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Effect of methylisothiazolinone on biological treatments: influence on the efficiency of SBR reactors and bioindicative studies

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Abstract

BACKGROUND: Methylisothiazolinone (MIT) is widely used as biocide in hair and skin care products and in high concentrations, more than 150 mg L⁻¹, in cooling towers in paper mill and refinery industry. This product has been recognized as potent contact allergen and is discharged to the WWTP from industrial processes or domestic usages. However, the role of MIT in biological treatment has not been characterized and therefore, the effect of MIT on the composition and performance of activated sludge has been analyzed.

RESULTS: For this purpose, a sequential batch reactor was fed with a synthetic solution containing 5 mg L⁻¹ of MIT. The addition of MIT had shown no significant effect on organic matter metabolization (DOC removal remained constant at 80-90%). On the contrary, a loss of efficiency in the nitrification process occurred (ammonium removal decreased from 90% to 20% per cycle throughout the experiment), first affecting the nitrite oxidizing bacteria (NOB) and, after around 40 cycles, also the ammonium oxidizing bacteria (AOB) disappeared, as shown by FISH analysis. Bioindicative studies based on counts of protozoa and sludge biotic index indicated that, at the end of the process, a stable and well colonized protozoa community was obtained. Finally, respirometric tests indicated some acclimatization of the heterotrophic bacteria to MIT, as shown by shorter lag periods when the sludge was previously fed with MIT.

CONCLUSION: The role of MIT in biological treatment has been characterized to increase knowledge about this pollutant effects on activated sludge in order to improve WWTP performance.

Highlights

MIT has no effect on the efficiency of activated sludge to treat organic matter.

Nitrification is severely affected by MIT, confirmed by FISH analysis

Respirometric and bioindicative tests show acclimatization of heterotrophic species

Keywords

Methylisothiazolinone; nitrification; activated sludge; bioindicative tests; SBR.

1. Introduction

Conventional wastewater treatment plants (WWTP) have been proven to be efficient in the depuration of domestic sewage as well as some industrial wastewaters (Avella et al., 2011, Jahan et al., 2012); however some pollutants are resistant to these processes. In particular, the development of new methods of wastewater analysis, able to decrease the limits of detection of pollutants to a few ng·L⁻¹ or even below, have shown that families of chemicals are not completely removed by WWTP and consequently become effluents or surface waters (Richardson et al., 200, De la Cruz et al., 2012). They are commonly referred to as emerging pollutants (EPs), as their potential effects have still not been determined. Among them, personal care products or household products can be included (Murray et al., 2010, Petrovic et al., 2006).

In particular, 2-methyl-4-isothiazolin-3-one or methylisothiazolinone (MIT) can be labeled as an emerging pollutant. This is a biocide compound used in cooling systems or as a preservative in cosmetics; in fact concentrations of this compound up to 150 mg L⁻¹ have been reported to be added in cooling towers (Critchley et al., 2009). MIT has shown some toxicity towards aerobic and anaerobic bacteria as well as fungi and other microorganisms (Torén et al., 1997) and its use is limited by its ability to cause skin irritation, allergic reactions and potential neurotoxicity (European Commisión, 2009). MIT has increasingly been used as a preservative in cosmetics, skin care products (Geier et al., 2012) and as preservative in surface waters to cooling towers in recent years (in Spain around 12% of total biocides used (Martinez et al., 2013).

As MIT has been found in trace amounts in surface waters (after household and industry wastewater treatment), it can represent a concern for water quality and its effect on microorganisms constituting activated sludge should be determined (Rafoth et al., 2007). In fact, some EPs have been reported to be potentially toxic towards the biomass and consequently, to induce changes in the efficiency of the biological treatment. In fact, this efficiency is closely related to the composition of the sludge, namely the bacterial, protozoan and metazoan population (Juliastuti et al., 2003). For instance, some bacterial families are responsible for the nitrification process and their absence in the sludge would result in ammonium accumulation (Bassin et al., 2012); furthermore, protozoa are considered to be important bioindicators for the activated sludge, as their population has been associated with operational parameters (Tocchi et al., 2012). As the protozoa sizes are some orders of magnitude larger than bacteria, they can be identified and quantified by optical microscopy; hence, they have been used as a fast and convenient way to define the biological quality of the sludge (e.g. sludge biotic index) (Arévalo et al., 2009).

Against this background, the aim of this work is to study whether MIT has some influence on the composition and performance of activated sludge, before a more detailed study can be performed; hence a high concentration of this pollutant was used in a synthetic sample, simulating a pulp and paper industry wastewater, in order to rule out, as far as possible, effects that are not attributable to this pollutant. For this purpose a sequencing batch reactor (SBR) was used. This approach was chosen because SBRs are able to remove not only organic carbon, but they can also deal with nitrogen (Gerardi, 2010). SBR can be programmed in consecutive cycles of aerobic and anoxic conditions to induce nitrification and denitrification processes (Singh et al., 2011). Furthermore, these reactors can be more flexible and efficient than biological

continuous reactors for the degradation of EPs, because the cycle can be modified to fit the biodegradation process (Salgado et al., 2012). Some important parameters were monitored to control SBR performance, namely organic matter and MIT removal, nitrification and denitrification processes as well as sludge composition.

2. Material and methods

2.1. Synthetic Wastewater and Initial Biomass

The biodegradable synthetic wastewater used to feed the activated sludge, consisted of a mixture of readily biodegradable organic substrates: 250 mg L⁻¹ of meat extract (BIOKAR), milk (175 mg L⁻¹) and ethanol (100 mg L⁻¹). Inorganic nutrients such as KH₂PO₄ (22.5 mg L⁻¹) and NH₄Cl (80 mg L⁻¹) were also present, together with trace amounts of other chemicals (MgSO4, NaHCO₃, KCl, CaCl₂ and FeCl₃ of ca. 250 mg L⁻¹). Synthetic wastewater was prepared daily by diluting 30 mL of this mixture with tap water to reach an initial chemical oxygen demand (COD) in the reactor of ca. 300 mg O₂ L⁻¹. Eventually, MIT (≥98% Fluka) was added to the synthetic influent in a concentration of 5 mg L⁻¹ to simulate a typical concentration in water production circuit of paper mill. According to Document on Best Available Techniques in the pulp and paper industry, there is an input of biocides to the water system (between 10 - 100 g/t of paper) together with other wet-end chemicals (European Commision, 2013).

The activated sludge was taken from a WWTP from the East of Spain fed with a mixture of household and industrial effluents. The sludge was first acclimatized to the biodegradable synthetic wastewater in the SBR reactor (20 cycles, 10 days) and then to MIT (55 cycles, 27 days). Initially, the volatile suspended solids (MLVSS) in the reactor were 2500 mg L^{-1} and the sludge retention time (SRT) was ca. 20 days; SRT was kept constant throughout the experiment considering the daily sludge extraction and total suspended solids in the reactor.

2.2. Laboratory-scale SBR

Experiments were performed in the set-up described in Figure 1. The SBR reactor has a volume of 7 liters and is equipped with a mechanical stirrer (50 rpm), peristaltic pumps to fill and empty the reactor, and a temperature control system (20°C). Oxygen was supplied through a compressor (1 L min⁻¹). Dissolved oxygen (DO) and pH (WTW probe) were measured on line and processed by a data acquisition card and software developed using LabView (National Instruments, USA).

The SBR was operated in 12 h automatic cycles consisting of the following consecutive phases: a) fill, b) aerobic reaction-nitrification c) anoxic reaction-denitrification and sludge extraction d) settle e) effluent discharge and f) idle. The sludge extraction was carried out in the last of 10 minutes of anoxic phase in order to control the sludge mass (at that time the concentration was known). This sequence and the length of each phase can be seen in Table 1.

2.3. Chemical analyses

Samples were periodically and directly taken from the reactor during the aerobic phase and filtered through glass fiber filter before analysis (nominal pore size 1.2 μm). MIT concentration was determined by HPLC analysis (Perkin Elmer Autosystem XL equipped with a diode-array detector and an autosampler). A LiChrosphere 100 RP-18 column was employed, and the eluent was an isocratic mixture of acetonitrile (90%) and formic acid 0.01N (10%). UV-Detection was fixed at 274 nm. Identification and quantitation was based on comparison with standards. DOC was measured with a Shimadzu model TOC-V CSH apparatus. COD was determined photometrically according to the dichromate method; digestions were performed at 148°C in a Thermoreaktor TR300 (Merck) and a Spectroquant NOVA 60 (Merck) was used for the photometric determination. Mixed liquor suspended solids (MLVSS) was also determined according to the standardized procedure (APHA, 1995). Nitrate and ammonia were measured potentiometrically (Crison NO₃⁻ 96 62, NH₄⁺ 96 63).

2.4. Biological measurements

Bioindicative studies and protozoan counts were performed with optical phase contrast microscopy (VWR microscope TR400-SW) using a Fluchs Rosenthal sample holder with bright line. Species identification and counts were done at $\times 200$ magnification *in vivo* conditions. Samples of activated sludge (25µL), used in protozoan counts, were analyzed immediately after sampling. Analyses were repeated four times and results are the average of those measurements.

The Sludge Biotic Index (SBI) was calculated; briefly, this is an indicator of the state of the activated sludge and can be used to estimate its depuration ability. It is based on the flagellates, testate, naked and some attached amoebae, and swimming and crawling ciliates populations. More detailed information of this methodology can be found elsewhere (Madoni et al., 1984).

Communities of nitrifying bacteria were identified and quantitated by means of the Fluorescence In Situ Hybridization (FISH) technique with image analysis (Manz et al., 1992). Measurements were carried out with probes labeled with fluorophores of the 16SrDNA of specific ammonium oxidizing ammonium bacteria (AOB) and nitrite oxidizing bacteria, NOB (probe Nso1225 for AOB and probes Ntspa662+Competing and NIT3+Competing for NOB). For the bacterial community analysis (Eubacteria domain) probes EUB 338I, II and III (EUB mix) were used. Matlab software (Borrás, 2008) was used for image analysis in order to carry out the quantification (X 10 \pm SD/ \sqrt{n}).

Respirometric analyses were performed in an electrolytic respirometer (Bioscience, BI2000). Glass bottles (1 liter) loaded with the samples to be analyzed were thermostatized and sealed with a device able to monitor the oxygen consumption and to regenerate it via electrolysis. As a consequence plots of oxygen consumption vs time can be calculated. Bottles were filled with 250 mL of sample containing MIT (5 mg L⁻¹) and glucose (700 mg L⁻¹). Parallel to this, blank experiments were also run (700 mg L⁻¹ of glucose dissolved in distilled water). Inoculum (2.5 mL of sludge) and the necessary nutrients (MgSO₄, CaCl₂, FeCl₃, Na₂HPO₄, NH₄Cl, KH₂PO₄, K₂HPO₄) were added to each bottle.

3. Results and discussion

3.1. Activated sludge acclimatization to readily biodegradable substrate

First, the sludge was subjected to an acclimation process for 17 days. The initial behavior of the aerobic phase of the SBR was evaluated by adding a readily biodegradable synthetic substrate, as described in the experimental section (initial COD = $300 \text{ mg} \cdot \text{L}^{-1}$). Variations in DOC and COD were monitored during each cycle (see figure 2 for an example): the mean removal in acclimation experiments was 87% and 92% respectively after three hours; beyond this point, those values remained constant. MLVSS were kept at ca. 2500 mg L⁻¹ throughout the acclimatization period. Furthermore, as shown in figure 2, ca. 65% ammonia removal was reached, indicating an acceptable nitrification in this acclimation process.

3.2. Activated sludge acclimatization to MIT

After seventeen days of initial acclimatization, MIT (5 mg L⁻¹) was added together with the biodegradable synthetic wastewater during 27 days (55 cycles). MLSS, DOC, COD, MIT concentration, ammonium and nitrate were monitored during eight cycles corresponding to different selected days (days 1, 4, 8, 11, 16, 22, 24 and 26).

The evolution of MLSS during the acclimatization period (fed with a biodegradable mixture without MIT) and the MIT treatment period can be found in Figure 3. Addition of MIT resulted in a remarkable decrease in MLSS during the first 22 cycles and, beyond this point, a recovery in this parameter was observed. This behavior in SBR has been described elsewhere (Choubert et al., 2006, Vaiopoulou et al., 2012) and can be associated to an initial toxic or inhibitory effect of the xenobiotic towards some species found in the activated sludge.

Variation of DOC throughout each cycle is shown in Figure 4. The trends are similar in all cases, reaching final DOC removal in the range 80-90% after 5 hours, which means that MIT is not able to decrease significantly the metabolization of the added organics. However, slight differences in the time-resolved data might be found: from cycles 1 to 22, longer periods are required to reach the final DOC values, as can be observed in Figure 4, inset, where the reduction reached after 30 min is shown; the opposite behavior is observed form cycle 22 to 44, when a faster decrease was monitored. Beyond this point, the process becomes slower again.

It is interesting to note that although some decrease in MLVSS was observed upon MIT addition in Figure 3, it did not result in noticeable loss of efficiency in DOC mineralization. This could be due to the fact that organic load is rather low and the hydraulic retention time is long enough to allow a good mineralization of organics despite the above mentioned kinetic differences [Orhon et al., 2009, Adrianus et al., 2012, Moussavi et al., 2009, Kim et al., 2013).

In order to gain further insight into the process, MIT concentration was also determined by HPLC analysis. Some differences can be observed in MIT removal yield after the aerobic phase (Figure 5): in the first cycles, there was a certain decrease in MIT elimination (from 80% to 55%), which can be associated to the period required for the acclimatization of the activated sludge to MIT (days 1-8); then, the yield was recovered to reach the initial values of ca. 80% (day 11). However, beyond this point, a slight loss of efficiency in dissolved MIT decrease was detected (70% at day 26), which might be due to some MIT accumulation in the reactor, (daily treated effluent discharged was 60% of total reactor volume) inducing some toxicity in the sludge. This is in agreement with the variation of the initial concentration of MIT in each cycle, also shown in figure 5, which beyond cycle 44 suffers a progressive increase.

It is interesting to study the effect of MIT on the nitrification process. For this purpose, ammonium and nitrate were analyzed in the aerobic cycle. Figure 6A shows that ammonium removal was high during the first cycles (70-95% yield in days 1-22). However, in the final days monitored (days 24-26), most of the ammonium remained in the reaction medium (less than 25% removal). Regarding nitrate, the high concentration of this ion formed in the first day

(close to their stoichiometric amount) indicated that a complete nitrification was achieved. In posterior cycles, the amounts of nitrate were systematically below 10% the expected stoichiometric concentration. This indicates that nitrification was not complete (Figure 6B).

Taking into account that the nitrification processes consists of two steps, namely nitrite formation and consequent oxidation to nitrate, the behavior above described can be explained by assuming that MIT was able to damage the bacteria responsible for the oxidation of nitrite to nitrate after a short period of exposure (a few days). On the other hand, more cycles were required to inhibit the transformation of ammonium into nitrite. In order to clarify this point, FISH measurements were carried out to better characterize bacteria population. Samples were taken at day 11, when ammonium transformation was still efficient but NO₃⁻ was not formed (partial nitrification) and at day 26, when nitrification was very poor.

FISH analysis for samples taken on day 11 showed the presence of ammonia oxidizing bacteria (AOB) with the typical formations of dense cell aggregates with spherical shape of ammonium oxidizing β -proteobacteria class. On the other hand, nitrite-oxidizing bacteria (NOB) were not found in these samples, as demonstrated by the absence of the typical formations of dense cell aggregates with spherical shape of genus Nitrospira and Nitrobacter. According to these data, there was a well-developed community of AOB which constitute a 5% ±0.8 of the total viable bacterial community of the activated sludge and an absence of NOB. This is in agreement with the sludge being able to oxidize ammonium into nitrite but not to further oxidize this anion to nitrate.

In sharp contrast, when samples taken in day 26 were submitted to FISH analyses, together with the absence of NOB, a decrease in the population of AOB was also observed (only $1\% \pm 0.2$ of the total viable bacterial community of activated sludge), which points to a significant loss of efficiency of the nitrification process (Figure 7). Although the system is too complex to attribute this effect to a single cause, MIT might play an important role in the inhibition of nitrification by autotrophic bacteria (Zanetti et al., 2012, Feng Wang et al., 2012), as other important operational parameters remained constant throughout the experiment (pH, temperature, dissolved oxygen concentration and alkalinity)

and nitrite concentration in the effluent was systematically below 2 mg L⁻¹ after the complete cycle. It is worth commenting that the low final values of nitrite could be attributed to the action of heterotrophic bacteria during the anoxic phase of the cycle.

3.3. Bioindicative studies

Evolution of protozoa population in the sludge when submitted to MIT might be also of interest in order to evaluate the effect of this chemical on the system. Figure 8 shows the total abundance of protozoa in samples taken throughout the experiment. The initial sludge mainly contained amoebae (*Arcella sp.* and *Mayorela sp.*), together with some *Vorticella Infosionum* sp. and *Epistylis sp.* Upon addition of MIT, the population of naked amoebae (*Mayorella sp.*) was observed to increase, reaching a maximum (day 8) and then decreased, nearly completely disappearing at the end of the experiment; the growth of naked amoebae in the startup of bioreactors followed by a decrease when the bioreactor is adapted to the medium is well-known behavior described elsewhere (Madoni, 2011). This is in general agreement with the chemical parameters reported above, which showed an initial period of adaptation to MIT (see above, Figure 5 and 6).

On the other hand, *Arcella sp.* becomes predominant after day 11, once the reactor is acclimatized to MIT. High relative populations of these species show good correlation with efficient pollutant removal, low organic load, high sludge retention time and high concentration of dissolved oxygen in the aeration tank, which are the conditions required to obtain a complete nitrification (Yusof et al., 2010); hence it is to be expected that problems in the nitrification process would result in a decrease in *Arcellas sp.* number. This was the behavior observed in our case, in which the number of *Arcellas* in the final samples decreased.

The sludge biotic index (SBI) was calculated based on the population of the five species identified. Attached ciliated protozoa were detected (*Vorticella Infusionum* or *Epistylis* sp.) throughout the experiment; on the other hand no rotifers (*Rotaria sp.*) were observed in the initial days although some of these

metazoa were found in the final samples (day 24 and 26). Taking also into account the amoebae (whose behavior is reported above), a SBI value of 7 was systematically calculated, which means that the sludge can be labeled as Class II (SBI 6 to 7). This means that the sludge was stable and well colonized, with optimal biological activity butshowing discrete efficiency in wastewater treatment. This is in general agreement with the results obtained on the behavior of the sludge (Salgado et al., 2012).

3.4. Respirometric Test

In order to determine the effect of MIT on acclimatized and non-acclimatized sludge, respirometric assays were performed with sludge taken from the SBR at the end of the process and from the WWTP (without adaptation to MIT); this is interesting to better understand the effect of MIT on the heterotrophic organisms that constitute the sludge. In both cases, the sludge was fed with a highly biodegradable mixture containing glucose and inorganic nutrients, to which MIT was later added. Figure 9 shows a plot of the oxygen uptake vs. time for all experiments. Although similar final values for oxygen uptake were obtained in all cases, significant differences can be found in the lag periods. In the presence of MIT results in longer lag periods: ca. 15 hours) and the SBR sludge and 60 hours for the WWTP sludge. This different behavior can be attributed to some inhibition caused by MIT to the sludge respiration, which is more noteworthy in that obtained from the WWTP, which has not been previously adapted to this pollutant.

4. Conclusions

MIT has been proven not to have a noteworthy influence on the ability of activated sludge to treat organic matter, and some acclimatization to this EP is indicated by respirometric tests. However, this compound has been shown to

completely inhibit the nitrification process, probably due to the toxicity of MIT towards NOB and, after longer exposure, also to AOB. Hence, further studies in order to better determine the potential effect of this pollutant in biological processes is necessary; furthermore, the development of an advanced treatment capable of removing MIT would be convenient when nitrification is required.

Bioindicative studies based on protozoa population are in general agreement with the observed behavior of the sludge. Hence, they could be a simple, quick and effective tool to predict the performance of an activated sludge.

The role of MIT in biological treatment has been characterized to increase knowledge about this pollutant effects on activated sludge in order to improve wastewater industrial treatment plant performance. In further works lower concentrations of MIT (around $ng \cdot L^{-1}$ or $\mu g \cdot L^{-1}$) should be employed to simulate SBR behavior when household effluents are treated.

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6. References

Avella, A.C., Görner, T., Yvon, J., Chappe, P., Guinot-Thomas, P., Donato, Ph. de, A combined approach for a better understanding of wastewater treatment plants operation: Statistical analysis of monitoring database and sludge and sludge physico-chemical characterization, Water Res. 45: 981-992 (2011).

Adrianus C. van Haander, J., van der Lubbe, G. M., Handbook of biological wastewater treatment: design and optimisation of activated sludge systems, 2012, London : IWA Publishing

APHA American Public Health Association. Standard methods for the examination of water and wastewater, 20th ed.,(1995) APHA, Washington.

Arévalo, J., Moreno, B., Pérez, J., Gómez, M.A., Applicability of the Sludge Biotic Index (SBI) for MBR activated sludge control, J. Hazard. Mat. 167: 784– 789 (2009).

Bassin, J.P., Kleerebezem, R., Muyzer, G., Rosado, A.S., Van Loosdrecht M.C.M., Dezotti, M., Effect of different salt adaptation strategies on the microbial diversity, activity, and settling of nitrifying sludge in sequencing batch reactors, Appl. Microbiol. Biot. 93: 1281-1294 (2012).

Borrás, L., 2008. Técnicas microbiológicas aplicadas a la identificación y cuantificación de microorganismos presentes en sistemas EBPR. Thesis. Universidad Politécnica de Valencia. Valencia. (2008) España.

Choubert, J.M., Racault, Y., Grasmick, A., Canler, J.P., Heduit, A., A simulationbased optimisation approach to control nitrogen discharge by activated sludge under winter seasonal peak load, Water SA 32 (2006) 561-566

Critchley M., Bentham, R., The efficacy of biocides and other chemical additives in coolingwater systems in the control of amoebae, J. Appl. Microbiol., 106: 784-789 (2009).

De la Cruz, N., Giménez, J., Esplugas, S., Grandjean, D.L., Alencastro, F. de, Pulgarín, C., Degradation of 32 emergent contaminants by UV and neutral photo-fenton in domestic wastewater effluent previously treated by activated sludge, Water Res. 46: 1947-1957 (2012).

European Commission, CosIng Cosmetic ingredients &Substances, (2009). http://ec.europa.eu/consumers/cosmetics/cosing/layout/CosIng_Manual.pdf

European Commission. Integrated Industrial Emissions Directive 2010/75/EU. Pollution Prevention and Control (IPPC). Best Available Techniques (BAT) Reference Document for the production of Pulp, paper and Borad. July 2013. (http://eippcb.jrc.ec.europa.eu/reference/BREF/PP_BREF_FD_07_2013.pdf).

Feng Wang, Yi Liu, Jinghan Wang, Yalei Zhang, Haizhen Yang. Influence of growth manner on nitrifying bacterial communities and nitrification kinetics in three lab-scale bioreactors. J. Ind. Microbiol. Biotechnol. 32: 595-604 (2012).

Geier, J., Lessmann H., Schnuch, A., Uter, W., Recent increase in allergic reactions to methylchloroisothiazolinone/methylisothiazolinone: is methylisothiazolinone the culprit?, Contact Dermatitis 67: 334–341 (2012).

Gerardi, M., Troubleshooting the sequencing batch reactor, Wiley, New Jersey, 2010.

Jahan, K., Hoque, S., Ahmed, T. Activated Sludge and other suspended culture processe, Water Environ. Res. 84: 1029-1080 (2012)

Juliastuti, S.R., Baeyens, J., Creemers, C., Bixio, D., Lodewyckx, E., The inhibitory effects of heavy metals and organic compounds on the net maximum specific growth rate of the autotrophic biomass in activated sludge, J. Hazard. Mater. 100: 271-283 (2003).

Kim, M., Guerra, P., Theocharides, M., Barclay, K., Smyth, S.A., Alaee, M., Parameters affecting the occurrence and removal of polibrominated diphenyl ethers in twenty Canadian wastewater treatment plants, Water Res. 47: 2213-2221 (2013).

Madoni, P, Protozoa in wastewater treatment processes: A minireview, Ital. J. Zool. 78(1): 3-11 (2011).

Madoni, P., Estimation of the size of freshwater ciliate populations by a subsampling technique, Hydrobiologia 111: 201-206 (1984).

Manz, W., Amann, R., Ludwig, W., Wagner, M., Schleifer, K.H., Phylogenetic oligonucleotide probes for the mayor sub-class of Proteobacteria: problems and solutions, System, Appl. Microbiol. 15: 593–600 (1992).

Martinez, G., Arias, G., Montero Rubio, J.C., Estudio de biocidas utilizados en Torres de Refrigeración y Condensadores Evaporativos. Área Sanitaria de Talavera de la Reina 2007-2012, XII Congreso Español de Salud Ambiental, Granada, Spain (2013). Moussavi, G., Mahmoudi, M., Barikbin, B., Biological removal of phenol from strong wastewater using a novel MSBR, Water Res., 43: 1295-1302 (2009).

Murray, K.E., Thomas, S.B., Badour A.A., Prioritizing research for trace pollutants and emerging contaminants in the freshwater environment, Environ. Pollut. 158: 3462-3471 (2010).

Orhon, D., Germirli, F., Karahan, O., Industrial wastewater treatment by activated sludge, 2009, London : IWA Publishing

Petrovic, M., Barceló, D., Liquid chromatography-mass spectrometry in the analysis of emerging environmental contaminants, Anal. Bioanal. Chem. 385: 422-424 (2006).

Rafoth, A., Gabriel, S., Sacher, F., Brauch, H.J., Analysis of isothiazolinones in environmental waters by gas chromatography–mass spectrometry, J. Chromatogr. A, 1164: 74–81 (2007).

Richardson, S.D., Environmental mass spectrometry: emerging contaminants and current issues, Anal. Chem. 80: 4373-4402 (2008).

Salgado, R., Oehmena, A., Carvalho, G., Noronhaa, J.P., Reis, M.A.M., Biodegradation of clofibric acid and identification of its metabolites, J. Hazard. Mat. 241–242: 182–189 (2012).

Singh, M., Srivastava, R. K., Sequencing batch reactor technology for biological wastewater treatment: a review, Asia-Pac. J. Chem. Eng. 6: 3–13 (2011).

Tocchi, C., Federici, E., Fidati, L., Manzi, R., Vincigurerra, V., Petrucciolia, M., Aerobic treatment of dairy wastewater in an industrial three-reactor plant: Effect ofaeration regime on performances and on protozoan and bacterial communities, Water Res. 46: 3334-3344 (2012).

Torén, K., Brisman, J., Meding, B., Sensitization and Exposure to Methylisothiazolinones (Kathon) in the Pulp and Paper Industry- A report of Two Cases, Am. J. Ind. Med. 31: 551-553 (1997).

Vaiopoulou, E., Gikas, P., Effects of chromium on activated sludge and on the performance of wastewater treatment plants: A review, Water Res. 46: 549-570 (2012).

Yusof, N., Hassan, M.A., Phang, L.Y., Tabatabaei, M., Othman, M.R., Morid, M., Wakisaka, M., Sakai, K., Shirai, Y., Nitrification of ammonium-rich sanitary landfill leachate, Waste Manage. 30: 100–109 (2010).

Zanetti, L., Frison, N., Nota, E., Tomizioli, M., Bolzonella, D., Fatone, F., Progress in real-time control applied to biological nitrogen removal from wastewater. A short-review. Desalination, 286: 1-7 (2012). Table 1. Operational setting of the SBR reactor and the duration of each phase of the cycle. There was some phases coincide in time with others (for example sludge extraction was performed at end of anoxic reaction). The feed volume was 4 L cycle⁻¹.

Phase	Time (min)
Fill	8
Aerobic reaction	420
Anoxic reaction	150
Sludge extraction (in the anoxic phase)	2
Settle	90
Effluent discharge	9
Idle	51

Figure Captions

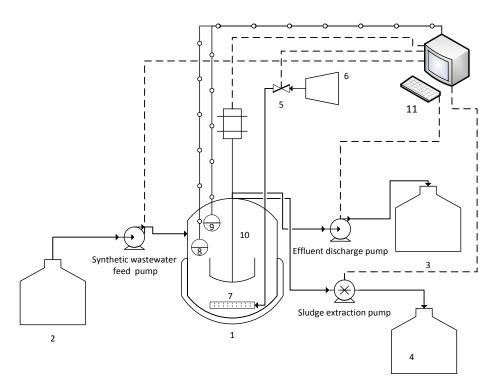


Figure 1. Scheme of the experimental set-up. (1) SBR, (2) Feed tank, (3) Treated water tank, (4) Sludge tank, (5) Air regulation valve, (6) Air Compressor, (7) Air diffusor, (8) pH probe, (9) Oximeter, (10) Stirrer, (11) Digital/analog I/O and Labview software to SBR control.

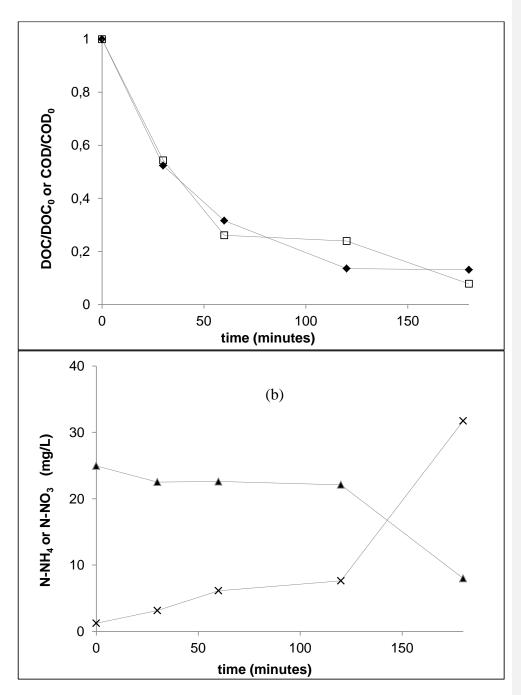


Figure 2. Example of the variations in different parameters vs time throughout an aerobic cycle in an SBR in the acclimatization period to a readily biodegradable substrate: (a) DOC (\blacklozenge) and COD (\Box);COD (t=0 min)= 230 mgL⁻¹, COD (t=180 min)= 18 mgL⁻¹, DOC (t=0 min)= 65 mgL⁻¹, DOC (t=180 min)= 8 mgL⁻¹ (b) Nitrate (x) and ammonium (\blacktriangle); N-NH₄ (t=0 min)= 25.0 mgL⁻¹, N-NH₄ (t=180 min)= 7.9 mgL⁻¹, N-NO₃ (t=0 min)= 1.2 mgL⁻¹, N-NO₃ (t=180 min)= 31.7 mgL⁻¹.

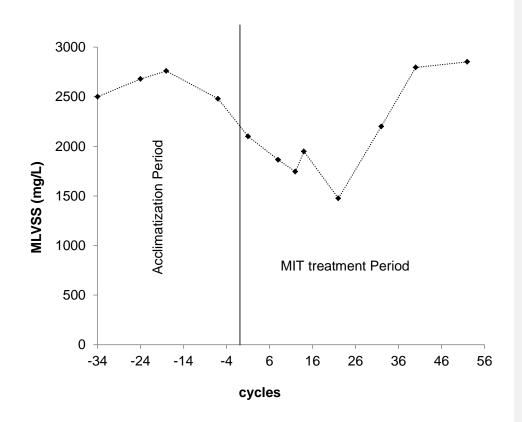


Figure 3. Plot of the mixed liquor suspended solids in the SBR experiment vs time. The negative values indicate the acclimatization period to a biodegradable mixture and the positive time represents the MIT treatment period. The line indicates the first day of MIT treatment period.

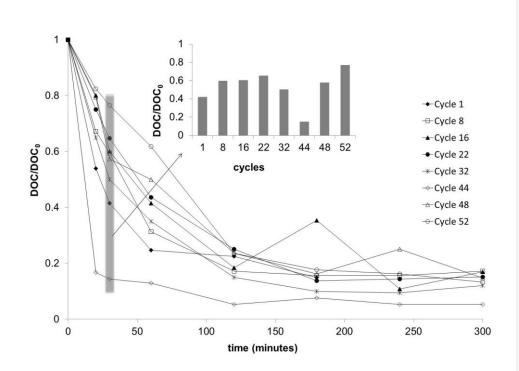


Figure 4. Relative DOC values (DOC/DOC_0) measured throughout aerobic phase. Data obtained after 30 minutes of treatment is shown in the inset.

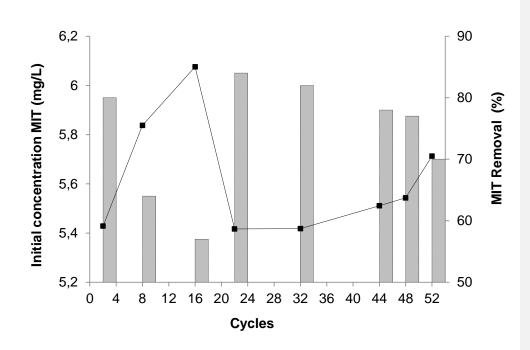


Figure 5. Concentration of MIT in an SBR at the beginning of a cycle in different days (■). The percentage of MIT removal for the corresponding cycle is also given (Grey column, right axis).

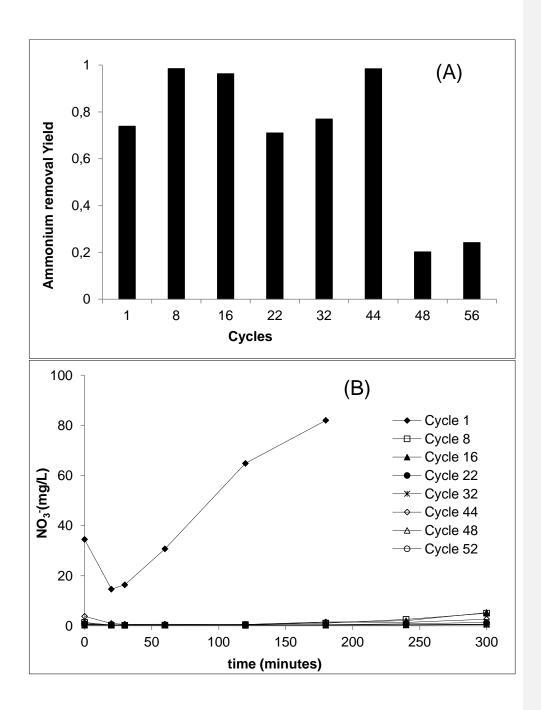


Figure 6. Nitification process in an aerobic cycle of the SBR submitted to MIT. A) Ammonium removal achieved in a cycle after different periods of contact of SBR to MIT. B) Nitrate (mg L^{-1}) detected along selected cycles.

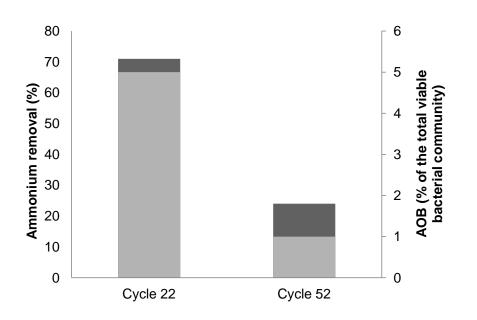


Figure 7. The figure shows the comparison between the AOB population (% of the total viable bacterial community), [grey column on the secondary axis] and ammonia removal efficiencies (%), [black column on the principal axis] in two cycles (22 and 52) of the MIT treatment.

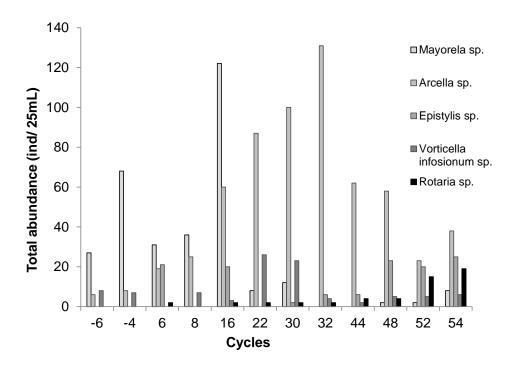


Figure 8. Composition of protozoa community present in an SBR. Results for the first adaptation period and MIT treatment are given.

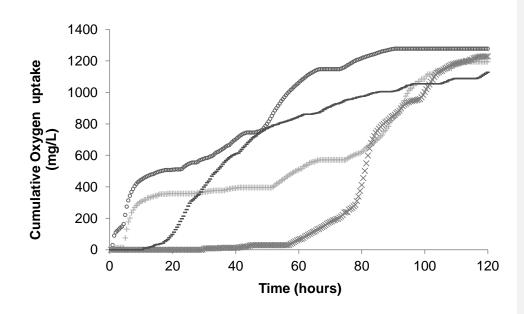


Figure 9. Plot of the accumulated oxygen uptake vs time obtained in the degradation of a readily biodegradable substrate (glucose, 700 mg L⁻¹) different respirometric tests: SBR biomass without MIT (ooo), SBR biomass with 5 mg L⁻¹ of MIT (---), WWTP biomass without MIT (+++); WWTP biomass with 5 mg L⁻¹ of MIT (××××).