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Additional Information

# Fuel Effect on the Liquid-Phase Penetration of an Evaporating Spray Under Transient Diesel-like Conditions

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# Abstract

Measurements of the maximum liquid-phase penetration have been performed injecting five different fuels through a single-hole nozzle in an optical engine under a large set of thermodynamic and injection conditions. The focus of this paper is twofold. First, it pretends to study fuel physical properties on liquid-phase fuel penetration. The choice made on Fischer-Tropsch diesel (FTD) and biodiesel fuels has been highly motivated by their potential to be, at short or middle term, possible substitutes to the conventional diesel fuel. Extensive characterization of fuel physical and chemical properties under ambient conditions are provided and related to the liquid-phase penetration in order to provide an accessible tool to predict liquid spray behavior based on cheap, off-engine measurements. Fischer-Tropsch fuels appeared to be the easiest to vaporize while biodiesel blends were getting always harder to vaporize as the Rapeseed Methyl Ester (RME) rate was increased. The second

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objective of this work is to study the time-response of liquid-phase penetration when subjected to density and temperature variations. Injections of 8 ms at three different pressures have been performed in transient diesellike conditions with density and temperature time derivatives up to 2000  $kg.m^{-3}.s^{-1}$  and 20000  $K.s^{-1}$ . In most cases, the spray appeared to closely follow predictions made from empirical models built out of steady-state air conditions, leading to the conclusion of an instantaneous adjustment of the spray to its environment, validating: (1) the hypothesis made in 1D spray models; (2) the use of empirical models in unsteady-state environment when obtained under steady-state conditions.

*Key words:* diesel engine, biodiesel, spray, Fischer-Tropsch, fuel properties, liquid length, correlation, unsteady conditions.

# 1 1. Introduction

During the past two decades, research on the effect of fuel properties 2 may not have received fervent interest by the automotive industry, perhaps 3 due to the long-standing establishment of conventional diesel and the lack of viable alternative solutions. Although the studies available on the subject 5 represent precious information for the validation of spray modeling hypoth-6 esis [1, 2, 3, 4], most of the research effort has been channeled into new 7 combustion concepts using complex injection strategies and high EGR levels 8 in order to reduce both  $NO_X$  and PM. More recently, worldwide environ-9 mental agencies have been inciting car constructors to find alternatives to 10 the exhaustible fossil fuel for a better sustainability of energy management 11 [5]. In this ambitious framework, biofuels and synthetic fuels represent an 12

interesting perspective, at least at short and middle term, for their capac-13 ity to be directly implanted in the actual car park with no major change 14 of the engine design. Their effect on combustion efficiency and emissions 15 is the result of a complex succession of physical and chemical processes [6]. 16 This study pretends to understand and assess which are the physical mech-17 anisms involved in the introduction of alternative fuels. For this objective, 18 various off-engine measurements have been performed on the five fuels be-19 fore their injection through a 82  $\mu$ m-single-hole nozzle, in an optical engine 20 [7] fed with pure nitrogen. The visualization of their respective maximum 21 liquid-phase penetration has been realized under a large set of operating con-22 ditions, including a sweep of air temperature at constant density, a sweep of 23 air density at constant temperature and three different injection pressures 24 have been performed for each fuel. High-speed imaging of the spray shadow 25 left on a highly lit background has been processed to measure the maximum 26 liquid-phase penetration as defined by Dec and Siebers in [9, 10]. In the first 27 instance, liquid length results and air conditions have been time-averaged as 28 in [11, 12, 13] and discussed. In a second instance, unsteadiness of air den-29 sity and air temperature during the fuel injection has been used as a way to 30 increase the number of experimental data and consequently the reliability of 31 statistics. For each image and so for each instant of the 8 ms injection event, 32 its corresponding air temperature and density were associated. Apart from 33 presenting clear advantages on the statistical point of view [14], these results 34 permitted to conclude on spray reactivity when submitted to variations of 35 ambient density and temperature. 36

#### <sup>37</sup> 2. Experimental Setup

## 38 2.1. Fuels

Five different fuels have been selected for their capacity and their poten-39 tial to be used in a diesel engine with no fundamental redesign of the engine 40 whilst having significant differences in both physical and chemical properties. 41 The first three fuels are widely known in the literature under the generic la-42 bel "first generation biodiesels". Indeed, they are partially or entirely issued 43 from cereal feedstock. RME (Rapeseed Methyl Ester) is a fuel issued from 44 the transesterification reaction between rape oil and methanol. B05 and B30 45 are blends of fossil diesel with respectively 5 and 30 mass percentage of the 46 same RME. 47

48

# [Table 1 about here.]

These three fuels have been previously used by the authors in a multi-hole 49 injector configuration under both reactive and non-reactive environments 50 [12, 13]. Finally, the two last fuels are Fischer-Tropsch fuels issued from gas, 51 coal or biomass liquefaction and will be referred as FT1 and FT2 in the fol-52 lowing study. Various measurements of fuel properties have been performed 53 off-engine. Thermodynamic properties, energetic content and equivalent for-54 mula have been measured following ASTM standards and are summarized 55 in Table 2. Results show that by increasing RME rate in biodiesel fuels, 56 both density and viscosity increase as well, whereas LHV reduces because 57 of the increasing oxygen content. Both Fischer-Tropsch fuels have a lower 58 density compensated by a higher energetic content, which is an important 59 data under a marketing point of view, since the energetic content of one 60

liter is pretty much the same between all these fuels. FT2 is singular by its 61 very low viscosity and its small extra oxygen content. Comparative trends 62 in fluid-mechanics properties were also observed in [15] for a similar selec-63 tion of fuels. Chemical equivalent formulas have been measured using gas 64 chromatography-FID and are also provided in Table 1. They appear to be 65 close to heptadecane  $(C_{17}H_{36})$  and dodecane  $(C_{12}H_{26})$  formulas respectively 66 for FT1 and FT2 while RME's closest pure surrogate could be methyl-oleate 67  $(C_{19}H_{36}O_2)$ . Distillation curves have been measured under the ASTMD86 68 standard. Besides, a weighing scale was measuring the collected mass simul-69 taneously, in order to detect a possible shift between mass and volume recov-70 ery percentage. Results are presented in Figure 1. On one hand, RME and 71 FT2 appear to have relatively flat distillation curves, which is the witness of 72 their homogeneity and their similitude to their corresponding surrogate. On 73 the other hand, B05, B30 and FT1 have similar trends in evaporation under 74 atmospheric pressure, starting from values close to FT2 and ending to values 75 close to RME. Consequently, it can be expected that B05, B30 and FT1's 76 lightest fractions are molecules heavier than FT2 ( $C_{12}H_{25}O_{0.2}$ ) and that their 77 heaviest fractions are close to RME's molecular weight  $(C_{18.95}H_{35.2}O_2)$ . For 78 B05 and B30, their RME content is expected to correspond to this heavy 79 fraction. No significative differences can be observed on the comparison be-80 tween mass and volume percentage recovery. This attests that no important 81 variations of density exist among the proper components of each fuel. 82

- <sup>83</sup> [Figure 1 about here.]
- While the fuel was getting to the temperature of its first boiling point, an important volume expansion has been observed, measured and traduced to

density as a function of temperature, considering mass conservation. Results 86 plotted in Figure 2 show linear trends with high  $R^2$ . Coefficients for a linear 87 regression  $\rho_f = B + A.T_f$  have been summarized in Table 2. ASTM D1298 88 measurements have been added to the plot as well for illustration, but have 89 not been used in the linear regressions for data consistency. A small offset 90 exists between the ASTM measurements and what would be the correspond-91 ing measurement by volume at 289 K. Such volume measurements are not as 92 accurate as the ASTM D1298 but authors believe that the trend is reliable 93 enough to be used as  $\rho_f = \rho_{ASTMD1298} + A.(T_f - 289)$ . It can be observed how 94 these coefficients (A) are all slightly inferior to the value for the US diesel  $\sharp 2$ 95 (0.9) referred by Siebers in [16]. 96

97	[Figure 2 about here.]
98	[Table 2 about here.]

2.2. Hot Spray Test Rig 99

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[Figure 3 about here.]

Tests have operated in a rapid cycling machine described in [7] and illustrated 101 in Figure 3. This facility is based on a modified loop-scavenged single cylin-102 der 2-stroke direct injection diesel engine with three liter displacement and 103 low rated rotational speed (500 rpm). This apparatus makes optical studies 104 on free sprays under inert or reactive diesel-like thermodynamic conditions 105 possible. Intake and exhaust being handled by transfers on the liner, opti-106 cal access to the high-pressure chamber can be easily achieved through the 107 cylinder head which encloses a cylindrical combustion chamber large enough 108

to avoid spray impingement against engine walls. This chamber has an up-109 per port where a single-hole injector equipped with a 82  $\mu m$  conical nozzle 110 is mounted, and four lateral orthogonal accesses. One of theses accesses is 111 used by a pressure transducer whereas the three other ones are equipped 112 with oval-shaped quartz windows, 88 mm long, 37 mm large, and 28 mm 113 thick. Although the use of a single-hole injector may produce faster pressure 114 build-up in the nozzle sac-hole, a faster needle lift and a higher pressure at 115 full needle lift [8], it still presented certain benefits compared to the multi-116 hole one previously used by the authors in the same facility [12]. First, it 117 impeded spray-to-spray interaction (aerodynamic + thermodynamic) and its 118 position relative to the chamber allowed a much larger field for spray de-119 velopment (80 mm vs. 35 mm). Above all, the mass injected was strongly 120 limited despite the performing of relatively long injections, so that no effect 121 on thermodynamic conditions alteration has been detected on the pressure 122 trace. Indeed, in [12], the use of a multi-hole injector with 130  $\mu m$  nozzle 123 hole had led the authors to consider the ambient temperature reduction due 124 to fuel vaporization energetic consumption. The window for time-averaging 125 had to be limited in order to consider steady-state environment. More de-126 tails about the nozzle and injection settings can be found in Figure 4. For 127 this study, the inert configuration has been set by feeding the engine with 128 pure nitrogen so that any reaction due to air oxygen content was avoided. 129 Consequently, outcomes relative to this work concern exclusively the physi-130 cal processes associated to fuel injection, atomization, mixing, heat transfer 131 and vaporization. The rig has been operated under a skip fire mode, i.e. 132 one injection event occurs every 20 engine cycles. This strategy is commonly 133

used to minimize windows fouling and to let the system filter the air andthen avoid air saturation with vaporized fuel.

137 2.3. Operating Conditions

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# [Figure 5 about here.]

The test matrix includes five different engine operating conditions which 139 have been selected in order to realize a sweep of three  $T_{max}$  at constant 140  $\rho_{max}$  (26 kg.m<sup>-3</sup>) and a sweep of three  $\rho_{max}$  at constant  $T_{max}$  (800 K) as 141 shown in Figure 5. The five operating conditions have been labeled NO, 142 LT, HT, LD, HD, standing respectively for NOminal, Low Temperature, 143 High Temperature, Low Density and High Density air setup. The five fuels 144 have been injected at three pressure levels (50, 100 and 150 MPa). The 145 injector was triggered at -16 °ATDC and energized during 8 ms ( $\approx 24CAD$ 146 depending on the instantaneous speed close to the TDC of each operating 147 condition). All information relative the the injector has been summarized 148 in Table 3. Each test has been repeated 10 times leading to a total number 149 of injections equal to 750 for the whole study (5 fuels x 5 OC x 3  $P_{inj}$  x 150 10 in j.). To determine the exact intake air condition required by the test 151 plan, an accurate characterization of the engine has been performed over 152 35 points covering its full range of operating conditions. Thermodynamic 153 conditions have been calculated from the cylinder pressure using a first-law 154 thermodynamic analysis considering blow-by, heat transfer and mechanical 155 stress. By a succession of interpolations, the exact air intake conditions for 156 the test plan are then calculated. A double-check is performed by setting 157

the resulting values to the engine and the reiteration of the same first-law analysis. Results of the engine characterization can be found in Table 3 and intake conditions to carry out the test plan are indicated in Figure 5. The resulting temperature and densities in the close to TDC region are plotted in Figure 6.

[Table 3 about here.]

<sup>163</sup> [Figure 6 about here.]

<sup>165</sup> 2.4. Optical Setup and Image Processing

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Diffused back-light images have been taken at  $8000 \ fps$ . Illumination was 166 provided by two 150 W quartz-halogen illuminators (Dolan-Jenner PL800), 167 supplied by 8 mm optic fiber bundles positioned at 60 mm from the diffuser 168 dispensing an illumination of 330  $W.m^{-2}$ . The optical setup is represented in 169 Figure 3. Exposure time of the high-speed CMOS camera (Photron Fastcam-170 Ultima APX) has been limited to 25  $\mu s$ . Imaging has been kept to this 171 relatively low speed in order to keep a reasonable spatial resolution of 8.9 172 pixels/mm. Camera bit depth of 10 bits allowed a good discretization of 173 digital levels for subsequent image segmentation. The camera was triggered 174 by a TTL signal synchronized with the injector start of energizing (SOE). 175 Each injection event was documented by 100 pictures, accommodating a 12.5 176 ms acquisition time from the SOE. 177

Images of the spray have been processed with a purpose-made C++ code described in [12, 17, 18]. Figure 8 shows two of these processing steps. After a background subtraction (a), a threshold is calculated based on a statistical analysis of each image background [12] and used for image segmentation. <sup>182</sup> Connectivity to the spray center of mass removes any imperfection left on <sup>183</sup> the segmented image. The distance between the injector tip and the front <sup>184</sup> part of the detected boundary is considered to be the maximum liquid-phase <sup>185</sup> penetration (b).

#### 186 3. Analysis Methodology

As commented in the introduction, data have been processed in two different ways to assess physical processes associated to engine operation and fuel physical properties. After a short theoretical review, the approach of the statistical analysis and its relation to the experiment will be presented.

#### <sup>192</sup> 3.1. Theoretical background

191

The computational cost of CFD motivated investigation for the under-193 standing and the assessment of the phenomena occurring in a diesel spray to 194 simplify the calculation of spray flow-field development. Thus, different 1D-195 models have been proposed [16, 19, 23] based on mixing-limited vaporization 196 control, in which hypothesis made are the following: - The spray reaches the 197 complete atomization regime very near the nozzle exit. - Local transfer rates 198 of momentum, mass and energy between liquid droplets and surrounding air 199 are fast in comparison to the rate of development of the flow field as a whole. 200 This means that an a priori complicated two-phase problem is treated from 201 the point of view of a single-phase flow where a fraction of fuel vaporizes 202 instantaneously once there is enough enthalpy in the surrounding gas to heat 203 it up and vaporize it. The appropriate mixture fraction where this energy 204

<sup>205</sup> balance is achieved is called  $Y_{f,evap}$ . Consequently, the liquid length, con-<sup>206</sup> sidered as the maximum liquid-phase penetration, could be defined as the <sup>207</sup> position on the spray axis where this specific  $Y_{f,evap}$  is reached. Following <sup>208</sup> this hypothesis, a scaling law for liquid length has been derived [24] based <sup>209</sup> on turbulent spray mixing considerations. The axial mass fraction within the <sup>210</sup> quasi-steady part of a diesel spray could be obtained from:

$$Y_f = K \cdot d_0 \sqrt{\frac{\rho_f}{\rho_{air}}} \cdot \frac{1}{X} \tag{1}$$

where K states for a spray constant,  $d_0$  is nozzle diameter,  $\rho_f$  and  $\rho_{air}$  fuel and ambient density and X is spray axial coordinate. Thus, liquid length is defined by:

$$LL = K \cdot \left[ d_0 \sqrt{\frac{\rho_f}{\rho_{air}}} \right] \cdot \frac{1}{Y_{f,evap}} \tag{2}$$

In Eqn. (2), the term in brackets is widely known in the literature as the equivalent diameter and is related to spray mixing scales (i.e momentum) while the last one, as stated before, is an energy term which takes into account vaporization processes. This last term could be written as in Eqn. (3), where  $T_{air}$  is ambient gas temperature,  $T_{f,0}$  is the initial fuel temperature and  $T_{evap}$  is the saturation temperature when the fuel is fully vaporized.

$$\frac{1}{Y_{f,evap}} = 1 + \frac{\Delta h_f(T_{evap}, T_{f,0})}{\Delta h_{air}(T_{air}, T_{evap})}$$
(3)

This parameter shows a complex dependence on both fuel properties and ambient conditions [16, 24] such as air temperature, fuel specific and latent heat, and fuel initial temperature.

#### 223 3.2. Statistical analysis

These theoretical considerations have been applied in a statistical study in order to analyse experimental results and check hypotheses reliability. This study aims at relating liquid length with operating conditions and fuel characteristics. The following model for the dependence of liquid length has been proposed:

$$LL \propto D^a_{noz} \cdot T^b_{air} \cdot P^c_{inj} \cdot \rho^d_{air} \cdot \rho^e_f \cdot \nu^f_f \cdot T^g_{10\%} \cdot T^h_{50\%} \cdot T^i_{95\%}$$
(4)

The classical correlations for liquid length in diesel sprays have been com-229 pleted with some factors particular to the fuel so that fuel fluid-mechanical 230 and vaporizing properties are accounted. Coefficients b, c, d from Eqn. (4) 231 have been previously evaluated independently for each fuel under both steady 232 and unsteady conditions. Nozzle diameter effect has not been studied so 233  $D^a_{noz}$  and will be consequently part of the constant factor. Injection pressure 234 exponent has been kept free, despite injection velocity (and thus injection 235 pressure) has theoretically no influence on liquid length. 236

#### 237 3.3. Steady-State Conditions Approach

The assumption of steady-state conditions has already been made by the authors in previous studies [12, 13] and so liquid length was considered to be constant around engine TDC and resolve exponents from Eqn. (4) in terms of average values. A window for time-averaging is selected on the stabilized liquid-length region. The engine first-law thermodynamic analysis showed that the engine reaches  $T_{max}$  between -2.8 and -3.1 °ATDC (4500 and 4625  $\mu s$  ASOE) and  $\rho_{max}$  between -0.1 and -0.5 °ATDC (5500 and <sup>245</sup> 5625  $\mu s \ ASOE$ ), depending on the engine operating conditions. Therefore, <sup>246</sup> time-averaging window has been limited between 3500 and 6500  $\mu s \ ASOE$ . <sup>247</sup> Figure 8 shows a plot of the ensemble average and its standard deviation. The <sup>248</sup> section used for time-averaging has been highlighted and the result plotted <sup>249</sup> in dashed line. Images from one of the ten corresponding sequences have <sup>250</sup> been added for illustration. Only one image out of two has been displayed <sup>251</sup> to simplify the figure.

#### [Figure 8 about here.]

Only the most relevant results of this analysis have been plotted in the Results and Discussion section but the whole set of numerical results is provided in an appendix table.

## 256 3.4. In-Cylinder Unsteady Conditions Approach

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In order to check if both empirical models based on results obtained un-257 der steady-state conditions and spray models based on a succession of quasi-258 steady evaporating states [16, 19] are extendable to real engine conditions, 259 most of the image sequence has been exploited by attributing to each image 260 of the spray its corresponding couple of  $T_{air}$  and  $\rho_{air}$  and resolve Eqn. (4) in 261 terms of time-resolved values. As commented in the experimental apparatus 262 description, the spray is exposed to important pressure variations. On Fig-263 ure 5, it can be observed how  $T_{air}$  fluctuates over more than 50 K and so does 264  $\rho_{air}$  by up to 7 kg.m<sup>-3</sup> during the injection event ( $\approx 24 \ CAD$ ). This is due 265 to the relative long injection timing (8 ms) compared to engine speed (500) 266 rpm). Figure 8 shows how the in-cylinder pressure leaves its mark on the 267 ensemble-averaged liquid length. Temperature and density time-derivatives 268

have been plotted in Figure 9. It is worthy to note that despite the temporal 269 variations seem to be small, they are of the order of expected variations in a 270 heavy-duty engine at 1200 rpm in the injection region for HCCI combustion 271 mode and in the close-to-TDC region for a conventional combustion mode. 272 For this analysis, the time window used for analysis had also to be restricted 273 to avoid the consideration of SOI and EOI penetration transients. As an 274 example, the case exposed in Figure 8, has been restricted between 1375 275 and 8875  $\mu s$  ASOE. The liquid length results have been reprocessed using 276 the same statistical method described above in order to assess the effect of 277 air temperature and air density. From a statistical point of view, such kind 278 of study is very interesting since it multiplies the combinations of  $T_{air}$  and 279  $\rho_{air}$ . Moreover, blow-by, heat transfer and mechanical stresses induce a delay 280 between both traces and reduce collinearity between both variables. 281

[Figure 9 about here.]

#### 283 4. Results and Discussion

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284

[Figure 10 about here.]

#### 285 4.1. Steady-State Conditions

The liquid length at different injection pressures has been plotted for the five studied fuels in Figure 10. Significant differences can be observed from one fuel to another given the reduction by more than a factor of two between RME and FT2 liquid lengths. Both of these fuels constructed the upper and lower boundaries of the tested fuels, respectively. Figure 10 shows similar trends regarding two fuels encasing the others by upper and

lower boundaries as in Figure 1, which illustrates the high influence of fuel 292 volatility. Such result was then expected since the association between liq-293 uid length and distillation curves is already widely assumed in the literature 294 [3, 10, 4]. The last works available on the subject still use this measurement 295 to explain both the shorter FTD liquid length [20] and the higher biodiesel 296 liquid length [21, 22] respective to the conventional diesel. A slight decrease 297 of the liquid length can be observed among all the fuels when injection pres-298 sure is increased. However, this effect is small enough to consider this result 299 in agreement with the "mixing-controlled" assumption. Although only the 300 NO-condition is represented, the same trends have been observed for the 301 four other operating points. Since it has just been confirmed that injection 302 pressure had no considerable effect on liquid-phase penetration, the effects 303 of air temperature and air density have been represented only for the 150 304 MPa injection pressure case in Figures 11 and 12. Again, the fuel hierarchy 305 is conserved and is quite consistent with the distillation curves at ambient 306 pressure. For all fuels, an increase on both air parameters leads to a reduc-307 tion of the liquid length. Likewise, the effect of  $T_{air}$  appears to be extremely 308 significant. Indeed, a 13% increase of air temperature affects up to a 43%300 decrease on the liquid length, while a 36% increase of air density only de-310 creases the liquid length by up to a 25%. It must be highlighted that the 311 100 K variation applied in this study is far from covering the whole range 312 of temperatures encountered in a diesel engine. Consequently, in early and 313 late injection strategies, where the ambient temperature is expected to be 314 even lower, the resulting liquid length, enhanced by the lower density as 315 well, could lead to an important liner-impingement if care is not taken dur-316

ing the hardware design. The purpose of the following section is precisely to
assess the weight of these parameters by means of the previously described
statistical analysis.

[Figure 12 about here.]

<sup>320</sup> [Figure 11 about here.]

321

4.2. Statistical regression for engine-depending physical processes assessment 322 In a first instance, the statistical analysis has been applied to each fuel 323 independently, introducing only the parameters which change with the oper-324 ating settings of the engine. In this way it is pretended to check if all fuels 325 have the same sensitivity to engine parameters.  $T_{air}$ ,  $\rho_{air}$  and  $P_{inj}$  effect 326 have been assessed and are presented in Table 5. Both temperature high 327 impact and injection pressure irrelevance are confirmed while air density ef-328 fect seems to be a bit higher than proposed by the scaling law. Moreover, 329 from one fuel to the other, slight differences are appreciated, indicating a 330 difference on fuel response to engine thermodynamic settings. Indeed, RME331 seems to be more gently affected by air conditions, way above the rest. If 332 results from steady and unsteady-state are now considered for comparison, it 333 can be observed that, exception made for RME, the resulting exponents are 334 remarkably close. It may be necessary to remind here that the "steady-state" 335 exponents have been obtained using a set of averaged data coming from a 336 sweep of three air density values at constant air temperature and from a 337 sweep of three air temperature values at constant air density, both fueled at 338 three injection pressures levels (15 values/fuel), while "unsteady state" con-339 siders air density and temperature values during the entire injection event 340

for both sweeps ( $\approx 900$  values/fuel). This parallelism in the results shows 341 how a spray under unsteady conditions behaves as a succession of sprays ob-342 tained under steady-state conditions, meaning that there is no delay in the 343 spray adjustment to its environment under the range of pressure derivatives 344 studied. This result is in agreement with recent studies [25] and validates 345 the use of theoretical 1D spray models [16, 19, 23] in unsteady conditions 346 as well as empirical models based on liquid length measurements obtained 347 in a steady-state environment. Such conclusions are supported by the high 348 correlations reliability that has been evaluated through the R-squared pa-349 rameter which is, apart for RME, consistent between steady and unsteady 350 state conditions. 351

[Table 4 about here.]

352

The differences observed on exponents for RME as well as the decay 353 observed on  $\mathbb{R}^2$  show that this fuel may not follow the same conclusions 354 depicted above and that the characteristic time of vaporization for a droplet 355 of such a dense, viscous and low volatility fuel may be significant compared 356 to the spray flow field development. In [21], Fisher *et al.* performed a similar 357 analysis as in [25] but using two biodiesel fuels. They also observed that 358 biodiesel liquid length is not directly related to instantaneous in-cylinder 359 temperature and density, and suggest that biodiesel may be subject to the 360 thermodynamic history. An attempt has been made to quantify the biodiesel 361 time-response. However the quality of the result showed to be highly affected 362 by our relatively low camera frequency. Yet, no clear trends were found when 363 this delay was correlated either with engine parameters or with the proper 364

liquid length. Thus, both data and correlations were not robust enough to 365 be presented in this manuscript and more investigation on the subject will be 366 needed. Finally, liquid length results from all the fuels have been introduced 367 to the statistical analysis simultaneously. As expected, if no dramatic effect 368 can be observed on exponents' values, the very low  $R^2$  shows that physical 369 parameters issued from the engine setup are not sufficient to predict liquid 370 length and that it is necessary to introduce fuel physical properties to achieve 371 a better prediction. 372

## 373 4.3. Statistical regression for fuel physics assessment

The same statistical tool has been applied, introducing data from the 374 measured fuel physical properties exposed in the upper corresponding sec-375 tion. They have been separated in 2 parts: fluid-mechanics and evaporative 376 properties. Fluid-mechanics properties are represented by density and vis-377 cosity while evaporative properties, in absence of specific and latent heat, 378 are represented by  $T_{10\%}, T_{50\%}$  and  $T_{95\%}$  from distillation curves. Indeed, the 379 purpose of the resulting correlations is to provide a tool that predicts liquid 380 length out of cheap, off-engine measurements. A set of selected correlations 381 are presented in Table 5 by using only some of the terms in Eqn. (4). In 382 order to compare correlations with a different number of parameters, reliabil-383 ity has been calculated using specific R-squared (  $R^2_{spe}). \ {\rm As}$  in the previous 384 section, no significative differences have been observed between steady and 385 unsteady-state considerations and therefore, only unsteady-state conditions 386 are reported in Table 6. First, physical properties issued from the engine 387 operation and fuel physical properties have been compared in correlations 388 (1) and (2). It appears that fuel properties are more important than physi-389

cal in the prediction of liquid length. However negative coefficients for  $T_{10\%}$ 390 and  $T_{95\%}$  are not physically reasonable. It is important then to identify, 391 among the five physical parameters, which are controlling the process. In 392 correlations (3) to (7), each fuel parameter has been associated to one phys-393 ical parameters issue from the engine. Fuel density seems to be the best 394 parameter for liquid length prediction, while no significant differences can 395 be observed separating the 3 distillation curve temperatures. However the 396 low  $R^2$  for  $T_{95\%}$  is unacceptable. In correlations (8) and (9), the fuel fluid 397 mechanics properties and fuel evaporative properties are respectively asso-398 ciated to engine physical properties. The result is that they are both good 399 groups of variables for empirical modelling, although, again, the negative 400 exponents for  $T_{10\%}$  and  $T_{95\%}$  are a physical non-sense. Finally, correlation 401 (10) shows the association of both fluid-mechanics and evaporative proper-402 ties using the most essential and reliable parameters. Correlation (11) has 403 been added to show the maximum reliability these parameters are capable 404 of, for comparison with upper correlations. 405

406

## 407 5. Summary and Conclusions

Measurements of the maximum liquid-phase penetration have been performed using five fuels with an interesting potential for diesel substitution, in an optical engine under a large set of thermodynamic and injection conditions. These measurements have been related to fuel properties measurements performed off-engine and to pressure variations similar to those found

[Table 5 about here.]

in a heavy-duty diesel engine, in order to assess the physical processes controlling the vaporization of a spray under such conditions. Relevant conclusions
are the following:

A database of fuel properties and time-averaged liquid-length results
 are provided for confrontation with modeling results (Cf. Appendix).

- 4182. Under all tested conditions, Fischer-Tropsch fuels showed to have a419shorter liquid length than biodiesel fuels, for which the liquid length420was increased as the RME percentage was increased as well. The fuel421hierarchy for liquid length was the following: FT2 < FT1 < B05 <422B30 < RME. This trend was maintained for all engine settings.
- 3. The qualitative effects of  $T_{air}$ ,  $\rho_{air}$  and  $P_{inj}$  already available in the literature for diesel fuel have been confirmed and could be extended to biodiesel and Fischer-Tropsch fuels.
- 426 4. A new method, based on time consideration, has been proposed for the
  427 processing liquid length high speed imaging. It permitted to multiply
  428 the number of samples for a more robust statistical analisis.
- 5. For 4 out of the 5 tested fuels, the comparison between two statistical approaches showed that the spray liquid-phase adjust instantaneously to the in-cylinder conditions. Such results confirms the hypothesis made by 1D spray models and allows the use of empirical models obtained under steady-state environment in unsteady conditions (with time-derivatives up to 20000  $K.s^{-1}$  and 2000  $kg.m^{-3}.s^{-1}$ ).
- 6. Fuel physical properties have been assessed against the physical properties resulting from engine operating conditions and traduced into correlations for empirical modeling.

<sup>438</sup> 7. A correlation based on low cost off-engine measurements is proposed <sup>439</sup> taking into account engine parameters, fuel fluid-mecanics properties <sup>440</sup> and evaporation properties:  $LL = T_{air}^{-2.63} \cdot P_{inj}^{-0.06} \cdot \rho_{air}^{-0.60} \cdot \rho_{f}^{4.39} \cdot T_{50\%}^{0.54}$ 

# 441 Acknowledgment

The authors wish to acknowledge the Spanish Ministry of Education and Science for the financial support through the OPTICOMB project (TRA2007-67961-C03-01). The authors would also like to thank Daniel Lerida for the management of the facility and his assistance in data acquisition.

Fuel	$T_{air}$	$ ho_{air}$	$P_{inj}$	$\Delta P$	$ ho_f$	$ u_f$	$T_{10\%}$	$T_{50\%}$	$T_{95\%}$	LL
B05	798.0	29.7	50	43.2	833	2.50	205	293	356	19.34
	798.0	29.7	100	93.2	833	2.50	205	293	356	18.42
	798.0	29.7	150	143.2	833	2.50	205	293	356	16.87
	845.2	25.8	50	43.7	833	2.50	205	293	356	17.83
	845.2	25.8	100	93.7	833	2.50	205	293	356	16.55
	845.2	25.8	150	143.7	833	2.50	205	293	356	15.84
	795.4	21.7	50	45.1	833	2.50	205	293	356	24.01
	795.4	21.7	100	95.1	833	2.50	205	293	356	22.68
	795.4	21.7	150	145.1	833	2.50	205	293	356	21.42
	747.5	25.9	50	44.5	833	2.50	205	293	356	25.64
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#### 447 Appendix

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Fuel	$T_{air}$	$ ho_{air}$	$P_{inj}$	$\Delta P$	$ ho_f$	$ u_f$	$T_{10\%}$	$T_{50\%}$	$T_{95\%}$	LL
	747.5	25.9	100	94.5	833	2.50	205	293	356	24.87
	747.5	25.9	150	144.5	833	2.50	205	293	356	22.90
	796.8	25.8	50	44.1	833	2.50	205	293	356	22.05
	796.8	25.8	100	94.1	833	2.50	205	293	356	20.17
B05	796.8	25.8	150	144.1	833	2.50	205	293	356	19.07
B30	798.0	29.7	50	43.2	849	3.10	223	304	347	24.53
	798.0	29.7	100	93.2	849	3.10	223	304	347	26.15
	798.0	29.7	150	143.2	849	3.10	223	304	347	24.28
	845.2	25.8	50	43.7	849	3.10	223	304	347	24.16
	845.2	25.8	100	93.7	849	3.10	223	304	347	23.07
	845.2	25.8	150	143.7	849	3.10	223	304	347	20.48
	795.4	21.7	50	45.1	849	3.10	223	304	347	30.58
	795.4	21.7	100	95.1	849	3.10	223	304	347	31.86
	795.4	21.7	150	145.1	849	3.10	223	304	347	31.17
	747.5	25.9	50	44.5	849	3.10	223	304	347	30.93
	747.5	25.9	100	94.5	849	3.10	223	304	347	32.08
	747.5	25.9	150	144.5	849	3.10	223	304	347	33.41
	796.8	25.8	50	44.1	849	3.10	223	304	347	28.49
	796.8	25.8	100	94.1	849	3.10	223	304	347	27.80
B30	796.8	25.8	150	144.1	849	3.10	223	304	347	26.03
RME	798.0	29.7	50	43.2	878	4.41	321	334	345	31.67
	798.0	29.7	100	93.2	878	4.41	321	334	345	29.57
Continued on next page										

Appendix – continued from previous page

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Fuel	$T_{air}$	$\rho_{air}$	$P_{inj}$	$\Delta P$	$ ho_f$	$ u_f$	$T_{10\%}$	$T_{50\%}$	$T_{95\%}$	LL
	798.0	29.7	150	143.2	878	4.41	321	334	345	30.20
	845.2	25.8	50	43.7	878	4.41	321	334	345	27.80
	845.2	25.8	100	93.7	878	4.41	321	334	345	27.57
	845.2	25.8	150	143.7	878	4.41	321	334	345	25.88
	795.4	21.7	50	45.1	878	4.41	321	334	345	39.14
	795.4	21.7	100	95.1	878	4.41	321	334	345	40.63
	795.4	21.7	150	145.1	878	4.41	321	334	345	39.36
	747.5	25.9	50	44.5	878	4.41	321	334	345	38.71
	747.5	25.9	100	94.5	878	4.41	321	334	345	45.85
	747.5	25.9	150	144.5	878	4.41	321	334	345	45.69
	796.8	25.8	50	44.1	878	4.41	321	334	345	36.73
	796.8	25.8	100	94.1	878	4.41	321	334	345	35.50
RME	796.8	25.8	150	144.1	878	4.41	321	334	345	34.03
$\mathbf{FT1}$	798.0	29.7	50	43.2	784	3.44	250	297	352	18.26
	798.0	29.7	100	93.2	784	3.44	250	297	352	17.33
	798.0	29.7	150	143.2	784	3.44	250	297	352	16.12
	845.2	25.8	50	43.7	784	3.44	250	297	352	17.07
	845.2	25.8	100	93.7	784	3.44	250	297	352	15.90
	845.2	25.8	150	143.7	784	3.44	250	297	352	15.72
	795.4	21.7	50	45.1	784	3.44	250	297	352	22.31
	795.4	21.7	100	95.1	784	3.44	250	297	352	20.79
	795.4	21.7	150	145.1	784	3.44	250	297	352	20.08
	Continued on next page									

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Fuel	$T_{air}$	$\rho_{air}$	$P_{inj}$	$\Delta P$	$ ho_f$	$ u_f$	$T_{10\%}$	$T_{50\%}$	$T_{95\%}$	LL
	747.5	25.9	50	44.5	784	3.44	250	297	352	23.28
	747.5	25.9	100	94.5	784	3.44	250	297	352	22.22
	747.5	25.9	150	144.5	784	3.44	250	297	352	20.92
	796.8	25.8	50	44.1	784	3.44	250	297	352	19.53
	796.8	25.8	100	94.1	784	3.44	250	297	352	18.48
FT1	796.8	25.8	150	144.1	784	3.44	250	297	352	17.54
FT2	798.0	29.7	50	43.2	773	1.29	177	200	242	13.53
	798.0	29.7	100	93.2	773	1.29	177	200	242	13.09
	798.0	29.7	150	143.2	773	1.29	177	200	242	12.15
	845.2	25.8	50	43.7	773	1.29	177	200	242	13.08
	845.2	25.8	100	93.7	773	1.29	177	200	242	12.36
	845.2	25.8	150	143.7	773	1.29	177	200	242	11.70
	795.4	21.7	50	45.1	773	1.29	177	200	242	16.69
	795.4	21.7	100	95.1	773	1.29	177	200	242	15.92
	795.4	21.7	150	145.1	773	1.29	177	200	242	15.58
	747.5	25.9	50	44.5	773	1.29	177	200	242	18.75
	747.5	25.9	100	94.5	773	1.29	177	200	242	16.92
	747.5	25.9	150	144.5	773	1.29	177	200	242	15.94
	796.8	25.8	50	44.1	773	1.29	177	200	242	15.21
	796.8	25.8	100	94.1	773	1.29	177	200	242	13.77
FT2	796.8	25.8	150	144.1	773	1.29	177	200	242	13.37

Appendix – continued from previous page

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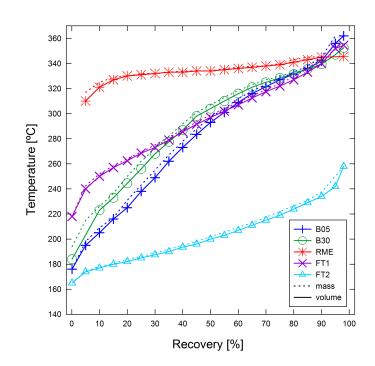


Figure 1: Distillation curves obtained by  $ASTM\ D86.$ 

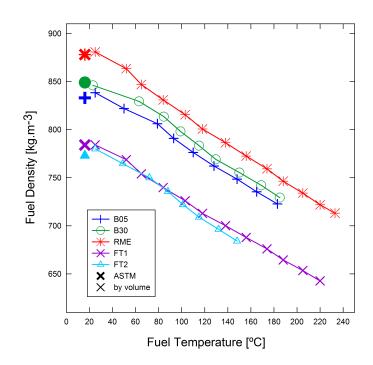


Figure 2: Temperature effect on fuel density under atmospheric pressure.

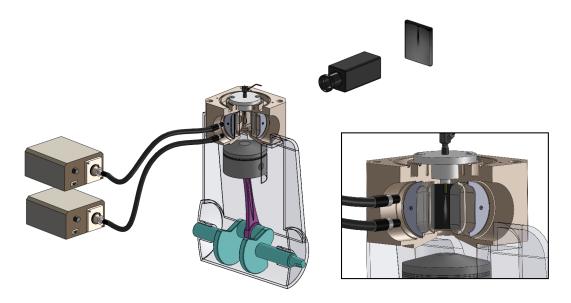


Figure 3: Hot spray test rig and diffuse back-lightening optical setup.

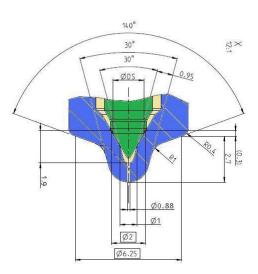


Figure 4: Cutaway view of the injector tip.

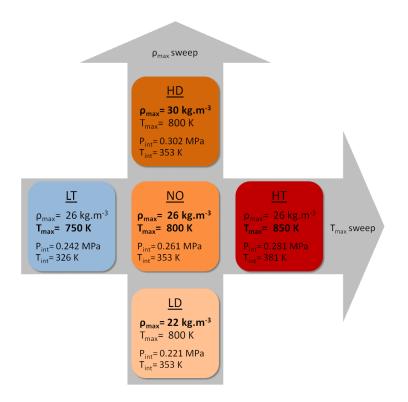


Figure 5: Schematic representation of the engine operating conditions.

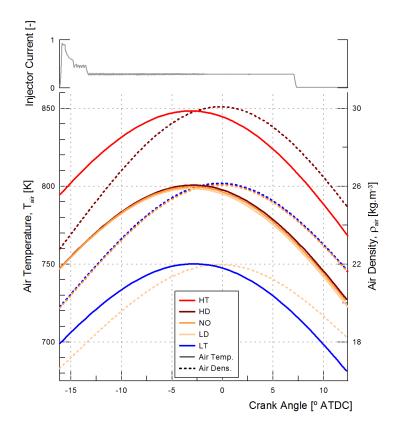


Figure 6: Results of in-cylinder first-law thermodynamic analysis for temperature and density calculation in the TDC region. 8 ms energizing time is represented by the injector current.

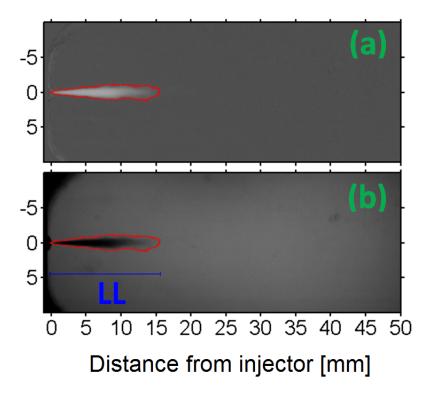


Figure 7: Intermediate processing images from FT2 at BT and  $P_{inj}=100 MPa$ . (a) Resulting image from original image subtraction to the background. (b) Overlay of the boundary resulting from the complete processing to the original image.

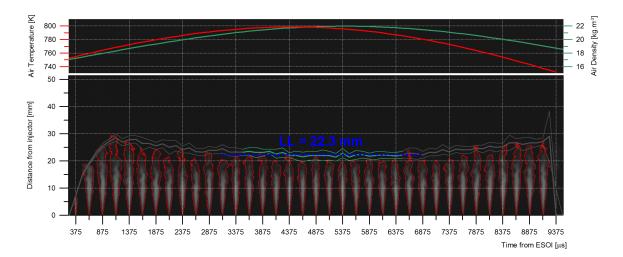


Figure 8: Representation of the cycle-to-cycle averaging and standard deviation (from 10 repetitions) for FT1, Low Density (22  $kg.m^{-3}$ ; 800 K) at 50 MPa injection pressure. Images (1 out of 2) from one cycle have been added for illustration. The time-averaging window (3500 to 6500  $\mu s$  ASOE) is represented in green and the time-averaged value dashed blue line.  $\rho_{air}(t)$  and  $T_{air}(t)$  are represented in the upper part of the figure.

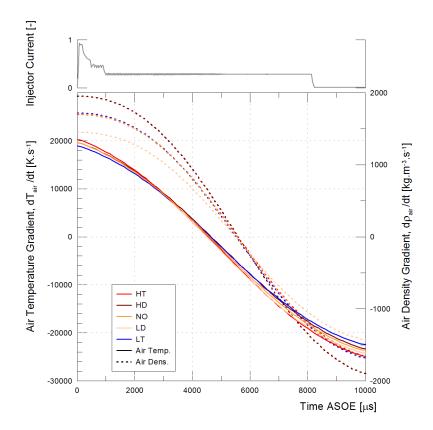


Figure 9: Temperature and density time-derivatives during the injection event.

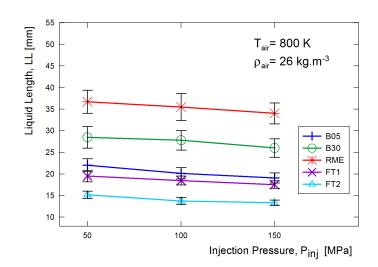


Figure 10: Injection pressure effect on liquid length for the five studied fuels at NO air conditions.

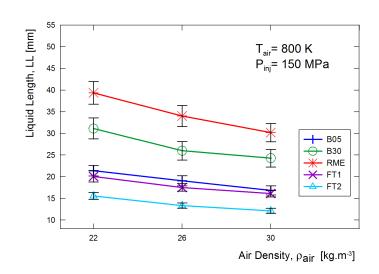


Figure 11: Air density effect on liquid length for the five studied fuels at 150 MPa injection pressure.

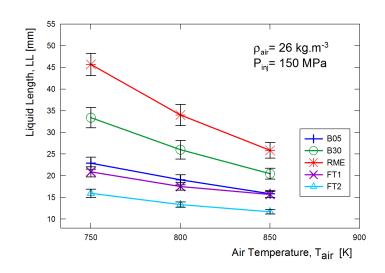


Figure 12: Air temperature effect on liquid length for the five studied fuels at 150 MPa injection pressure.

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Table 2: Fuel relevant properties.

	10010	<b></b>	01 1010	rane pi	oper tree.		
Fuels Properties	Unit A	ASTM Ste	$d.\mathbf{B05}$	$\mathbf{B30}$	RME	$\mathbf{FT1}$	FT2
Density	$[kg.m^{-3}]$	D1298	833	849	878	784	773
Kinematic Viscosity	$[mm^2.s^{-1}]$	D445	2.5	3.1	4.4	3.4	1.3
Lower Heating Value	$[MJ.kg^{-1}]$	D240	42.11	41.77	38.24	44.76	44.24
Equivalent Chemica Formula	- l	D5291	-	-	$C_{18.95}H_{35.2}O_2$	$C_{17}H_{35.5}$	$C_{12}H_{25}O_{0.2}$
C/H ratio	-	-	-	-	0.538	0.479	0.480
$A/F_{st}$ (20.9% $XO_2$ )	-	-	-	-	12.398	14.748	14.388

Table 3: Linear regression coefficients for fuel density dependency to temperature (  $\rho_f = B + A.T_f$ ).

Coefficients	B05	B30	RME	$\mathbf{FT1}$	FT2
А	-0.747	-0.759	-0.815	-0.726	-0.804
В	859.5	871.2	900.6	801.8	803.6
$\mathbb{R}^2$	99.8%	99.4%	99.8%	99.8%	99.4%

Table 4: Injector characteristics.

Bosch Solenoid
Mini-Sac & Single Hole
$80/82~\mu m$
K 1.5
8 ms
50, 100, 150 MPa

Table 5: Results from the statistical analysis for assessment of engine physical conditions under both steady and unsteady-conditions. Non-significative exponents (p-value>0.05) appear in grey.

Parameter		Cte	$d_0$	$T_{air}$	$P_{inj}$	$\rho_{air}$	-	-
Exponents		-	a	b	с	d	$\mathbb{R}^2$	RMSE
Theoretical		-	1	-1.58	0	-0.5	-	-
B05	9)	$3.0324E{+}11$	-	-3.11	-0.10	-0.68	99.0	0.28
B30	$tat_{i}$	3.7266E + 10	-	-2.80	-0.02	-0.70	92.6	0.91
RME	-S-	3.2664E + 15	-	-4.39	-0.04	-0.82	99.0	0.59
FT1	adj	5.3889E + 09	-	-2.55	-0.09	-0.63	99.3	0.19
FT2	Steady-State	$1.1733E{+}10$	-	-2.68	-0.10	-0.67	97.9	0.26
All fuels		$1.2095E{+}11$	-	-2.98	-0.06	-0.69	15.3	6.95
B05	te	$3.0139E{+}11$	-	-3.12	-0.10	-0.66	96.5	0.60
B30	Sta	$3.9238E{+}10$	-	-2.81	-0.02	-0.69	89.4	1.23
RME	ly	7.9248E + 12	-	-3.55	-0.01	-0.75	88.9	2.02
FT1	eau	4.4816E + 09	-	-2.54	-0.09	-0.60	97.4	0.43
FT2	Unsteady-State	3.1336E + 09	-	-2.53	-0.09	-0.58	95.5	0.44
All fuels	U	$9.9393E{+}08$	-	-2.42	-0.05	-0.39	11.3	6.82

Table 6: Results from the statistical analysis for assessment of engine physical conditions and fuel physical properties under unsteady-state conditions.

Parameter			Cte	$d_0$	$T_{air}$	P <sub>inj</sub>	$\rho_{air}$	$\rho_f$	$\nu_f$	$T_{10\%}$	$T_{50\%}$	$T_{95\%}$	-	-
Exponents		#	-	a	b	с	d	e	f	g	h	i	$R_{spe}^2$	RMSE
Theoretical			-	1	-1.58	0	-0.5	0.5	-	-	-	-	-	-
		(1)	9.9393E + 08	-	-2.42	-0.05	-0.39	-	-	-	-	-	11.3	6.82
		(2)	1.0000E + 00	-	-	-	-	0.71	0.22	-1.14	5.41	-4.53	78.2	3.45
	0)	(3)	8.2699E-08	-	-2.78	-0.06	-0.61	5.99	-	-	-	-	88.7	2.57
	$tat_{\epsilon}$	(4)	6.8209E + 09	-	-2.72	-0.06	-0.55	-	0.62	-	-	-	79.5	3.15
	-S-	(5)	9.4517E + 06	-	-2.68	-0.06	-0.54	-	-	1.27	-	-	75.1	3.80
All fuels	Jnsteady-State	(6)	3.2668E + 06	-	-2.66	-0.06	-0.54	-	-	-	1.39	-	69.0	4.10
	ste	(7)	1.5327E + 06	-	-2.51	-0.06	-0.47	-	-	-	-	1.27	45.4	5.01
	$U_{n}$	(8)	2.2874E-03	-	-2.74	-0.06	-0.61	4.39	0.26	-	-	-	94.6	1.62
		(9)	1.8131E + 10	-	-2.85	-0.07	-0.63	-	-	-1.23	6.94	-5.45	97.9	0.97
		(10)	6.1213E-05	-	-2.63	-0.06	-0.60	4.39	-	-	0.54	-	94.4	1.57
		(11)	1.0000E + 00	-	-2.85	-0.07	-0.63	6.61	1.70	-0.90	-2.89	-0.06	98.1	0.93

	Table 7: Nomenclature
B05/B30	Fossil diesel with $5\%/30\%$ RME (in mass)
RME	Rapeseed Methyl-Ester
FT(D)	Fischer-Tropsch (Diesel)
LD/HD	Low/High Density condition (at 800 $K$ )
LT/HT	Low/High Temperature condition (at 26 $kg.m^{-3}$ )
NO	Nominal condition
OC	Operating Conditions
Subscripts	
0	relative to initial conditions
f	fuel
air	relative to the air surrounding the spray
inj	injection
max	maximum
evap	evaporation
back	relative to the spray counter-pressure
Abbreviations	
Р	pressure
$\Delta P$	pressure drop= $P_{inj} - P_{back}$
Т	temperature
ρ	density
h	enthalpy
Κ	constant value
Х	spray axis
Y	mixture fraction
LL	Liquid Length
1D	One-dimensional
(A)SOI/E, EOI	(after) start of injection/energizing, end of injection
(A)TDC	(After) Top Dead Center
ASTM	American society for testing and materials
CAD	crank angle degree
CFD	computational fluid dynamics
CMOS	complementary metal-oxide semiconductor
EGR	Exhaust Gas Recirculation
FID	flame ionization detector
HCCI	homogeneous charge compression ignition
LHV	Lower Heating Value $[MJ.kg^{-1}]$
$NO_X$	mono-nitrogen oxides
PM	particulate Matter
rpm	revolutions per minute
$R^2_{(spe)}$	(specific) coefficient of determination
RMSE	root mean square error
TTL	transistor-transistor logic