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Al₂O₃-3YTZP-Graphene multilayers produced by tape casting and spark plasma sintering

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ABSTRACT

This work aims to establish a colloidal route to obtain laminates of alumina-zirconia combining layers with and without graphene. Green tapes of alumina, alumina with 5vol.% of 3Y-TZP and alumina with 5vol.% of 3Y-TZP and graphene-oxide (2vol.%) were obtained by aqueous tape casting. It is

possible to design materials for different structural applications with a controlled microstructure with a high number of different layers. The tapes were punched into 20-mm discs, joined to form laminates alternating up to 18-layers, and sintered in one-step by spark plasma sintering (SPS) at 1400 °C. It has demonstrated that there is a significant graphite diffusion provoked by the required graphite holders into the SPS-furnace. Dense laminates with layer thicknesses ~100 μm and good cohesion between layers were obtained.

Nanoindentation results showed that hardness and elastic modulus values were higher than 27 GPa and 300 GPa, respectively, and similar for all layers.

Keywords: Multilayers; Mechanical properties; Tape casting; Spark plasma sintering, Graphene

1. Introduction

The incorporation of carbon derivatives, such as carbon nanotubes (CNTs), carbon nanofibers (CNFs), diamond nanoparticles and graphene as secondary phases in composites, is receiving increasing attention due to their unique combination of properties that enhance when they are added in small concentrations.[1-5] CNTs and CNFs have been largely used as suitable reinforcing agents of different materials for many applications such as prostheses fabrication, lithium ion batteries, proton exchange membranes, etc.[6-9] Single-wall carbon nanotubes (SWCNTs) have been successfully used for these purposes, although the difficult homogenisation within the matrix has still important limitations.[10,11] Consequently, the studies regarding the use of multi-wall carbon nanotubes (MWCNTs), which are much easier to disperse in a

1 matrix (especially in water), have received greater attention. The most widely
2 studied ceramic matrices are oxides such as alumina, zirconia, mullite, etc.,[12-
3 16] and also non-oxidic ceramics such as silicon nitride in which new properties
4 are achieved, for example electrical conductivity, as well as mechanical
5 reinforcement.[17-19] The use of diamond nanoparticles has huge interest for
6 improving the contact-mechanical and tribological properties of the
7 composites.[20,21]

16 Graphene is considered one of the most singular materials for reinforcing
17 polymers, metals and ceramics. Among the wide number of applications of
18 graphene those referring to ceramics have not yet achieved the expected
19 consideration, but it has been already demonstrated its beneficial effect in the
20 enhancement of conductivity and reinforced mechanical properties, to mention
21 only a couple.[22-26] In spite of these clear advantages, there are two main
22 topics that are far to be already solved: the synthesis of the graphene sheets,
23 on one hand, and the preparation of homogeneous mixtures of graphene in a
24 ceramic powder matrix, on the other hand. Graphene is usually obtained from
25 graphite flakes by liquid exfoliation, milling or sonication and by the so-called
26 Hummers' method,[27] in which the oxidation of graphite to graphitic oxide is
27 accomplished by treating graphite with essentially a water-free mixture of
28 concentrated sulphuric acid, sodium nitrate and potassium permanganate.
29 Graphene is obtained by reduction of this graphitic oxide for which hydrazine is
30 often used. Moreover, the use of a hydrophilic form of graphite (as an oxide)
31 has important advantages from the point of view of processing and
32 dispersibility, as it allows an easy mixing in aqueous suspensions if colloidal
33 interactions are adequately controlled.[28,29]

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The polar character of graphene oxide (GO) allows the dispersion by electrostatic or electrosteric forces facilitating its dispersion in polar solvents such as water. Since the properties of GO are worse than those of the reduced graphene, materials obtained with GO must be reduced in a further step, normally by using sintering cycles under reducing atmosphere.[30]

As stated earlier, the most studied ceramic matrices for such GO-reinforced ceramics have been alumina and Y-TZP, as it was expected that the carbon nanodispersoids could provide the required strengthening. It is well known that small zirconia additions can enhance the mechanical behaviour of an alumina matrix by transformation toughening.[31,32]

From a structural point of view, the major concern is to increase the mechanical performance by either the action of reinforcing mechanisms or by reducing the flaws effect with tailored microstructures, which make the material to become flaw tolerant. In this sense, laminates have demonstrated their suitability for obtaining materials with superior properties to those of the individual components.[33-36] Laminates can improve mechanical performance using either laminates with a weak interface that deflects cracks, thus preventing catastrophic failure or laminates with strong interfaces designed on the basis of residual stress development in the layers during cooling from the sintering temperature. Furthermore, from an electrical features point of view, to achieve the configuration of a series of electrical and non electrical conductive layered material it should be very interesting to develop microelectronic devices with high performances.

1 The most widely studied laminate system is that formed by alumina-zirconia, in
2 which a broad interval of thermal strains can be reached by modulating the
3 thermal expansion coefficient of the constituent layers and/or the extent of
4 zirconia phase transformation. The manufacture of these laminates needs the
5 use of techniques allowing a simple processing and lamination process. For
6 such purpose colloidal shaping techniques have been extensively used,
7 especially simple processes such as sequential slip casting and tape
8 casting.[37-40]

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19 In contrast with the extensive bibliography about alumina-zirconia
20 laminates and the increasing effort on the use of carbon nanodispersoids on
21 single phase materials, the effect of graphene additions on the processing, on
22 mechanical performance and electrical conductivity of alumina-zirconia
23 composites is very scarce.[41]

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31 In this context the objective of this work was to study the processing of
32 alumina-zirconia laminates comprising layers with and without graphene,
33 obtained by spark plasma sintering, in order to evaluate the effect of the
34 graphene addition on the properties of alumina-zirconia composite layers.
35 Furthermore, by means of stacking layers with and without graphene, it has been
36 achieved the construction of a ceramic compound formed by electrical layers
37 isolated by non electrical layers. In a previous paper, the preparation of the
38 individual layers constituting the laminate was studied in terms of suspensions
39 stabilisation, rheological behaviour and tape casting performance.[42] This
40 paper focuses on the lamination of those tapes into specific microarchitectures
41 combining layers of alumina-zirconia with and without graphene.

2. Experimental procedure

Two commercial ceramic powders were used as starting materials: an α -alumina powder (Ceralox, Condea HPA05, USA) with an average particle size of 0.35 μm and a specific surface area of $9.5 \text{ m}^2\cdot\text{g}^{-1}$, and a nanosized zirconia doped with 3 mol.% of Y_2O_3 (TZ3YE, Tosoh, Japan) with an average particle size of 100 nm and a specific surface area of $14.5 \text{ m}^2\cdot\text{g}^{-1}$. A monolayer graphene oxide (Nanoinnova Technologies, Spain), simply referred to as GO, with average lengths and thicknesses in the range of 1 to 4 μm and 0.7 to 1.2 nm, respectively, and a surface area of $\sim 103 \text{ m}^2\cdot\text{g}^{-1}$ was also used as a secondary reinforcing phase.

Multilayer materials were prepared by simple lamination (under atmospheric conditions and hand pressure gluing with a few water drops) of dry layers obtained by aqueous tape casting of concentrated alumina suspension containing zirconia as secondary phase (referred to as AZ), and layers of AZ containing 2 vol.% (0.5 wt.%) with regard to dry ceramic powders of graphene oxide as a dispersed phase (referred to as AZGO). Alumina (A) layers were also prepared to allow comparison with AZ ones. Since alumina is very well known no further rheological studies were done, so that the tapes were prepared as described in previous work.[42] The relative volume ratio of alumina to zirconia was 95:5 vol.% (92.5:7.5 wt.%). The suspensions were prepared to solids loadings of 53 vol.% for A and AZ, and 40 vol.% for AZGO by sonication in pulsed mode (0.5 cycles). The details of the rheological studies and the optimisation of tape casting performance were reported in a previous work.[42] Tape casting suspensions were prepared with an acrylic latex emulsion with a content of active matter of 55 wt.% (Duramax B-1000, Rohm &

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Haas, PA, USA), which was used as a binder adding 15 wt.% (weight of as-supplied product related to the total solid content, i.e. by 8.3 wt.% of active matter), and a defoamer (KS 1115, Zschimmer & Schwarz, Germany) to allow the elimination of air bubbles originated during homogenisation, which was added in a concentration of 0.15 wt.%. The so-prepared mixtures were kept under mechanical stirring for 20 min to achieve homogenisation before casting. Tape casting was conducted onto a Mylar film using a single blade geometry casting head; the blade height was set at 200 μm with a casting speed of 50 $\text{mm}\cdot\text{s}^{-1}$.

Three families of laminates were produced for this study: 1) alternating layers of A and AZ, the laminate being referred to as A/AZ; 2) alternating layers of AZ and AZGO, which was referred to as AZ/AZGO, and 3) alternating the three types of layers, A/AZ/AZGO. In Fig. 1 a schematic flow chart of the lamination process is shown. Tapes were first cut into discs with 2 cm in diameter, and the discs were then laminated applying a small pressure and using a few drops of water as gluing agent. Symmetric laminates with 9-layers were obtained.

To obtain the sintered materials two stacks of 9 layers each were introduced into a 20-mm-diameter graphite die. The architecture of the different green laminates formed from the tailored stacks can be seen in the picture. The stacks were sintered using a spark plasma sintering (SPS) apparatus HP D25/1 (FCT Systeme GmbH, Rauenstein, Germany) at temperature of 1400 $^{\circ}\text{C}$ and 80 MPa of pressure to obtain fully sintered bulk materials. The tests were carried out under vacuum at a heating rate of 100 $^{\circ}\text{C}\cdot\text{min}^{-1}$ with a 1 min dwelling time at the maximum temperature.

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Sintered samples were fractured on their cross-section and were polished to 0.5 mm using SiC paper and diamond suspension. Resultant layers were observed with optical microscope Nikon LV-100 under dark field illumination. The fracture surface was analysed by using a field emission gun scanning electron microscope (FE-SEM, HITACHI S-4800, SCSIE of the University of Valencia).

Raman spectra were determined on the cross-section of polished samples except for a sample that was analysed after fracture to diminish the possible contamination of the polishing procedure or graphene loss. The measurements were performed at room temperature with a Renishaw inVia spectrometer with $\lambda=514.5$ nm laser. A 100x objective was used to locate the 3 μm diameter spot area of analysis on sample. Two spectral ranges were scanned for each layer, 100-800 cm^{-1} and 1000-3200 cm^{-1} with 2 cm^{-1} step resolution. The incident power was carefully programmed to avoid laser induce heating.

Vickers microhardness (HV) was acquired using a Shimadzu HMV tester under a 0.2 Kg indentation load. Hardness (H) and modulus (E) were analysed by a nanoindenter G-200 of Agilent Technol. (Inc., Santa Clara, CA) under a 1200 nm constant indentation depth program. A Berkovich tip was used after calibration of the function area in fused silica. Stiffness was recorded in depth by Continuous Stiffness Measurement (CSM). The Poisson's coefficient was 0.23 for all calculations considering a fully dense material. In this regard, the wide range of reported Poisson's coefficient for alpha alumina and tetragonal zirconia depends on the porosity and the level of defects in the

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microstructure.[43] The resulting reduced modulus has been also reported in this manuscript in order to visualise the influence of this coefficient.

3. Results and discussion

Homogeneous, dense laminates were obtained after heating at 1400 °C by SPS. Fig. 2 shows the microarchitecture of the three multilayers as observed by optical microscopy. Obviously the external surfaces resulted damaged after the cycle due to the interaction with graphite dies, but the layers configuration remains and the interfaces are sharp and well defined. The picture of Fig. 3 corresponds to a higher magnification image showing a detailed view of the microstructure at the interfaces between A, AZ and AZGO layers. This picture demonstrates the clear, sharp interfaces and the uniformity of the layers.

The microstructure of the fracture surfaces of the different layers (A, AZ, and AZGO) in the A/AZ/AZGO laminate can be observed in Fig. 4. FE-SEM observations of these fracture surfaces show a high degree of densification without apparent macroscopic defects or porosity. As observed in the microphotographs, faceted grains of about 0.25 μm are obtained in any of the layers. These images corroborate that the conditions employed for the SPS procedure were adequate. The presence of some isolated nanosheets within the A/AZ/AZGO layers was captured (indicated by an arrow). These films were observed at higher magnification (Fig. 5) and were revealed as sheets with planar dimensions of several microns but only few nanometres thin. Similar morphological aspect has been reported for few-layered graphene or graphene oxide nanoplatelets.[44] Nevertheless, the identification of graphene flakes is a

1 major hurdle and scanning electron microscopy technique is not an
2 appropriated tool for this aim.
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4 In order to confirm that graphene remains in the reduced form inside the
5 AZGO layers, Raman spectroscopy analysis was performed for such layers.
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7 The α -Al₂O₃ and tetragonal zirconia (3YTZP) have been well characterized by
8 Raman spectroscopy showing bands centred at 378 cm⁻¹, 416 cm⁻¹, 429 cm⁻¹,
9 576 cm⁻¹, 644 cm⁻¹ and 750 cm⁻¹ for alumina and 144 cm⁻¹, 268 cm⁻¹, 314 cm⁻¹,
10 458 cm⁻¹ and 650 cm⁻¹ for tetragonal zirconia.[45,46] Fig. 6 shows the spectra
11 obtained on the cross-section of A and AZ layers indicating that the unique
12 phases present in these layers are α -Al₂O₃ and 3YTZP+ α -Al₂O₃, respectively,
13 as expected.
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26 Nevertheless, graphene bands are centred at the same wavelength than
27 those of graphite. For a 514.5 nm laser these bands are found at 1370 cm⁻¹,
28 also named D-band that is characteristic of disorder, at 1583 cm⁻¹ or G-band
29 that is assigned to a graphitic structure and it is good indicator of graphene
30 layers; and the 2700 cm⁻¹ band or 2D band that can be also used to determine
31 the number of graphene layers. It has been shown [46] that as the number of
32 layers increases, the G-band position shifts to lower frequencies. Fig. 7 screens
33 the spectra obtained in the spectral range of Raman wavelengths in which
34 graphene must be detected. These results reveal the presence of the
35 characteristic G, D and 2D bands on all analyzed layers and samples, even
36 inside the layers that did not contain any carbon source (A and AZ layers). This
37 fact demonstrates that there is a significant graphite diffusion provoked by the
38 required graphite holders into the SPS furnace. This hypothesis was
39 corroborated when the same spectra was acquired on the grey coloured
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1 samples' surfaces instead of their cross-sections. The diffusion mechanism of
2 graphite inside the ceramic material in SPS is not yet well understood.
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4 Consequently, it was not possible to distinguish the graphene signal by Raman
5 analysis because the graphite signal or by the carbon formed from
6 the decomposition of the polymer masks the possible graphene response.
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8 Furthermore, Centeno et al. [25] showed that graphene basal plane is
9 preferentially oriented perpendicularly to the direction of the pressure applied
10 during SPS. Consequently, the surface of graphene exposed to the Raman's
11 source is very small when samples are analyzed on their cross-section.
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13 However mechanical properties seem to be affected by the addition of
14 graphene while graphite contamination does not negatively affect, as shown
15 below.

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17 Mechanical properties of single layers were determined by
18 microhardness measurements and by nanoindentation technique. Fig. 8 shows
19 several arrays of Berkovich imprints performed at a constant 1200 nm depth by
20 nanoindentation technique. The microstructure shows that A and AZ layers
21 have densified to suitable values, whereas the layers containing GO have
22 considerably higher porosity. The different layers were obtained at similar tape
23 casting speed and blades height, but there is a clear difference, the A layer
24 being the thickest while the AZGO one is the thinnest. There is an inverse
25 relation between viscosity of the slurry and final thickness. The viscosities of the
26 A, AZ and AZGO tape casting suspensions were 819, 480 and 523 mPa·s (at a
27 shear rate of 250 s^{-1} , which was roughly calculated to be the shear rate when
28 the slurry is passing through the blade). The corresponding average
29 thicknesses are 75, 60, and 50 μm , respectively.

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2 Fig. 9 shows the resulting H and E curves as a function of the indentation
3 depth obtained by nanoindentation technique. Vickers hardness was also
4 acquired for each layer and these results are summarised in Table 1.
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7 It must be remarked that the laminated material displays high
8 nanomechanical properties that can be explained by the high densification and
9 small grain size achieved by tapes lamination and further non-conventional
10 sintering by SPS technique. The content of either ZrO_2 or graphene slightly
11 diminishes the hardness and modulus with regard to that of the pure α -alumina
12 and these results are in good agreement with those reported for fully dense
13 materials.[47-50] The nanomechanical properties were also investigated along
14 all the width of the laminate in order to detect possible gradients in densification.
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17 Fig. 10 shows the elastic modulus (Fig. 10a) and the nanohardness (Fig. 10b)
18 profiles for indentations performed at 300 nm maximum depth (Fig. 10c). These
19 curves demonstrate that the density along the cross-section of sample is
20 constant and the mechanical properties remain constant inside the entire
21 sample, without any differences at the various layers comprising the laminate.
22 The dispersion of values for a single layer is explained because at low
23 penetration depths the response of single phases is distinguished. 3Y-TZP is
24 found in the form of small aggregates in the alumina layer (inset figure).
25 Consequently, the variability of results indicates the presence of some
26 aggregates of 3Y-TZP grain or dense alumina beyond the indenter. This
27 tendency to form nanoaggregates is, unfortunately, very usual in the
28 preparation of AZ composites, in spite of the enhancement of the dispersion
29 using colloidal processing.
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In this work, the electrical conductivity of the A, AZ and AZGO layers were confirmed by a very simple method. Using a two probe electrical multimeter device, it was confirmed that the electrical resistance was below 10 Ohm on the AZGO layers while open circuit (Megaohm) was registered on the surface on the A and AZ layers. In order to ensure a good electrical contact very thin gold wire probes were used. In an incoming paper, the electrical conductivity performance of these laminates will be further studied using more adequate techniques.

In summary the results shown in this paper demonstrate that it is possible to tailor ceramic laminates comprising any number of layers comprising graphene or not, as desired. While the mechanical properties at every layer are not so different, the electrical conductivity is only found in layers with graphene.

4. CONCLUSIONS

This paper reports for the first time the preparation of Al₂O₃-3YTZP-graphene multilayer materials combining tape casting and fast spark plasma sintering technique with good cohesion and high mechanical properties.

Raman spectroscopy has been used to study in detail the layers and to evaluate and optimize the in situ graphene thermal reduction process in the SPS. During the composite fabrication process, the material sintering and the GO thermal reduction occurs simultaneously, and also, we have demonstrate that there is a significant graphite diffusion provoked by the required graphite holders into the SPS furnace. Therefore, it was not possible to distinguish the graphene signal by Raman analysis because the graphite signal masks the possible graphene response.

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2 From these results it can be concluded that it is possible to design
3 materials with a controlled microstructure with a high number of different layers
4 comprising graphene or not, as desired. The nanomechanical properties at the
5 different layers of the laminate are rather similar, so that the presence of
6 graphene is not deleterious for the laminate integrity and the layers properties.
7 This could be very advantageous for the macroscopic mechanical behavior of
8 the laminate, in which the presence of the interfaces could contribute to
9 enhance the laminate behavior. This would require the evaluation of bending
10 strength, which is out of the scope of the present investigation.
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25
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Tables:

Table 1. Vickers hardness and mechanical properties obtained by nanoindentation technique of the A/AZ/AZGO sample. Nanoindentation results were averaged out in the range between 500 to 1000 nm in depth.

Layer	H _{IT} (GPa)	Std. Dev	E (GPa)	Std. Dev.	Er (GPa)	HV _{2kg}
A	27.0	0.3	409.8	2.9	395.8	2145
AZ	25.8	0.4	400.0	3.7	317.8	2020
AZGO	23.5	0.3	373.9	3.1	298.0	1856

Captions to figures:

Fig. 1. Schematic flow chart showing the preparation and configuration of the green multilayer compositions, including the following steps: (1) punching of green discs; (2) lamination of the layer to obtain symmetric 9-layers stacks at room conditions by pressing using water drops as gluing agent; 3) sintering of materials at 1400 °C by SPS under vacuum and 80 MPa of pressure.

Fig. 2. Optical microscopy images of the three types of laminates sintered by SPS; A/AZ (a), AZ/AZGO (b), and A/AZ/AZGO (c), taken under dark field illumination. The brightest layers in (c) correspond to the AZGO composition while darkest ones are α -Al₂O₃.

Fig. 3. Optical microscopy picture showing the interfaces between the A, AZ and the AZGO layers in the sintered A/AZ/AZGO sample.

Fig. 4. FE-SEM images showing the fracture surfaces of the A layer (a), AZ layer (b), and AZGO layer (c). The arrow in (c) shows a possible few-layered graphene or graphite-detached film. This analysis was performed on the cross-section of fractured A/AZ/AZGO sample.

Fig. 5. FE-SEM image of a fracture surface of an AZGO layer in the A/AZ/AZGO composite, in which two thin films of graphene or few-layered graphite were captured.

1 **Fig. 6.** Raman spectra of Al₂O₃ (A) and Al₂O₃+5vol.% ZrO₂ (AZ) layers.

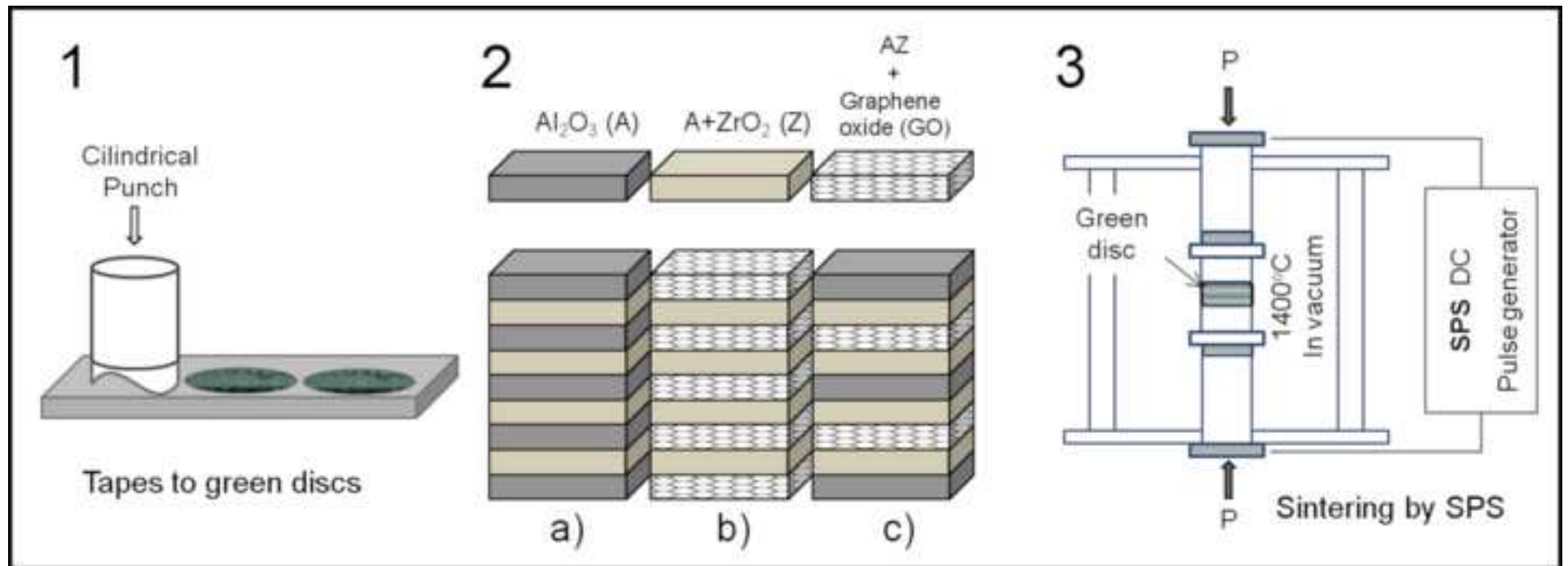
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5 A/AZ/AZGO sample. Analysis was performed on the cross-section of the
6 polished and fractured sample except when the surface of the sintered disc was
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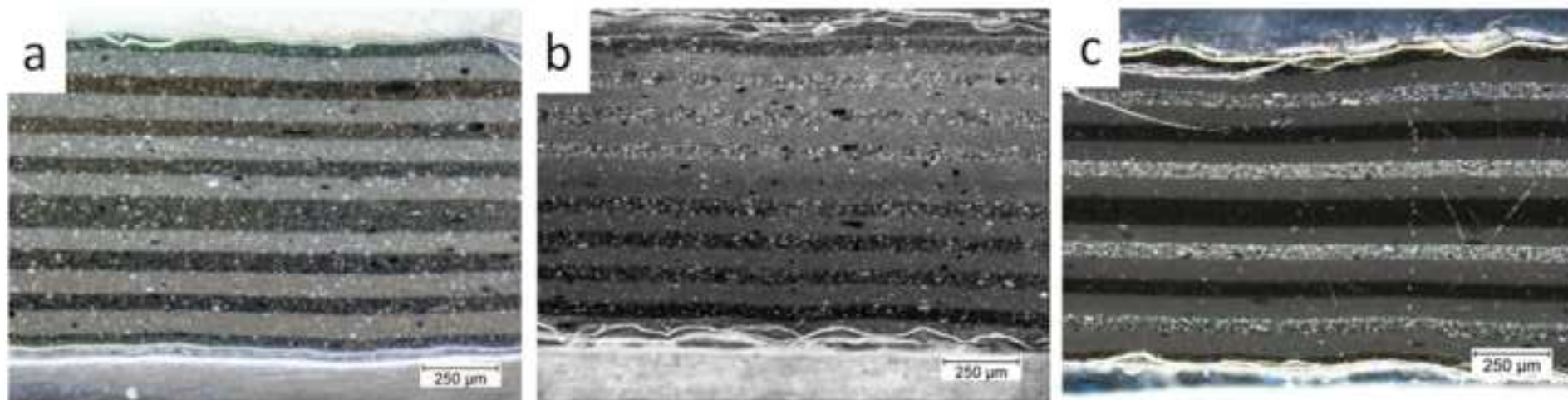
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35 Fig. 10c shows the array of Berkovich imprints distanced by 10 μm.
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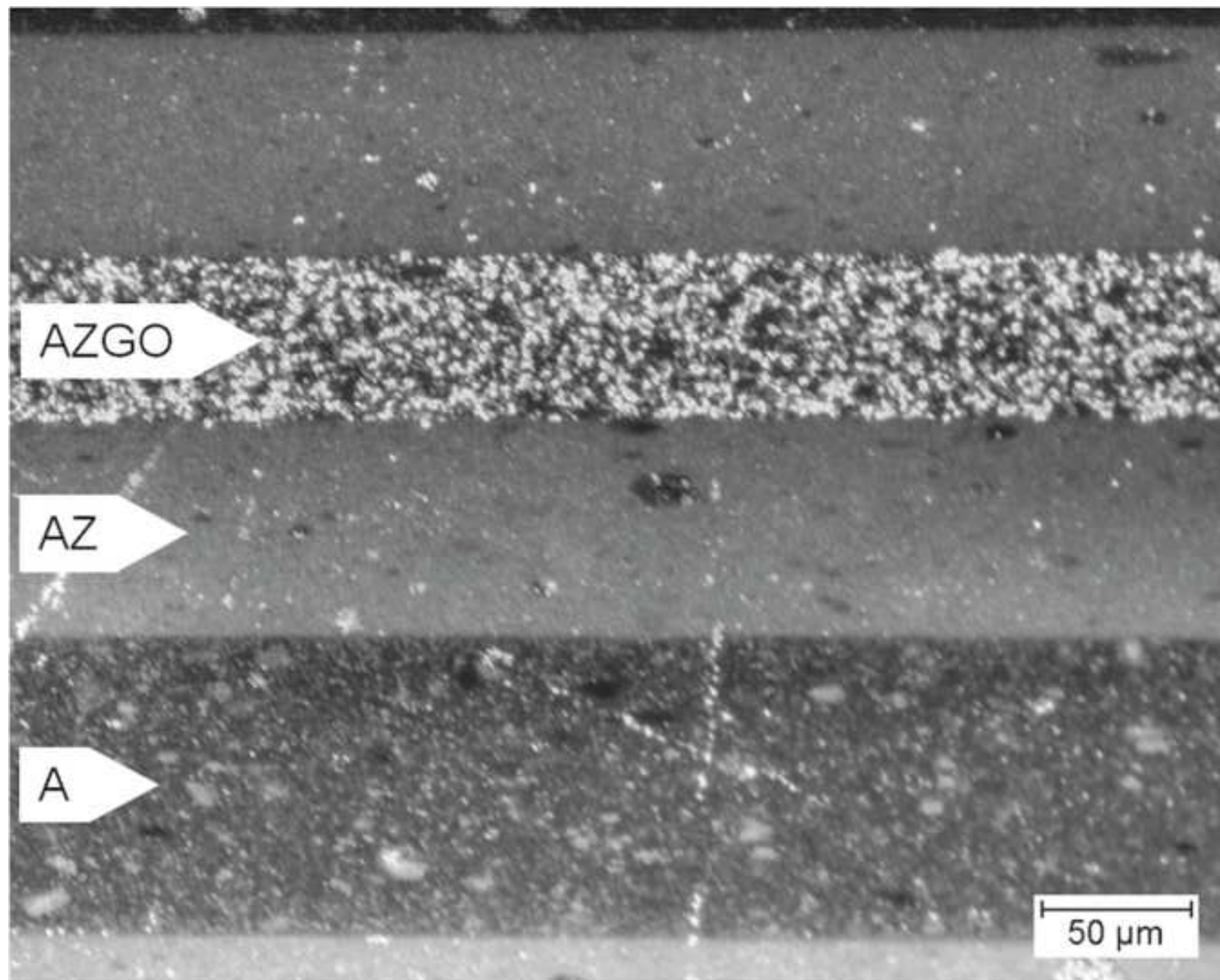
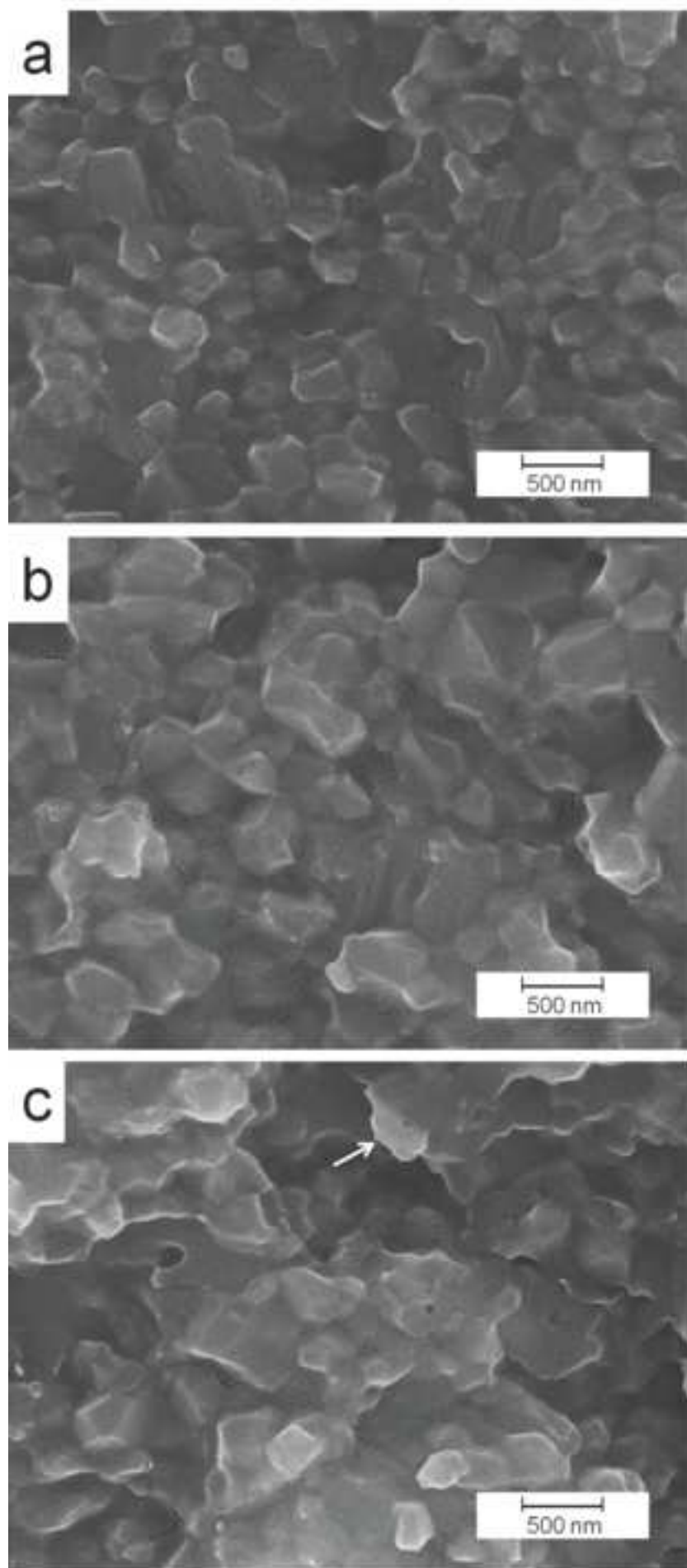


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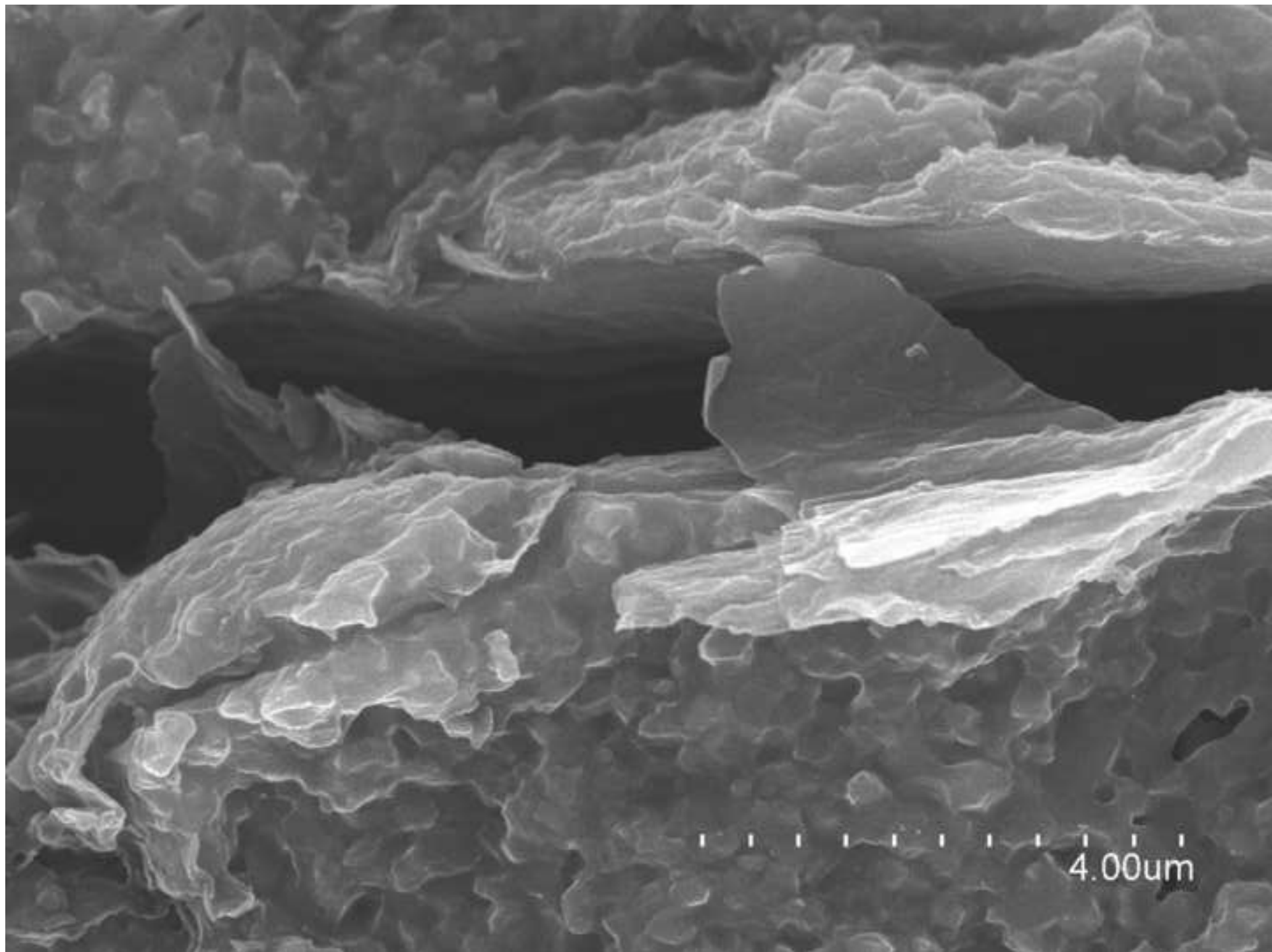


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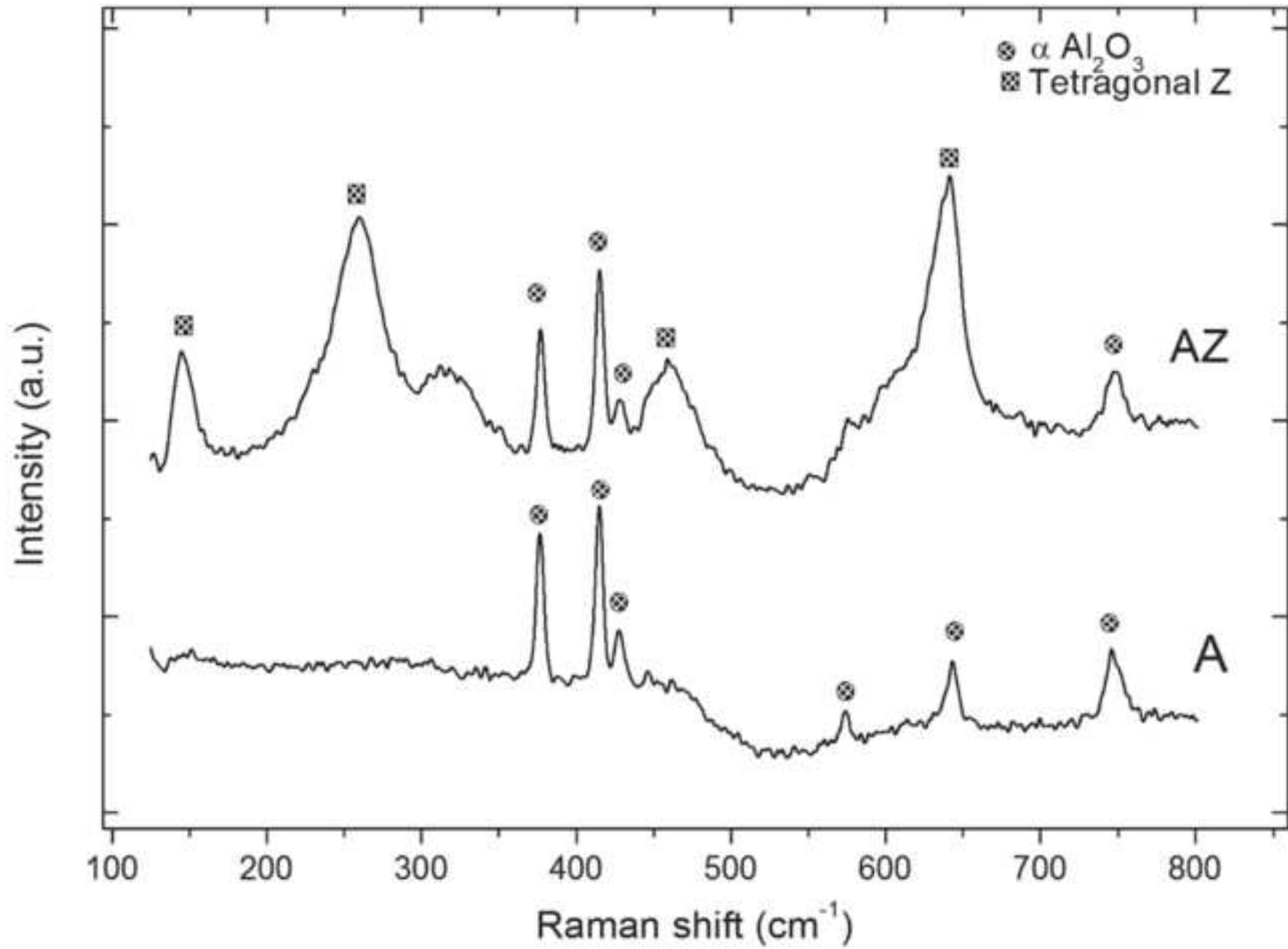


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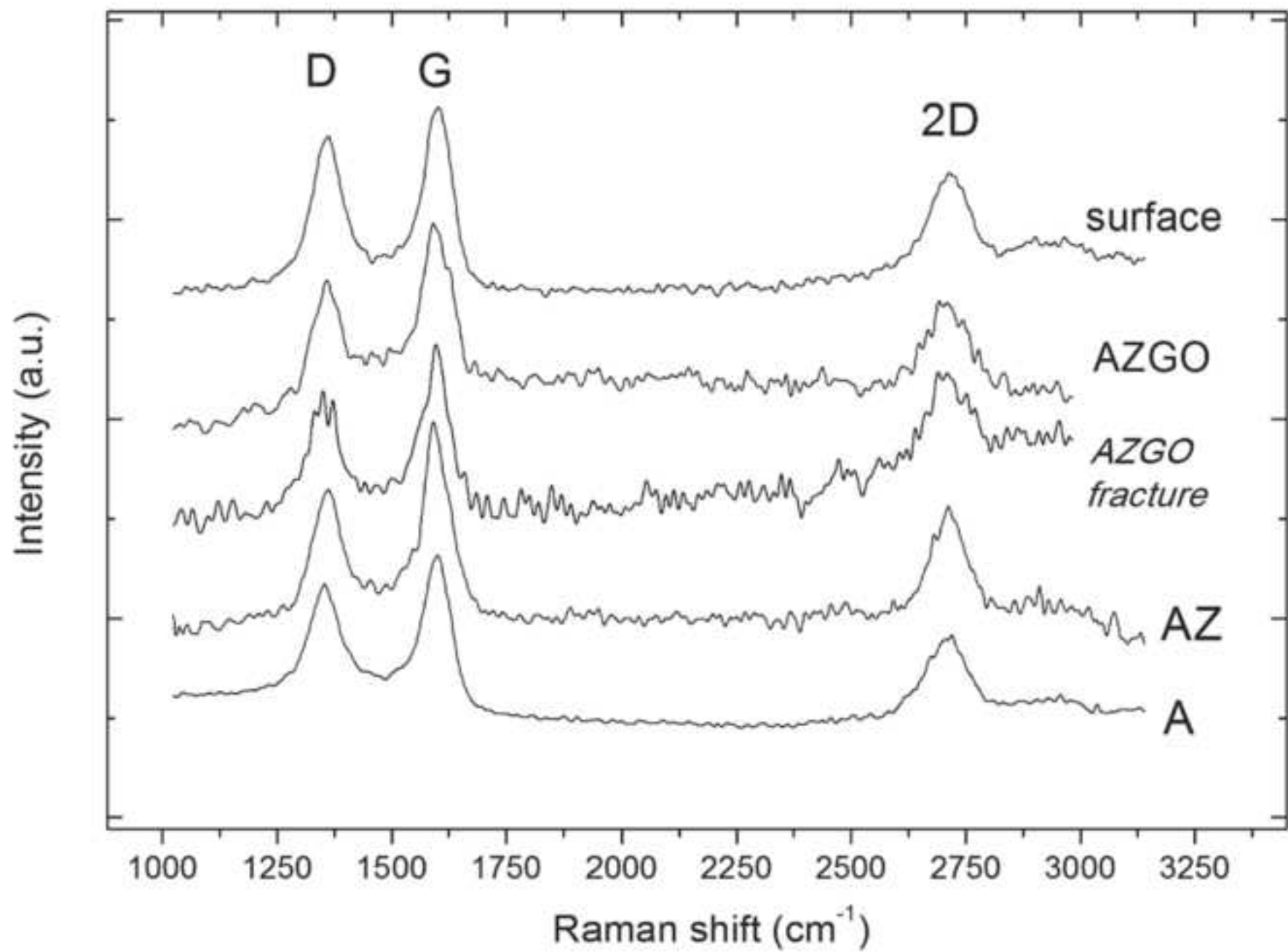


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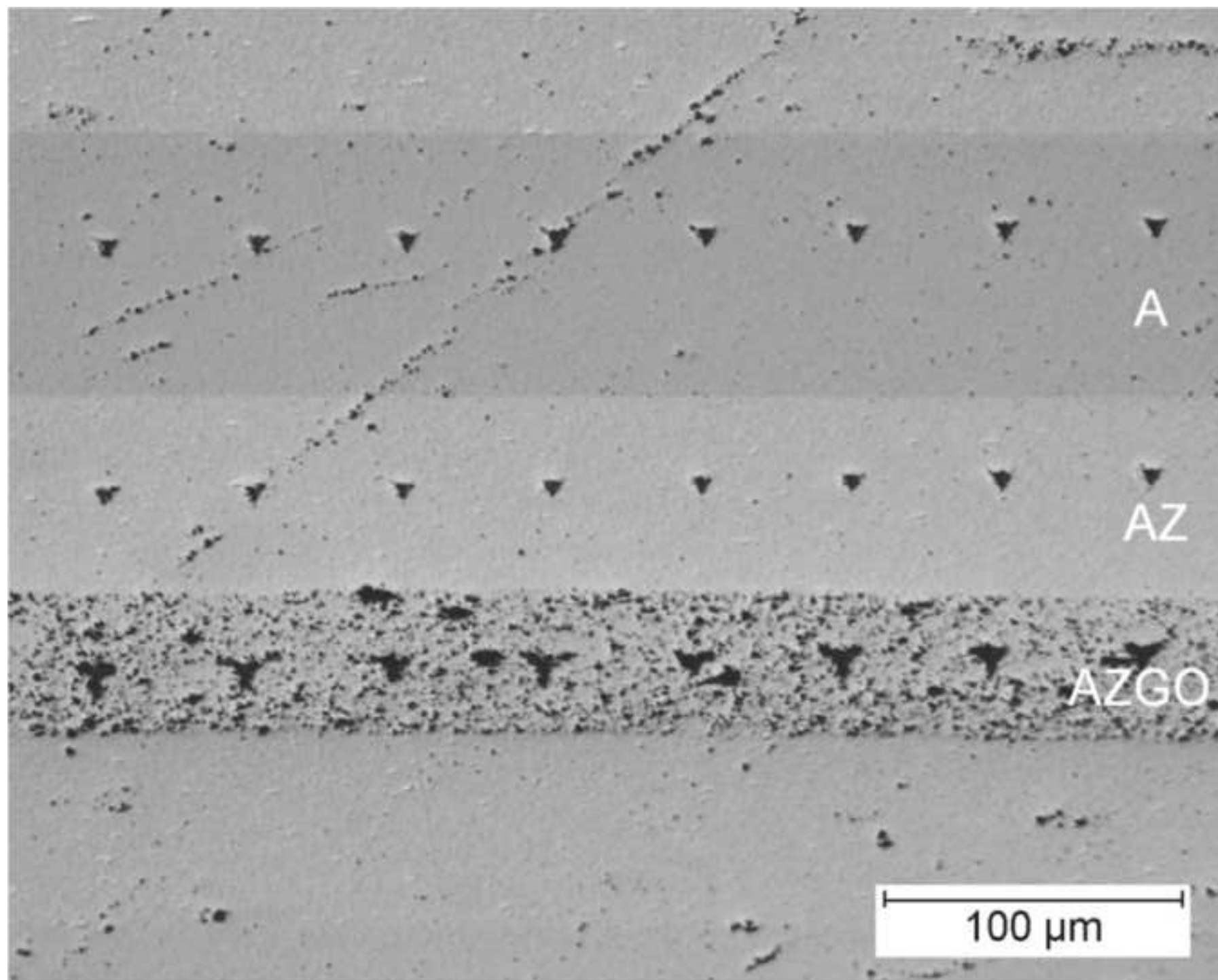


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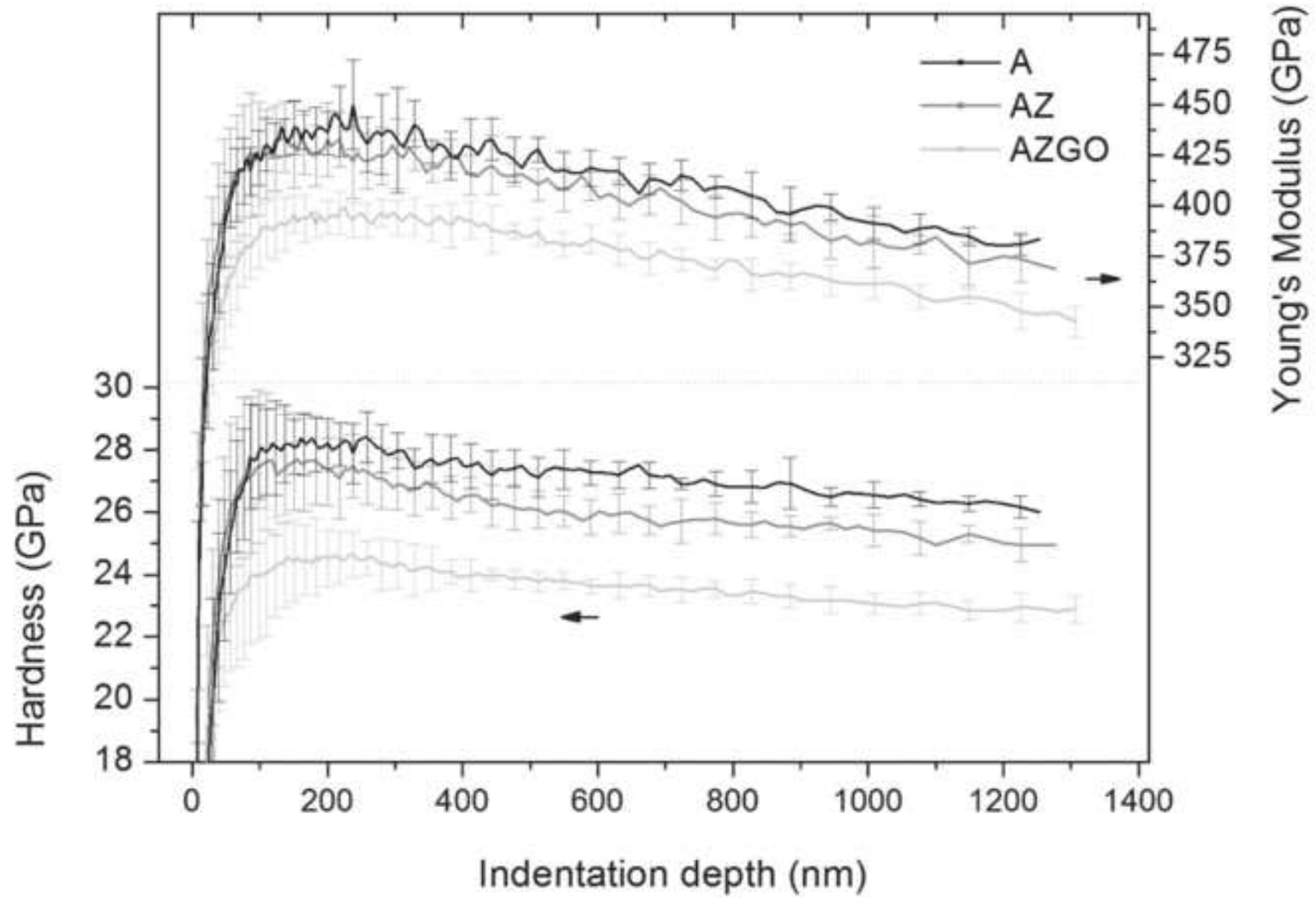
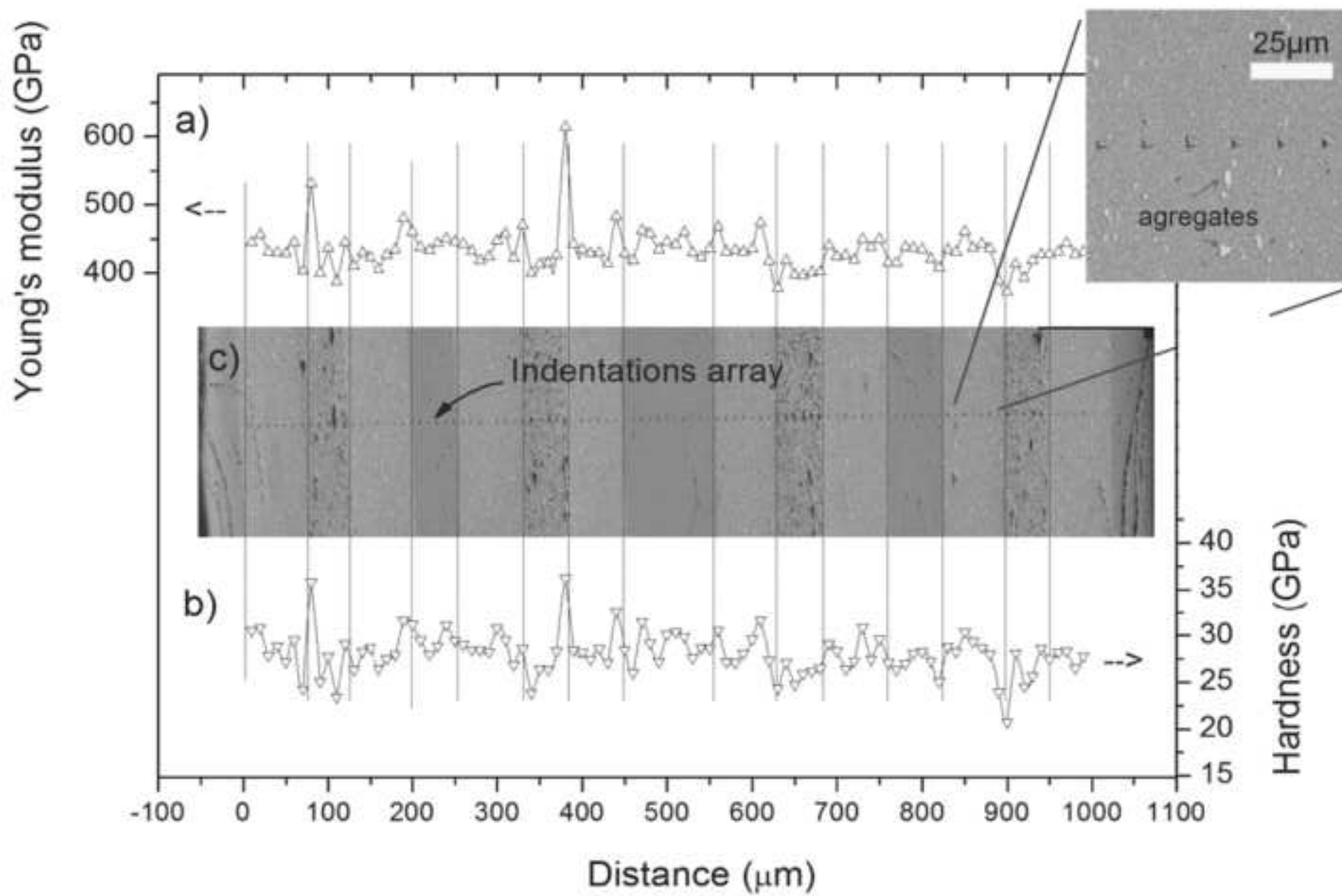


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