with similarities to that of Ga₂S₃ have also been recently
418 reported for the mc64 and mC16 polymorphs of CsGaS₂, constituted by corner-sharing and edge-
419 sharing GaS₄ units, respectively, and showing no cation vacancies in their structures.⁸⁶

420 To better understand the vibrational modes attributed in **Figure 6**, we have plotted the atomic
421 displacements of some of the more representative Raman-active modes of α'-Ga₂S₃ in **Figs. S10**
422 to **S19** in the SI. In short, we have observed that modes below 200 cm⁻¹ are mainly a mixture of
423 Ga-S bending modes and external modes of the GaS₄ units due to translation and rotation of the
424 tetrahedra (see **Figs. S10 to S14**). Between 200 and 300 cm⁻¹, modes are mainly Ga-S bending
425 modes with some admixture of Ga-S stretching modes. In particular, the strongest Raman mode
426 is the breathing mode near 230 cm⁻¹ and corresponds to the A'(6) mode; a mixture of Ga-S
427 bending modes leaded by displacements of S1, S2 and S3 atoms (see **Fig. S15**). The concerted
428 displacements of S atoms lead to a strong symmetric Ga-S stretching mode of S atoms vibrating
429 around the vacancy and to a small symmetric Ga-S stretching ν₁ mode of the GaS₄ units.⁷⁷ This
430 is the reason why this mode is also known as the breathing mode of the vacancy. Finally, modes
431 above 300 cm⁻¹ are mainly asymmetric Ga-S stretching modes (deriving from the ν₃ mode of GaS₄
432 units)⁷⁷ with some admixture of Ga-S bending modes (see **Figs. S16 to S19**).

433 We now turn to analyze the contribution of the different atoms to each vibrational region. For
434 that reason, we have studied the one-phonon density of states (1-PDOS) at 0 GPa (see **Fig. S20**).
435 Notably, our 1-PDOS is similar to that found in the Materials Project Database.⁸⁷ The 1-PDOS
436 shows that the high-frequency region (above 200 cm⁻¹) can be subdivided into three parts: low
437 (200-300 cm⁻¹), medium (300-350 cm⁻¹), and high (above 350 cm⁻¹). The contributions of Ga and
438 S atoms are rather similar in the low-frequency region (below 200 cm⁻¹) and in the low part of
439 the high-frequency region (200-300 cm⁻¹). However, in the medium and high parts of the high-
440 frequency region there is a much larger contribution of S atoms than of Ga atoms. Concretely, S1
441 and S2 atoms (both with three-fold coordination) are predominant in the medium frequency
442 region, meanwhile the S3 atoms (with two-fold coordination) contribute mainly in the high part
443 of the high-frequency region. On the other hand, it is remarkable that the S3 atoms vibrate with
444 low amplitudes when S1 and S2 atoms vibrate with higher amplitudes and conversely. This can
445 be understood as a way to minimize the electrostatic repulsion between the double LEs from
446 S3 atoms and the single LEs of S1 and S2 atoms.

447 The same three phononic gaps observed on the 1-PDOS of α'-Ga₂S₃ are seen in β-Ga₂Se₃, whose
448 1-PDOS can be found in the Materials Projects database.⁸⁷ A further comparison can be made if
449 we take into account other OVCs of AGa₂S₄ stoichiometry (A = Zn, Cd and Hg) with tetragonal
450 defect chalcopyrite structure (S.G. $I\bar{4}$, No. 82, Z=2),⁸⁷ which exhibit only two phononic gaps. On
451 the other hand, the wurtzite type-structure such as ZnS, with no vacancies, only exhibits one
452 phononic gap in its 1-PDOS.^{87, 88} In this context, we can assume that the presence of vacancies
453 in a compound gives rise to more inequivalent atoms in the asymmetric unit cell (five for α'-

454 Ga₂S₃, four for AGa₂S₄ and two for wurtzite type-structure). Curiously, the more inequivalent
455 atoms, the more phononic gaps can occur in the 1-PDOS. The same trend cannot be observed in
456 OVCs with AGa₂Se₄ stoichiometry,⁸⁷ that even do not have a single phononic gap. The reason is
457 the overlapping of the different regions due to the reduced range of frequencies where all
458 phonons occur as a consequence of the higher mass of Se than S. Therefore, the appearance of
459 these phononic gaps seems to clearly depend on the mass of the atoms.

460 **Figure 7** shows the pressure dependence of the experimental and theoretical Raman-active
461 mode frequencies of the α' phase. Note that theoretical mode frequencies refer only to the TO
462 counterpart. In order to do the tentative assignment of the experimental Raman-active modes
463 shown both in **Fig. 7** and **Table S1**, we have compared our experimental frequencies and
464 pressure coefficients with those for the TO components of the theoretical vibrational modes and
465 also with the frequencies at room pressure reported in previous works. As observed, the rather
466 good agreement of experimental and theoretical frequencies and pressure coefficients has
467 allowed us to tentatively assign the symmetry of each experimental mode observed in the α'
468 phase (see **Table S1**). It must be recalled that all A' and A'' modes can show a TO-LO splitting, so
469 two modes can be expected for each theoretically predicted mode in both RS and IR spectra.
470 Therefore, the relatively similar frequencies shown for several A' and A'' modes, coming from
471 Raman- and IR-active modes if symmetry would be higher,⁷⁸ could be the TO and LO component
472 of a single mode. In this respect, we have tentatively assigned the LO counterparts of some
473 phonons to modes whose pressure coefficients do not match well with theoretical TO modes of
474 similar frequencies.

475 As regards the pressure coefficients, we have found that the $A'(6)$ mode has the highest pressure
476 coefficient (~ 8.0 cm⁻¹/GPa) of all vibrational modes, as occurs for the breathing mode of
477 isostructural β -Ga₂Se₃ (at 155 cm⁻¹)⁷² and for the breathing mode (A^1 mode) of defect
478 chalcopyrites as ZnGa₂S₄, CdGa₂S₄ and HgGa₂S₄.^{31, 83, 84} In the low-frequency region, most modes
479 of α' -Ga₂S₃ show a very small pressure coefficient. Moreover, the $A'(1)$ and $A''(1)$ modes; i.e.
480 those with the lowest frequency, exhibit negative pressure coefficients. This has also been
481 observed in a number of compounds, including adamantine-type AB_2X_4 OVCs.^{31, 47, 83, 84} The
482 anomalous decrease of the frequency with increasing pressure of many Raman-active low-
483 frequency modes cannot be explained by an increase of the cation-anion distances with
484 increasing pressure in all these compounds. Instead, it is likely related to an instability of those
485 structures to hydrostatic compression that is evidenced by negative pressure coefficients of low-
486 frequency vibrational modes at the Brillouin zone edge of the cubic lattice (mainly coming from
487 acoustic branches folded into the zone center of low-symmetry structures).⁸⁹ In the low-
488 frequency region, the overlapped Raman peaks are easily discerned thanks to their intensities.
489 According to the idealised P.G. proposed by Lucazeau for the α' phase,⁷⁸ the A' modes derive
490 from Raman-active modes and the A'' modes from IR-active, therefore giving rise a higher
491 intensity for the former than for the latter. As an exception, the $A''(3)$ mode is identified as more
492 intense than the $A'(3)$ mode and both modes have been confirmed by the agreement between
493 their experimental and theoretical pressure coefficients (see **Table S1**). In terms of pressure
494 coefficients, the mode at the right side of the $A'_{TO}(5)$ mode does not match well with the
495 theoretically predicted for the $A''(4)$ mode; therefore it has been labelled as $A'_{LO}(5)$. On the other
496 hand, the high-frequency region shows a massive overlapping of the Raman peaks, and their low
497 intensity hampers their uncoupling even at high pressure except for the $A'(12)$ mode that was
498 also previously observed.^{20, 25, 78, 90} In this region, we consider that the theoretically predicted
499 $A''(7)$, $A''(9)$, $A''(10)$, $A''(12)$, $A''(14)$ modes have not been experimentally observed either in this

500 work or in previous ones. Instead, we have tentatively assigned the LO counterparts of the $A'(8)$,
501 $A''(11)$, $A'(11)$ and $A'(13)$ modes in view of the mismatch between their pressure coefficients
502 with those theoretically predicted and the higher frequencies observed. The modes of the high-
503 frequency region exhibit higher pressure coefficients than in the low-frequency region (see
504 **Table S1**). It must be noticed that the theoretical mode frequencies in the high-frequency range
505 underestimate those observed experimentally, mainly above 325 cm^{-1} from the $A''(11)$ to the
506 $A'_{\text{LO}}(13)$ modes (**Figure 7**). Moreover, it can be noticed that, in general, the modes at the lower
507 part of the high-frequency region show larger pressure coefficients than those at the high part
508 of the high-frequency region and are more separated at 0 (162 cm^{-1}) than at 12 GPa (135 cm^{-1}).
509 These features of these modes, related to TO and LO modes from the cubic structure,
510 respectively, indicate a decrease of the LO-TO splitting on increasing pressure as expected from
511 the more covalent character of the compound on increasing pressure as is also observed in
512 adamantane-type AB_2X_4 OVCs.^{31, 47, 83, 84}

513 In order to evaluate the anharmonicity of each vibrational mode, we have calculated the
514 isothermal mode Grüneisen parameter, γ_i^T (**Eq. 1** in the SI). In general, we observe a good
515 agreement between theoretical and experimental γ_i^T . Excluding the $A'(1)$ and $A''(1)$ modes,
516 which exhibit negative γ_i^T , most modes show positive γ_i^T values between 0 and 1. The only
517 exception is the breathing mode $A'(6)$ that features the highest γ_i^T with a value above 1. This
518 result highlights the high anharmonicity of the stretching vibration of S atoms around the cation
519 vacancies. Taking into account the microscopic definition (**Eq. 3** in the SI), the isothermal
520 averaged Grüneisen parameter, γ_{av}^T , obtained for the α' phase from our experimental and
521 theoretical data are 0.49 and 0.44, respectively. These values are in a good agreement with the
522 common range for γ_{av}^T in tetrahedral compounds (between 0.5 and 1.5).⁹¹

523 B) Vibrational properties at HT

524 **Figure 8** shows the normalized Raman spectra in the α' phase at room pressure and at selected
525 temperatures. To assign the corresponding symmetry to each mode observed at HT, we have
526 used the mode frequencies calculated at 0 GPa. However, due to the lack of lattice-dynamical
527 calculations at HT, we were not able to decouple $A''(3)$ and $A'(3)$ modes. For that reason, we
528 have noted them as the $A''(3)/A'(3)$ mode. From 300 to 923 K, all observed modes shift to lower
529 frequencies, even the the $A'(1)$ and $A''(1)$ modes that show softening at HP. At 948 K, the mode
530 intensities decrease significantly and only the $A''(3)/A'(3)$, $A'(4)$, $A'(5)$, $A'(6)$, $A'(12)$ modes from
531 the α' phase can be discerned. However, new peaks arise at 160, 200, 320 and 410 cm^{-1} at this
532 temperature. At temperatures above 1000 K, all modes of the α' -phase disappear but the new
533 modes persist (see **Fig. S21**). The disappearance of the Raman-active modes of the α' phase is
534 consistent with its melting point at about 1300 K.^{28, 29} Therefore, we attribute the new Raman
535 peaks above 948 K to the impurity of Ga_2O_3 present in the sample (see discussion in the SI in
536 relation to **Fig. S21**).

537 **Figure S22** shows the temperature dependence of the observed mode frequencies in α' - Ga_2S_3
538 and temperature coefficients are summarized in **Table S2**. The mode frequencies show a normal
539 shift to lower frequencies as temperature increases, even the $A'(1)$, $A'(7)$, $A''(8)$, $A'(9)$ and
540 $A''_{\text{LO}}(11)$ modes. However, these modes exhibit a positive linear temperature coefficient but a
541 negative nonlinear coefficient, dominating their temperature dependence behaviour.

542 Moreover, we can observe that the modes from the high-frequency region are more sensitive
543 to temperature than those in the low-frequency region.

544 According to the Klemens model (**Eq. 5** in the SI), we have obtained the cubic and quartic
545 anharmonic contributions, i.e. A and B coefficients, and their absolute ratio $|A/B|$ for all the
546 experimentally observed modes under HT. As can be observed in **Table S2**, low-frequency modes
547 show $|A/B|$ values above 1; i.e. with cubic anharmonic contributions dominating over quartic
548 contributions. On the contrary, the $|A/B|$ ratio is lower than 1 in the high-frequency region,
549 leading to a major contribution of quartic anharmonic contributions.⁹² Therefore, we can
550 confirm that high-frequency modes have an important nonlinear temperature dependence
551 (higher b_2 coefficients), where fourth-order processes are quite relevant. As an exception, the
552 low-frequency $A'(1)$ and $A''(5)$ modes have a similar $|A/B|$ ratio as the high-frequency modes,
553 thus indicating a relevant contribution of fourth-order processes in these modes. In contrast,
554 the high-frequency $A'(8)$ and $A'(10)$ modes have a similar $|A/B|$ ratio as the low-frequency
555 modes, with a clear third-order processes dominance. Concerning the $A'(6)$ mode, it shows the
556 highest $|A/B|$ ratio so it has the highest contribution of third-order processes.

557 Since the $A'(6)$ mode is not overlapped with other modes, we have analysed its linewidth (full
558 width at half maximum, FWHM) under HT (**Eq. 6** from SI). The fit of FWHM data of the $A'(6)$
559 mode at HT to **Eq. 6** in the SI yields $\Gamma_0 = 4.15 \text{ cm}^{-1}$, $C = 0.002 \text{ cm}^{-1}$ and $D = 5.62 \cdot 10^{-7} \text{ cm}^{-1}$, where C
560 and D are the cubic and quartic anharmonic contributions, respectively. These values again
561 indicate a dominance of third-order processes in the phonon-phonon coupling of the breathing
562 mode at low temperatures. However, fourth-order processes become dominant in the FWHM
563 shift above 300 K (see **Fig. S23**).

564 On the other hand, we have calculated the isobaric and isochoric mode Grüneisen parameters,
565 γ_i^P and γ_i^V , in order to estimate the implicit and explicit effects that govern the frequency shifts
566 observed at HT (**Eq. 7** and **8** in the SI). **Table S2** summarizes γ_i^P , γ_i^T and γ_i^V for each experimental
567 Raman-active mode at room temperature. If the quasiharmonic approximation is valid, each γ_i^V
568 must be equal to 0, which means a negligible explicit effect. In fact, we observe that γ_i^V values
569 are quite similar to those of γ_i^P for the α' phase, so that there is a high contribution of the
570 phonon-phonon coupling to the total thermal frequency shift.

571 In order to visualize the contribution of the implicit and explicit effects for each mode,^{93, 94} we
572 have plotted in **Figure 9** the isobaric temperature derivative (the total thermal effect, the left-
573 hand side of the **Eq. 7** in the SI) vs the isothermal pressure derivative (the implicit effect, the first
574 term in the right-hand side of the **Eq. 7** in the SI). The dashed lines are labelled with its
575 corresponding implicit fraction, η_i (**Eq. 9** in the SI), which specifies the ratio of the implicit and
576 total effects. For $\eta_i = 0$ ($\gamma_i^T = 0$) there is only contribution of the explicit effect to the mode
577 frequency shift at HT. On the other hand, for $\eta_i = 1$ ($\gamma_i^V = 0$) the implicit effect is responsible for
578 the mode frequency shift at HT. Finally, for $\eta_i = 0.5$ ($\gamma_i^T = \gamma_i^V$), both effects are comparable and
579 for values $\eta_i > 1$ ($\gamma_i^V < 0$) both effects have opposite signs. Notice that negative η_i is similar to
580 $\eta_i > 1$, but only changing which effect is negative ($\gamma_i^T < 0$). **Table S2** shows η_i for each mode
581 observed. In view of **Figure 9**, we can see that most modes have $\eta_i < 0.5$, implying a
582 predominantly explicit effect. The modes of the low-frequency region ($0.01 < \eta_i < 0.09$) exhibit
583 a higher explicit effect than those of the high-frequency region ($0.07 < \eta_i < 0.30$). A high
584 contribution of the explicit effect has also been observed in several transition metal
585 dichalcogenides.^{94, 95} Concerning the lowest frequency modes, the $A'(1)$ and $A''(1)$ modes have

586 negative η_i , the latter with a higher explicit effect than the former. These negative η_i stem from
587 their softening behaviour under HP, as has been observed for many modes on As_2S_3 , $\beta\text{-GeSe}_2$,
588 SnGeS_3 and SnSe .^{43, 96-98} On the other hand, the $A'(9)$ and $A''(8)$ modes from the high-frequency
589 region show $\eta_i > 1$, with a highly relevant implicit effect, but small explicit effect of the opposite
590 sign. For the $A'(6)$ mode (the breathing mode), the explicit contribution is almost three times
591 higher than the implicit one ($\eta_i \approx 0.25$).

592 In summary, we have studied the pressure and temperature dependence of the Raman-active
593 modes of the α' phase, evidencing the onset of the pressure-induced phase transition at 15.1
594 GPa and its thermal stability until reaching the melting point at about 1300 K. Moreover, the
595 combination of the experimental measurements and theoretical calculations has allowed us to
596 assign tentatively the symmetries of the experimentally observed Raman-active modes. On the
597 other hand, our calculations have lead us to highlight the mixed bending-stretching character of
598 the Raman-active modes, with a remarkable vibration amplitude of Ga and S1 and S2 (S3) atoms
599 in the low-frequency region and lower (high) part of the high-frequency region, respectively.
600 Finally, the combination of HP-RS and HT-RS measurements has enabled us to address the
601 anharmonic effects, yielding a relevant contribution of third (fourth)-order processes in the low
602 (high)-frequency region and a high dominance of the implicit effect in most modes. Particularly,
603 the breathing mode features the highest anharmonicity at HP (γ_i^T) and the highest contribution
604 to the third-order processes to its frequency and FWHM at HT.

605 **Topological analysis of the electron density under compression**

606 Bader's quantum theory of atoms in molecules (QTAIM)⁹⁹ is based on a partition of any
607 electronic system in real space into non-overlapping basins, whose interatomic surfaces satisfy
608 the zero-flux condition. Critical points (CPs) are those points where $\nabla\rho(r)$ vanishes. For the
609 chemical point of view, the bond critical points (BCPs) are CPs linking every pair of atoms along
610 a unique path, the bond path. The nature and strength of these interatomic bonds can be
611 evaluated using the properties at the BCPs. The basin partition also allows the integration of
612 atomic properties, like atomic volumes and atomic charges. Concerning the α' phase, **Figure S24**
613 **a), b) and c)** show the evolution of atomic volumes, Bader atomic charges (Q_i) and charge
614 transfers (CT_i) for S and Ga atoms under compression, respectively. Due to the more diffuse
615 distribution of $\rho(r)$ in anions than cations,¹⁰⁰ the S basins have larger volume than Ga basins,
616 around twice at 0 GPa (see **Figure S24 a)**). Furthermore, S3 atoms (with double LEPS) have a
617 larger atomic volume in comparison to S1 and S2 atoms (with single LEPS). Since all basin
618 properties are additive and these basins fill the space to give the total unit-cell volume,
619 macroscopic properties, such as the bulk modulus, can be obtained by the volume-weighted
620 sum of the contributions in each basin (**Eq. 10** in the SI).

621 According to this fraction, the S atoms represent 75% of the unit-cell volume and, therefore,
622 their contribution dominates the crystal compressibility, in the same way as oxygen and halogen
623 anions dominate in oxide spinels and alkali halides.¹⁰¹ By fitting each $V_i(P)$ to a BM3-EoS (not
624 shown), we have obtained for S1(2), S3, Ga1, and Ga2 basins a bulk modulus of 38.5, 31.9, 80.0,
625 and 66.4 GPa, respectively. In this way, a global bulk modulus of 45.0 GPa is obtained for the α'
626 phase in reasonable agreement with experimental and theoretical values (see **Table 1**). The
627 above results show that the Ga basins are less compressible than the S basins. This is consistent
628 with the high bulk modulus of both GaS_4 tetrahedra as compared to the total volume bulk
629 modulus, as previously discussed. On the other hand, the smaller bulk modulus of S3 atoms
630 compared to S1(2) atoms is explained by the greater compressibility of the double LEPS than the
631 single LEPS. This point reinforces our previous statement that the compressibility of the α' phase

632 is dominated by the channels, whose behaviour under compression is determined by single and
633 double LEPS inside the channels.

634 The Bader charges of the different basins, Q_i , under compression (see **Figure S24 b**) are far from
635 their nominal oxidation states (OS_i). These deviations increase under increasing pressure, thus
636 pointing to a decrease of the ionicity of the compound at HP. It is worth noting the difference in
637 terms of atomic charge between S1(2) and S3 atoms, probably due to the different number of
638 LEPS. In order to emphasize the change in ionicity, we have plotted the charge transfer (**Figure**
639 **S24 c**), using the **Eq. 11** in the SI. The CT of Ga1, Ga2 and S1 basins decrease almost linearly. For
640 the S2 basins, the CT flattens above 8 GPa, and for the S3 basins it decays nonlinearly. Again, the
641 presence of the double LEPS on the S3 atoms is seen, not only in terms of CT, but also in the
642 pressure dependence of this dimensionless parameter. Moreover, the fact that S2 atoms face
643 the S3 atoms along the channels could explain the CT flattening of the S2 basins above 8 GPa.
644 To get a global idea of the ionicity of the whole unit cell, we can average the CT for all the
645 basins.¹⁰² As a result, we have tabulated an average CT of 39.2 and 38.5 at 0 and 16 GPa,
646 respectively. Both averaged CTs fall in the range of polar compounds (30-60), including most III-
647 V crystals and nitrides,¹⁰² and suggests a slight decrease in the ionicity of α' -Ga₂S₃ at HP.

648 To obtain relevant information about the chemical bonds present in the α' phase, we now turn
649 to inspect the topological properties of the BCPs (**Figure S25**). While Ga-S BCPs are located along
650 Ga-S bonds, S-S BCPs fill the channels, except for the S3-S3 BCPs. The appearance of the S-S BCPs
651 occurs because anionic basins share their surfaces not only with cation basins, but with nearest
652 anionic basins as well. In this sense, the S-S BCPs found are associated to the shared IASs among
653 the S basins. Most of the S-S BCPs involve contacts between S3 atoms with other S3 atoms and
654 with S1 and S2 atoms. The BCPs related to the S3 atoms indicate a distorted form of the S3
655 basins, which is likely associated with the double LEPS present inside the S3 basins.

656 **Figure 10 a**) and **b**) show the pressure dependence of $\rho(r)$ and its Laplacian, $\nabla^2\rho(r)$, for every
657 BCP, hereafter named ρ_b and $\nabla^2\rho_b$. We can distinguish the chemical interactions into two great
658 realms depending on the sign of the Laplacian:⁹⁹ $\nabla^2\rho_b < 0$ for shared-shell (SS) interactions
659 (covalent and polar bonds) and $\nabla^2\rho_b > 0$ for closed-shell (CS) interactions (ionic bonds, H-bonds
660 and van der Waals (vdW) interactions). According to these criteria, all BCPs of α' -Ga₂S₃ are found
661 to exhibit CS interactions. In particular, Ga-S interactions are categorized with ionic bonding due
662 to the high ρ_b and $\nabla^2\rho_b$ of Ga-S BCPs. Conversely, S-S interactions are typical vdW interactions
663 due to their low values of these topological properties in S-S BCPs. Due to the non-directionality
664 of the vdW interactions, these S-S BCPs at certain positions in the crystal inform about vdW
665 interactions inside the channels. For every BCP, the ρ_b and its $\nabla^2\rho_b$ increase at HP, as expected.
666 However, the $\nabla^2\rho_b$ for the Ga2-S3 BCP increases smoothly until it decreases slightly above 14
667 GPa. We discuss this result below. For S-S BCPs, this increase indicates the increase of the vdW
668 interactions inside the channels. Among these BCPs, S3-S3 BCPs experiment the smallest
669 increase of the vdW interaction on compression since they are located between the S3 atoms
670 and not along the channels per se.

671 Other topological properties than $\rho(r)$ and its Laplacian, $\nabla^2\rho(r)$, can be considered to describe
672 the nature of the chemical bonds such as the local energy densities, $G(r)$, $V(r)$ and $H(r)$
673 referring to the kinetic, potential and total energy densities, respectively. On the other hand and
674 considering the BCPs, where $\nabla\rho(r) = 0$ and the Kirzhnits¹⁰³ approximation holds, the H_b/ρ_b ,
675 G_b/ρ_b and $|V_b|/G_b$ ratios give adimensional numbers that can be used for the chemical bonding

676 classification. Furthermore, these ratios allow comparing with other systems since they are
677 intrinsic properties and not extrinsic properties as $\rho(r)$ and its Laplacian, $\nabla^2\rho(r)$.^{100, 104-109} The
678 first was introduced by Espinosa¹⁰⁸ and establishes $H_b/\rho_b < 0$ for SS interactions (covalent and
679 polar bonds) and $H_b/\rho_b > 0$ for CS interactions (ionic bonds, H-bonds and vdW interactions);
680 the second was proposed to distinguish in a more generic way between SS interactions
681 ($G_b/\rho_b < 1$) and CS interactions ($G_b/\rho_b > 1$).¹⁰⁷ Lastly, the $|V_b|/G_b$ ratio establishes three
682 chemical bonding regions: a) a pure CS interaction ($|V_b|/G_b < 1$, $\nabla^2\rho_b > 0$); b) a pure SS
683 interaction ($|V_b|/G_b > 2$, $\nabla^2\rho_b < 0$); and c) a transit CS interaction ($1 < |V_b|/G_b < 2$, $\nabla^2\rho_b >$
684 0). Therefore, knowing these density energies and their ratios we can evaluate the pressure
685 dependence of G_b and V_b and properly characterize the different chemical bonds and their
686 variation under compression.

687 **Figure 11 a), b) and c)** shows the pressure dependence of the H_b/ρ_b , G_b/ρ_b and $|V_b|/G_b$ ratios.
688 In view of these ratios at zero pressure and having in mind **Figure 10**, we can discern the nature
689 of the chemical bonding according to the classification scheme of **Ref.**¹⁰⁰ In this way, Ga-S bonds
690 can be described as polar bonds (small ρ_b , positive $\nabla^2\rho_b$, $H_b/\rho_b < 0$, $G_b/\rho_b > 0$ and $1 <$
691 $|V_b|/G_b < 2$), meanwhile S-S BCPs evidence the vdW interactions along the channels (smaller
692 ρ_b than Ga-S BCPs, positive $\nabla^2\rho_b$, $H_b/\rho_b > 0$, $G_b/\rho_b > 0$ and $|V_b|/G_b < 1$). Among the Ga-S
693 bonds, we can distinguish two types: Ga1(2)-S3 and Ga1(2)-S1(2) bonds. The former exhibit the
694 highest absolute values of these ratios, the highest ρ_b and the highest $\nabla^2\rho_b$ for the Ga2-S3 and
695 a similar $\nabla^2\rho_b$ for the Ga1-S3 in comparison to other Ga-S bonds. On other hand, Ga1(2)-S1(2)
696 bonds exhibit intermediate absolute values of the ratios, as well as intermediate values of ρ_b
697 and $\nabla^2\rho_b$. We must stress that using the values of ρ_b and the sign of $\nabla^2\rho_b$ as a criterion to
698 classify the chemical bonding is critical to establish a clear difference between ionic interactions
699 (CS) and polar bonds (SS). We can easily compare these interactions with those found in other
700 systems, like cation-cation polar bonds in CaAl_2Si_2 -type structures¹⁰⁴ or Ti-S bonds in layered
701 TiS_2 ¹¹⁰, and anion-anion vdW interactions in layered SnS_2 ¹⁰⁴ and TiS_2 .¹¹⁰

702 Concerning the Ga-S bonds and their ratios under compression, their values are inside the range
703 considered for polar Ga-S bonds. We can observe an increase of the absolute value of the H_b/ρ_b
704 ratio due to the increase of the prevalence of V_b under compression (**Figure 11 a)**), at the same
705 time that G_b increases (**Figure 11 b)**). However, the increase of G_b overcomes the increase of V_b
706 for most Ga-S bonds (**Figure 11 c)**). Again, the Ga2-S3 bond stands out as an exception, where
707 the increase of V_b is the most relevant. This explains the almost flat pressure behaviour of its
708 Laplacian and its decrease above 16 GPa. Indeed, this fact shows that the Ga1-S3 and Ga2-S3
709 bonds are not equivalent and they could be related with the symmetrisation of the relative z
710 coordinate of the Ga2 atoms at HP (see **Figure S7**), which drives to the phase transition observed
711 in **Ref. 38**. On the other hand, since Ga-S bonds are considered as transit CS interactions ($1 <$
712 $|V_b|/G_b < 2$, $\nabla^2\rho_b > 0$), the greater is the absolute value of H_b/ρ_b ratio, the more covalent
713 and stronger is the bond.¹⁰⁸ In this sense, Ga1(2)-S3 bonds are more covalent and stronger than
714 the rest of the Ga-S bonds.

715 More surprising are the S-S interactions and their behaviour under pressure. As was observed in
716 Ga-S bonds, both V_b and G_b increase under compression (**Figure 11 a)** and **Figure 11 b)**), but
717 contrary to Ga-S bonds, G_b is dominant at zero pressure, resulting in $H_b > 0$ (**Figure 11 a)**). In
718 the same trend of Ga2-S3 bonds, the increase of V_b under compression is higher than the
719 increase of G_b (**Figure 11 c)**) for the S-S interactions, but with $|V_b|/G_b < 1$ at 0 GPa. It is between
720 10 and 14 GPa where the interactions between S-S atoms change, from $H_b/\rho_b > 0$ to < 0 and

721 $|V_b|/G_b < 1$ to > 1 . This change means an increment of the charge concentration in the
722 channels, increasing the vdW interactions due to the decrease of S-S distances. Indeed, the
723 increasing charge concentration in the channels might promote the phase transition observed
724 at 16 GPa. Again, we can find an exception in the S3-S3 interactions, whose ratios are in the
725 whole range of pressures inside the range of values attributed to vdW interactions.

726 Concerning the proposed dative bonds, Ga1-S1* and Ga2-S2*, they exhibit the smallest ρ_b ,
727 intermediate positive $\nabla^2\rho_b$, highest H_b/ρ_b , smaller G_b/ρ_b and $|V_b|/G_b$ compared to the other
728 Ga-S bonds. Moreover, the trend of these properties on compression for both proposed dative
729 bonds is not as soft as it can be seen for the other Ga-S bonds (**Figure 11 a, b**) and **c**). Therefore,
730 in terms of these topological properties and their pressure dependence, we can establish
731 differences on these properties to tentatively assign the Ga1-S1* and Ga2-S2* bonds as dative.
732 In order to highlight other relevant differences between covalent and dative bonds, we analysed
733 the ELF of the α' phase. In particular, we evaluated the one-dimensional ELF profile (1D-ELF)
734 along the Ga-S bonds in both GaS₄ tetrahedra (**Figure S26**). Thanks to the ELF, we can observe
735 the different shell structure of atoms.¹¹¹ In an ionic compound, each maximum in the 1D-ELF
736 corresponds to a closed shell of an atom. Conversely, along the covalent bond between two
737 atoms, the 1D-ELF has a maximum at the centre of the bond. In our case, where polar bonds are
738 present, the valence shells are deformed.¹¹² The 1D-ELF along the Ga1(2)-S3 bond is the only
739 one that looks like different to other 1D-ELFs; however, the double LEP of S3 atoms impede
740 them to form dative bonds with the bonded Ga. For the rest of the Ga-S bonds, the 1D-ELF does
741 not allow us to distinguish between covalent and dative bonds existing in the α' -Ga₂S₃.

742 The maxima of the ELF can also be used to locate the position of the LEPs associated to the sulfur
743 atoms. **Figure 12 a**) shows the representation of the ELF ($\eta = 0.87$) isosurface at 0 GPa, with the
744 single and double LEPs of S1(2) and S3 atoms located inside the channels. At the ELF isosurface
745 level chosen, we have plotted the LEPs with their respective maxima at several pressures, as can
746 be seen in **Figure 12 b**). As pressure increases, the volume of the single and double LEPs is
747 reduced. This observation supports our assertion that the compressibility of the channels, and
748 of the overall α' phase, is ruled by the compressibility of the LEPs. The shape of the LEPs also
749 changes with pressure. In particular, the ELF maxima corresponding to the double LEP are more
750 separated under compression. This separation is evidenced by the angle between the double
751 LEP and the S3 atom (**Figure 13**) that increases from 118.5 to 131.6 degrees from 0 to 14 GPa,
752 and is followed by a decrease at 16 GPa. This decrease at 16 GPa may suggest that the system
753 undergoes an instability around this pressure which is related to the phase transition. On the
754 other hand, the overall increase of LEP-S3-LEP angle with increasing pressure can be explained
755 by the increasing repulsion of the S3 LEPs. The opening of the LEP-S3-LEP angle is a distortion of
756 the S3 environment that affects to the surrounding Ga1(2) atoms. In fact, the Ga1-S3-Ga2 angle
757 varies from 104.9 to 97.5 degrees from 0 to 16 GPa showing also a small jump between 14 and
758 16 GPa indicative of the instability (**Figure 13**).

759 In summary, QTAIM analysis of the electronic structure has allowed to explore the properties of
760 the basin volumes. From the individual basin volumes we have obtained an averaged bulk
761 modulus in agreement with the experimental and theoretical values reported in **Table 1**.
762 Additionally, the pressure behaviour of the atomic charge and their charge transfer show a slight
763 decrease in the ionicity of the compound with increasing pressure in agreement with the
764 decrease of the LO-TO splitting of the Raman-active modes. Concerning the BCPs, both the
765 charge density and its Laplacian, in addition with the three H_b/ρ_b , G_b/ρ_b and $|V_b|/G_b$ ratios,
766 drives a clear classification of the Ga-S bonds and S-S interactions as polar bonds and vdW

767 interactions, respectively. The ELF and its isosurface has helped us locate the single and double
768 LEPs along the channels and to visualize how anion LEPs in α' -Ga₂S₃ evolve under compression.
769 In particular, the double LEP of S3 atoms separate with increasing pressure, distorting the
770 environment of S3-bonded Ga1(2) atoms.

771 **Electronic properties under compression**

772 The electronic band structure of α' -Ga₂S₃ at 0 GPa has been already reported⁸⁷ and its behaviour
773 at HP was partially studied by means of the electronic total density of states (EDOS).^{38, 41}
774 Specifically, both a decrease of the band gap from 1.9 to 1.6 eV between 0 and 9.3 GPa³⁸ and a
775 increase of the band gap from 1.7 to 2.1 eV between 0 to 8 GPa⁴¹ have been reported. Curiously,
776 contradictory those results were strikingly different to those previously reported for
777 isostructural β -Ga₂Se₃, that showed an increase of the band gap with pressure from optical
778 absorption measurements¹¹³ and that were supported by theoretical calculations.⁷⁴
779 Furthermore, those calculations additionally predicted a decrease of the direct band gap above
780 7.5 GPa in β -Ga₂Se₃. The unusual nonlinear behaviour of the direct band gap of β -Ga₂Se₃ was
781 explained by a conduction band anticrossing of the two lowermost conduction bands (CBs) at
782 the Γ point, resulting from the presence of the ordered vacancies in the structure of β -Ga₂Se₃.⁷⁴
783 In fact, the same behavior of the direct band gap was observed in OVCs, like CdGa₂Se₄,⁷⁴
784 HgGa₂Se₄,⁷⁴ MgGa₂Se₄¹¹⁴ and CdGa₂S₄.¹¹⁵ Therefore, it was proposed that this nonlinear
785 behaviour must be rooted in the presence of ordered vacancies in OVCs, regardless of the
786 composition and the structure. Consequently, on the basis of the reported contradictory data
787 we have studied the behavior of the electronic band structure of α' -Ga₂S₃ under compression by
788 means of theoretical calculations.

789 **Figure 14** shows the electronic band structure together with the EDOS and partial electronic
790 density of states (pEDOS) of α' -Ga₂S₃ at zero pressure (plotted with the SUMO package¹¹⁶). High
791 symmetry directions in the BZ have been chosen according the Seek-path tool.^{117, 118} The first BZ
792 and its relevant high symmetry points of α' -Ga₂S₃ are shown in **Figure S27**. From the electronic
793 band structure, we observed that the valence band (VB) shows the valence band maximum
794 (VBM, green points) at the Γ point. Similarly, the CB minimum (CBM, red points) is also at the Γ
795 point, thus yielding a direct band gap of 1.81 eV. Our direct band gap energy is quite similar to
796 the reported 1.9 and 1.7 eV from **Refs. 38** and **41**, respectively. As expected for conventional
797 DFT calculations, these theoretical direct band gaps are smaller than the experimental values
798 that fall in the range of 2.4-3.5 eV.^{14, 119}

799 The contribution of the atomic orbitals to the EDOS is shown in **Figure 14**. It is well known that
800 the anion *p* (cation *s*) orbitals have a major contribution to the top (bottom) of the VB (CB) in
801 heteropolar semiconductors. This is in agreement with our results for α' -Ga₂S₃. The VB is mainly
802 composed by S 3*p* orbitals, specifically the high energy orbitals with the single and double LEPs,
803 which have a strong *p* character. On the other hand, the Ga 4*s*, 3*d* and S 3*p* orbitals contribute
804 equally to the CB. The contribution of the atomic orbitals of S1, S2, S3, Ga1 and Ga2 atomic to
805 the pEDOS at 0 GPa is shown in **Figure S28 a**). The top of the VB is mainly contributed by S3 3*p*
806 orbitals, exhibiting a higher contribution of the double LEPs of the S3 atoms over the single LEPs
807 of S1(2) atoms. On the other hand, the bottom of the CB has dominant contributions from Ga
808 4*s* and S 3*p* orbitals and the topmost of the CB is mainly contributed by Ga 4*p* orbitals and S 3*p*
809 orbitals.

810 The theoretical pressure dependence of the direct and indirect band gap of the α' -Ga₂S₃ is shown
811 in **Figure 15 a**). The behaviour of the direct band gap under HP resembles quite well to that
812 observed in other OVCs, especially in β -Ga₂Se₃.^{74, 113} It increases up to around 7.5 GPa, and then
813 decreases above that pressure. Concerning the direct-to-indirect band gap crossover, it occurs
814 at around 5.5 GPa in α' -Ga₂S₃. A similar direct-to-indirect band gap crossover was also predicted
815 to occur around 6.0 GPa in β -Ga₂Se₃.⁷⁴ **Figure 15 b**) depicts the pressure dependence of the VBM
816 and the lowest three CBs at the Γ point, together the lowest CBM at the V'_2 point. We can clearly
817 observe the direct-to indirect band gap crossover near 5.5 GPa and the conduction band
818 anticrossing at the Γ point at around 7.5 GPa. It is curious that both α' -Ga₂S₃ and β -Ga₂Se₃ pass
819 through the same changes in their band gap at similar pressures (**Figure 15 a**) and **Figure 4 a**) of
820 **Ref. 74**), despite the difference of the anion and that the nature of the indirect band gap and its
821 pressure dependence are different. Finally, concerning the theoretical pressure dependence of
822 the band gap in α' -Ga₂S₃ reported in **Ref. 38**, we think that the pressure behavior reported in
823 **Ref. 38** likely corresponds to the indirect band gap (see **Figure 15 a**)), which seems to be more
824 underestimated within LDA³⁸ than with our GGA-PBEsol calculations. On the other hand, the
825 pressure dependence of the band gap reported up to 8 GPa in **Ref. 41** is quite similar to our
826 results because of the use the GGA functional. However, they have not reported neither the
827 direct-to-indirect band gap crossover at around 5.5 GPa nor the decrease of the direct band gap
828 above 8 GPa showed in **Figure 15 a**). In this context, we think that the reason why we have
829 observed the indirect band gap is because we have used the complex k path suggested by the
830 Seek-path tool^{117, 118} to calculate the complex electronic band structure of the monoclinic Cc
831 phase instead of the simple path used in the Materials Project Database.⁸⁷

832 In order to quantitatively analyze the changes of the topmost VB and lowermost CBs, we have
833 tabulated the relative pressure coefficients for the band energies, yielding 8.1, 6.0, 3.9 and 3.0
834 GPa⁻¹ for the 1st VB, 1st and 2nd CB at the Γ point and 1st CB at the V'_2 point, respectively. In view
835 of these relative pressure coefficients, the energy of the 1st VB at the Γ point (VBM) is the most
836 sensitive under compression. This should not be surprising since the single and double LEPS
837 occupy the high energy levels of the VB. At increasing pressure, the closing of the channels
838 increases the repulsion of the LEPS, especially among those of the double LEPS, driving a strong
839 increase of the VBM energy. We must also stress the larger relative pressure coefficients of the
840 1st and 2nd CB at the Γ point in comparison to that of the 1st CB at the V'_2 point leads to the direct-
841 to-indirect band gap crossover at 5.5 GPa. In the same way, the larger relative pressure
842 coefficient of the 1st in comparison with the 2nd CB at the Γ point leads to the anticrossing of the
843 two lowermost CB bands of the direct band gap near 7.5 GPa, thus resulting in a change of
844 pressure coefficient for the direct band gap at this pressure. In summary, our results for α' -Ga₂S₃
845 show a nonlinear pressure dependence of the direct band gap in good agreement with the
846 expected behavior for all OVCs, as first stated in **Ref. 74**.

847 The electronic band structure at 6 GPa is shown in **Figure S29** to evidence the indirect band gap
848 Γ - V'_2 on α' -Ga₂S₃. A more detailed view of the pEDOS at 6 GPa of **Figure S29** is illustrated **Figure**
849 **S28 b**) for each atom. We can observe a more distributed contribution of Ga 4s, 3d and 4p
850 orbitals throughout the CB, as S 3s and 3p orbitals do. This important contribution is more
851 remarkable at 16 GPa (**Figure S28 c**)), where the atomic orbitals from Ga and S atoms are more
852 equally distributed over the CB.

853 As noted above, the most sensitive pressure behaviour of the VBM is rooted in the strong
854 repulsion of the LEPS located in the channels, as a consequence of their approach with increasing

855 pressure. According to **Figure 5 c)** and **d)**, the Ga-S bond compressibilities decay to a greater
856 extent (except for the Ga1-S1* bond) below 7.5 GPa. At the same time, the charge transfers
857 decrease smoothly (**Figure S24 c)**) as a consequence of the increased covalent nature and
858 strength of the Ga-S bonds (**Figure 11 a)** and **c)**) under compression. Moreover, the repulsion
859 between the LEPs of the S3 atoms is moderate below 7.5 GPa (**Figure 13**). Therefore, it is around
860 7.5 GPa where various changes in the trends mentioned are observed. These results allow us to
861 tentatively suggest that the nonlinear pressure dependence of the direct band gap near 7.5 GPa
862 could be related to the slight decrease of the compressibilities, as a result of the prevalence of
863 the repulsive forces between the nuclei, and the increase of the vdW interactions along the
864 channels (**Figure 11 a)** and **c)**). These latter are caused by the higher increase of the repulsion
865 between the LEPs of the S3 atoms as a consequence of the closing of the channels.

866 In summary, we have reported in this section the pressure dependence of the direct and indirect
867 band gaps of α' -Ga₂S₃. Our results evidence a clear similarity between the behaviour of the
868 electronic band structure of α' -Ga₂S₃ under compression and those of other OVCs, especially
869 with isostructural β -Ga₂Se₃. In particular, we have found a nonlinear behavior of the direct band
870 gap and a direct-to-indirect band gap crossover, as in β -Ga₂Se₃. This similarity was not reported
871 in two previous works that also analysed the electronic band structure of α' -Ga₂S₃ under
872 compression. Additionally, we have emphasized the role of the anion LEPs, contributing mainly
873 to the topmost VB, in the changes of the electronic band structure of α' -Ga₂S₃ under
874 compression.

875 **Conclusions**

876 In this work, we have performed HP-XRD, HP-RS and HT-RS measurements on α' -Ga₂S₃, in
877 addition to *ab initio* calculations. With our experimental HP-XRD measurements and theoretical
878 results, we have revised the pressure dependence of the volume and lattice parameters of α' -
879 Ga₂S₃ in comparison with previously data.³⁸ In particular, we have shown that the unit-cell
880 volume vs. pressure is better fitted to a BM3-EoS, with a softer behaviour under compression
881 than previously reported³⁸ and in agreement with the pressure behaviour β -Ga₂Se₃.⁷⁹ We have
882 reported the isothermal compressibility tensor between 0 and 14 GPa and its principal axes with
883 the maximum, minimum and intermediate compressibilities and their relation with the channels
884 of the structure. In particular, the maximum and intermediate compressibilities are related to
885 the cross-sections of the channels. The role of the channels in the anisotropic pressure
886 behaviour of the unit cell is highlighted. In particular, we have found that the channel
887 compressibility governs the unit-cell volume compressibility and that the two GaS₄ tetrahedra
888 become more distorted at increasing pressure, with the distortion of both tetrahedra being
889 equal at 16 GPa. This result, in addition to the symmetrisation of the z coordinate of Ga2 atoms,
890 points out to the pressure induced phase transition observed at 16 GPa, as already reported in
891 **Ref. 38** and **41**.

892 HP-RS measurements in α' -Ga₂S₃ have allowed us to tentatively assign the symmetries of the
893 observed Raman-active modes, to give the pressure coefficients of the Raman-active modes and
894 to reveal structural changes above 15.1 GPa, in agreement with the pressure-induced phase
895 transition reported previously.^{38, 41} Besides, HT-RS measurements have shown the thermal
896 stability until reaching the melting point at about 1300 K. Our lattice-dynamic calculations and
897 the Lucazeau's idealized model⁷⁸ have allowed us to assign the symmetry of most of the
898 observed Raman modes. Additionally, our calculations have lead us to highlight the mixed
899 bending-stretching character of the Raman-active modes, with a remarkable vibration

900 amplitude of Ga and S1 and S2 (S3) atoms in the low-frequency region and lower (high) part of
901 the high-frequency region, respectively. The anharmonic effects of the different experimentally
902 observed Raman modes have been studied, showing their different cubic and quartic
903 anharmonic contributions, with a predominance of the implicit effect in most modes. In
904 particular, the $A'(6)$ mode, the breathing mode related to the vacancies, has the highest
905 anharmonicity and cubic anharmonic contribution at HP and HT, respectively.

906 The study of the topology of the electron density at HP within the QTAIM method has allowed
907 us to explore the properties of the basin volumes and confirm the bulk modulus as well the
908 increase of the covalency (decrease of the ionicity) of the compound with increasing pressure.
909 Additionally, it has allowed us to classify the different chemical bonds in the structure and has
910 evidenced the strengthening of the Ga-S bonds at increasing pressure and the increase of the
911 vdW interactions along the channels, especially above 10 GPa. On the other hand, we have
912 tentatively assigned the Ga1-S1* and Ga2-S2* bonds as dative bonds attending to the notable
913 differences in bond lengths, compressibilities and topological properties of these two bonds
914 with respect to the other Ga-S bonds. Thanks to the ELF, we have visually evaluated the
915 decreasing volume of single and double LEPs and the increasing angle between the double LEPs
916 of the S3 atoms at HP. This angle shows a decrease around 16 GPa that may suggest an incipient
917 instability related to the first-order phase transition already commented. Moreover, we have
918 presented the use of the topological properties as ρ_b , $\nabla^2\rho_b$ and the H_b/ρ_b , G_b/ρ_b and $|V_b|/G_b$
919 ratios as an easy methodology to evaluate the chemical interactions within a system and
920 enabling the comparison with others, in addition to studying the pressure dependence of these
921 interactions.

922 Finally, our calculated electronic band structures at HP have evidenced a strong nonlinear
923 pressure behaviour of the direct band gap and a pressure-induced direct-to-indirect band gap
924 crossover. Both features are similar to those previously reported for isostructural β -Ga₂Se₃ and
925 other OVCs. Moreover, we have shown that, as in other OVCs, the VBM in α' -Ga₂S₃ shows a very
926 large pressure coefficient that is related to the strong contribution of anionic LEPs. In this
927 context, there is a greater relevance for the topmost VB of the double anion LEPs in α' -Ga₂S₃
928 than in less complex OVCs. In α' -Ga₂S₃, the strong compression of the anion LEPs leads to a strong
929 increase of the VBM energy so LEPs play a similar role as in adamantane-based OVCs, like
930 CdGa₂S₄. The strong compression of these LEPs explains the strong increase of the VBM energy
931 and the nonlinear dependence of the direct band gap.

932 In summary, the present work has been aimed at understanding in detail the behaviour of the
933 low-pressure phase of α' -Ga₂S₃ under compression. Our study has shown that the HP properties
934 of α' -Ga₂S₃ are similar to those of β -Ga₂Se₃ and other adamantane-based OVCs. Therefore, this
935 work has contributed to establish a general behaviour under compression of materials with
936 unpaired anion electrons in the crystalline structure.

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1240 **FIGURES**

1241 **Figure 1.** (Color online) a) Perspective view of α' -Ga₂S₃. Green, yellow and pink balls represent
1242 Ga atoms, S atoms and vacancies, respectively. Ga atoms are four-fold coordinated while S1 and
1243 S2 atoms on one hand and S3 atoms on the other hand are three- and two-fold coordinated,
1244 respectively. b) Lewis dot diagrams for Ga and S atoms and their nearest bonded neighbour
1245 atoms. c) View of the α' phase perpendicular to the c axis. The arrangement of the vacancies
1246 forms the channels along the c axis. d) View of the α' phase perpendicular to the b axis. Layers
1247 based on GaS₄ tetrahedra parallel to the a - b plane are stacked along the c axis; i.e. coincident
1248 with the direction of the channels. When vacancies are included, the strong resemblance of the
1249 monoclinic structure of the α' phase with the wurtzite structure can be observed.

1250 **Figure 2.** (Color online) Selected XRD patterns of α' -Ga₂S₃ at several pressures. Tick marks
1251 corresponding to the α' phase, impurity and Cu reflections are added at a few pressures to
1252 enlighten the appearance/disappearance of diffraction peaks. Down arrows show the
1253 disappearance of several reflections of the α' phase at 16.1 GPa.

1254 **Figure 3.** (Color online) a) Experimental (black solid circles) and theoretical GGA-PBEsol (black
1255 solid line) pressure dependence of the unit-cell volume of α' -Ga₂S₃. Experimental Run-2 (blue
1256 empty circles) and theoretical LDA (black dash line) from **Ref. 38** are also plotted for comparison.
1257 Both experimental data are fit to a BM3-EoS (dotted lines). b) Pressure dependence of the lattice
1258 parameters a , b and c of α' -Ga₂S₃. The inset shows the pressure dependence of the monoclinic
1259 β angle.

1260 **Figure 4.** (Color online) a) View of α' -Ga₂S₃ perpendicular to c axis where ev_2 is located along b
1261 axis, covering the major section of the vacancy channels. ev_2 refers to the intermediate direction
1262 of compression. b) View of α' -Ga₂S₃ perpendicular to b axis, parallel to a - c plane. ev_1 intersects
1263 the small cross-section of the vacancy channels, meanwhile ev_3 is almost normal to the GaS₄-
1264 based layers. ev_1 and ev_3 refer to the maximum and the minimum direction of compression,
1265 respectively. Ga and S atoms and vacancies are shown in green, yellow and pink, respectively.

1266 **Figure 5.** (Color online) Pressure dependence of Ga-S bond lengths and compressibilities in the
1267 two GaS₄ polyhedra: a) and c) for Ga1S₄ and b) and d) for Ga2S₄ polyhedra. Ga1-S1* and Ga2-
1268 S2* refer the longest bond lengths between the Ga1-S1 and Ga2-S2 atoms, respectively.

1269 **Figure 6.** (Color online) RS spectra of α' -Ga₂S₃ at selected pressures. The different A' and A''
1270 modes have been tentatively identified in the RS spectra.

1271 **Figure 7.** (Color online) Pressure dependence of the theoretical (solid lines) and experimental
1272 (symbols) Raman-active mode frequencies of α' -Ga₂S₃.

1273 **Figure 8.** (Color online) RS spectra of α' -Ga₂S₃ at selected temperatures. Up arrows show the
1274 appearance of new peaks at 948 K.

1275 **Figure 9.** (Color online) Correlation between the temperature and pressure derivatives of the
1276 experimental Raman-active mode frequencies. Dashed lines are labelled by the corresponding
1277 implicit fraction.

1278 **Figure 10.** (Color online) Pressure dependence of the charge density ($\rho(r)$) and its laplacian
1279 ($\nabla^2\rho(r)$) for the BCPs.

1280 **Figure 11.** (Color online) Pressure dependence of the a) H_b/ρ_b , b) G_b/ρ_b and c) $|V_b|/G_b$ ratios
1281 for the BCPs.

1282 **Figure 12.** (Color online) a) View of α' -Ga₂S₃ perpendicular to the *c* axis with the single and
1283 double LEPs along the channels at 0 GPa. b) Detail of the single and double LEP volume of the S1
1284 and S3 atoms, respectively, at selected pressures (ELF isosurfaces at $\eta = 0.87$).

1285 **Figure 13.** (Color online) Pressure dependence of the LEP-S3-LEP and Ga1-S3-Ga2 angles.

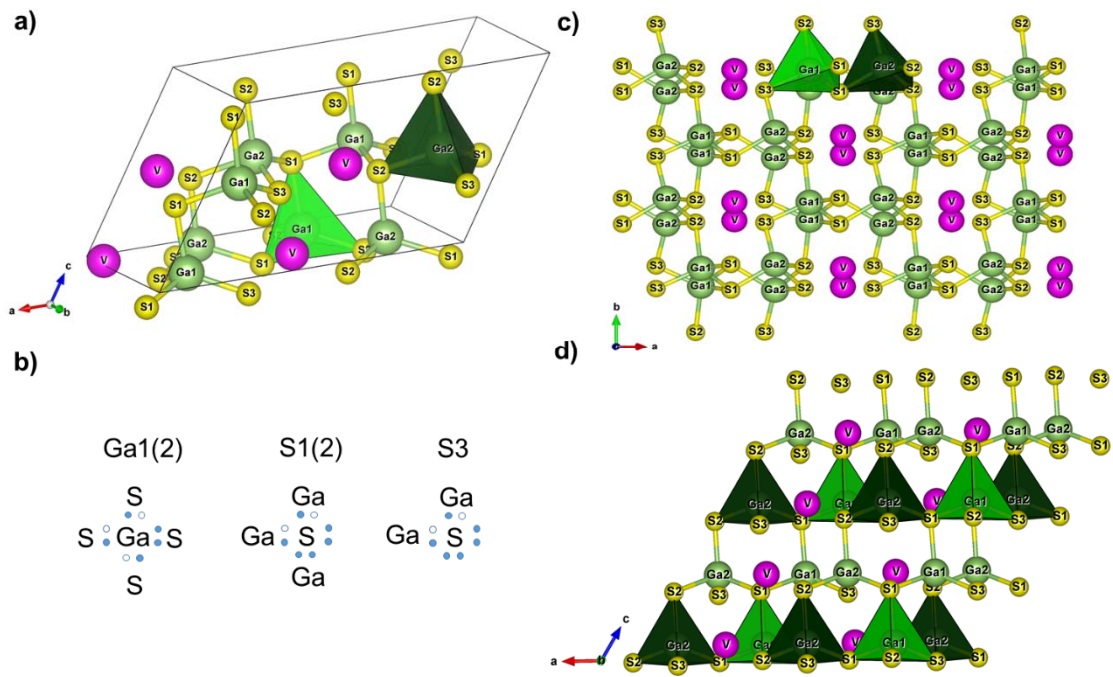
1286 **Figure 14.** (Color online) Electronic band structure and PDOS of α' -Ga₂S₃ at 0 GPa. Red and green
1287 points mark the position of the conduction band minimum and valence band maximum,
1288 respectively. The 1st valence band (VB) and the 1st and 2nd conduction bands (CB) are labelled.

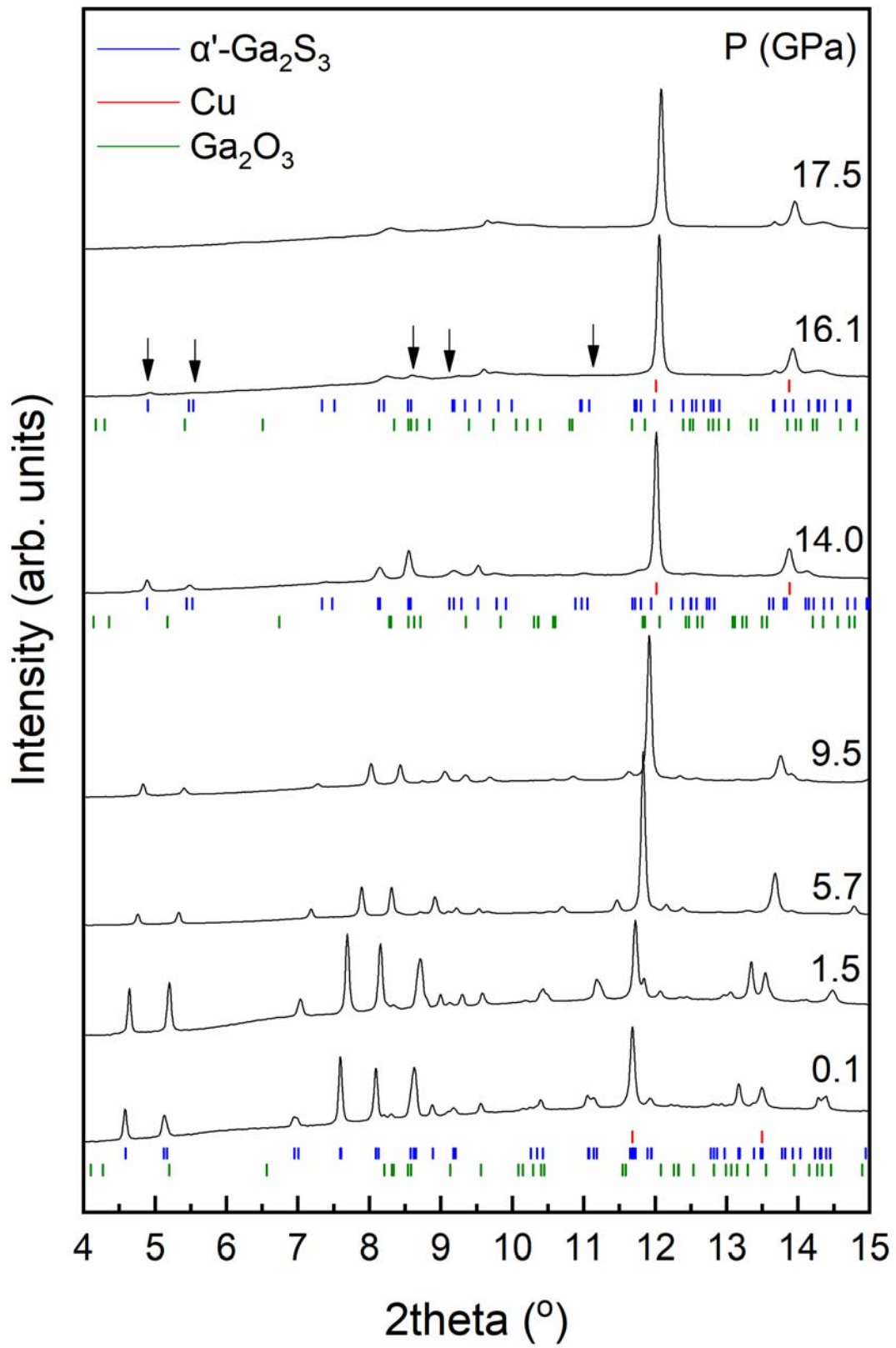
1289 **Figure 15.** (Color online) a) Pressure dependence of the theoretical direct and indirect band gaps
1290 in α' -Ga₂S₃. b) Pressure dependence of the calculated energies of the three lowest conduction
1291 band (CB) minima and valence band (VB) maximum at the Γ point, together with the CB minimum
1292 at the V'_2 point. Blue solid (dash) line indicate the pressure for the the direct-to-indirect band
1293 gap crossover (conduction band anticrossing of the direct band gap) at 5.5. (7.5) GPa.

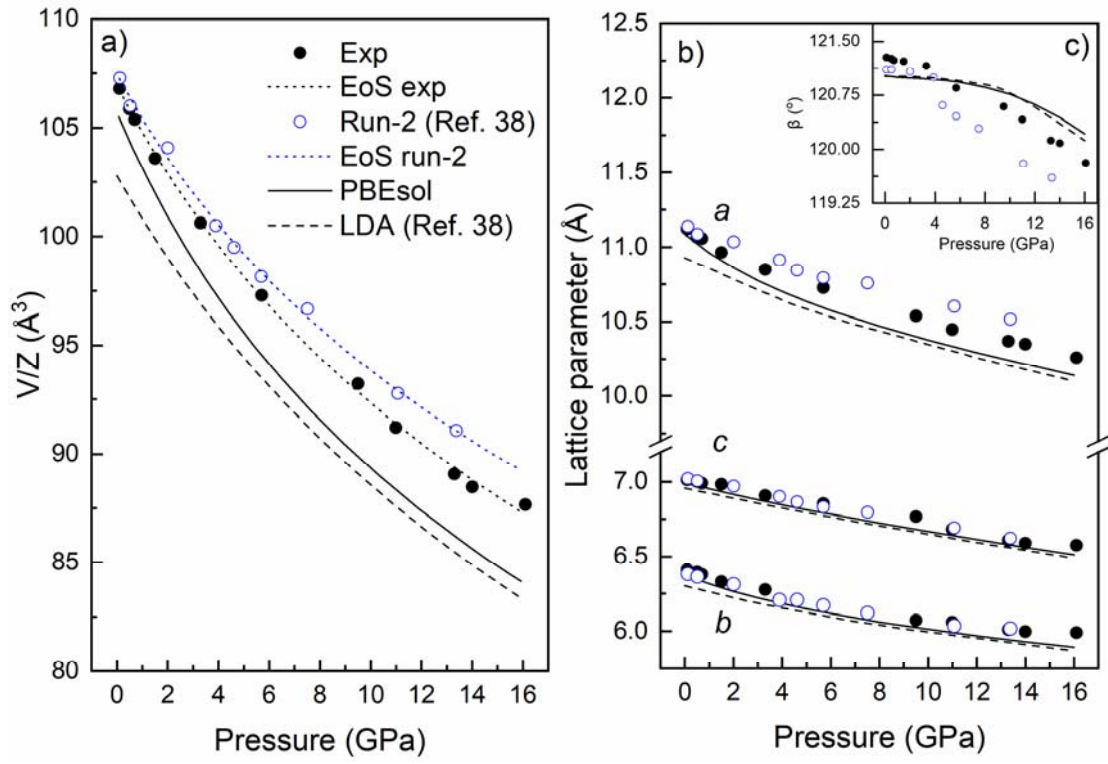
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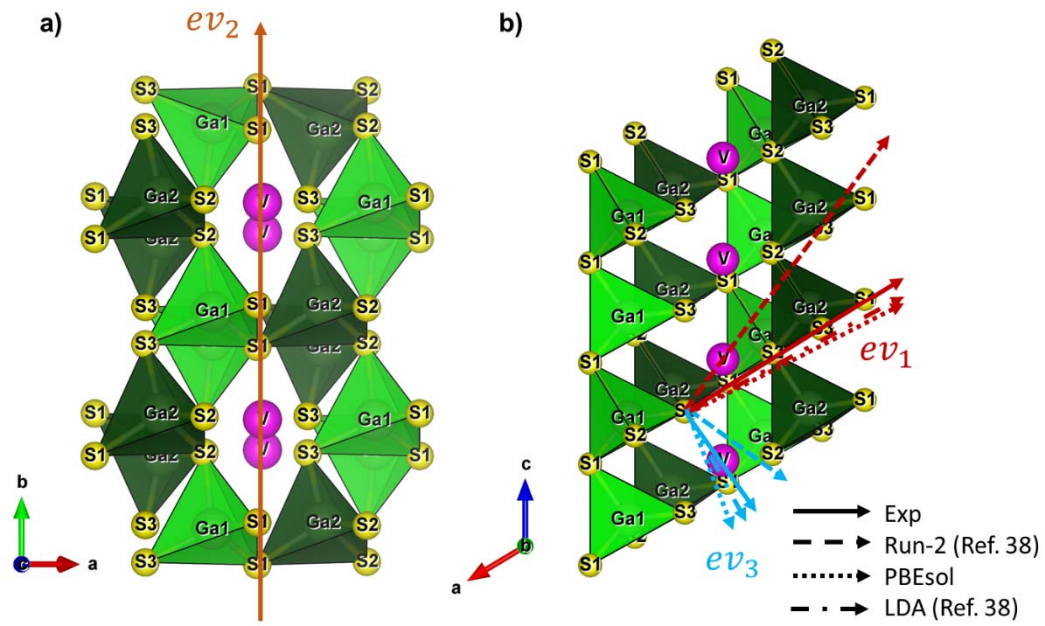




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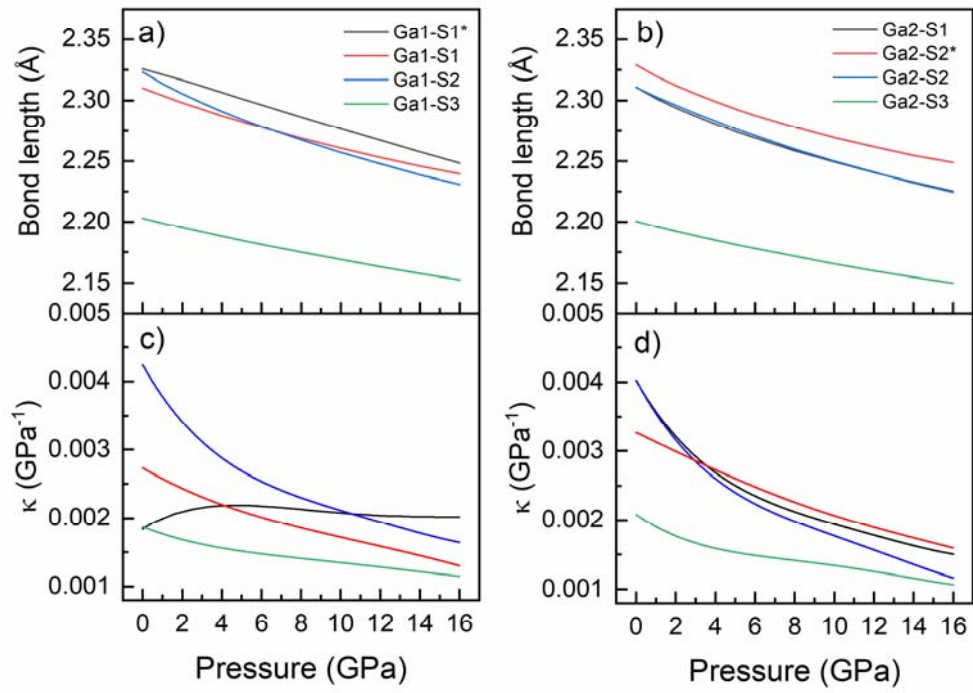
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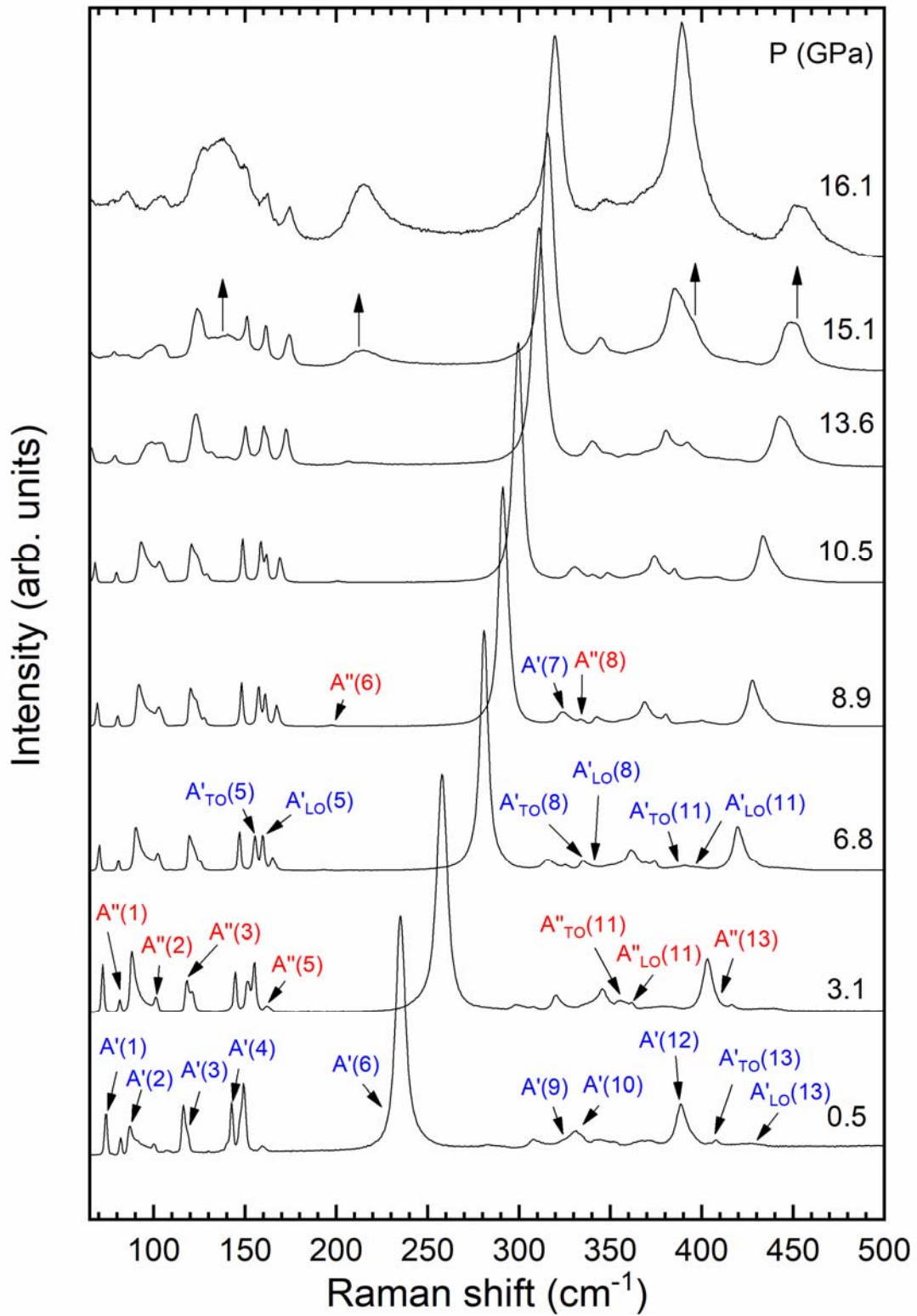
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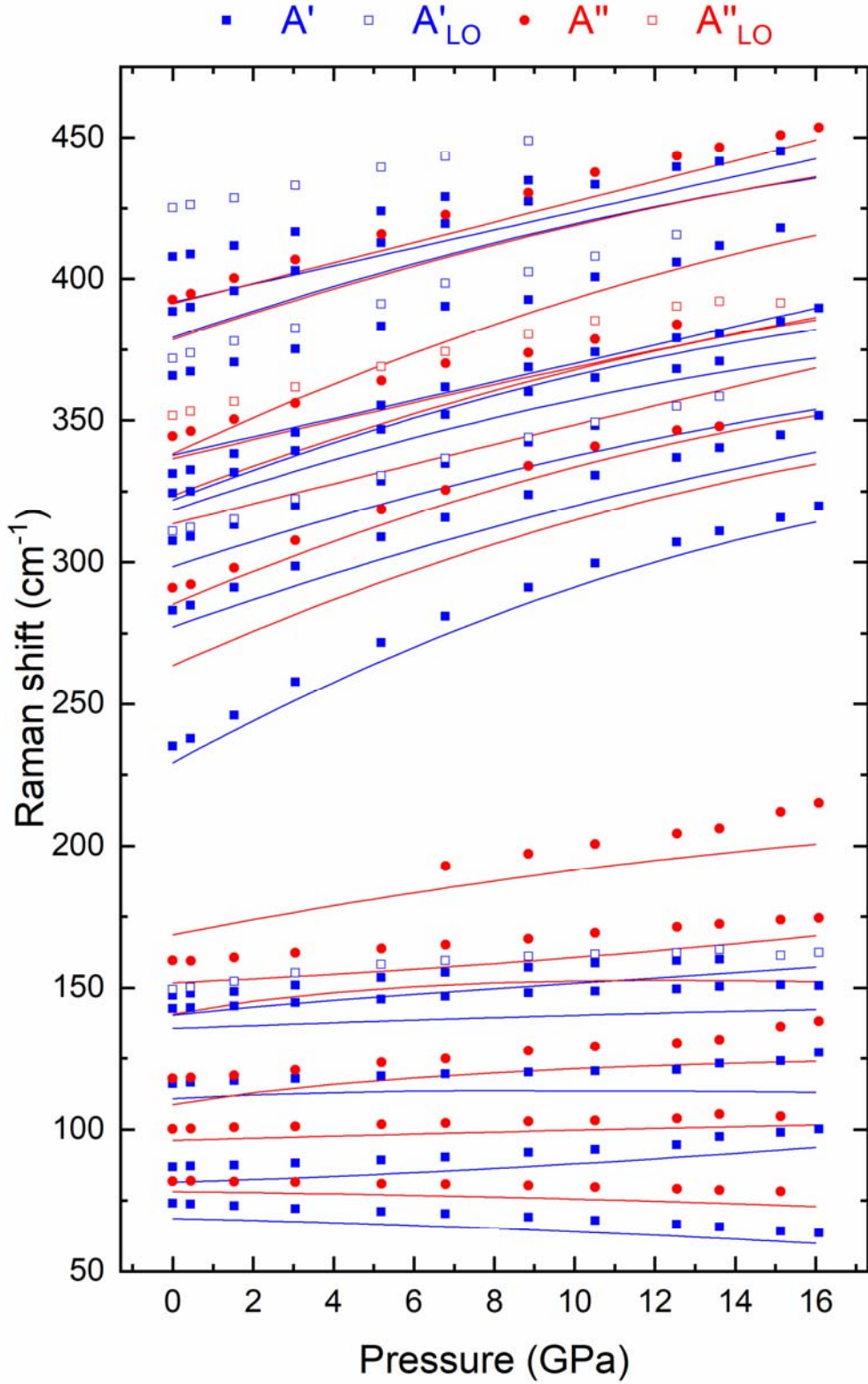
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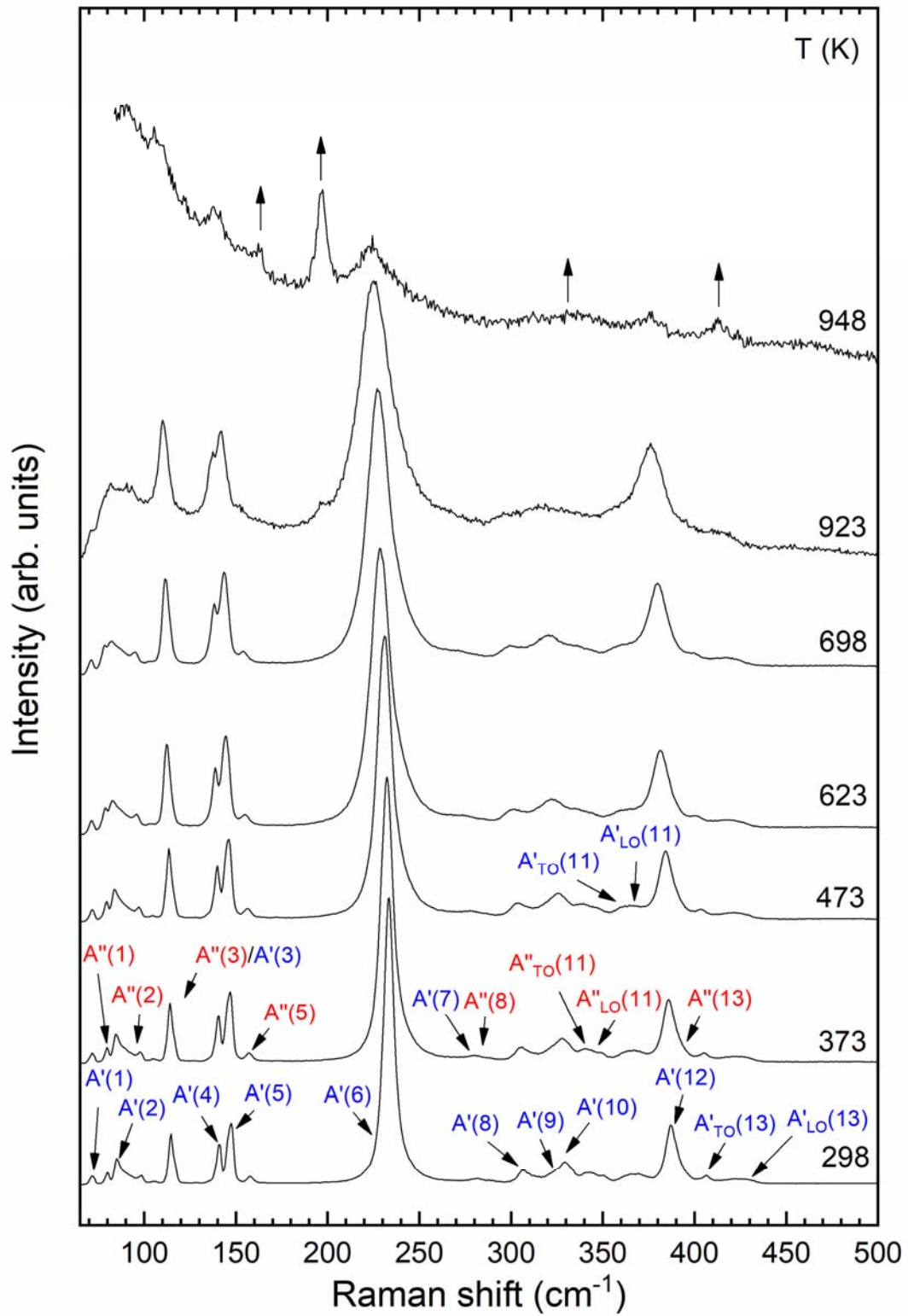


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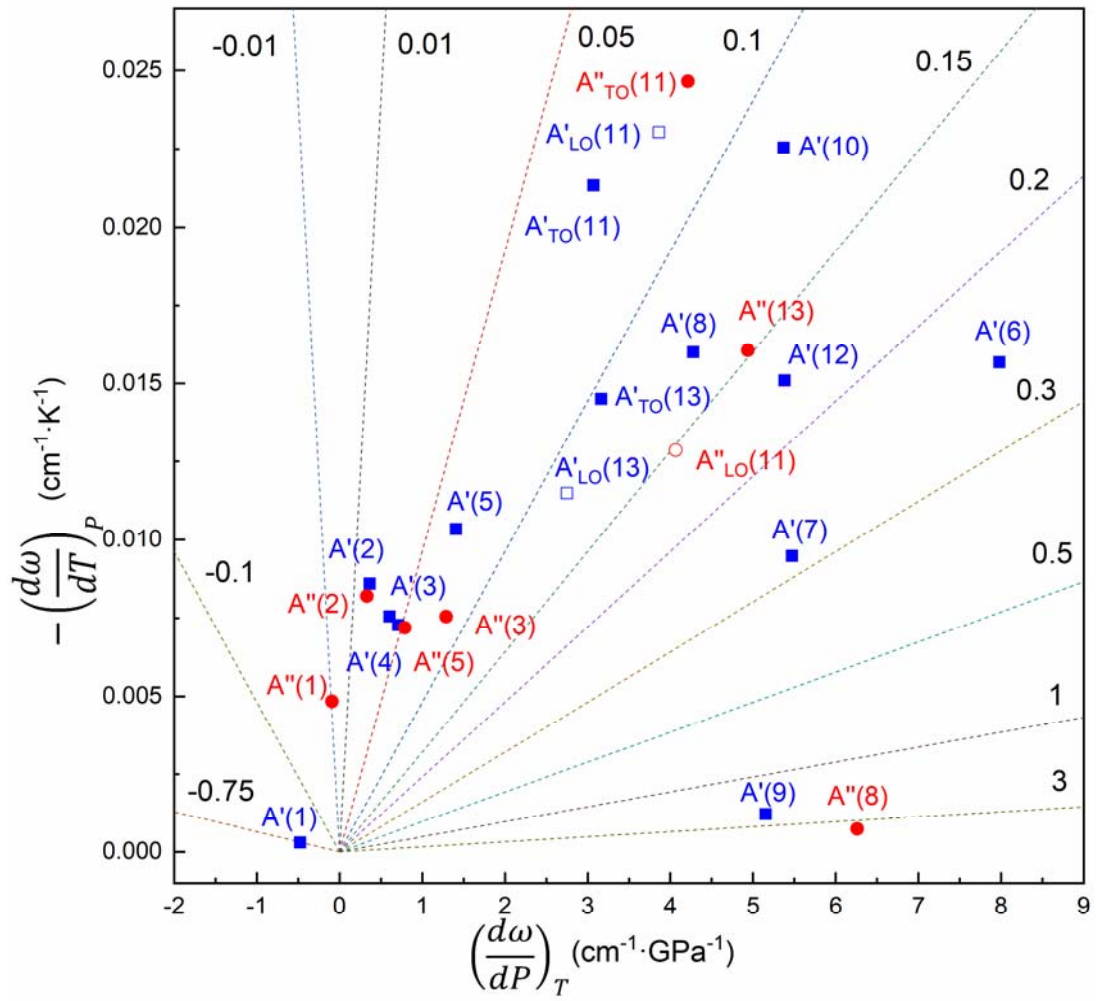
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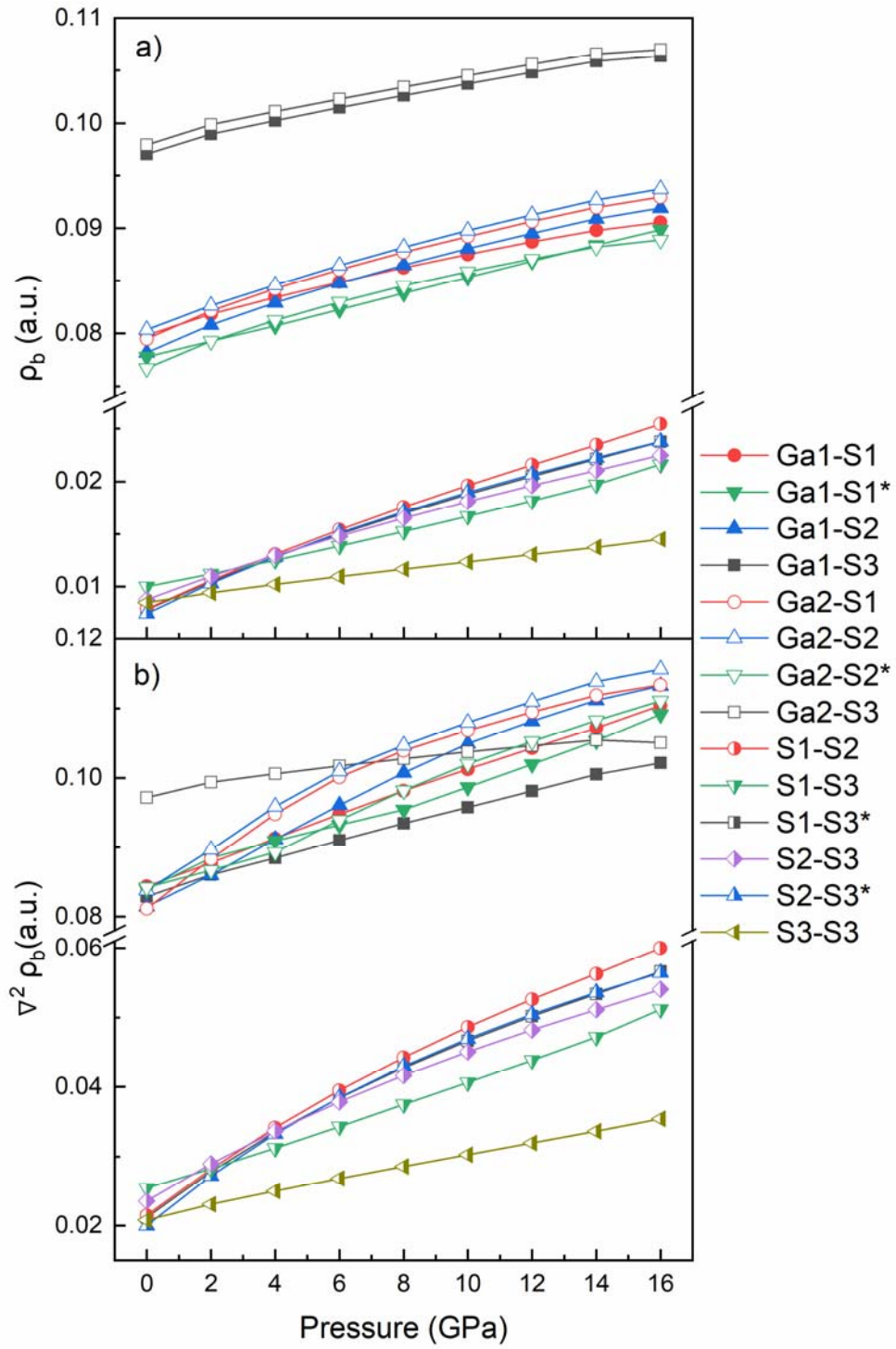
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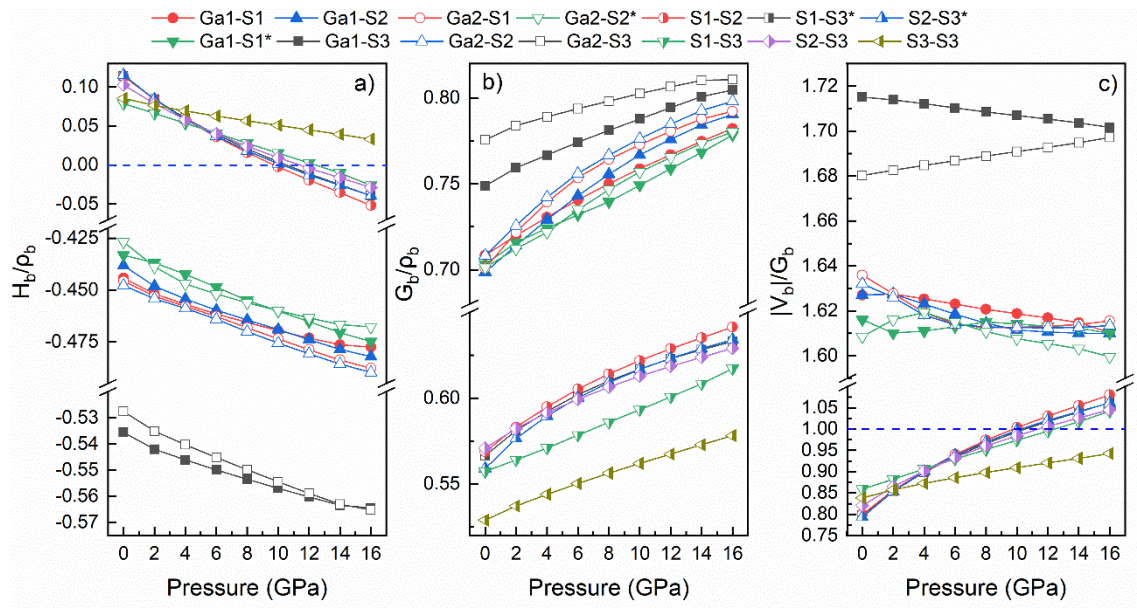


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1331 **Figure 11**

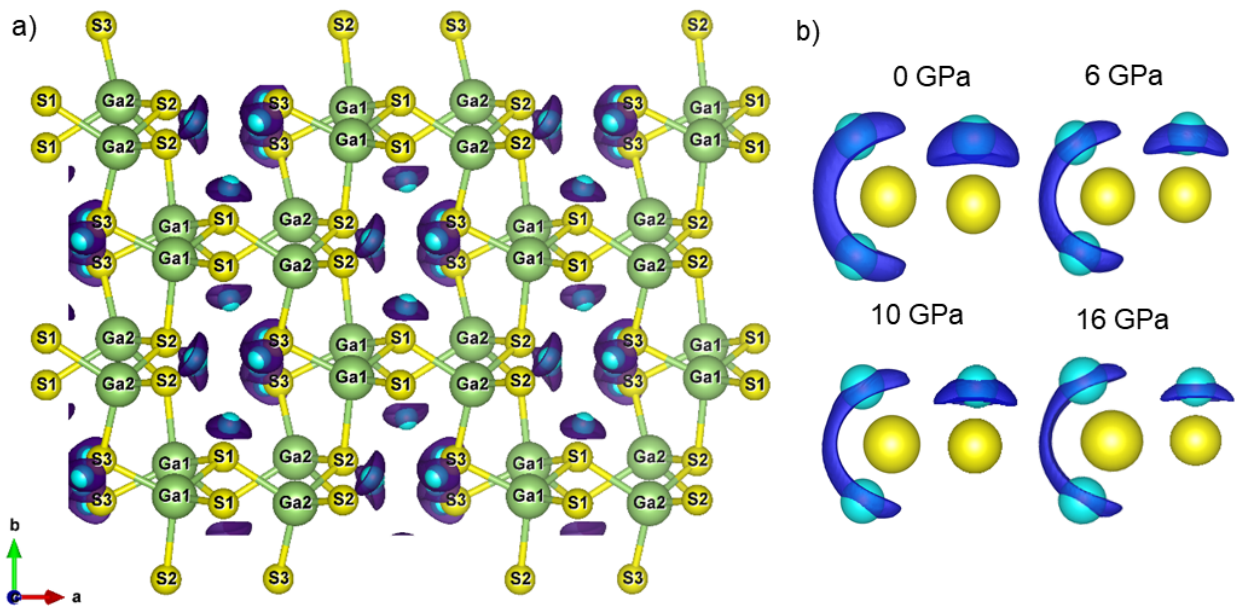


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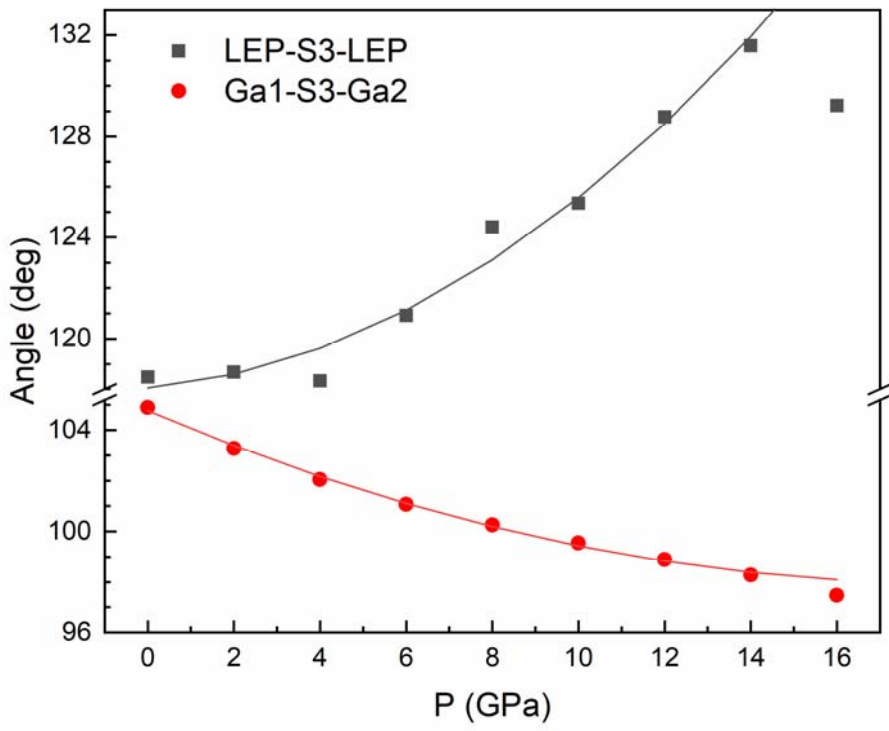
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1335 **Figure 12**



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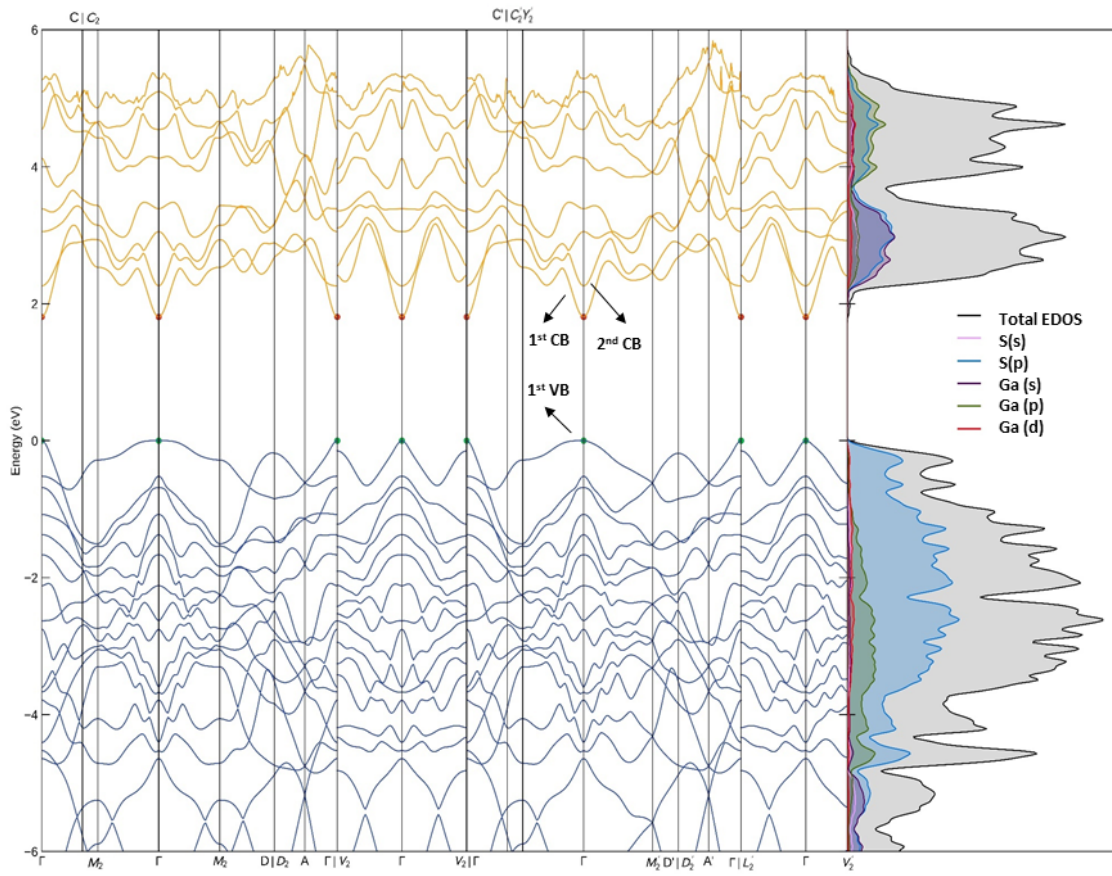
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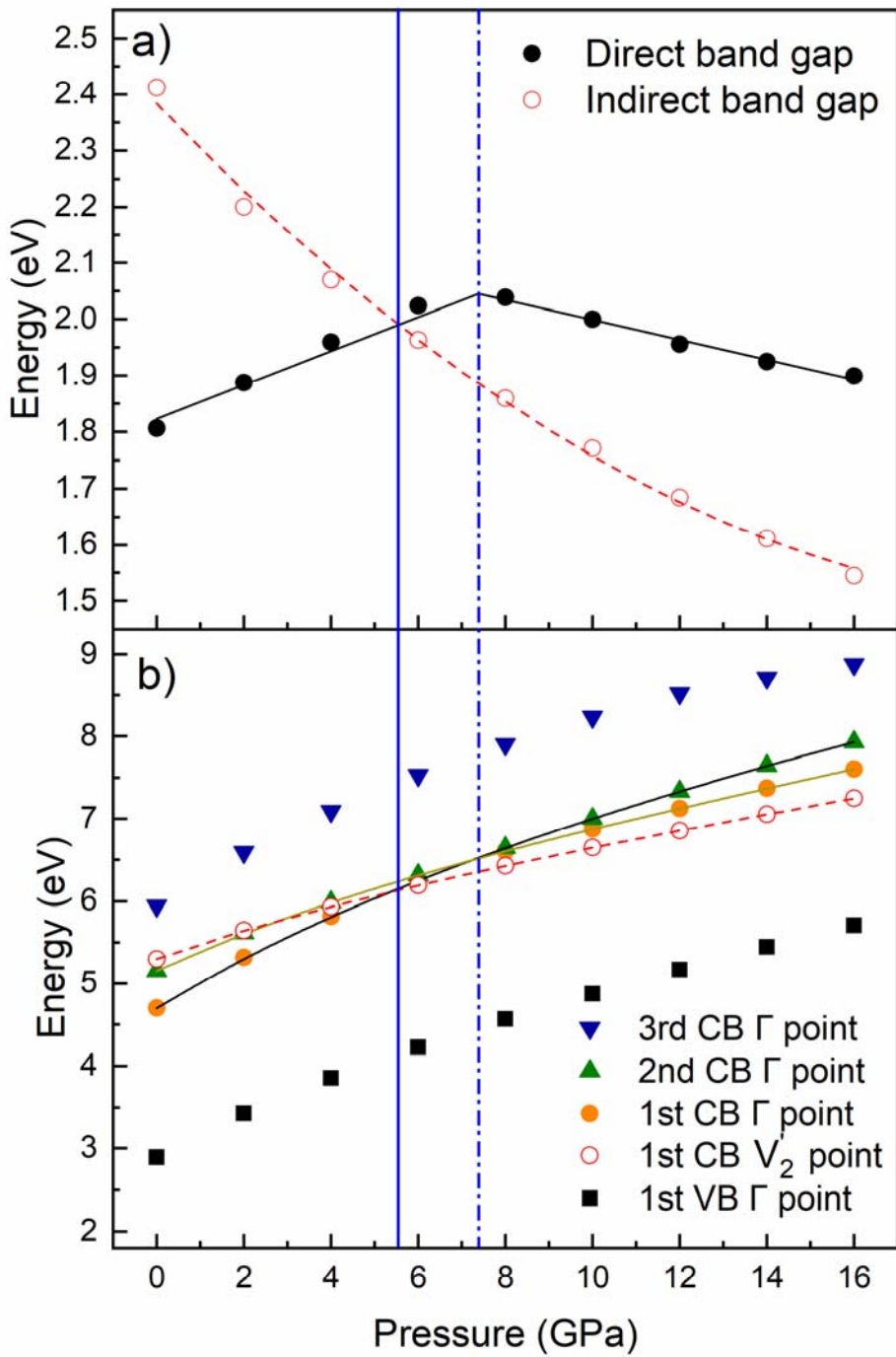
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1341 **Figure 14**



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1346 **Table 1.** Zero-pressure volume (V_0), bulk modulus (B_0) and pressure derivative (B'_0) from BM3-
 1347 EoS. Our experimental (Exp) and theoretical (GGA-PBEsol) values are compared to experimental
 1348 (Run-2) and theoretical (LDA) data from **Ref. 38**. The same parameters are shown for the
 1349 isostructural β -Ga₂Se₃.

| | α' -Ga ₂ S ₃ | | | | β -Ga ₂ Se ₃ | |
|-------------------------|---|---------------------|------------|-------------------|--|-----------------------|
| | Exp | Run-2 ³⁸ | GGA-PBEsol | LDA ³⁸ | Exp | GGA-PBE ⁷⁴ |
| V_0 (Å ³) | 427.8 (3) | 429 (2) | 423.1 (3) | 411.1 (3) | 488.4 ¹²⁰ | 467.5 |
| B_0 (GPa) | 47 (5) | 49 (5) | 38.0 (6) | 49.6 (8) | 51 ⁷³ | 39.9 |
| B'_0 | 5 (1) | 6 (1) | 5.3 (1) | 4.1 (2) | | 4.7 |

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1374 **Table 2.** Axial compressibilities, principal axes (ev_i) and their compressibilities (λ_i), and angles
 1375 of maximum (ψ_{max}) and minimum compression (ψ_{min}) in α' -Ga₂S₃ from 0 to 14 GPa. Our
 1376 experimental (Exp) and theoretical (GGA-PBEsol) data are compared to experimental (Run-2)
 1377 and theoretical (LDA) data from **Ref. 38**. ψ_{max} and ψ_{min} are measured to the c axis (from c to
 1378 a).

| | Exp | Run-2 ³⁸ | GGA-PBEsol | LDA ³⁸ |
|--------------------------------------|---------------------|---------------------|---------------------|---------------------|
| κ_a (10 ⁻³ GPa) | 9.05 | 6.36 | 10.98 | 7.49 |
| κ_b (10 ⁻³ GPa) | 12.76 | 8.86 | 9.62 | 6.85 |
| κ_c (10 ⁻³ GPa) | 4.317 | 4.93 | 5.39 | 4.82 |
| λ_1 (TPa ⁻¹) | 5.53 | 4.81 | 5.04 | 5.02 |
| ev_1 | (-0.963, 0, -0.270) | (-0.833, 0, 0.553) | (-0.940, 0, -0.341) | (-0.958, 0, -0.286) |
| λ_2 (TPa ⁻¹) | 5.05 | 4.57 | 4.52 | 4.35 |
| ev_2 | (0, 1, 0) | (0, 1, 0) | (0, 1, 0) | (0, 1, 0) |
| λ_3 (TPa ⁻¹) | 3.29 | 2.52 | 3.88 | 3.68 |
| ev_3 | (-0.231, 0, -0.973) | (-0.436, 0, -0.900) | (-0.201, 0, -0.980) | (-0.224, 0, -0.975) |
| ψ_{max} (°) | 68.6 | 43.0 | 71.7 | 69.4 |
| ψ_{min} (°) | 158.2 | 132.5 | 164.6 | 159.2 |

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