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Additional Information

19 **Abstract**

20 Green methods of modification, such as ozone, can bring new functionalities to starch. In
21 this study, starch-based plastics were produced by extrusion, using cassava starch
22 ozonated with a gas flow of $1 \text{ L}\cdot\text{min}^{-1}$ at a concentration of $41 \text{ mg O}_3\cdot\text{L}^{-1}$ resulting in
23 doses of 78.9 and 145.3 $\text{mg O}_3/\text{g}$ starch. Pre-mixes (starch, glycerol, and water) were
24 processed in a co-rotating twin-screw extruder. The main results showed that ozonated
25 sheet produced with the treatment of 78.9 $\text{mg O}_3/\text{g}$ starch resulted in an increase in Young
26 Modulus by 43 % and a decrease in hydrophilicity by 37 % in comparison to the native
27 one. Conversely, treatment with 145.3 $\text{mg O}_3/\text{g}$ starch did not increase the bio-based
28 plastic mechanical properties or surface characteristics, showing a specific behavior
29 tendency between ozonation and extrusion processes. The ozonation of starch showed to
30 be an alternative for producing by extrusion bio-based plastics with enhanced properties,
31 by selecting adequate processing conditions.

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33 **Keywords:** biodegradable plastics; ozonation, green technologies, starch modification

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Nomenclature

<i>A</i>	area (m ²)
<i>E</i>	Young's Modulus (MPa)
<i>p</i>	pressure (kPa)
<i>RC</i>	Relative Crystallinity
<i>t</i>	time (h)
<i>TS</i>	Tensile Strength (MPa)
<i>WVP</i>	Water Vapor Permeability (g·mm/m ² ·day·kPa)
<i>x</i>	thickness (mm)
<i>Y</i>	opacity (%)

43

44 **Subscripts**

<i>b</i>	black standard
<i>w</i>	white standard

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46 **Greek**

ε	elongation at break (%)
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47

48 **1. Introduction**

49 Starch-based plastics have been developed as an alternative to petroleum-based
50 ones, being cassava starch an interesting raw material for this application [1]. However,
51 starch-based plastics can show lower performance when compared with traditional
52 plastic, such as worse mechanical properties, low transparency, and high-water
53 absorption [2,3] mainly due to the hydrophilic character of this polymer[4].

54 Different modification techniques can be used to overcome those issues, being the
55 so-called “green technologies” interesting approaches to enhance performance for
56 attaining sustainability. The ozonation process (i.e., processing using the ozone gas) is
57 considered an environmentally friendly technology that promotes three main
58 modifications in starch molecules, achieving different properties: partial
59 depolymerization, oxidation of hydroxyl groups into carbonyl and carboxyl groups [5].
60 In fact, ozonation was able to enhance the performance of the bio-based plastics of
61 cassava [6,7] and potato[8] bio-based plastics produced by the casting technique.

62 However, although casting is an interesting technology to understand the
63 polymeric interactions with some other compounds or process conditions, as a first
64 approach, it is a laboratory-scale technique, **difficult to apply** on an industrial scale.
65 Extrusion technology is widely used in the polymer industry [2], but studies on starch-
66 based plastics are still scarce in the literature. The extrusion technique has many
67 advantages, being a continuous and low time-consuming process, and efficient to produce
68 viscous materials [9,10].

69 When both processes are compared (casting vs. extrusion), drying conditions are
70 quite different. Casting plastics are slowly dried (~12 h) at low temperatures and rest (i.e.,
71 at negligible stress), while extruded plastics are quickly dried at high temperatures and
72 high shear stress, resulting in a different matrix conformation and recrystallization ability

73 [11,12]. These differences affect the mechanical, barrier, and physical properties of the
74 material, highlighting the importance of this research.

75 Therefore, in this study, ozonation, a “green emerging technology”, was used to
76 modify cassava starch before bio-plastic production. Ozonated starch-based plastics were
77 produced by extrusion, reported for the first time in the literature. Those new plastics
78 were characterized by correlating their mechanical, barrier, physical and thermal
79 characteristics with the starch modification process.

80 **2. Material and Methods**

81 *2.1. Materials and ozonation process*

82 Native cassava starch (Amilogill 1500; 28.94 g/100 g of amylose) provided by
83 Cargill Agrícola – Brazil, glycerol P.A. grade (Sigma-Aldrich, Brazil), and Milli-Q water
84 were used to produce the ozonated starch-based plastics.

85 The ozonation process and the full characterization of cassava starch were described
86 in a previous study [13]. Briefly, a rich stream was produced with a gas flow of $1 \text{ L} \cdot \text{min}^{-1}$
87 with an ozone concentration of $41 \text{ mg O}_3 \cdot \text{L}^{-1}$ going through the samples for 15 min or 30
88 min at $25 \pm 1 \text{ }^\circ\text{C}$, which resulted in the ozone doses (consumption during the reaction) of
89 78.9 and 145.3 $\text{mg O}_3/\text{g starch}$, respectively (and hereafter called O78.9 and O145.3,
90 respectively). This was followed by decantation, centrifugation, and air-drying in a
91 circulation oven (Marconi, MA 035, Brazil) for $\sim 12 \text{ h}$ at $35 \text{ }^\circ\text{C}$, until reaching a moisture
92 content of $\sim 12 \text{ g}/100 \text{ g w.b.}$ The O78.9 and O 145.3 samples resulted in carbonyl,
93 carboxyl, and amylose content of (0.080; 0.020) groups/100 glucose units, 25.85 g/100 g
94 and (0.12; 0.040) groups/100 glucose units, 23.79 g/100 g, respectively [13].

95 *2.2. Bio-based plastic production*

96 Premixes of starch, glycerol (43.1 g/100 g of starch), and water (25.9 g/100 g of
97 starch)[14] were homogenized in a mixer (Kitchen Aid, 525, Brazil) and then extruded in

98 a corotating twin-screw extruder (Thermo Fisher Scientific, Process 11, Germany) L/D
99 40, and 11-mm screw diameter to produce pellets. The premixes were fed through a
100 double screw feeder (Brabender, Volumetric Minitwin Process 11, Germany) at 0.05 kg/h
101 and a thread speed of 80 rpm. The temperature profile to obtain pellets from the feed zone
102 (1) to the extrudate outlet (zone 8) was based on Vedove et al. [14], being: 60, 75, 90,
103 100, 110, 110, 105, and 100 °C. The cylindrical material obtained was then cut to 2 mm
104 to obtain the pellets.

105 Those pellets were fed from a conical type feeder (Brabender, Kulturstrasse 55-73,
106 Germany), with a feed flow of 0.2 kg/h to the same co-rotating twin-screw extruder. The
107 sheets were obtained by the extrusion process using the same temperature profile, the die
108 (3 cm x 1 mm) with the mold to form the sheets at 105 °C and the thread speed at 80 rpm.

109 For each level of ozone applied to cassava starch (78.9 and 145.3 mg O₃/g starch),
110 the pellets and their respective sheets were produced by extrusion in duplicate. The pellets
111 and sheets were also produced from native cassava starch (no ozonated) for comparison.

112 *2.3. Bio-based plastic characterization*

113 The materials were conditioned for 7 days in desiccators (75 % RH) at room
114 temperature (~ 25 °C) and then evaluated **according to** their morphological, structural,
115 mechanical, barrier, physical, and thermal characteristics.

116 *2.3.1. Morphology*

117 The morphology of the bio-based plastics was visualized by a scanning electron
118 microscope (SEM). The specimens were coated with 10-nm platinum and observed using
119 the microscope (FEI, model Quanta 600FEG, Netherlands) at 15 kV and the
120 magnification ranged from 1500× (surface) to 150× (cross-section) images.

121

122 *2.3.2. Relative Crystallinity*

123 The crystallinity of the materials was analyzed by an X-ray diffractometer
124 (PANalytical, model X'Pert PRO, Netherlands) at 45 mA and, 40 kV. The scanning rate
125 was 0.02°/s with 2 θ angles from 2° to 40°. The Origin 2020 software was used for data
126 analysis (OriginLab Corporation, Massachusetts, USA).

127 2.3.3. Fourier-transform infrared (FTIR) analysis

128 Sheet spectra were investigated by Fourier transform infrared spectroscopy (Pekin
129 Elmer Frontier IR-4700, USA) in attenuated total reflectance (ATR) mode using a
130 diamond ATR crystal in a wavenumber range from 540 to 4000 cm⁻¹; 32 scans were
131 recorded per sample. The software Origin 2020 was employed for data obtained from the
132 spectra (OriginLab Corporation, Massachusetts, USA).

133 2.3.4. Mechanical Properties

134 Plastic thickness was determined at five random positions using a digital
135 micrometer (precision 0.001 mm) (MITUTOYO, Japan). Their mechanical properties
136 were evaluated using a uniaxial tensile assay according to the ASTM D882-12 standard
137 method [15] on a texture analyzer (Stable Micro Systems, TAXT2i, UK). Tensile strength
138 (*TS*) and elongation at break (ϵ) were evaluated for ten strips (100 mm x 25 mm) using
139 the A/TGT self-tightening roller grips at a speed of 1 mm·s⁻¹ and grip separation of 50
140 mm. Young's modulus (*E*) was calculated as the slope of the initial linear portion of the
141 stress *versus* strain curve.

142 2.3.5. Barrier Properties

143 Water vapor permeability (*WVP*) was determined through the gravimetric method
144 according to the ASTM E96/E96M-16 Standard Method [16], in triplicate. The specimens
145 were placed in circular permeation cells containing silica gel and then in a desiccator (75
146 % RH) in a climate chamber (TIRA CLIMA, model TCC 7034, Germany) at a temperature

147 of 25 °C. The weight gain was recorded at intervals of 1 h. The WVP was calculated
148 according to Eq. (1).

$$149 \quad WVP = \frac{\Delta m \cdot x \cdot 24}{\Delta t \cdot A \cdot \Delta p} \quad (1)$$

150 wherein: $\Delta m/\Delta t$ is the moisture gain per unit of time ($\text{g}\cdot\text{h}^{-1}$); A is the film area exposed to
151 permeation ($4.9\cdot 10^{-4} \text{ m}^2$); x is the film thickness (mm); Δp is the difference in vapor
152 pressure (kPa) of the atmosphere over silica gel and pure water.

153 2.3.6. Wettability

154 The wettability of the bio-based plastic surface was determined by the contact angle
155 with Milli-Q water, according to ASTM D7334 08 standard methods [17] through an
156 optical system (Dataphysics, OCA 15, Germany). The water contact angle was
157 determined in six small specimens per sample. Angle measurements were quickly carried
158 out to avoid water vaporization.

159 2.3.7. Solubility and moisture content

160 The bio-based plastic solubility in water was evaluated in triplicate with discs
161 (diameter of 20 mm) immersed in water at 25 °C, being agitated at 120 rpm for 24 h,
162 according to Gontard et al. [18]. The solubility was calculated according to Eq. (2).

$$163 \quad S = \frac{m_f - m_o}{m_o} \cdot 100 \quad (2)$$

164 wherein: m_o and m_f are the initial and the final mass of the discs (g), respectively.

165 Moisture of bio-based plastics was determined in triplicate, according to Gontard
166 et al. [18] in an oven (Marconi, M030/2057, Brazil) at 105 °C, until constant mass.

167 2.3.8. Opacity and Light Transmittance

168 The transmission of UV to visible light of the materials was measured with a
169 spectrophotometer from 200 nm to 800 nm (FEMTO, model 700.Plus, Brazil). The bio-

170 based plastics opacity was determined, in triplicate, using a colorimeter (HUNTERLAB,
171 model ColorQuest XE, USA) using Eq. (3).

$$172 \quad Y = \frac{Y_b}{Y_w} \quad (3)$$

173 wherein: Y_b is the opacity on the black standard and, Y_w is the opacity on the white
174 standard.

175 *2.3.9. Thermogravimetric analysis (TGA)*

176 The thermogravimetric analysis was performed in duplicate using a thermal
177 analyzer (TA Instrument, model TGA-Q500 V20.13, EUA), using ~ 10 mg of samples,
178 which were heated from 25 °C to 700 °C, at a rate of 10 °C/min under nitrogen flow. [The](#)
179 [sample mass was continuously measured to describe its reduction over temperature.](#)

180 *2.4. Statistical analyses*

181 All data were evaluated by the Analysis of Variance (ANOVA), applying Tukey's
182 test ($p < 0.05$) using the Statgraphics Centurion XV software (StatPoint, Inc., USA).

183 **3. Results and Discussion**

184 Figure 1 presents the visual aspect of the bio-based plastics produced by extrusion,
185 which were classified as “sheets” due to their thickness (> 0.250 mm) according to the
186 standard terminology of ASTM D882-18 [15].

187 [Different performances were achieved in each treatment, while the ozonation](#)
188 [conditions resulted in a hormetic pattern \(increasing performance at low doses and](#)
189 [decreasing it at high doses\) for some properties. This indicates there is an interesting range](#)
190 [of starch molecular modification levels from which the ozonation process becomes](#)
191 [deleterious to the bio-based plastic – therefore highlighting the importance of the present](#)
192 [evaluation.](#)

193 For instance, while the sheet produced with cassava starch ozonated at 78.9 mg
194 O₃/ g starch seems more homogenous and thinner, with adequate transparency to be used
195 as packaging material, the plastics produced with the native and 145.3 mg O₃/g ozonated
196 starches were rougher and heterogeneous - the latter (145.3 mg O₃/g starch) presented
197 higher opacity. More details are discussed in the following.

198 *3.1.Morphology*

199 Figure 2 presents the morphology of the sheets obtained. In general, the 15-min
200 ozonated sheets present a compact and smooth structure on the surfaces and cross section,
201 while for the O145.3, some irregularities are visualized. Higher heterogeneity and
202 roughness are observed in the sheet produced from native cassava starch, which could be
203 attributed to the formation of bubbles/drops in the extruder exit section [19]. A smoother
204 surface for ozonated cassava films in comparison to native films produced by casting was
205 reported by La Fuente et al. [6]. According to Biduski et al.[20], the oxidation of hydroxyl
206 groups allows a higher penetration of the plasticizer in the starch chains, improving the
207 interactions between molecules. However, in this study, it is important to highlight the
208 difference in the drying process, which can affect the final conformation of the matrix.
209 Actually, extrusion drying favors the organization of the molecules in the flow direction
210 [21], while in casting, molecules can be entangled due to slow drying under steady
211 conditions, resulting in morphology with different characteristics.

212 The presence of micro-voids and cracks in extruded bio-based plastics is
213 associated with the accelerated drying process [22,23], as well as the formation of
214 bubbles/droplets at the exit of the die, due to the pressure and temperature drop [19].

215 *3.2.Crystallinity and Fourier-Transform Infrared (FTIR) spectra*

216 Figure 3A shows the X-ray diffractograms for the native and ozonated cassava
217 sheets, in which peaks were observed at 5.1 °, 17.0 °, 20.0 °, 22.0 °, 31.0 °, and 33.9 °
218 (2θ), indicating a type-B crystalline structure. The peaks at 5.1 °, 17.0 °, 20.0 °, and 22.0
219 ° were also observed by La Fuente [6] in ozonated cassava starch films produced by
220 casting, while Valencia [24] also observed peaks near 31.0 ° in native cassava films
221 produced by casting. Vedove [14] observed peaks near 20.0 ° and 22.3 ° in native cassava
222 starch sheets produced by extrusion, under conditions similar to those applied in this
223 study.

224 Ozonated sheets presented higher values of relative crystallinity (Table 1) in
225 comparison with the native starch, the same behavior was observed in ozonated potato
226 [8] and cassava [6] starch films produced by casting. This suggests that depolymerization
227 occurred during ozonation and the partial replacement of the hydroxyl groups by carbonyl
228 and carboxyl groups [5,13], resulting in molecules with better organization, increasing
229 the crystalline regions.

230 Figure 3B presents the FTIR spectra of the sheets obtained. In general, the spectra
231 showed a typical behavior for cassava starch films plasticized with glycerol [25–27]. The
232 absorbance at 3300 cm⁻¹ is attributed to O–H stretching, while C–H stretching appeared
233 at 2922 cm⁻¹. The band near 1650 cm⁻¹ is assigned to the water adsorbed by starch
234 molecules [5,26], near 1455 cm⁻¹ to C–H bending, near 1242 cm⁻¹ to C–O stretching, or
235 the C–OH stretching [25]. Then, at 1000 cm⁻¹ to the C–O stretching, while near 925 cm⁻¹
236 designated to the C–O and the C–H stretching [25,26].

237 Satmalawati et al. [28] observed a difference in the peak sharpness at 1780–1680
238 cm⁻¹ between native and ozonated cassava starch, which they attributed to the carbonyl
239 groups due to the ozone oxidation process. However, no difference in this region among
240 samples was visualized in Figure 3. Vicentini et al. [27] discussed that different molecular

241 changes could occur in other regions, which probably explains why carboxyl and
242 carbonyl groups were not visualized.

243 According to the literature [5,26,27], starch shows a fingerprint region within
244 1200–900 cm⁻¹ bands, which is attributed to changes in starch structure. However, once
245 again, no differences were visualized in this region among samples. At this point, it is
246 important to note that, in ATR spectra (mode used in this study), the penetration depth of
247 infrared light in the sample is about 15 μm and, probably, the higher thickness of the
248 sheets (Table 1) could interfere with visualization of this functional groups in all ranges
249 studied herein.

250 *3.3.Mechanical properties*

251 Table 1 presents the mechanical properties of the materials. Similar values have
252 been reported for native cassava sheets [14]. As shown, there was no significant variation
253 ($p>0.05$) in tensile strength between the native and the O78.9 cassava sheet, while for the
254 O145.3 sheet, this property was ~ 20 % lower. Similarly, the elongation at break was ~25
255 % lower for ozonated starch sheets compared to the plastic produced with the native
256 starch. Table 1 allows verifying that the Young Modulus follows the same behavior as
257 the tensile strength, as expected. It increased by 43 % for O78.9 in comparison to the
258 O145.3 and native cassava sheets, indicating that O78.9 increased the mechanical
259 properties, while for O145.3 this property is similar to that of native starch.

260 **Table 1.**

261 Relative crystallinity, thickness, tensile strength, elongation at break, Young modulus's, moisture content, and opacity of native and ozonated
262 cassava starch sheets.

Ozone dose (mg O₃/g starch)	RC (%)	Thickness (mm)	Tensile Strength (MPa)	Elongation at break (%)	Young Modulus (MPa)	Moisture (g/100 g)	Opacity (-)
0	3.81 ± 0.59 ^a	1.19 ± 0.00 ^b	1.13 ± 0.11 ^b	38.82 ± 1.32 ^b	5.23 ± 0.70 ^a	20.47 ± 0.35 ^a	49.33 ± 1.15 ^a
78.9	5.68 ± 0.17 ^a	0.96 ± 0.10 ^a	1.29 ± 0.09 ^b	29.84 ± 4.39 ^a	7.50 ± 0.46 ^b	19.65 ± 0.29 ^a	79.33 ± 1.53 ^b
145.3	5.69 ± 0.81 ^a	1.17 ± 0.01 ^b	0.88 ± 0.08 ^a	29.41 ± 2.36 ^a	4.90 ± 0.61 ^a	19.23 ± 1.22 ^a	93.33 ± 0.58 ^c
Tukey HSD*	2.46	0.15	0.26	7.46	1.50	0.02	2.90

263 The means with different letters in the column are significantly different ($p < 0.05$).

264 * Tukey's honestly significant difference (HSD).

265

266 The increase in stiffness and the decrease in the elongation of cassava sheets,
267 depending on the process conditions, can be interesting since we could produce materials
268 with unique characteristics according to the market demand. This behavior can be
269 attributed both to the new matrix conformation obtained after determining the ozonation
270 process and to the drying conditions in the extruder. Probably, the higher degree of
271 depolymerization of this material (O145.3 cassava starch), in combination with higher
272 temperatures and shear stress during extrusion, was drastic enough to weaken the
273 polymeric matrix, thus resulting in the lower values of Young Modulus, while for the
274 O78.9 material, the degree of depolymerization reached, in combination with the
275 extrusion process (temperature and shear stress) was good enough to increase the
276 toughness of the polymeric matrix.

277 3.4. Barrier and physical properties

278 As shown in Figure 4, there is no statistical difference in water vapor permeability
279 between native and the ozonated cassava sheets. In addition, an increase of 13 % in the
280 solubility of the O78.9 sheet can be observed compared to the other materials. The values
281 are similar to those reported by Pérez-Vergara et al. [29] for native cassava starch films.

282 Although a decrease in water vapor permeability was expected due to the decrease
283 in hydroxyl groups, leading to a decrease in hydrophilic properties [30], this did not occur.
284 A possible explanation may be the carboxyl groups formed as a result of ozonation, which
285 are electronegatively charged, change the polarity of the polymeric matrix, increasing the
286 repulsive forces, favoring the mobility of water, as Biduski et al. [20] exposed. In fact, an
287 increase in the water vapor permeability was observed in cassava [6] and potato [8]
288 ozonated starch films produced by casting. These electrostatic repulsion forces could also
289 be responsible for the increase in the solubility of the O78.9 sheets. However, for O145.3
290 sheets, the greater ozonation might cause the opposite behavior. Additionally, it is also

291 possible that the intense shear stress and high temperature during the extrusion process
292 resulted in a molecular conformation that reduces the effect of these repulsive forces,
293 without affecting the water vapor permeability or the solubility.

294 Figure 5 presents the water contact angle formed with the first drop of water
295 released onto the surface of the material, indicating the hydrophobicity or hydrophilicity
296 of the surface of the material. According to Chimonyo et al. [31], hydrophobic
297 interactions occur when non-polar starch groups escape from the water to interact with a
298 hydrophobic surface. In this sense, the ozonated cassava sheet for O78.9 could be
299 considered a hydrophobic material with an increase of 37 % in contact angle in
300 comparison to the native cassava starch sheet. On the contrary, we observed a decrease
301 in the contact angle for the O145.3 sheet, which could be attributed to the strong
302 polarization and resonance stabilization of the carboxyl-OH bonds formed (higher for the
303 O145.3 starch), which enhances the level of polarity on the polymeric matrix [31],
304 resulting in the decrease in the water contact angle.

305 There was no statistical difference in the moisture content of the bio-based plastics
306 (Table 1), showing that ozonation did not interfere with this parameter, being the the
307 values similar to those reported for native cassava sheets [14].

308 *3.5.Opacity and transmittance*

309 As shown in Table 1, the native cassava starch sheet showed lower values of
310 opacity in comparison to the O78.9 sheet, similar to those reported by La Fuente et al.
311 [6,7] for ozonated cassava films produced by casting. However, for the O145.3 material,
312 while the films obtained by casting were less opaque [6] the sheets were more opaque.
313 This could be attributed to several factors, such as the thickness of the material (Table 1),
314 different molecules' conformation due to the extrusion process, and also the

315 bubbles/droplets formation, which could interfere with the measurement of this property.
316 Figure 6 presents the spectra of transmittance of the sheets, confirming that the ozonated
317 sheets presented less transmission from 200 to 800 nm. Specifically, in the UV region
318 (200 to 400 nm), considerably lower transmittance values can be observed, which means
319 the starch ozonation treatment on starch improves the ability of films to protect against
320 UV light. Note that this is important for different applications, such as for food packaging.
321 Other studies reported this achievement but after the addition of other compounds, such
322 as nano-encapsulated lycopene, β -carotene, or blueberry and pomace extracts [32,33]. It
323 is worth highlighting that the ozonated starch sheets do not need the addition of any
324 components to achieve this good behavior. This result can be a consequence of the new
325 polymeric matrix formed by ozonation, which, according to Fakhouri et al. [22], can also
326 be related to the high relative crystallinity of these sheets (Table 1).

327 *3.6. Thermogravimetric analysis (TGA)*

328 The thermogravimetric analyzes and their respective differential
329 thermogravimetric (DTG) profiles are presented in Figure 7, where different profile
330 slopes can be observed. According to Medina-Jaramillo et al. [34], the first stage (100–
331 150 °C) is related to the evaporation of water and/or volatiles. The second stage (180–
332 260) °C is correlated with the decomposition of the glycerol-rich phase, which also holds
333 starch; and the third stage (250–350 °C) is related to the degradation of the components
334 of the starch. As observed in Figure 7 (A to C), no differences are observed between
335 sheets of (350-500 °C), which implies that ozonation did not interfere with starch
336 degradation. Finally, the region from (350-500) °C corresponds to carbon burning [35].

337 From the weight loss profile (Figure 7D), faster degradation of native and O78.9
338 cassava starch sheets can be observed within the step from (100–150) °C and (250–350)
339 °C. According to Medina-Jaramillo et al. [34], the degradation at lower temperatures

340 suggests a reduction in the molecular interactions between starch-starch chains and
341 strengthening of hydrogen bonding interactions between hydroxyl groups and other
342 molecules. Therefore, ozonation could reduce the molecular interactions between the
343 starch–starch chains and favor the interactions between the hydroxyl groups of the starch
344 and carbonyl and carboxyl groups, thus resulting in a decrease in the degradation
345 temperature for the O78.9 sheet. This was not observed for the O145.3 sheet, probably
346 due to the higher degree of ozonation, which can promote other interactions between the
347 hydroxyl, carbonyl, and carboxyl groups, as well as the different molecular size
348 distribution (molecular depolymerization).

349 **4. Conclusions**

350 The starch ozonation process resulted in the formation of the carbonyl and carboxyl
351 groups, as well as partial molecular depolymerization of the amylose and amylopectin
352 molecules. Consequently, a polymeric matrix with different molecular size distribution,
353 charge distribution, and thus, packing properties, was achieved.

354 In this work, first reported in the literature, it was possible to produce cassava sheets
355 with unique characteristics, from ozonated starch processed by the extrusion process. The
356 level of ozone processing was shown to be a key factor in selecting the desired properties.

357 The sheet produced with ozonated starch for O78.9 resulted in a homogeneous
358 structure, with improved mechanical behavior (increased Young Modulus of 43 % and
359 decreased elongation at break of 23 %), increase hydrophobicity (37 %), without
360 interfering with the water vapor permeability. Furthermore, a lower transmittance was
361 achieved in the UV-region (200 to 400 nm) was achieved.

362 Conversely, for the O145.3 starch sheet, the higher degree of ozonation in
363 combination with the conditions of the extrusion process did not improve the mechanical
364 properties, the water vapor permeability, or the wettability characteristics. This behavior

365 could be attributed to the combination of the molecular modification during the pre-
366 treatment with ozone, and the network formation under the shear-temperature conditions
367 during extrusion. Therefore, an adequate combination of ozonation and extrusion
368 processes yields materials with specific characteristics, with a wide range of applications
369 for starch-based bioplastic production.

370 We suggest future studies including other starch sources, processing conditions,
371 formulations, and a combination of processes to obtain bio-based plastics with better
372 performance.

373

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381

382 **Author's Contribution / CRediT author statement**

383 **Carla I. A. La Fuente:** Conceptualization, Methodology, Validation, Formal analysis,
384 Investigation, Resources, Data Curation, Writing - Original Draft, Writing - Review &
385 Editing, Project administration.

386 **Larissa do Val Siqueira:** Methodology, Formal analysis, Investigation, Data Curation,
387 Writing - Original Draft, Writing - Review & Editing.

388 **Pedro E.D. Augusto:** Conceptualization, Methodology, Formal analysis, Writing -
389 Review & Editing, Supervision, Project administration, Funding acquisition.

390 **Carmen C. Tadini:** Conceptualization, Methodology, Formal analysis, Resources,
391 Writing - Review & Editing, Supervision, Project administration, Funding acquisition.

392

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516

517 **Figure Caption**

518 **Figure 1.** Images of bio-based plastics produced with native and ozonated cassava
519 starch by extrusion.

520 **Figure 2.** Morphology of native and ozonated cassava starch plastics observed by
521 scanning electron microscopy (SEM).

522 **Figure 3.** (A) X-ray diffraction patterns and (B) FTIR spectra of native and ozonated
523 cassava starch sheets.

524 **Figure 4.** Water vapor permeability and solubility of native and ozonated cassava starch
525 sheets produced by extrusion.

526 **Figure 5.** The water contact angle of native and ozonated cassava starch sheets produced
527 by extrusion.

528 **Figure 6.** Transmittance of native and ozonated cassava starch sheets produced by
529 extrusion.

530 **Figure 7.** Thermo gravimetical analysis of native (A) and ozonated (B) and (C) cassava
531 starch sheets produced by extrusion. (D) Comparatives profiles.

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