

Document downloaded from:

<http://hdl.handle.net/10251/46856>

This paper must be cited as:

López Paz, J.L.; González Martínez, M.Á.; Escorihuela Fuentes, J.; Bañuls Polo, M.J.; Puchades Pla, R.; Maquieira Catala, Á. (2014). Direct and label-free monitoring oligonucleotide immobilization, non-specific binding and DNA biorecognition. *Sensors and Actuators B: Chemical*. 192:221-228. doi:10.1016/j.snb.2013.10.110.



The final publication is available at

<http://dx.doi.org/10.1016/j.snb.2013.10.110>

Copyright Elsevier

1 Direct and label-free monitoring oligonucleotide  
2 immobilization, non-specific binding and DNA biorecognition

3  
4 José Luis López-Paz, Miguel Ángel González-Martínez\*, Jorge  
5 Escorihuela, María-José Bañuls, Rosa Puchades, Ángel Maquieira

6 *Centro de Reconocimiento Molecular y Desarrollo Tecnológico, Departamento de Química, Universidad*  
7 *Politécnica de Valencia, Camino de Vera s/n, 46022 Valencia, Spain*

8  
9  
10 ABSTRACT

11  
12 DNA binding chemistry on silicon surface has been investigated. Aminated  
13 oligonucleotide probes were immobilized on the chip surface by chemical silanization and  
14 further covalent attachment. The chemistries employed were the classical 3-  
15 aminopropyltriethoxysilane/glutaraldehyde and, for comparison purposes, the novel 3-  
16 isocyanatepropyltriethoxysilane, that allows the direct attachment of the aminated probe.  
17 Alternatively, a thiolated oligonucleotide was also photochemically immobilized by means of  
18 a thioether linkage. The experiments were followed label-free by Dual Polarization  
19 Interferometry. All chemical and photochemical methods gave rise to a probe density  
20 immobilization in the order of  $1.0\text{-}2.5 \cdot 10^{10}$  molecules/mm<sup>2</sup>, similar to the values reported for  
21 other chemistries. The obtained data suggest that DNA strands are anchored in a different  
22 conformation depending on the immobilization method employed. In order to avoid non-  
23 specific binding of target molecules, ethanolamine and inert proteins were assayed, and  
24 successful results were obtained when using small size proteins such as gelatine. Hybridization  
25 efficiency was around 20% for aminosilane-based immobilized probes, and more than 4-fold  
26 this value when the other immobilization methods were employed. The ability for recognition  
27 complementary DNA strands discriminating non-complementary ones was applied for species  
28 identification in mixtures.

29  
30 *Keywords:* DNA Immobilization; Silicon; Surface chemistry; Dual Polarization Interferometry; Biosensing

31  
32 E-mails: [jolopa@qim.upv.es](mailto:jolopa@qim.upv.es)  
33 [mgonzall@qim.upv.es](mailto:mgonzall@qim.upv.es)  
34 [joresful@upvnet.upv.es](mailto:joresful@upvnet.upv.es)  
35 [mbpolo@upvnet.upv.es](mailto:mbpolo@upvnet.upv.es)  
36 [rpuchades@qim.upv.es](mailto:rpuchades@qim.upv.es)  
37 [amaquieira@qim.upv.es](mailto:amaquieira@qim.upv.es)

38  
39 \* Corresponding author: Miguel Ángel González-Martínez. Tel. +34 96 3877342; E-mail: [mgonzall@qim.upv.es](mailto:mgonzall@qim.upv.es)

## 1 **1. Introduction**

2  
3 The studies and uses of nucleic acids have been growing up for decades, and the trend  
4 is to keep on in first line. Thus, basic research on nucleic acids provides fundamental knowledge  
5 for *in vivo* and *in vitro* applications. DNA-based methods have also been employed in many  
6 areas, examples being forensic science [1], environmental control [2] and biosensing [3].  
7 Additionally, new issues addressed to both array-like platforms [4] and lab-on-a-chip  
8 arrangements [5] are currently under development.

9 Nearly all nucleic acid-based methods employ the recognition element immobilized on  
10 a surface, so DNA immobilization has been extensively studied in the last years, and many  
11 attachment chemistries have been described [6]. Among all the supports used in biosensing,  
12 silicon is one of the most popular because this is a highly versatile solid, readily functionalized,  
13 biocompatible, and its derivatization for attaching biomolecules has been deeply studied [7].

14 Many strategies have been developed for detecting DNA hybridization processes. The  
15 employment of labels such as fluorophores is well-known, facilitated by the PCR amplification  
16 of the target and recommendable to routine high-load analysis. However label-free detection  
17 retrieves better information and the processes to be monitored are more similar to what is  
18 happening in life or biological systems.

19 There are several label-free technologies for biosensing and related applications, e.g.  
20 Quartz Crystal Microbalance (QCM) [8], Surface Plasmon Resonance (SPR) [9] and  
21 interferometry [10]. Among the latter ones, Dual Polarization Interferometry (DPI) is a very  
22 interesting option.

23 DPI is considered a reliable technique for monitoring binding events happening on a  
24 silicon oxynitride surface in real time. Basically [11], light from a laser is passed through a  
25 sandwiched waveguide structure, and an interference pattern is detected on the opposing side  
26 by a CCD camera. The phase shift of transverse magnetic (TM) and transverse electric (TE)  
27 modes is recorded in real time, and data are resolved; so that only one value of thickness and  
28 absolute refractive index at a given timepoint  $t$  will satisfy Maxwell's equations of  
29 electromagnetism for both TM and TE polarizations. Differences in the waveguide mode  
30 dispersion between the TE and TM modes allow unique solutions for adlayer thickness and  
31 refractive index, and these data can be converted to surface mass and density. Thus, this  
32 technique provides label-free precise measurements, in real time, enabling details of the  
33 structure and function of biomolecules to be elucidated. Hence, high-quality information on the  
34 orientation, distortion, and efficiency of immobilization procedures as well as the interaction  
35 events can be obtained. It is remarkable that a DPI instrument is a quantitative analytical tool,  
36 not a sensor. DPI routinely and reproducibly provides simultaneous quantitative data on real-  
37 time changes in dimension (resolution  $< 0.1 \text{ \AA}$ ) and density (resolution  $< 0.1 \text{ pg/mm}^2$ ) of  
38 biomolecules immobilized on the chip surface, without recourse to tagging, but it is not intended

1 as a sensor system for the highly sensitive determination of the agents that provoke these small  
2 changes [12].

3 DPI is frequently compared to SPR [13] because both are based on similar principles.  
4 However, SPR utilizes only the TM mode, while DPI takes advantage of measuring both the  
5 TM and the TE polarizations [14], which allows the simultaneous determination of both  
6 effective refractive index and thickness values. That is clearly a great advantage over SPR and  
7 other optical biosensing techniques, which can only report relative changes of refractive index  
8 [15]. Combining the thickness, mass and density data allows not only to monitor the binding  
9 but also to retrieve structural information at molecular level, and this is the main benefit of DPI.

10 Another difference between DPI and SPR or QCM relies on the different solid substrate  
11 employed. QCM and SPR use gold as platform, while DPI employs silicon-based chips, this  
12 material having the advantage of its low cost, versatility and applicability [7]. Nevertheless,  
13 both DPI silicon [16] and SPR gold [17] can be coated with a polymer in order to expand and  
14 improve the attaching chemistry.

15 Applications of DPI can be found in any area where microscopic surface properties need  
16 to be monitored, provided that silicon-based chips act as a supports. Interesting reports are the  
17 monitoring of lipids [18], polyelectrolytes [19] and other multilayers [20], events happening in  
18 liposomes [21], as well as the study of induced conformational changes in proteins [22],  
19 molecular aggregations [23], basic protein-protein interactions [24,26] and protein  
20 crystallization [27].

21 Nucleic acids chemistry and their interactions have also been monitored by DPI, in  
22 terms of DNA multilayer films formation [28], interactions of DNA aptamers with small  
23 molecules [29,30] and even with metal ions [31]. However the studies of DNA hybridization  
24 are scarce. Two interesting pioneer papers describe a basic study about monitoring DNA  
25 immobilization and hybridization. In the first one [32], single stranded DNA immobilization by  
26 passive adsorption, covalent linkage and avidin-biotin interaction are studied, and hybridization  
27 is monitored. In the other [33], authors covalently immobilize aminated oligonucleotides on  
28 commercial amine-derivatized chips, then block with BSA the remaining active sites, and  
29 finally add a complementary oligonucleotide for hybridization, all the three processes being  
30 monitored. Both papers are focused on monitoring DNA immobilization and hybridization,  
31 with useful discussions, but other issues such as unspecific binding or practical applications are  
32 scarcely addressed, there being little or none good contribution papers on this topic.

33 The aim of the present paper is to describe a basic study of DNA immobilization and  
34 hybridization by means of DPI including variables such as chip derivatization (silanization) as  
35 well as different modes of DNA anchoring including photochemical activation [34]. Attention  
36 is especially paid to the use of different blocking agents to avoid unspecific binding but  
37 maintaining specific hybridization. The applicability of the DPI for DNA screening and other  
38 practical issues is also discussed. The paper pretends to show that DPI can be a very useful tool

1 for understanding and helping in the designing of reliable biosensors, due to the conformational  
2 information provided.

## 3 4 5 **2. Materials and Methods**

### 6 7 *2.1. Materials*

8  
9 Synthetic oligonucleotides for immobilization, hybridization and control are shown in  
10 Table 1, and were supplied by Sigma-Aldrich (Madrid, Spain) and TibMolbiol (Berlin,  
11 Germany). Silanizing agents: 3-aminopropyltriethoxysilane (APTES), 3-(triethoxysilyl)propyl  
12 isocyanate (ICPTS) and 3-glycidoxypropyltrimethoxysilane (GOPTS), as well as blocking  
13 proteins: bovine serum albumin (BSA), gelatine, lysozyme and casein, were supplied by Sigma-  
14 Aldrich. Ethanolamine, 2-mercaptoethanol and 4,7,10-trioxatridecane-1,13-diamine supplied  
15 by Sigma-Aldrich were also used as a blocking agents. All other common chemicals were  
16 analytical grade and Milli-Q purified water was employed in all stages.

17 Phosphate buffered saline (PBS, 10 mM phosphate, 137 mM NaCl, 2.7 mM KCl, pH  
18 7.4), sodium saline citrate (SSC, 150 mM NaCl, 15 mM sodium citrate, pH 7) and carbonate  
19 buffer (CB, 100 mM sodium carbonate, pH 9.12) solutions were used as a carrier or medium in  
20 all assays. They were previously filtered through a 0.45 µm PVDF filter and degassed by  
21 sonication under vacuum prior to be employed.

22  
**Table 1**

Sequence of the oligonucleotides used

Name	Sequence (5' to 3')	5' end	3' end	MW
Probe A	(T) <sub>10</sub> TTTGATTACAGCCGGTGTACGACCCT	-NH <sub>2</sub>	none	11162
Target A	(T) <sub>10</sub> AGGGTCGTACACCGGCTGTAATCAAA	biotin	none	11618
Probe B	(T) <sub>15</sub> CCCGATTGACCAGCTAGCATT	-SH	none	11115
Target B	AATGCTAGCTAATCAATCGGG	biotin	none	6859
Probe C	(T) <sub>12</sub> AATGCTAGCTGGTCAATCGGG	-NH <sub>2</sub>	none	10315
Target C	(T) <sub>15</sub> CCCGATTGACCAGCTAGCATT	biotin	none	11334
Probe D	(T) <sub>10</sub> CAAAGCATAGTATCTCGCAATG	-NH <sub>2</sub>	none	9948
Target D	(T) <sub>10</sub> CATTGCGAGATACTATGCTTTG	biotin	none	10351
Probe E	CGCCGATAACTCTGTCTCTGTA	-NH <sub>2</sub>	none	6840
Target F	CGCCGATAACTCTGTCTCTGTA	biotin	none	7066

23 Oligonucleotide target A is a *Salmonella spp* DNA fragment, targets B and C correspond to different  
24 *Plum pox virus* serotypes, target D is a *Campylobacter jejuni* strand, while oligonucleotides E and F come from  
25 *Escherichia coli*.

## 2.2. Instrumentation

All experiments were performed at 20 °C with an Analight Bio200 Dual Polarization Interferometer (Farfield Scientific Ltd, Crewe, UK). The apparatus consisted of a helium neon laser ( $\lambda = 632.8$  nm), a sensor chip ( $24 \times 5.8$  mm) clamped inside a thermostated block (temperature control within 1 mK) and a 1024 x 1024 element-imaging device.

Silicon oxynitride chips (FB 100, Farfield, UK) were used for measurements. The sensor surface has two fluidic interfaces, 2  $\mu$ L dead volume each, named channel 1 and channel 3, with an additional waveguide reference area, channel 2, having a dielectric cover with constant refractive index.

Carrier and other solutions were flowed using a double-channel precision syringe pump (Harvard Apparatus PHD 2000 Infusion, Kent, UK) and injections were carried out by means of a dual injection valve. The state of polarization of the input beam was switched (50 Hz) between TE and TM using a ferroelectric crystal before passing through the sandwiched waveguide structure, the interference pattern being detected on the opposing side by the element-imaging device. Analight DAQ and Analight Explorer (Farfield) software packages were used for system managing/data acquisition and data treatment, respectively.

## 2.3. Chip silanization and oligonucleotide immobilization

For silanization using APTES, the chemical vapour deposition or dry method was employed. It consisted of exposing the chip surface to silane vapour at room temperature in a dessicator [35,36], followed by baking at 160 °C for 30 min. Silanized chip was placed in the DPI device, starting the data collecting, and the surface was activated by injecting glutaraldehyde (5% v/v in PBS, 250  $\mu$ L at 50  $\mu$ L/min). Then, on-line oligonucleotide immobilization was carried out by injecting the aminated probes (0.1  $\mu$ M in PBS, 150  $\mu$ L at 10  $\mu$ L/min).

For ICPTS chip derivatization, silanization was accomplished as described for APTES, but changing the silanizing agent. Once the chip was placed in the DPI, glutaraldehyde was not necessary and oligonucleotides were directly immobilized by injecting the aminated probes exactly in the same conditions, using CB as a buffer.

On the other hand, chip silanization with GOPTS for photochemical oligonucleotide anchoring was carried out by immersion in a solution of this compound in toluene at 2% (v/v) at room temperature, followed by rinsing with dichloromethane, air-drying and finally baking at 160 °C. Immobilization of thiolated probes was performed off-line, according to a procedure already described [34].

The chemical reactions involved in the three protocols can be found in Figure 1. X-ray photoelectron spectroscopy was applied to characterize the chip surface after silanization, in the three protocols.

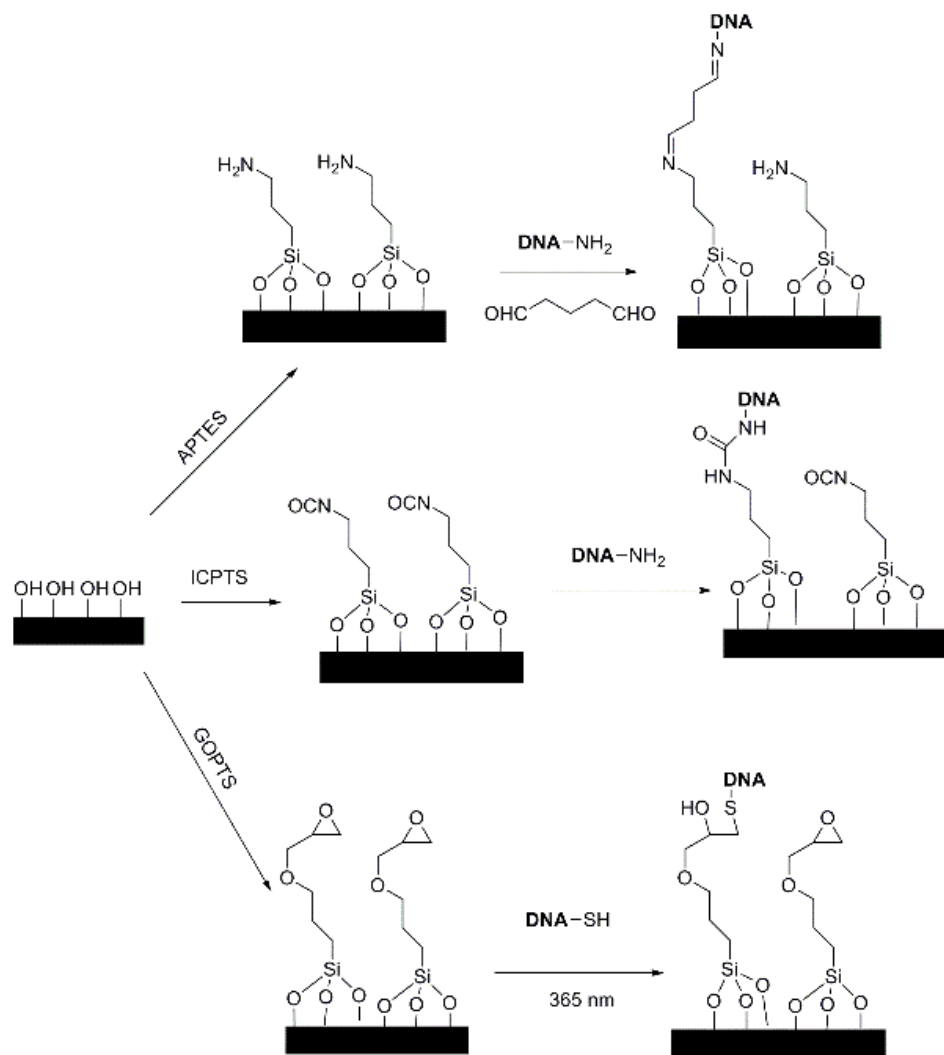


Figure 1. Scheme of the chemical processes taking place on the chip surface.

#### 2.4. DPI experiments

After oligonucleotide immobilization, the remaining binding sites were neutralized by injecting a blocking agent (100  $\mu\text{L}$  at 10  $\mu\text{L}/\text{min}$  in all cases). The materials assayed with that purpose were ethanolamine and 4,7,10-trioxatridecane-1,13-diamine (0.1 M in PBS, pH 9.0) and the proteins: BSA, casein, lysozyme and gelatine (1000  $\mu\text{g}/\text{mL}$  in carrier buffer).

The hybridization experiments were carried out by injecting a complementary biotin-ended oligonucleotide (0.5  $\mu\text{M}$  in SSC, 250  $\mu\text{L}$  at 10  $\mu\text{L}/\text{min}$ ).

Finally, a DPI calibration procedure was carried out, consisting of three stages [26,28]: *linearisation*, performed by injecting an 80% v/v ethanol/water mixture over the chip followed by stabilization back to the buffer baseline; *chip calibration*, by registering absolute signal of the ethanol/water mixture; and *bulk calibration*, performed by recording signals when flushing pure water.

## 2.5. Chip regeneration and reuse

In order to employ the same chip for more than one experiment, a cleaning-regeneration procedure was performed. It consisted of immersion in chromic mixture (100 g/L  $K_2Cr_2O_7$  in 85%  $H_2SO_4$ ) for 12 h, followed by rinsing with water, cleaning thoroughly the surface with isopropanol, and immersion in chromic mixture again for another 12 hours. Finally, the chip was rinsed with isopropanol and water, and air-dried. This protocol was useful for more than 90 % of used chips, and allowed them to be reused up to four times with no loss of performance. Figure 2 shows a brand new chip, one regenerated three times with the chromic mixture protocol, and a badly regenerated one (only rinsed with water and isopropanol).

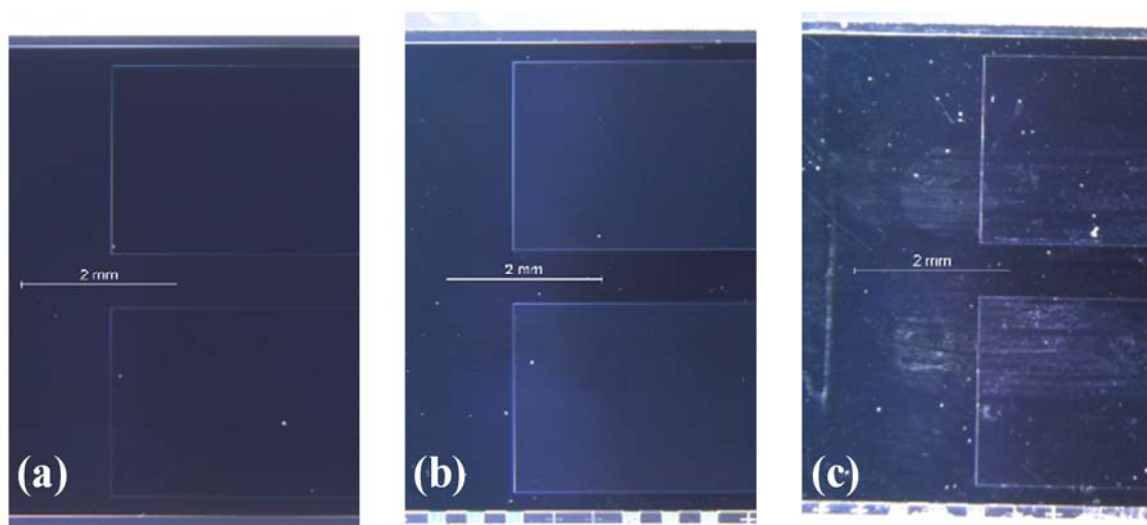


Figure 2. Magnified photograph of a brand new chip (a), a reused chip applying the optimized regeneration protocol (b) and a chip rinsed with water and isopropanol (c).

## 3. Results and discussion

DPI was used to investigate biorecognition on a silicon-based surface. The events studied were: covalent attachment of a DNA probe, application of a blocking agent for saturating unspecific binding sites and hybridization.

### 3.1. APTES and ICPTS chemical probe immobilization

Organosilanes are widely used to covalently attach promoters or cross-linkers on silicon-based materials, and the dry chemical vapour deposition is an easy, clean and effective methodology for relatively volatile silanes such as APTES and ICPTS. When applying the former, an amine-derivatized surface is formed, and the binding of the aminated oligonucleotide



1 is carried out using glutaraldehyde as a cross-linking agent prior to probe injection. For ICPTS,  
 2 surface ends in accessible isocyanate groups that can react with the aminated oligonucleotides  
 3 forming a urea linkage. In both cases, two silanization times were applied: 2 h as common  
 4 procedure and 12 h for ensuring maximal silane coating. The results of chip surface  
 5 characterization achieved by X-ray photoelectron spectroscopy (see Supplementary material,  
 6 Figure 1A) show that silicon was correctly functionalized with both silanes.

7 Probes A and E (Table 1) were injected for recognition and control, respectively. The  
 8 immobilization was monitored by recording thickness and mass (See curve plot in  
 9 Supplementary material Figure 2A). Table 2 shows the increment in these parameters for the  
 10 different silanization chemistries and times assayed.

11 **Table 2**

Oligonucleotide chemical immobilization assays

		APTES 2 h	APTES 12 h	ICPTS 2 h	ICPTS 12 h
Probe A	Thickness (nm)	0.53 ± 0.14	0.44 ± 0.09	0.64 ± 0.18	0.61 ± 0.15
(36-mer)	Mass (pg/mm <sup>2</sup> )	350 ± 100	330 ± 80	260 ± 40	340 ± 80
Probe E	Thickness (nm)	0.28 ± 0.08	0.24 ± 0.07	0.63 ± 0.06	0.54 ± 0.10
(22-mer)	Mass (pg/mm <sup>2</sup> )	160 ± 40	120 ± 40	170 ± 10	290 ± 90

12 Each value is the mean ± S.D. of three replicates

13  
 14 In all cases, the immobilization process seems to be effective, although there are some  
 15 significant observations. First of all, regarding inter-assay reproducibility, standard deviation  
 16 values are high, achieving variation coefficients greater than 25% in some instances, and  
 17 variability being higher in thickness than in mass. Thus, results of Table 2 should be considered  
 18 as semi-quantitative tendencies.

19 Regarding silanization time, APTES and ICPTS behave different. Thickness and mass  
 20 after 2 h APTES silanization are similar to those for 12 h. For ICPTS, 12 h reaction leads to a  
 21 higher mass than 2 h, although thickness results to be similar. Thus, 12 h vapour deposition  
 22 time was selected for further experiments so as to ensure the best probe coating.

23 The effect of the immobilized DNA size is to be remarked. Clearly, the larger the DNA  
 24 strand is, the higher mass increment, but quantification is not easy. Regarding mean values for  
 25 the two silanization procedures, anchoring probe A gives rise to a mass increment around 300  
 26 pg/mm<sup>2</sup>, while the shorter probe E increases the mass by approx. 150 pg/mm<sup>2</sup> (except ICPTS  
 27 12 h achieving a higher value). Mass increment values are slightly lower to those reported in  
 28 the literature, although differences are low (for 18 and 20-mer probes, around 200 pg/mm<sup>2</sup> mass  
 29 values were achieved [32,33]), and could be attributed to the chemistry employed.

30 Immobilization yield could be estimated from mass increment data. As an example, 330-  
 31 340 pg/mm<sup>2</sup> probe A is equivalent to  $1.8 \cdot 10^{10}$  molecules/mm<sup>2</sup> and assuming a maximum  
 32 packaging density of  $3 \cdot 10^{11}$  molecules/mm<sup>2</sup> [33], coverage value results to be around 6%.

1 Similarly, for probe E, 120 pg/mm<sup>2</sup> is equivalent to 1.05·10<sup>10</sup> molecules/mm<sup>2</sup>, while 290  
2 pg/mm<sup>2</sup> corresponds to 2.54·10<sup>10</sup>, so the coating varies from 3.5 to 8.5%. The values are similar  
3 to that reported for other DPI chips [32,33], but lower than those achieved for other silicon-  
4 based supports [5]. If comparing with SPR, immobilization of thiolated probes on gold surface  
5 results in a probe density from 5·10<sup>10</sup> to 4·10<sup>11</sup> molecules/mm<sup>2</sup>. These values are higher than  
6 those obtained in our experiments. Considering that a high probe density can lead to a decrease  
7 in hybridization efficiency and selectivity [37], our results would allow obtaining better  
8 hybridization yields.

9 Thickness data have also to be commented. For 12 h APTES silanization, 330 pg/mm<sup>2</sup>  
10 immobilization produces a thickness increase of 0.440 nm. As the measured thickness is a mean  
11 value, and considering the contour length (~12.2 nm for probe A) and the estimated 6%  
12 coverage, the maximal thickness increment should be 0.72 nm. Experimental values suggest  
13 that the oligonucleotides are not straight perpendicular to the surface, but partly horizontal and  
14 folded. This statement has been discussed previously with the same conclusion [5,33]. In a  
15 hypothetical situation with oligonucleotides placed only in vertical (12.2 nm height) or  
16 horizontal (2 nm height) position, it is estimated that 52% of strands are placed vertical.

17 Clearly, thickness increment is higher for 12 h ICPTS silanization (0.610 nm) than for  
18 APTES immobilization (0.440 nm for nearly the same mass), which indicates that the position  
19 of oligonucleotides on the surface is different. Consequently, it can be concluded that  
20 oligonucleotides are placed more “vertically” when immobilized via ICPTS, so the  
21 hybridization should be boosted due to the better availability. Indeed, registered density values  
22 for ICPTS immobilization (around 0.5 g/cm<sup>3</sup>) were lower than for APTES anchoring (0.7  
23 g/cm<sup>3</sup>), indicating that the DNA strand orientation is more erect for ICPTS. The obtained data  
24 suggest an 80% of standing oligonucleotides, clearly higher than the result obtained with  
25 APTES.

26 The difference in orientation can be interpreted in terms of surface treatment. APTES  
27 silanization leads to an amine-ended surface, and further glutaraldehyde treatment changes  
28 amine to aldehyde. If this transformation is not complete, residual surface amine groups are  
29 protonated at the neutral pH of carrier buffer, and the resulting positive charge attracts the  
30 oligonucleotide phosphate backbone. ICPTS results in isocyanate groups on the surface, which  
31 are not protonated, so the surface has not positive charge, allowing the standing of the probes.  
32 Hydrolysis of isocyanate groups could transform them in terminal amine ones, but this process  
33 is slow on the chip surface, and it should not to be taken into account along a single experiment.  
34

### 35 *3.2. Blocking and hybridization for chemical probe immobilization*

36

37 After immobilization of probes A and E (see above), the unspecific binding sites were  
38 saturated by injecting a blocking agent. Ethanolamine and 4,7,10-trioxatridecane-1,13-diamine  
39 (0.1 M in PBS at pH 9.0), as well as the inert proteins BSA, casein, lysozyme and gelatine

(1000  $\mu\text{g/mL}$  in PBS) were tested. Following, target A was injected in both channels in order to monitor the specific hybridization and non-specific binding, simultaneously assessing the efficacy of the blocking treatment. Hybridization step was carried out in SSC medium, pH 7.0, which is the typical buffer for DNA hybridization assays in microarray and other formats [34]. Table 3 shows the increment in thickness and mass for both channels after assaying the most representative blocking agents: ethanolamine (ETA), BSA and gelatine.

**Table 3**

Hybridization assays. Comparison between specific and non-specific binding after blocking step

Blocking agent	APTES				ICPTS			
	Thickness (nm)		Mass ( $\text{pg/mm}^2$ )		Thickness (nm)		Mass ( $\text{pg/mm}^2$ )	
	Specific	Control	Specific	Control	Specific	Control	Specific	Control
None	0.440	0.450	210	190	0.730	1.130	280	280
ETA	0.290	0.460	150	140	0.390	0.370	150	120
BSA	0.120	0.130	77	64	0.240	-0.06	39	25
Gelatine	0.130	0.010	17	-33	0.730	0.060	340	50

ETA ethanolamine

Data from Table 3 show clearly that blocking step is necessary for both APTES and ICPTS silanization, because target A binding is nearly the same in control and specific channel.

The use of ethanolamine led to poor results since strong unspecific binding was also recorded. Similar results (see data in Supplementary material, Table 1A) were achieved with 4,7,10-trioxatridecane-1,13-diamine. These results suggest that unspecific binding depends on physical adsorption phenomena rather than the presence of non-occupied active covalent binding sites (carbonyl for APTES/glutaraldehyde, isocyanate for ICPTS).

The employment of inert proteins provided interesting results. BSA is a commonly used blocking protein, even in DPI experiments involving DNA binding [28]. However, our results with it were not satisfactory; mass and thickness values registered were lower than those obtained without blocking, and nearly the same for the specific and the unspecific binding. This result suggests that BSA covers the whole available surface, including the immobilized DNA binding sites.

Blocking with gelatine afforded better results, since there was a clear difference between hybridization and control for both thickness and mass. Unspecific binding was low, so gelatine was effective as blocking agent. It is worth mentioning that gelatine is a mixture of a wide range size fibrous proteins [38], so peptides having the appropriate size can fill any pore present in the surface. Casein and lysozyme, two relatively small sized proteins, were also assayed (see data in Supplementary material, Table 1A), however gelatine resulted to be the best option.

Considering the difference between specific and control values, net thickness and mass increments were 0.120 nm and 50  $\text{pg/mm}^2$ , respectively, for APTES probe immobilization and

1 optimal blocking with gelatine. The mass values corresponded to  $2.6 \cdot 10^9$  molecules/mm<sup>2</sup>,  
2 equivalent to about 15% hybridization efficiency.

3 With ICPTS silanization and gelatine blocking, the results obtained were quite different.  
4 The absolute mass increment was 290 pg/mm<sup>2</sup>, equivalent to  $1.5 \cdot 10^{10}$  molecules/mm<sup>2</sup>, with an  
5 absolute thickness increase of 0.670 nm. The estimated hybridization yield was around 80%,  
6 much higher than the reported for APTES silanized chips, and similar or higher than those  
7 achieved in previous works (the reported values are 35% for direct attachment and 85%  
8 employing a cross-linker [33]). The target amount detected by ICPTS-immobilized probe  
9 ( $1.5 \cdot 10^{10}$  molecules/mm<sup>2</sup>) is also analogous to that obtained in recent SPR approaches ( $2 \cdot 10^{10}$   
10 molecules/mm<sup>2</sup> activity [17]). Thus, the higher hybridization yield obtained was expectable,  
11 because the ICPTS immobilization led to more vertical DNA strands and therefore more  
12 available for hybridization, and the probe coating density on the surface (Table 2) was adequate,  
13 in the  $10^{10}$  molecules/mm<sup>2</sup> order [37]. Indeed, the hybridization yield was nearly the same value  
14 as the percentage of oligonucleotide probes hypothetically placed in vertical when ICPTS was  
15 used (see previous section), although this correspondence is not so clear in the case of APTES.  
16  
17

### 18 *3.3. Blocking and hybridization for photochemical probe immobilization*

19

20 The assayed immobilization chemistries provided good results, but surface blocking  
21 was critical in both cases. It was interesting to assay a new immobilization method not leading  
22 to a reactive (isocyanate) or electrically charged (amine) surface. Silanization with GOPTS was  
23 promising because terminal epoxy or its hydrolysis product (alcohol) are neutral. X-ray  
24 photoelectron spectroscopy showed that derivatization with GOPTS was effective (see  
25 Supplementary material, Figure 1A).

26 Thiolated oligonucleotides were attached to the GOPTS-derivatized surface, this  
27 process being induced by UV radiation, with good efficacy in short time [34].

28 Thiolated probe B was immobilized off-line in both channels. The rest of steps -blocking  
29 with inert proteins and hybridization- were carried out on-line and monitored by DPI, using  
30 always SSC as a medium buffer. The specific hybridization was monitored using target B, while  
31 unspecific binding was controlled by injecting target F, both being the same size (see Table 1).  
32 Results of the target binding are shown in Table 4, for the different blocking agents assayed.

33 Regarding hybridization data, no blocking or the use of mercaptoethanol led to non-  
34 specific binding, but specific values were higher than control ones, so there was a net signal  
35 corresponding to hybridization. The net values were 0.190 and 0.220 nm thickness, and 120  
36 and 140 pg/mm<sup>2</sup> mass, for no blocking and mercaptoethanol, respectively. Thus, hybridization  
37 was more effective when mercaptoethanol blocking is carried out, but the difference is almost  
38 negligible. According to surface chemistry (see Figure 1), no blocking leads to epoxy-finished  
39 surface, that can be slowly converted into alcohol by hydrolysis, while reaction with

1 mercaptoethanol transforms epoxy into terminal alcohol. Both surfaces show neither electrical  
 2 charge nor reactivity with oligonucleotides and proteins, so the unspecific binding is only due  
 3 to physisorption on a neutral surface, and a net specific binding of target oligonucleotide can  
 4 be measured.

5 **Table 4**

Blocking and hybridization assays for photochemically immobilized oligo B

Blocking agent	Hybridization			
	Thickness (nm)		Mass (pg/mm <sup>2</sup> )	
	Specific	Control	Specific	Control
None	0.320	0.130	240	120
Mercaptoethanol	0.350	0.130	390	250
BSA	0.099	-0.012	47	-3
Casein	0.190	0.020	210	32
Gelatine	0.240	0.072	130	-29

6 Mercaptoethanol is equivalent to ethanolamine assayed in chemical immobilization experiments, and  
 7 blocking was carried out off-line and under UV-radiation [35].

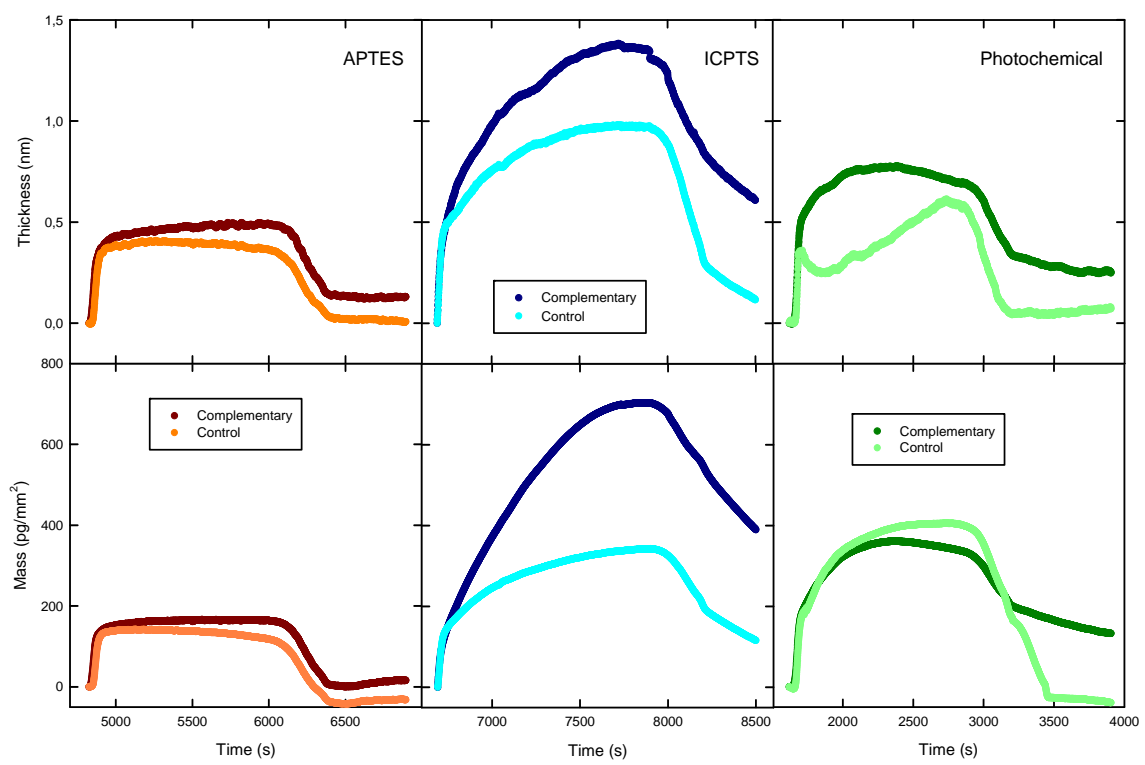
8  
 9 Blocking with inert proteins reduced the non-specific binding, because control results  
 10 are very low or even negative. The net difference signals were 0.170 nm and 180 pg/mm<sup>2</sup> for  
 11 casein and similar values for gelatine as well as 0.110 nm and 50 pg/mm<sup>2</sup> for BSA. The net  
 12 values obtained with casein and gelatine were the same order as those achieved with no  
 13 blocking or mercaptoethanol treatment, suggesting that these proteins do not cover the active  
 14 centres and allow the hybridization to take place. However, the lower net values obtained with  
 15 BSA suggest that the higher size of blocking protein hinders the approach of oligonucleotides  
 16 to the surface, so its use is disapproved. Therefore, it can be concluded that photochemical  
 17 immobilization is advantageous over APTES and ICPTS anchoring, regarding surface  
 18 blocking, because this step can be avoided maintaining hybridization net results, although the  
 19 use of casein or gelatine to totally remove unspecific binding is recommended in order to lower  
 20 background signals.

21 Mass increments for hybridization can be traduced to a maximum of  $1.56 \cdot 10^{10}$  (casein  
 22 blocking) and a minimum of  $4.4 \cdot 10^9$  (BSA blocking) molecules/mm<sup>2</sup>. This result is higher than  
 23 the best achieved with APTES immobilization, but similar to those for ICPTS. The  
 24 hybridization yield cannot be directly estimated, because no experimental data about probe  
 25 density on the chip has been measured. On the basis on the value reported in the literature [34]  
 26 for photochemical immobilization ( $1.5 \cdot 10^{10}$  molecules/mm<sup>2</sup>), hybridization efficiency would  
 27 be 70-100%, i.e. similar or higher than the best achieved with ICPTS, except when using BSA  
 28 for blocking (30%).

29 According to data shown in Table 4, and target B contour length (~8 nm), a theoretical  
 30 thickness increment of 0.280 nm can be estimated. The experimental net value was 0.190 nm,

1 which indicates that hybridization takes place in a partially flattened oligonucleotide  
 2 conformation. When using blocking agents, the observed trend was the same. This result agrees  
 3 with the probe placed in part horizontally, as reported [34], and is similar to other  
 4 immobilization chemistries employed.

5 Figure 3 compares the efficiency in hybridization achieved employing the three  
 6 anchoring methods. It can be seen that ICPTS and photochemical DNA probe immobilization  
 7 show to be better performing than classic APTES/glutaraldehyde, because mass and thickness  
 8 increments achieved are higher. ICPTS anchoring provides the highest amounts of detected  
 9 DNA target, improving sensitivity, but a surface blocking step with gelatine is needed.  
 10 Photochemical immobilization also allows a good detectability of DNA target, and in this case,  
 11 the surface blocking step can be avoided. The information provided by DPI has been useful to  
 12 develop high performance DNA microarray sensing systems. In those sensors, the low limits  
 13 of detection achieved (0.2 nM [34] and 0.02 nM [5]) were competitive with those obtained  
 14 using SPR methods [37]. In addition, the sensor selectivity was also successfully addressed,  
 15 and a maximum discrimination ratio of 14.5 was achieved for a single mismatch target [34].  
 16



17  
 18 Figure 3. Evolution of thickness (top) and mass (bottom) with time during the hybridization  
 19 step. Comparison between the different probe immobilization modes. Gelatine was employed  
 20 as blocking agent in all cases.  
 21  
 22  
 23

### 3.4. Application to species identification in multi-target mode

Hybridization of complementary DNA strands is widely used with analytical purposes. DPI has shown to be able of monitoring this binding event and providing valuable structural information, this being the main strength of this technique. As a weak point, DPI is able to process simultaneously only two events (two measuring channels), so its applicability as screening methodology is limited. Although the technique is not intended to be used as a sensor [12], the possibility of qualitative identification of specific DNA fragments was planted on the basis on a multiplex format.

For assessing the recognition and discrimination ability in a multi-target mode, probes A, C and D were anchored on both chip channels, in separate and in mixtures. ICPTS immobilization was used due to the higher DPI response, simplicity and ability to monitor on-line the whole process. After blocking with gelatine, combinations of target oligonucleotides, including the non-complementary (F), were passed through, recording the mass and thickness increment in different experiments. SSC buffer was always used as hybridization medium for being adequate in all probe-target combinations. Table 5 shows the mass and thickness display for single and multiple recognition events.

**Table 5**  
Hybridization assays in multiplex modes

Experiment No.	Immobilized Probes <sup>a</sup>	Injected targets <sup>b,c</sup>				
		A+C	A+F	C+F	A+C+D	F
1	Ch 1: A	0.530	-	-	-	0.067
	Ch 3: C	160	-	-	-	11
2	Ch 1: A + C	0.480	0.340	-	-	0.046
	Ch 3: A + C	88	63	-	-	12
3	Ch 1: A + C	-	-	0.250	-	0.043
	Ch 3: A + C	0.500	-	69	-	12
4	Ch 1: A + C + D	110	-	-	-	0.054
	Ch 3: A <sup>d</sup>	-	-	-	0.760	0.074
		-	-	-	210	-6
		-	-	-	0.160	0.090
					86	3

<sup>a</sup>Immobilization is carried out at 0.1  $\mu\text{M}$  for an only probe, 0.075  $\mu\text{M}$  each when two probes are immobilized together, and 0.05  $\mu\text{M}$  for the three probes simultaneously.

<sup>b</sup>Oligonucleotide targets at 0.5  $\mu\text{M}$  are mixed at the same solution and injected.

<sup>c</sup>Numeric values in each cell correspond to thickness increment (top) expressed in nm and mass increment (bottom) in  $\text{pg}/\text{mm}^2$ .

<sup>d</sup>Probe A immobilized at 0.05  $\mu\text{M}$ .

1 In the first experiment, the ability of each single probe for recognizing its target was  
2 checked. High mass and thickness increments for both specific bindings, and negligible values  
3 for the non-specific one, were observed.

4 Further, the recognition of one or two targets simultaneously was carried out using two  
5 probes immobilized in each channel and mixtures of strands containing both targets or only one  
6 (experiments 2 and 3). The registered mass and thickness increment for the double recognition  
7 was always higher (thickness  $\sim 0.500$ , mass  $\sim 115$ ) than for a single hybridization (thickness  
8  $\sim 0.300$ , mass  $\sim 65$ ), confirming the system potential for a duplex target identification.

9 Finally, applicability of DPI in a triple analyte sensor was tested by immobilizing three  
10 probes, each against a different target, and comparing with the hybridization for only one  
11 system (experiment 4). Results showed that the three mixed probes can bind all targets together  
12 or in separate. This concept could be used, for instance, in a microorganism survey for food  
13 safety; the three-target identification (A+B+C) could be applied for screening a bacteria family  
14 e.g. *Salmonella*, while assaying a single organism (A) could differentiate a key microorganism  
15 e.g. the dangerous *Salmonella typhimurium*. This application could be also performed by means  
16 of a typical DNA array [4], and although DPI philosophy is different from that of the screening  
17 analytical methods, this potential application has been also demonstrated.

#### 18 19 20 **4. Conclusions**

21 DPI has shown to be a useful tool for monitoring biomolecule binding events and  
22 associated conformational dynamics happening on a silicon oxynitride chip, with high  
23 accuracy. DPI chips allow employing different immobilization chemistries for covalent  
24 anchoring oligonucleotides, the classical APTES/glutaraldehyde method for amine-derivatized  
25 biomolecules, and the newer and simpler ICPTS and photochemical protocols. In order to  
26 minimize the effect of unspecific binding of oligonucleotide targets, the application of gelatine  
27 has proven to be superior to the use of other blocking agents such as ethanolamine, or BSA.  
28 With the adequate binding sites blocked, specific hybridization of up to three complementary  
29 oligonucleotides can be monitored, allowing the screening of DNA targets in a multiplex  
30 format. Chip coverage and binding yields can be determined at real time, and conformational  
31 dynamics can be elucidated from mass and thickness data.

32 The novel oligonucleotide immobilization methods, ICPTS-based and the  
33 photochemical procedure, have proven to provide better results of hybridization yields than  
34 APTES/glutaraldehyde method, which is an important conclusion for surface genomic analysis.  
35 The photochemical method avoids the need of blocking, while using ICPTS it is possible to  
36 monitor the probe immobilization and higher DPI responses are achieved. Furthermore, the  
37 same chip can be reutilized up to four times before losing its properties. The results obtained  
38 indicate the suitability of the methodology applied for the study of surface chemistries in order  
39 to be transferred to nanobiosensors development.



1  
2  
3  
4  
5  
6  
7  
8  
9  
10  
11  
12  
13  
14  
15

### Acknowledgements

Research projects MASCREEN CTQ2010-15943 from the Spanish Ministerio de Ciencia e Innovación, and PROMETEO 2010/008 from the Generalitat Valenciana are gratefully acknowledged for financial support.

### Appendix A. Supplementary data

Supplementary material related to this article can be found, in the online version

### References

- 
- [1] A.J. Hopwood, C. Hurth, Yang J, Z. Cai, N. Moran, J.G. Lee-Edghill, A. Nordquist, R. Lenigk, M.D. Estes, J.P. Haley, C.R. McAlister, X. Chen, C. Brooks, S. Smith, K. Elliott, P. Koumi, F. Zenhausern, G. Tully, Integrated microfluidic system for rapid forensic DNA analysis: sample collection to DNA profile, *Anal. Chem.* 82 (2010) 6991-6999.
  - [2] I. Palchetti, M. Mascini, Nucleic acid biosensors for environmental pollution monitoring, *Analyst* 133 (2008) 846-854.
  - [3] I.E. Sendroiu, L.K. Gifford, A. Lupták, R.M. Corn, Ultrasensitive DNA microarray biosensing via in situ RNA transcription-based amplification and nanoparticle-enhanced SPR imaging, *J. Am. Chem. Soc.* 133 (2011) 4271-4273.
  - [4] L. Tortajada-Genaro, S. Santiago-Felipe, S. Morais, J.A. Gabaldón, R. Puchades, A. Maquieira, Multiplex DNA detection of food allergens on a digital versatile disk, *J. Agric. Food. Chem.* 60 (2012) 36-43.
  - [5] J. Escorihuela, M.J. Bañuls, J. García-Castelló, V. Toccafondo, J. García-Rupérez, R. Puchades, A. Maquieira, Chemical silicon surface modification and bioreceptor attachment to develop competitive integrated photonic biosensors, *Anal. Bioanal. Chem.* 404 (2012) 2831-2840.
  - [6] M.J. Bañuls, R. Puchades, A. Maquieira, Chemical surface modifications for the development of silicon-based label-free integrated optical (IO) biosensors: A review, *Anal. Chim. Acta* 777 (2013) 1-16.
  - [7] M.S. Luchansky, R.C. Bailey, High-Q optical sensors for chemical and biological analysis, *Anal. Chem.* 84 (2012) 793-821.
  - [8] N.C. Pesquero, M.M. Pedroso, A.M. Watanabe, M.H.S. Goldman, R.C. Faria, M.C. Roque-Barreira, P.R. Bueno, Real-time monitoring and kinetic parameter estimation of the affinity interaction of jArtinM and rArtinM with peroxidase glycoprotein by the electrogravimetric technique, *Biosens. Bioelectron.* 26 (2010) 36-42.
  - [9] J.S. Yuk, M. Trnavsky, C. McDonagh, B.D. MacCraith, Surface plasmon-coupled emission (SPCE)-based immunoassay using a novel paraboloid array biochip, *Biosens. Bioelectron.* 25 (2010) 1344-1349.
  - [10] D.D. Nolte, *Optical Interferometry for Biology and Medicine*, first ed., Springer, New York, 2000.
  - [11] G.H. Cross, A.A. Reeves, S. Brand, J.F. Popplewell, L.L. Peel, M.J. Swann, N.J. Freeman, A new quantitative optical biosensor for protein characterisation, *Biosens. Bioelectron.* 19 (2003) 383-390.
  - [12] <http://www.farfield-group.com/pdfs/027.pdf>, accessed October 11, 2013.
  - [13] A.W. Sonesson, T.H. Callisen, H. Brismar, U.M. Elofsson, A comparison between Dual Polarization Interferometry (DPI) and Surface Plasmon Resonance (SPR) for protein adsorption studies, *Colloids Surf. B* 54 (2007) 236-240.
  - [14] M.J. Swann, L.L. Peel, S. Carrington, N.J. Freeman, Dual-Polarization interferometry: an analytical technique to measure changes in protein structure in real time, to determine the stoichiometry of binding events, and to differentiate between specific and nonspecific interactions, *Anal. Biochem.* 329 (2004) 190-198.

- 
- [15] H.N. Daghestani, B.W. Day, Theory and Applications of Surface Plasmon Resonance, Resonant Mirror, Resonant Waveguide Grating, and Dual Polarization Interferometry biosensors, *Sensors* 10 (2010) 9630–9646.
- [16] G.W. Platt, F. Damin, M.J. Swann, I. Metton, G. Skorski, M. Cretich, M. Chiari, Allergen immobilisation and signal amplification by quantum dots for use in a biosensor assay of IgE in serum, *Biosens. Bioelectron.* 52 (2014) 82–88.
- [17] J.B. Wood, M.W. Szyndler, A.R. Halpern, K. Cho, R.M. Corn, Fabrication of DNA microarrays on polydopamine-modified gold thin films for SPR imaging measurements, *Langmuir* 29 (2013) 10868–10873.
- [18] J.A. Ho, T.Y. Kuo, L.G. Yu, Dual polarization interferometric and capillary electrophoretic analysis of supported lipid bilayer constructed on silicabased surface: Evaluation of its anti-protein adsorption effect, *Anal. Chim. Acta* 714 (2012) 127–133.
- [19] A. Shovskiy, I. Varga, R. Makuska, P.M. Claesson, Adsorption and solution properties of bottle-brush polyelectrolyte complexes: effect of molecular weight and stoichiometry, *Langmuir* 28 (2012) 6618–6631.
- [20] S. Johnson, A. Bronowska, J. Chan, D. Evans, A.G. Davies, C. Wälti, Redox-induced conformational change in mercaptoalkanoic acid multilayer films, *Langmuir* 28 (2012) 6632–6637.
- [21] J.F. Poplewell, M.J. Swann, N.J. Freeman, C. McDonnell, R.C. Ford, Quantifying the effects of melittin on liposomes, *Biochim. Biophys. Acta* 1768 (2007) 13–20.
- [22] K.E.D. Coan, M.J. Swann, J. Ottl, Measurement and differentiation of ligand-induced calmodulin conformations by Dual Polarization Interferometry, *Anal. Chem.* 84 (2012) 1586–1591.
- [23] J. Zhai, T.H. Lee, D. Small, M.I. Aguilar, Characterization of early stage intermediates in the nucleation phase of A $\beta$  aggregation, *Biochemistry* 51 (2012) 1070–1078.
- [24] J. Moore, M.A. Pérez-Pardo, J.F. Poplewell, S.J. Spencer, S. Ray, M.J. Swann, A.G. Shard, W. Jones, A. Hills, D.G. Bracewell, Chemical and biological characterisation of a sensor surface for bioprocess monitoring, *Biosens. Bioelectron.* 26 (2011) 2940–2947.
- [25] H.Y. Song, X. Zhou, J. Hobley, X. Su, Comparative study of random and oriented antibody immobilization as measured by Dual Polarization Interferometry and Surface Plasmon Resonance spectroscopy, *Langmuir* 28 (2012) 997–1004.
- [26] D. Giménez-Romero, M.A. González-Martínez, M.J. Bañuls, I.S. Monzó, R. Puchades, A. Maquieira, Modeling of the role of conformational dynamics in kinetics of the antigen–antibody interaction in heterogeneous phase *J. Phys. Chem. B* 116 (2012) 5679–5688.
- [27] A. Boudjemline, E. Saridakis, M.J. Swann, L. Govada, I.M. Mavridis, N.E. Chayen, Use of Dual Polarization Interferometry as a diagnostic tool for protein crystallization, *Anal. Chem.* 83 (2011) 7881–7887.
- [28] L. Lee, A.P.R. Johnston, F. Caruso, Manipulating the salt and thermal stability of DNA multilayer films via oligonucleotide length, *Biomacromolecules* 9 (2008) 3070–3078.
- [29] Y. Wang, J. Wang, F. Yang, X. Yang, Probing biomolecular interactions with Dual Polarization Interferometry: real-time and label-free coralyne detection by use of homo adenine DNA oligonucleotide, *Anal. Chem.* 84 (2012) 924–930.
- [30] V.C. Özalp, Dual-Polarization Interferometry for quantification of small molecules using aptamers, *Anal. Bioanal. Chem.* 402 (2012) 799–804.
- [31] Y. Wang, Y. Zheng, F. Yang, X. Yang, Dual Polarisation Interferometry for real-time, label-free detection of interaction of mercury(II) with mercury-specific oligonucleotides, *Chem. Comm.* 48 (2012) 2873–2875
- [32] H. Berney, K. Oliver, Dual Polarization Interferometry size and density characterisation of DNA immobilisation and hybridisation. *Biosens. Bioelectron.* 21 (2005) 618–626.
- [33] B. Lillis, M. Manning, H. Berney, E. Hurley, A. Mathewson, M.M. Sheehan, Dual Polarisation Interferometry characterisation of DNA immobilisation and hybridisation detection on a silanised support. *Biosens. Bioelectron.* 21 (2006) 1459–1467.
- [34] J. Escorihuela, M.J. Bañuls, R. Puchades, A. Maquieira, Development of oligonucleotide microarrays onto Si-based surfaces via thioether linkage mediated by UV irradiation. *Bioconjugate Chem.* 23 (2012) 2121–2128.
- [35] J. García-Rupérez, V. Toccifondo, M.J. Bañuls, J. García Castelló, A. Griol, S. Peransi-Llopis, A. Maquieira, Label-free antibody detection using band edge fringes in SOI planar photonic crystal waveguides in the slow-light regime, *Opt. Express* 18 (2010) 24276–24286.
- [36] C. Shi, S. Mehrabani, A.M. Armani, Leveraging bimodal kinetics to improve detection specificity, *Opt. Lett.* 37 (2012) 1643–1645.
- [37] H. Šípová, J. Homola, Surface plasmon resonance of nucleic acids: A review, *Anal. Chim. Acta* 773 (2013) 9–23.
- [38] Gelatin, CAS RN 9000-70-8. Product Information Sheet, Sigma-Aldrich web page.

1  
2  
3  
4  
5  
6  
7  
8  
9  
10  
11  
12  
13  
14  
15  
16  
17  
18  
19  
20  
21  
22  
23  
24  
25  
26  
27  
28  
29  
30  
31  
32  
33  
34  
35  
36  
37  
38  
39  
40  
41

## Biographies

**Jose-Luis López-Paz** received his PhD degree in Chemistry from the University of Valencia in 1996. He worked in the sensors group led by Dr. Joseph Wang at New Mexico State University in 1998, for electrochemical detection of nucleic acids. He is Assistant Professor in the Department of Chemistry at the Universidad Politécnica de Valencia since 2005 and joined Institute of Molecular Recognition and Technological Development in 2012. His current research interests are mainly focused on optical sensors and biomolecular recognition.

**Miguel Angel González-Martínez** received his Ph.D. degree in Chemistry from the Universidad Politécnica de Valencia in 1998 and is Associate Professor of Analytical Chemistry at the Universidad Politécnica de Valencia since 2002. His research is centred on automation in Analytical Chemistry, bioreagent immobilization and biosensing. He has co-authored around 25 research papers in international journals. He is also member of the Spanish Society of Analytical Chemistry.

**Jorge Escorihuela** received his Ph.D. degree in Green Chemistry from the Universitat Jaume I (Castellón, Spain) in 2009, his doctoral research being focused on developing new chiral catalysts derived from amino acids and their application in asymmetric reactions. In 2010 he joined the Centro de Reconocimiento Molecular y Desarrollo Tecnológico at the Universidad Politécnica de Valencia as research scientist, their activities being centered on biosensing on silicon materials and DNA microarrays.

**María-José Bañuls** received the European Ph.D. degree in Organic Chemistry from the Universitat de València in 2004. She is Assistant Professor in the Department of Chemistry at the Universidad Politécnica de Valencia. Her research interest is focused on chemical modification of polymeric and inorganic materials for biosensing, especially for the development of microarray methodologies to be applied in Genomics and Proteomics. She has co-authored about 30 papers in international journals and holds 3 patents.

**Rosa Puchades** is Full Professor of Analytical Chemistry at the Chemistry Department of the Universidad Politécnica de Valencia. She received her Ph.D. degree from the University of Valencia (1982). Her research interests include automation of analytical methods, sample treatment, development of immunochemical reagents, immunoanalytical methods, screening methodologies for organic pollutants, biosensors and microarraying. She is currently participating in several research projects in the analytical and bioanalytical field. Dra. Puchades has co-authored more than 170 papers in international journals and conferences. She is also member of the Spanish Society of Analytical Chemistry and the American Chemical Society.

**Ángel Maquieira** is Full Professor of Analytical Chemistry at the Chemistry Department of the Universidad Politécnica de Valencia. He received his Ph.D. degree from the University of Valencia, Spain (1980). His research interest topics include development of immunochemical methods, massive screening methodologies, surface functionalization chemistry for biosensing and probe-target biointeraction. The newest study topic is related to developing microarraying protocols operating on compact disc supports and nanobiosensing systems applied in the life sciences area.